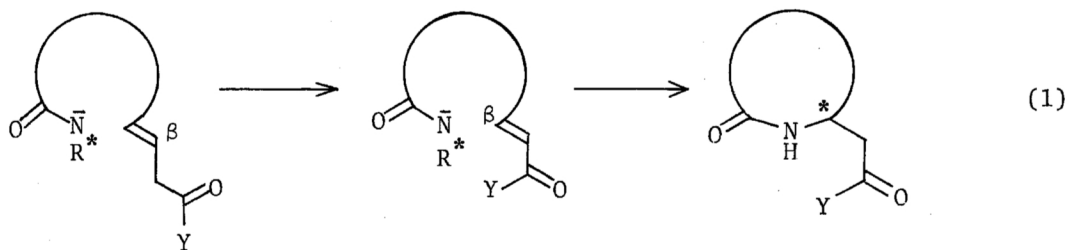


A GENERAL ASYMMETRIC CYCLIZATION.  
ASYMMETRIC SYNTHESIS OF  
OPTICALLY ACTIVE 2-OXO-5-PYRROLIDINEACETIC ACID DERIVATIVES

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Novel asymmetric intramolecular Michael addition by chiral amide anion to  $\alpha,\beta$ -unsaturated ester was performed for the asymmetric synthesis of (S)-2-oxo-5-pyrrolidineacetic acid. The acid was related to (S)-(-)-ecgoninic acid in order to determine its absolute configuration. Much higher diastereoselectivity of the chiral amide than that of the chiral ester was also demonstrated.

Although different approaches for asymmetric cyclization in carbon-carbon bond formation to yield optically active alicyclic compounds have been explored,<sup>1</sup> a study on an asymmetric cyclization in carbon-nitrogen bond formation has few examples.<sup>2</sup> We describe herein the first asymmetric intramolecular Michael Addition of the chiral amide anion onto the  $\beta$ -carbon of an  $\alpha,\beta$ -unsaturated carbonyl compound or a  $\beta,\gamma$ -unsaturated carbonyl compound which is convertible to  $\alpha,\beta$ -unsaturated one. This concept is outlined in eq. 1.

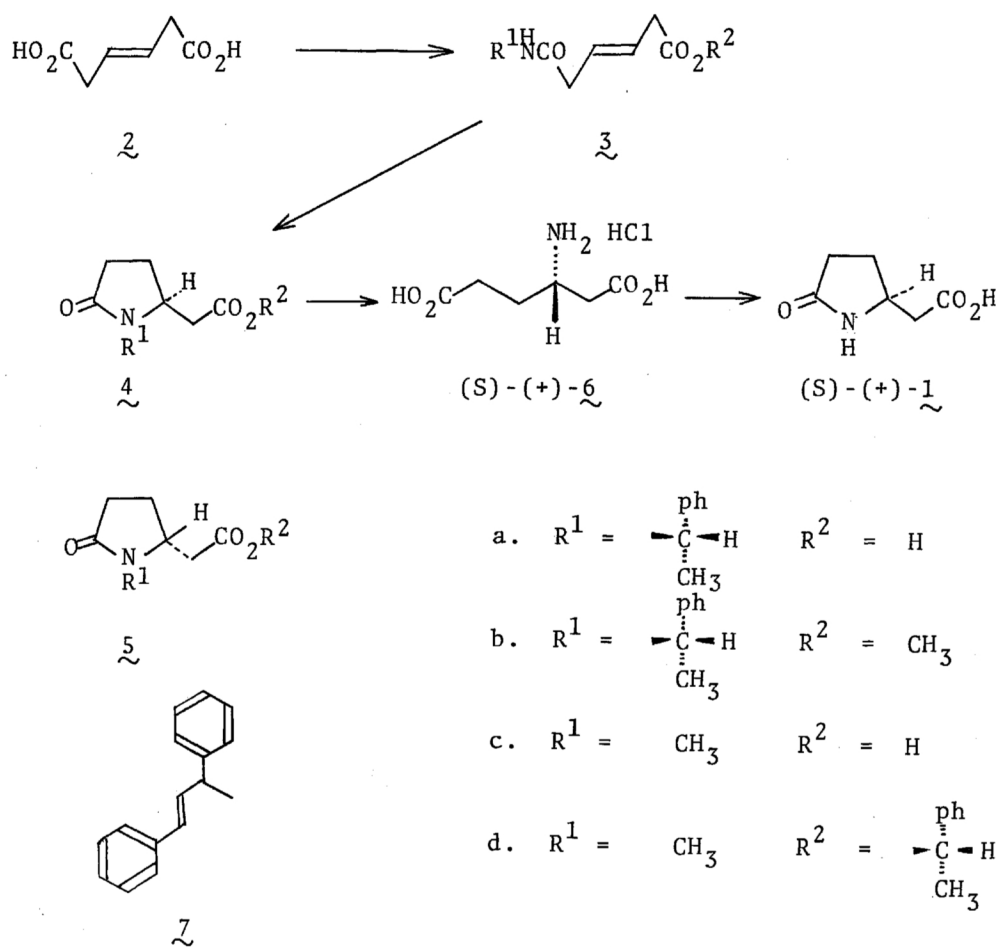


R\* : chiral controlling group

This report deals with the asymmetric synthesis of (S)-(+)-2-oxo-5-pyrrolidineacetic acid (**1**)<sup>3</sup>, which is a potential intermediate for optically active pyrrolidine derivatives, by the cyclization process utilizing a readily available and efficient chiral controlling group.

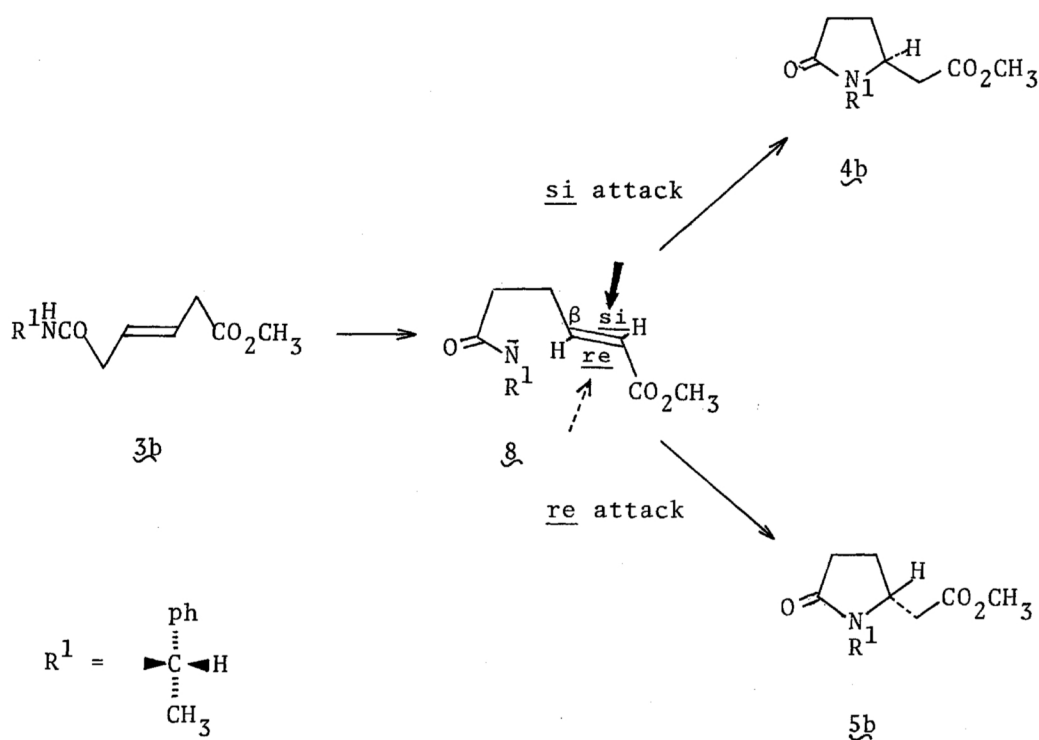
Treatment of  $\beta$ -hydromuconic acid (**2**) with (S)-(-)- $\alpha$ -phenethylamine and diethyl phosphorocyanidate (DEPC)<sup>4</sup> in dimethylformamide (DMF) gave the optically active acid **3a**, mp 66-67°,  $[\alpha]_D^{23}$  - 63.5° (c=1.0<sup>5</sup>, EtOH), in 41% yield. Initial attempts of the asymmetric cyclization of the acid **3a** with sodium hydride (NaH) or triethylamine, or with p-toluene sulfonic acid gave neither the amide **4a** nor the amide **5a**. Then, the acid **3a** was esterified with diazomethane to the ester **3b**, which was treated with 0.2 equiv. of NaH in THF at 4°C for 19 hr to give a mixture of methyl ester **4b** and its diastereomer **5b** in 72% yield. To determine the ratio of

Scheme I



4a and 5a the corresponding mixture of methyl esters 4b and 5b prepared by treatment with diazomethane was submitted to an nmr assay. The 100-MHz nmr spectra of the above mixture of methyl esters 4b and 5b in deuteriochloroform showed chemical shift differences (3.53 and 3.61 ppm) in the methyl ester region and showed 44% diastereomeric purity<sup>6</sup> for 4b (4b : 5b = 72 : 28). Hydrolysis of the latter mixture of the methyl ester 4b and 5b with aqueous methanolic KOH gave a mixture of amides 4a and 5a which yielded 96% optically pure 4a, mp 205-206°,  $[\alpha]_D^{27} - 159.9^\circ$  (EtOH), after recrystallizations from ethanol. The above transformation of the ester 3b to the diastereoisomeric mixture of the amides 4b and 5b is outlined in Scheme II. Migration of the double bond of the ester 3b by NaH generates the probable intermediate 8 which has a chiral amide anion functionality discriminating si-face from re-face<sup>7</sup> at the  $\beta$ -carbon to afford the amide 4b diastereoselectively.<sup>8</sup>

Scheme II





Saponification of the ester 10 afforded (S)-(-)-ecgoninic acid (9a), mp 90-93°,  $[\alpha]_D^{23}$  - 0.9° (EtOH) in 2% optical purity and in 68% yield from 3d.

The above results demonstrate that the 1,6-asymmetric induction by the amide anion in 8 at the  $sp^2$  carbon atom is more efficient than 1,5-asymmetric induction by that from 3d. Thus, the geometry of the transition state for the above 1,6-asymmetric induction seems to resemble the product 4b.

In conclusion, it should be pointed out that the new asymmetrically induced intramolecular Michael reaction using a chiral amide has major potential for biomimetic asymmetric alkaloid syntheses.<sup>11</sup>

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