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## DAKIN - WEST REACTION WITH CYSTEINE,

#### CYSTINE, AND SERINE

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The C-acylation of cysteine (I) and serine (II) (Dakin-West reaction) lies at the base of some metabolic processes [1-3].

In order to create chemical models of these biochemical transformations we studied in the present paper the Dakin-West reaction of (I), (II), and cystine (III) with  $Ac_2O$  and either pyridine (Py) or  $\gamma$ -picoline:

 $\begin{array}{c|c} \text{RCH}_2\text{CH}(\text{NH}_2)\text{COOH} & \xrightarrow{\text{AcsO, Py}} & \text{AcSCH}_2\text{CH}(\text{NHAc}) \text{Ac} \\ \text{R} = \text{SH} (I), \text{OH} (II), & (IV) \\ \text{S}_2\text{CH}_2\text{CH}(\text{NH}_2)\text{COOH} (III) & (\text{AcNH}(\text{Ac})\text{CHCH}_2)_2\text{S} & \text{AcSH} \\ & (VII) & \text{AcSH} & \text{EtsN} \\ & \text{AcSH} & \text{EtsN} \\ \text{AcNHCH}_2\text{Ac} & \xrightarrow{(\text{CH}_1\text{O})_x} & \text{AcNHCH}(\text{CH}_2\text{NC}_5\text{H}_{10}) \text{Ac} \rightarrow \text{AcNHC}(=\text{CH}_2)\text{Ac} \\ & (VI) & (V) & (V) \end{array}$ 

When (I) was reacted with Ac<sub>2</sub>O and Py we obtained, as the result of the C-, N-, and S-acetylation and decarboxylation of (I), 3-acetamido-4-acetylmercapto-2-butanone (IV) in 46% yield, whose structure was proved by the elemental analysis, IR and NMR spectra (see Experimental Part), and also by counter synthesis from 3-acetamido-2-butenone (V). The latter added AcSH under the influence of catalytic amounts of Et<sub>3</sub>N to give (IV) in 20% yield. The reaction of 3-acetamido-4-piperidino-2-butanone (VI) with AsSH led to di ( $\beta$ -acetamido- $\gamma$ -ketobutyl) sulfide (VII), whose structure was confirmed by the IR and NMR spectra. The same Dakin-West reaction product (IV) was formed by reacting Ac<sub>2</sub>O and Py or  $\gamma$ -picoline with either (II) or (III) in the presence of AcSH.

The conversion of (II) and (III) to (IV) includes, together with the processes of decarboxylation and Cand N-acetylation, replacement of the  $\beta$ -substituents of (II) and (III) by the acetylmercapto group and the liberation of water and respectively sulfur and hydrogen sulfide.

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## EXPERIMENTAL

The IR spectra were taken as KBr pellets on a UR-10 instrument, while the NMR spectra were taken on a a DA-60-IL instrument (internal standard = HMDS).

Reaction of (I) with Ac<sub>2</sub>O and Py. With stirring, to 1 g of (I) as the hydrochloride monohydrate was gradually added 6 ml of Ac<sub>2</sub>O, and then 4 ml of Py, after which the mixture was stirred for another 30 min at ~20°C, let stand at ~20° for 48 h, poured into excess Na<sub>2</sub>CO<sub>3</sub> solution, and extracted with ethyl acetate (EA). The extract was dried over MgSO<sub>4</sub>, evaporated in vacuo, and the residue was chromatographed on a column filled with SiO<sub>2</sub> (160 mesh), followed by elution with benzene and EA. From the EA eluate we isolated 0.53 g (46%) of (IV), mp 80-81° (from cyclohexane), R<sub>f</sub> 0.44 (TLC, Silufol UV-254, 1: benzene – EA mixture, detection of spots with I<sub>2</sub> vapors and in UV light). Infrared spectrum ( $\nu$ , cm<sup>-1</sup>): 1660 (CON), 1690 (AcS), 1725 (C=O). NMR spectrum (CDCl<sub>3</sub>,  $\delta$ , ppm): 2.00 s and 2.27 s (CH<sub>3</sub>CON and CH<sub>3</sub>COS); 2.34 s (CH<sub>3</sub>CO); 3.39 m (CH<sub>2</sub>); 4.76 m (CH); 6.57 m (NH). Found: C 47.27; H 6.47; N 6.94; S 15.58%. C<sub>3</sub>H<sub>13</sub>NO<sub>3</sub>S. Calculated: C 47.27; H 6.40; N 6.89; S 15.76%. Similar results were obtained when (I) is reacted with Ac<sub>2</sub>O and Py in the presence of AcSH.

Addition of AcSH to 3-Acetamido-2-butenone (V). With stirring, to 1 ml of AcSH +2 drops of Et<sub>3</sub>N was gradually added 0.9 g of (V) [4], after which the mixture was kept at 20° for 140 h. After the above described workup we isolated 0.29 g (20%) of (IV).

<u>Reaction of 3-Acetamido-4-piperidino-2-butanone (VI) with AcSH.</u> A mixture of 6 g of acetamidoacetone, 1.6 g of paraform, and 8 ml of piperidine in 24 ml of alcohol was kept for 6 days at 20°, after which it was evaporated in vacuo, to the residue with stirring was added 11 ml of AcSH, and the mixture was kept for another 7 days at 20°. After treating the reaction mixture with n-heptane the precipitate was filtered and washed with EA to give 3.01 g (40%) of di-( $\beta$ -acetamido- $\gamma$ -ketobutyl) sulfide (VII), mp 143-144° (from isoporpanol). Infrared spectrum ( $\nu$ , cm<sup>-1</sup>): 1640 (CON), 1618 (CO). NMR spectrum (CF<sub>3</sub>COOH,  $\delta$ , ppm): 1.90 s (2CH<sub>3</sub>CON); 1.95 s (2CH<sub>3</sub>CO); 2.72 t (2CH<sub>2</sub>); 4.42 m (2CH); 8.14 m (2NH); Found: C 50.01; H 6.66; S 11.15%. C<sub>12</sub>H<sub>20</sub>N<sub>2</sub>SO<sub>4</sub>. Calculated: C 50.00; H 6.94; S 11.11%.

<u>Reaction of Cystine (III) with Ac<sub>2</sub>O and Py</u>. With stirring, to 1 g of cystine (III) dihydrochloride were consecutively added 6 ml of Ac<sub>2</sub>O, 1 ml of Py, and 1 ml of AcSH, after which the mixture was kept at 20° for 48 h and then treated with excess Na<sub>2</sub>CO<sub>3</sub> solution. The product was isolated as described above. We obtained 0.61 g (47%) of (IV). The yield of (IV) was 39% when (II) was reacted with Ac<sub>2</sub>O and AcSH in either Py or  $\gamma$ -picoline under analogous conditions (9 days, 20°).

# CONCLUSIONS

3-Acetamido-4-acetylmercapto-2-butanone is formed when cysteine, cystine or serine is reacted with Ac<sub>2</sub>O and either pyridine or  $\gamma$ -picoline in the presence of thioacetic acid.

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