

Syntheses and Potato Tuber-inducing Activity of Coronafacic Acid Analogues

Hiroaki TOSHIMA, Shinji NARA, Akitami ICHIHARA, Yasunori KODA,*
and Yoshio KIKUTA*

Department of Bioscience and Chemistry, and *Department of Botany, Faculty of Agriculture,
Hokkaido University, Sapporo 060-8589, Japan

Received September 5, 1997

Coronafacic acid (**1**) is an acid component of coronatine, and has been isolated from several pathovars of *Pseudomonas syringae*. Syntheses of C₆-non- and C₆-alkyl-substituted analogues of **1** were accomplished *via* intramolecular 1,6-conjugate addition as the key step. Among them, **1** and four C₆-alkyl-substituted analogues exhibited potato tuber-inducing activity, but the C₆-non-substituted analogue did not. It was revealed that a certain length of the C₆-alkyl group was necessary to exhibit activity.

Key words: coronafacic acid; coronatine; jasmonic acid; 1,6-conjugate addition; tuber-inducing activity

Coronafacic acid (**1**) is an acid component of the phytotoxin, coronatine (**2**), and has been isolated from several pathovars of 2-producing *Pseudomonas syringae*.¹⁾ There has been strong interest recently shown in **1** and **2** for their unique biological activities, similar to those of (3*R*, 7*S*)-*epi*-jasmonic acid (**3a**),²⁾ as a kind of plant hormone.³⁾ It is likely that the structural similarity of **1** and **2** to **3a** would be the cause for their common biological activities. In particular, their cyclopentanone moieties with two stereogenic centers and a carboxyl group are closely similar. Furthermore, the C₄-unit of **1** might correspond to the (Z)-2-pentenyl group (C₅-unit) of **3a**. The bicyclic structure of **1** contributes to the retention of C_{3a}- and C_{7a}-stereogenic centers as *cis*-relationship; however, monocyclic compound **3a** readily undergoes epimerization to provide the thermodynamically stable *trans*-isomer, (3*R*, 7*R*)-jasmonic acid (**4a**). Therefore, naturally occurring **1** might be regarded as a configurationally and conformationally restricted bicyclic analogue of **3a** (Fig. 1). From the fact that natural **1** exhibited equal or slightly weak **3a**-like activities in several bioassays characteristic for **3a** and **4a** [sometimes used as an equilibrium mixture of minor (±)-**3a** and major (±)-**4a**; or of minor (±)-**3b** and major (±)-**4b**],^{2e)} **1** is a useful probe to investigate plant physiological effects relating to jasmonoids. Under the mentioned background, in connection with the construction of a functional 1-hydrindanone framework *via* intramolecular 1,6-conjugate addition as the key step, C₆-non- and C₆-alkyl-substituted analogues of **1** have been synthesized,⁴⁾ and their biological activities have been examined. We have already synthesized **1** in both its racemic and optically active form.⁵⁾

A commercially available starting material, 2-cyclopenten-1-one, was converted to racemic ester **5** *via*

Michael addition of the acetic acid-ester synthon⁶⁾ and subsequent acetalization with ethyleneglycol (Scheme).^{5b)} The desired precursors (α,β,γ,δ-unsaturated esters; **8a**, **8b**, **8c** and **8d**) of the intramolecular 1,6-conjugate addition were obtained by the following four-step manipulation: (1) aldol condensation between the lithium enolate of **5** and acrolein derivatives, (2) mesylation of the resulting β-hydroxyl group with mesyl chloride, (3) subsequent β-elimination with 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU), and (4) acidic deacetalization.^{5b)} Only in the case of using acrolein, aldol condensation gave a mixture of hydroxy esters **6a** and **6e**, each as a mixture of two diastereomers, in a 90% yield (**6a**:**6e** = *ca.* 4:1), which could be separated by column chromatography. Since the relative stereochemistry between the ethoxycarbonyl and hydroxyl groups could not be determined at this stage, major **6a** was converted into unsaturated ester **7a**, and minor **6e** into **7e**, each as the sole product. It was deduced that the elimination of the mesyloxy group would proceed *via* the E2 mechanism under the conditions used. The geometries of **7a** and **7e** were determined from their ¹H-¹H-NOE difference spectra (Fig. 2). The observed NOE enhancements were indicative of **7a** and **7e** being (*E*)- and (*Z*)-isomers, respectively. Therefore, *syn*- and *anti*-relationships of **6a** and **6e** were respectively proved. Deacetalization of **7a** gave desired (*E*)-isomer **8a**. In the cases of using 2-alkylacroleins possessing methyl-, ethyl- and *n*-butyl groups, *syn*-hydroxy esters **6b**, **6c** and **6d**, each as a mixture of two diastereomers, were obtained. Their relative stereochemis-

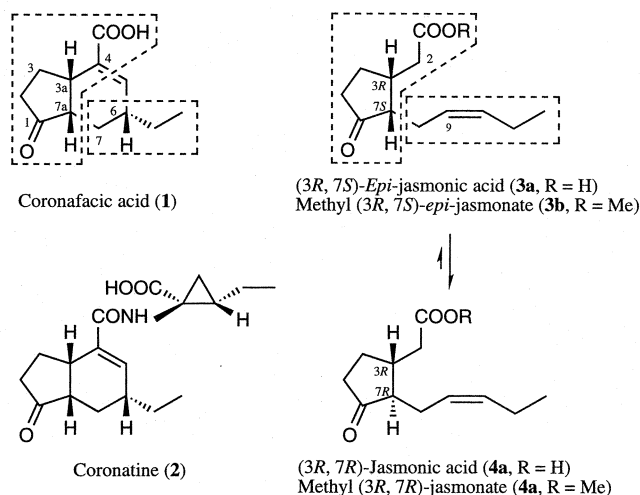
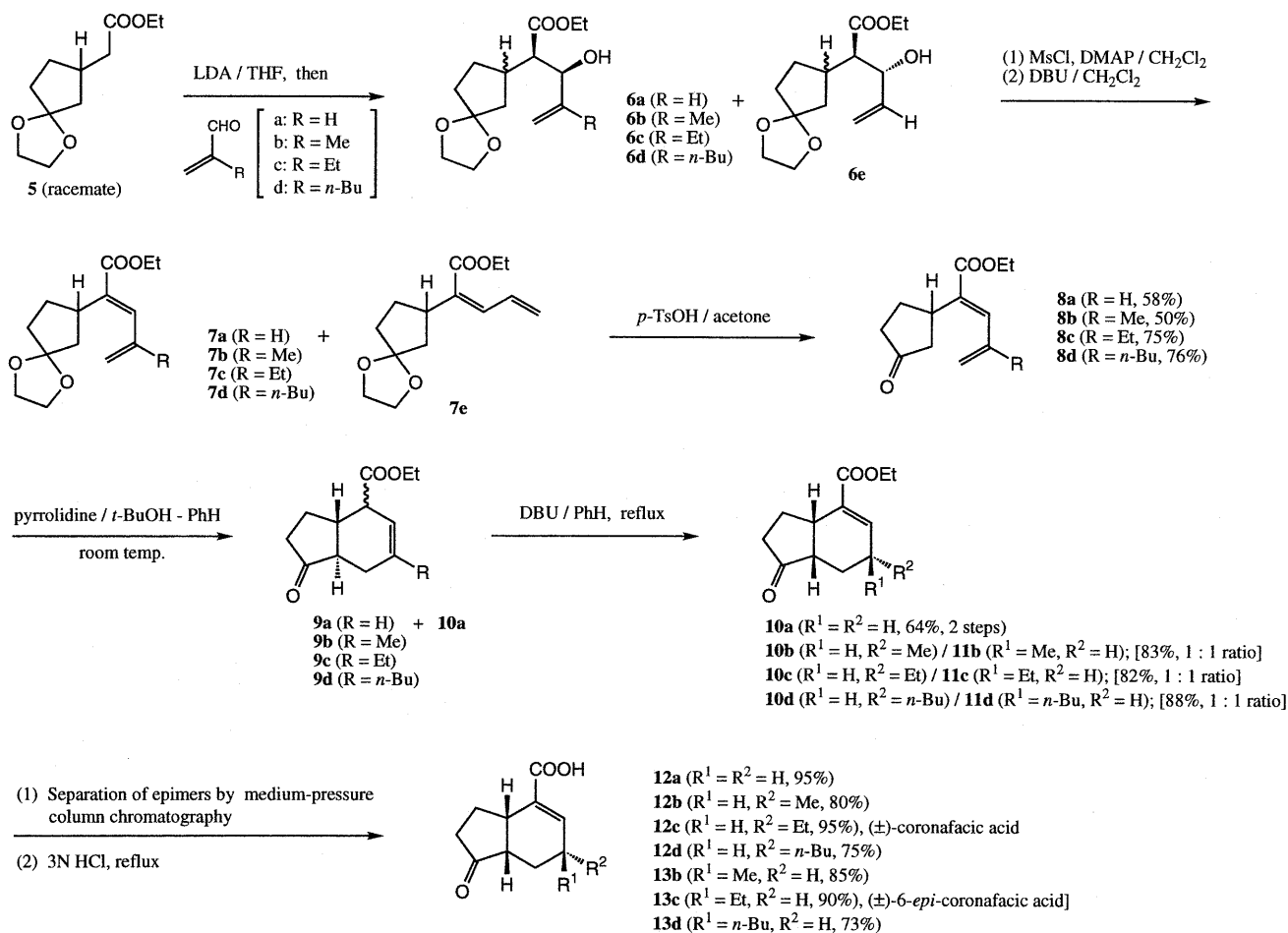


Fig. 1. Structures of Coronafacic Acid and Related Compounds.



Scheme. Syntheses of Coronafacic Acid Analogues.

try was also revealed by converting to unsaturated esters **7b**, **7c** and **7d**, each as the sole product. In a ^1H - ^1H -NOE experiment on **7c**, NOE enhancement was observed between the C_7 - and C_5 -protons indicating that **7c** possessed (*E*)-geometry and *s-cis* conformation. In contrast to **7a** and **7e** existing in the *s-trans* conformation, the ethyl group at the C_4 -position in **7c** would contribute to maintaining the *s-cis* conformation, which is more favorable for intramolecular cyclization than the *s-trans* one. In the ^1H -NMR spectra of **7b** and **7d**, the signals of $\text{C}_3\text{-H}$, $\text{C}_5\text{-H}_a$ and $\text{C}_5\text{-H}_b$ were closely similar to those of **7c** ($\text{C}_3\text{-H}$, δ 7.03 ppm; $\text{C}_5\text{-H}_a$, δ 4.88 ppm; $\text{C}_5\text{-H}_b$, δ 5.08 ppm), respectively. Furthermore, the mechanism for the aldol condensation between the lithium enolate of **5** and acrolein derivatives is reasonably explicable for producing predominantly or exclusively *syn*-hydroxy esters (Fig. 2). The (*E*)-geometry of the lithium enolate was proved from that of corresponding silyl enol ether **5a**, which had been obtained from **5** by treating with LDA and then TMSCl in THF, exhibiting NOE-enhancement of the olefinic proton by irradiating the methyl protons. In two possible 6-membered transition states (**T1** and **T2**), **T1** predominates over **T2** due to steric hindrance between the bulky cyclopentane moiety of the enolate of **5** and the alkenyl moiety of an aldehyde in **T2**. When acrolein lacking an alkyl group was used, the aldol condensation would proceed to some extent *via* **T2** to give

6a and **6e**, because of the relaxation of steric hindrance. When 2-alkylacroleins were used, the aldol condensation would proceed exclusively *via* **T1** to give *syn*-hydroxy esters **6b**, **6c** and **6d**, and further E2 elimination could explain the (*E*)-geometry of **7b**, **7c** and **7d**. Deacetalization of **7b**, **7c** and **7d** gave desired (*E*)-isomers **8b**, **8c** and **8d**, respectively.

Intramolecular 1,6-conjugate addition of **8a** with pyrrolidine gave a mixture of β,γ -unsaturated ester **9a**, itself as a mixture of two diastereomers (*ca.* 5:1 ratio), and α,β -unsaturated ester **10a** in a 7:1 ratio (based on integration of the ^1H -NMR spectrum), which was further treated with DBU to give only **10a**. In the ^1H -NMR spectrum of *C*_{7a}-*epi*-coronafacic acid, whose relative stereochemistry has been determined by an X-ray analysis,^{1a)} $\text{C}_{3a}\text{-H}$ and $\text{C}_{7a}\text{-H}$ were observed at δ 2.75 and 2.25 ppm, respectively. However, in the ^1H -NMR spectrum of **10a**, $\text{C}_{3a}\text{-H}$ and $\text{C}_{7a}\text{-H}$ were observed at δ 3.22 and 2.33–2.47 ppm, similar to those of natural **1** (δ 3.08 and 2.37 ppm),^{5b)} respectively. Therefore, the juncture of **10a** is considered to be *cis*-relationship. Treatment of **8b** with pyrrolidine exclusively gave β,γ -unsaturated ester **9b** as a mixture of two diastereomers (*ca.* 5:1 ratio), which could be isomerized to a mixture of α,β -unsaturated esters (**10b** and **11b**, 1:1 ratio) with DBU. In the same two-step manner, **8c** and **8d** were also converted to α,β -unsaturated esters (**10c/11c** and **10d/11d**, each in a 1:1

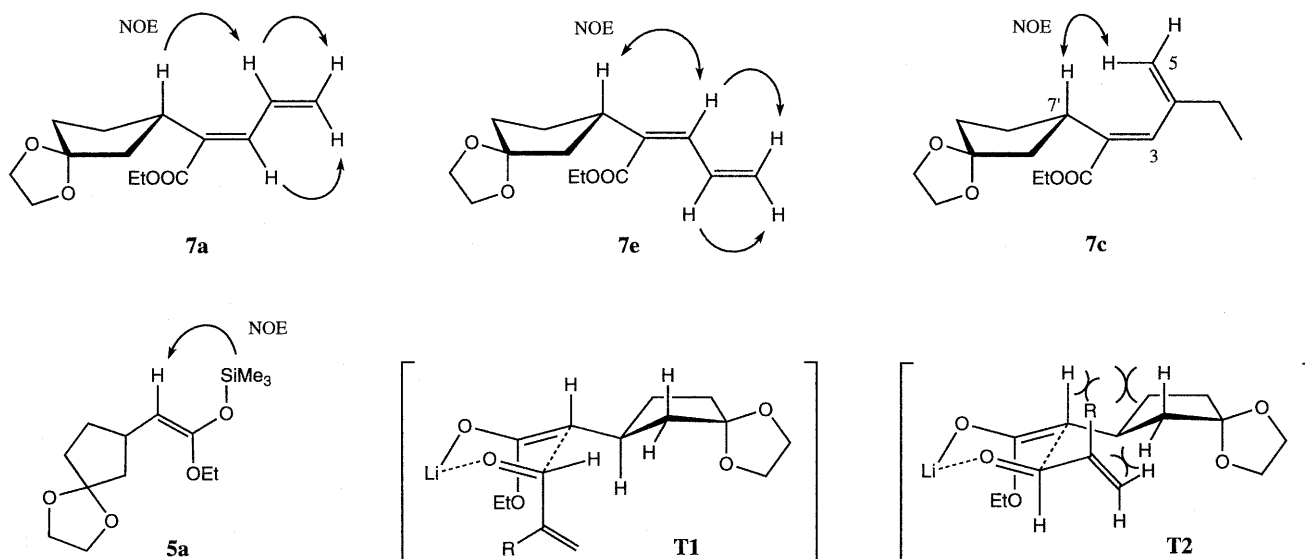


Fig. 2. Observed NOEs of **7a**, **7e**, **7c** and **5a**; and Two Possible Transition States of Aldol Condensation.

ratio). The ratios of the two diastereomers in **9c** and **9d** were *ca.* 5:1, as well as those in **9a** and **9b**. The relative stereochemistry of the major diastereomers of **9a–9d** would probably be similar and useful for considering the reaction mechanism. However, the two diastereomers of **9a–9d** were inseparable and were subjected to the next isomerization to give α,β -unsaturated esters (**10a–10d** and **11b–11d**). A possible mechanism for intramolecular 1,6-conjugate addition is illustrated in Fig. 3. When pyrrolidine was used, an enamine (**IM1**) generated *in situ* would act as an electron donor. The resulting dienolate (**IM2**) would undergo kinetically controlled protonation at the α -position to give β,γ -unsaturated esters. The stereochemistry at the C_{7a} -position would be controlled thermodynamically *via* an enamine (**IM3**) to give **9a–9d**. The presence of alkyl substituent at the C_6 -position might also contribute to keeping the β,γ -position of the double bond in **9b–9d**. In practice, **8a** only gave **9a** and **10a** under the same conditions. Pyrrolidine cannot deprotonate from **IM3** or **9a–9d** to provide a dienolate. DBU acts as base for **9a–9d** to provide a bis-enolate. The dienolate part would undergo protonation at the γ -position to give α,β -unsaturated esters (**10** and **11**). In **10** and **11**, deprotonation at the C_6 -position with DBU is also possible to give thermodynamically stable α,β -unsaturated esters **10** and **11** as a 1:1 equilibrium mixture. The position of the double bond influences the relative stereochemistry of the juncture. Based on the chemical shifts (C_{3a} -H and C_{7a} -H) in the ^1H -NMR spectra already described, all the β,γ -unsaturated esters existed as the *trans*-1-hydrindanone, and all the α,β -unsaturated esters as the *cis*-1-hydrindanone. Three pairs of epimers (**10b/11b**, **10c/11c** and **10d/11d**) with respect to the C_6 -position could be separated by column chromatography. The relative stereochemistry of **11c** was determined from ^1H - ^1H -NOESY spectra at 500 MHz (Fig. 4). The observed NOESY correlations could reasonably explain its stereochemistry. The correlation between C_{3a} -H and C_{7a} -H revealed the *cis*-1-hydrindanone frame-

work. The correlations between C_{7a} -H and the methylene protons of the C_6 -ethyl group and between C_{7a} -H and C_7 -H $_{\beta}$ resulted from the preferential half-chair-like conformation in **11c**. These observations mean that the relationship between C_6 -H and C_{7a} -H would be *trans*. Furthermore, acidic hydrolysis (3 N HCl, reflux) of **10c** gave (\pm)-coronafacic acid (**12c**) in a 95% yield, whose spectral data were identical with those of natural **1**.^{5b)} Therefore, **11c** was the C_6 -epimer of **10c**. In the ^1H -NMR spectra of **10b** and **10d**, the characteristic signals of C_{3a} -H (each of δ 3.07 ppm) were closely similar to those of **10c** (δ 3.08 ppm), while the other corresponding signals were slightly similar. In the ^1H -NMR spectra of **11b** and **11d**, the characteristic signals of C_{3a} -H (δ 3.23 and 3.27 ppm) were closely similar to those of **11c** (δ 3.26 ppm), while the other corresponding signals were slightly similar. In this way, the relative stereochemistry of **10b–10d** and **11b–11d** was unequivocally determined. The final acidic hydrolysis of ethyl esters **10a**, **10b**, **10d**, **11b**, **11c** and **11d** gave the analogues of **1**: C_6 -non-substituted analogue **12a** and C_6 -alkyl-substituted analogues **12b**, **12d**, **13b**, **13c** and **13d**, respectively. In this way, we obtained six new analogues of **1** with respect to the C_6 -position.

In addition to the weak potato cell expansion-inducing activity of natural **1** ($>10^{-5}$ M), the potato tuber-inducing activity⁸⁾ of natural **1** and of the positive standard [a mixture of minor (\pm)-**3a** (*ca.* 7%) and major (\pm)-**4a**], were almost equivalent ($>10^{-7}$ M) with respect to the rate of tuberization.^{2e)} These results suggested that a tuber-inducing assay would be more sensitive than a cell expansion-inducing assay for **1** and its analogues. Therefore, the tuber-inducing activity was examined, and the results (except for **12b** and **13b**) are shown in the Table. We used a racemic positive standard and analogues in the tuber-inducing assay and assumed the following background information: the tuber-inducing activities of **3a** and **3b** were almost equivalent,⁹⁾ and (3*R*, 7*S*)-**3b** exhibited the strongest tuber-inducing activity

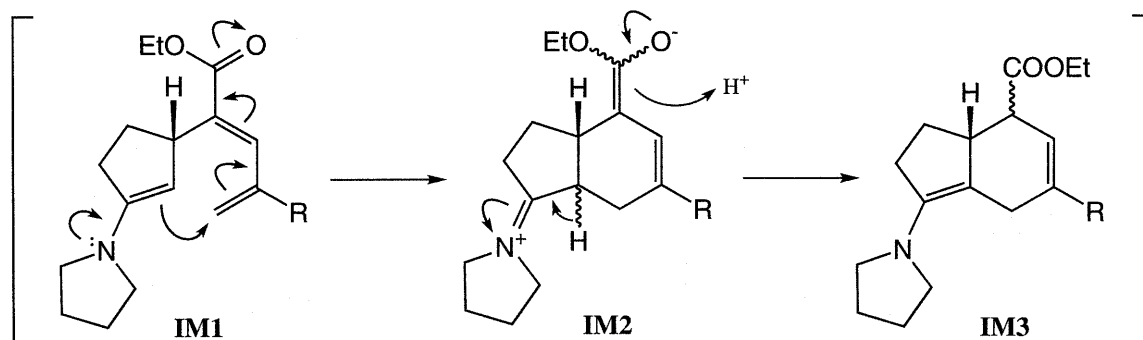
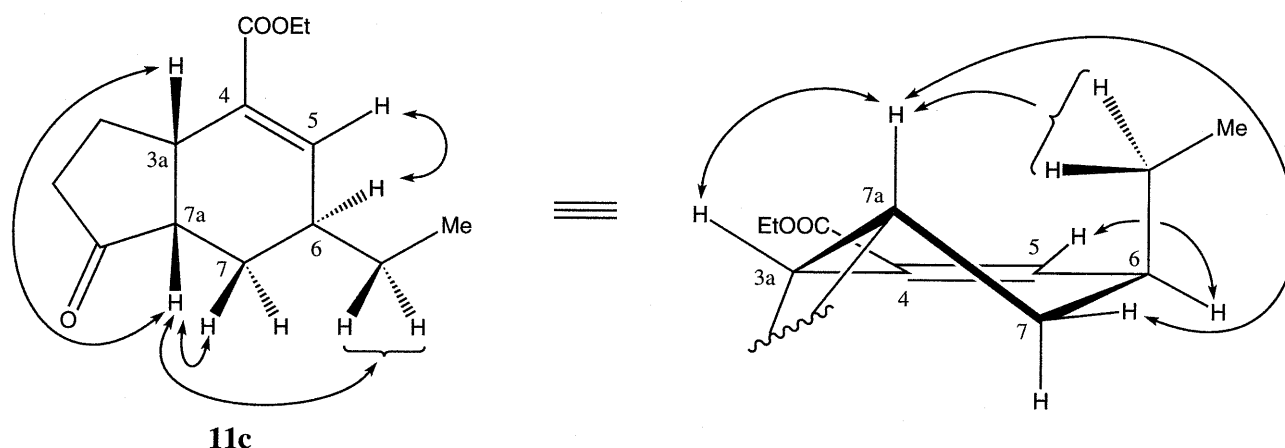


Fig. 3. Possible Mechanism for Intramolecular 1,6-Conjugate Addition.

Fig. 4. Selected NOESY Correlations of **11c** Around the Cyclohexene Ring.

among the four stereoisomers of **3b** and **4b**, while the other three stereoisomers also exhibited this activity.¹⁰⁾ Isomerization by enolization was not evaluated in the assay. However, even the positive standard exhibited reasonably similar activity to that of **3b** in the range from 10^{-7} to 10^{-5} M concentration. Sessile tubers are usually induced by the positive standard at a 10^{-5} M concentration with 80–100% rate of tuberization (strong induction), and tubers are induced at the end of elongated lateral shoots below 10^{-6} M concentration (weak induction) from our experience. The racemic analogues of **1** might also be considered to exhibit tuber-inducing activity without competitive inhibition. Only the C₆-non-substituted analogue (**12a**) exhibited no tuber-inducing activity; however, **12c** and three other analogues (**12d**, **13c** and **13d**) each exhibited tuber-inducing activity in the range from 10^{-6} to 10^{-4} M concentration (weak induction). For such an example, tuberonic acid¹¹⁾ [12-hydroxyl-jasmonic acid that was used as a synthetic mixture of major (3*R**, 7*R**)- and minor (3*R**, 7*S**)-isomers] did not exhibit cell expansion-inducing activity and showed only strong tuber-inducing activity (H. Matsuura, T. Yoshihara, and Y. Koda, unpublished result). The tuber-inducing activities of **12c** and **12d** were almost equivalent to each other at all attempted concentrations, and about one-half that of the positive standard at a 10^{-5} M concentration. By considering the tuber-inducing activity of natural **1**, unnatural enantiomers in **12c** and **12d** might be regarded as exhibiting no activity. However, there is the possibility that unnatural enan-

tiomers in **12c** and **12d** exhibited tuber-inducing activity, because all four stereoisomers of **3b** exhibited this activity.¹⁰⁾ The tuber-inducing activities of **13c** and **13d** (corresponding to C₆-epimers of **12c** and **12d**, respectively) were weaker than those of **12c** and **12d** themselves. The syntheses of C₆-methyl analogues **12b** and **13b** were carried out at different times, so all the analogues could not be subjected to this bioassay at the same time. Since 11-month old tubers were used for the bioassay of **12b** and **13b**, spontaneous tuberization occurred even with the control (20%) on White's medium,⁸⁾ and the results were judged after 2 weeks. In such a preliminary assay, although C₆-epimer **13b** exhibited no tuber-inducing activity, **12b** exhibited tuber-inducing activity (60%) at a rather high concentration (10^{-4} M). By considering the control result (20%), the tuber-inducing activity of **12b** was weakest among all the analogues exhibited this activity.

In conclusion, a certain length of the alkyl group (ethyl or *n*-butyl) attached at the C₆-position was essential for exhibiting **3a**-like activity in tuber-inducing assay. The activities of C₆-epimers **13b**, **13c** and **13d** were weaker than those of **12b**, **12c** and **12d**, respectively. The C₄-unit of **12c**/**13c** and the C₆-units of **12d**/**13d**, including the two carbons at the C₆- and C₇-positions, are mimics of the (*Z*)-2-pentenyl group (C₅-unit) of **3a**. Therefore, **1** and its homologue **12d** might have been configurationally also conformationally restricted analogues of **3a**. These compounds would be useful for analyzing the active conformation of **3a** in plant regula-

Table Effect of Coronafacic Acid and Its Analogues on Potato Tuber-inducing Activity *in vitro*

Compound	Rate of tuberization at each concentration (%) ^a		
	10 ⁻⁶ M	10 ⁻⁵ M	10 ⁻⁴ M
control		0	
positive standard ^b		92 ^c	
12a	0	0	0
12c	12	42	65
13c	0	12	51
12d	15	46	65
13d	0	12	30

^a The bioassay was carried out by following the method in ref. 8. The results were judged after 3 weeks.

^b A mixture of minor (\pm)-**3a** (ca. 7%) and major (\pm)-**4a** was used.

^c Tuber-inducing activity of the positive standard at a 10⁻⁵ M concentration was only examined at this time. The sessile tubers were induced.

tory processes. Design and syntheses of the other analogues of **3a** are in progress.

Experimental

General Methods. ¹H- and ¹³C-NMR spectra were recorded with a JEOL JNM-EX-270 (¹H, 270 MHz; ¹³C, 67.5 MHz) or Bruker AM-500 spectrometer (¹H, 500 MHz; ¹³C, 125 MHz), and chemical shift values are reported in δ (ppm) relative to internal tetramethylsilane or the residual proton of the deuterated solvent. IR spectra were measured with a Perkin Elmer System 2000 FT-IR spectrometer, and mass spectra were recorded with a JEOL JMS-AX500 or JEOL JMS-SX102A spectrometer. Melting point values were obtained with Yanaco micro-melting point apparatus and are uncorrected. Column chromatography was carried out with Silica gel 60 (spherical, 70–140 mesh ASTM, Kanto Chemical). Medium-pressure column chromatography was carried out with a pre-packed glass column, Lobar® (LiChroprep® Si 60, 40–63 μ m, Merck). All chemical yields are summarized in the Scheme.

$\alpha,\beta,\gamma,\delta$ -Unsaturated esters (**7a–7e**, and **8a–8d**). The synthetic manner has been described as the typical procedure for preparing **8c**.^{5b} In the case of preparing **8a**, desired *syn*-hydroxyester **6a** was separated by medium-pressure column chromatography (hexane:EtOAc = 7:3) after aldol condensation. The spectral data for **7a–7e** and **8a–8d** are described next.

Ethyl (2E)-{1,4-dioxaspiro[4.4]non-7-yl}-2,4-pentadienoate (7a). IR ν_{\max} (film) cm⁻¹: 3082, 2980, 1707, 1626, 1587, 1463, 1368, 1328, 1255, 1170, 1119, 1029, 1024, 947, 775; ¹H-NMR (CDCl₃) δ : 1.31 (3H, t, J = 7.3 Hz, CO₂CH₂Me), 1.73–1.83 (2H, m, C₈-H, C₉-H), 1.87 (1H, dd, J = 7.9, 12.2 Hz, C₆-H), 1.90–2.16 (2H, m, C₈-H, C₉-H), 2.23 (1H, t, J = 12.2 Hz, C₆-H), 3.35 (1H, dq., J = 7.9, 12.2 Hz, C₇-H), 3.85–3.96 (4H, m, OCH₂CH₂O), 4.21 (2H, q, J = 7.3 Hz, CO₂CH₂Me), 5.45 (1H, br.d, J = 10.2 Hz, C₅-H), 5.57 (1H, br.d, J = 16.5 Hz, C₅-H), 6.79 (1H, ddd, J = 10.2, 11.2, 16.5 Hz, C₄-H), 7.11 (1H, d, J = 11.2 Hz, C₃-H); ¹³C-NMR (CDCl₃) δ : 14.25, 28.57, 35.64, 35.78, 40.54, 60.41, 64.03, 64.55, 117.75, 124.76, 131.57, 134.02, 139.10,

166.54; EIMS m/z : 253 (6, MH⁺), 252 (36, M⁺), 99 (66), 86 (100); HRMS m/z (M⁺): calcd. for C₁₄H₂₀O₄, 252.1361; found, 252.1380.

Ethyl (2E)-{1,4-dioxaspiro[4.4]non-7-yl}-4-methyl-2,4-pentadienoate (7b). IR ν_{\max} (film) cm⁻¹: 3082, 2955, 2877, 1715, 1622, 1456, 1435, 1310, 1241, 1160, 1113, 1075, 1036, 949, 910, 771; ¹H-NMR (CDCl₃) δ : 1.32 (3H, t, J = 7.3 Hz, CO₂CH₂Me), 1.75–1.83 (3H, m, C₆-H, C₈-H, C₉-H), 1.88 (3H, s, Me), 1.90–2.15 (2H, m, C₈-H, C₉-H), 2.27 (1H, t, J = 12.4 Hz, C₆-H), 3.44 (1H, m, C₇-H), 3.87–3.95 (4H, m, OCH₂CH₂O), 4.22 (2H, q, J = 7.3 Hz, CO₂CH₂Me), 4.94 (1H, br.s, C₅-H), 5.12 (1H, br.s, C₅-H), 7.01 (1H, s, C₃-H); ¹³C-NMR (CDCl₃) δ : 14.23, 22.82, 28.93, 35.58, 35.94, 40.88, 60.38, 63.99, 64.55, 117.05, 117.86, 134.18, 140.41, 141.58, 167.55; EIMS m/z : 266 (13, M⁺), 99 (97), 86 (100); HRMS m/z (M⁺): calcd. for C₁₅H₂₂O₄, 266.1518; found, 266.1523.

Ethyl (2E)-{1,4-dioxaspiro[4.4]non-7-yl}-4-ethyl-2,4-pentadienoate (7c). The spectral data for **7c** have been described elsewhere.^{5b}

Ethyl (2E)-4-butyl-{1,4-dioxaspiro[4.4]non-7-yl}-2,4-pentadienoate (7d). IR ν_{\max} (film) cm⁻¹: 3082, 2957, 2875, 1715, 1623, 1464, 1324, 1238, 1118, 1028, 901, 773; ¹H-NMR (CDCl₃) δ : 0.90 (3H, t, J = 6.9 Hz, CH₂CH₂CH₂Me), 1.32 (3H, t, J = 7.3 Hz, CO₂CH₂Me), 1.24–1.43 (4H, m, CH₂CH₂CH₂Me), 1.73–1.83 (3H, m, C₆-H, C₈-H, C₉-H), 1.94–2.16 (4H, m, C₈-H, C₉-H, CH₂CH₂CH₂Me), 2.26 (1H, t, J = 12.4 Hz, C₆-H), 3.44 (1H, m, C₇-H), 3.87–3.95 (4H, m, OCH₂CH₂O), 4.22 (2H, q, J = 7.3 Hz, CO₂CH₂Me), 4.89 (1H, br.s, C₅-H), 5.08 (1H, br.s, C₅-H), 7.02 (1H, s, C₃-H); ¹³C-NMR (CDCl₃) δ : 13.87, 14.23, 22.23, 28.93, 30.26, 35.62, 35.99, 36.37, 40.86, 60.36, 63.99, 64.51, 114.70, 117.92, 134.81, 141.58, 144.92, 167.37; EIMS m/z : 309 (3, MH⁺), 308 (17, M⁺), 99 (100), 86 (74); HRMS m/z (M⁺): calcd. for C₁₈H₂₈O₄, 308.1987; found, 308.1988.

Ethyl (2Z)-{1,4-dioxaspiro[4.4]non-7-yl}-2,4-pentadienoate (7e). ¹H-NMR (C₆D₆) δ : 1.02 (3H, t, J = 7.3 Hz, CO₂CH₂Me), 1.73 (1H, m, C₈-H), 1.89–2.37 (4H, m, C₆-H, C₈-H, C₉-H₂), 2.25 (1H, ddd, J = 1.4, 7.3, 11.9 Hz, C₆-H), 3.26 (1H, m, C₇-H), 3.51–3.62 (4H, m, OCH₂CH₂O), 4.07 (2H, q, J = 7.3 Hz, CO₂CH₂Me), 5.20 (1H, dd, J = 2.0, 10.9 Hz, C₅-H), 5.26 (1H, dd, J = 2.0, 16.9 Hz, C₅-H), 6.35 (1H, d, J = 9.9 Hz, C₃-H), 7.52 (1H, ddd, J = 9.9, 10.9, 16.9 Hz, C₄-H); ¹³C-NMR (C₆D₆) δ : 14.17, 29.93, 36.39, 41.62, 42.34, 60.24, 64.14, 64.37, 117.38, 121.98, 134.31, 135.60, 136.00, 167.37.

Ethyl (2E)-2-(3-oxocyclopent-1-yl)-2,4-pentadienoate (8a). Colorless oil (888 mg); IR ν_{\max} (film) cm⁻¹: 3082, 2980, 1741, 1702, 1626, 1588, 1463, 1422, 1404, 1370, 1257, 1179, 1086, 1024, 990, 934, 777, 757; ¹H-NMR (C₆D₆) δ : 1.01 (3H, t, J = 7.9 Hz, CO₂CH₂Me), 1.69 (1H, q, J = 8.9 Hz, C₅-H), 1.90 (1H, dd, J = 8.9, 16.5 Hz, C₄-H), 2.08–2.31 (3H, m, C₂-H, C₄-H, C₅-H), 2.78 (1H, dd, J = 8.9, 17.5 Hz, C₂-H), 3.11 (1H, quint.,

$J=8.9$ Hz, C_7 -H), 4.02 (2H, q, $J=7.9$ Hz, CO_2CH_2Me), 5.20 (1H, br.d, $J=10.2$ Hz, C_5 -H), 5.30 (1H, br.d, $J=16.8$ Hz, C_5 -H), 6.42 (1H, ddd, $J=10.2$, 11.6, 16.8 Hz, C_4 -H), 7.30 (1H, d, $J=11.6$ Hz, C_3 -H); ^{13}C -NMR (C_6D_6) δ : 14.12, 27.97, 35.51, 38.23, 42.43, 60.42, 124.95, 131.52, 133.62, 139.46, 166.54, 215.58; EIMS m/z : 208 (100, M^+), 209 (18, $M^+ + H$); HRMS m/z (M^+): calcd. for $C_{12}H_{16}O_3$, 208.1099; found, 208.1082.

Ethyl (2E)-4-methyl-2-(3-oxocyclopent-1-yl)-2,4-pentadienoate (8b). Colorless oil (901 mg); IR ν_{max} (film) cm^{-1} : 3082, 2979, 1744, 1714, 1627, 1456, 1404, 1368, 1242, 1157, 1096, 1026, 903, 775; 1H -NMR (C_6D_6) δ : 1.01 (3H, t, $J=7.3$ Hz, CO_2CH_2Me), 1.64 (3H, s, *Me*), 1.70–1.93 (2H, m, C_4 -H, C_5 -H), 2.16 (1H, dd, $J=9.9$, 17.5 Hz, C_2 -H), 2.20–2.35 (2H, m, C_4 -H, C_5 -H), 2.81 (1H, dd, $J=9.9$, 17.5 Hz, C_2 -H), 3.52 (1H, quint., $J=9.9$ Hz, C_7 -H), 4.02 (2H, q, $J=7.3$ Hz, CO_2CH_2Me), 4.84 (1H, br.s, C_5 -H), 4.96 (1H, br.s, C_5 -H), 7.25 (1H, s, C_3 -H); ^{13}C -NMR (C_6D_6) δ : 14.11, 22.61, 28.45, 35.60, 38.23, 42.97, 60.47, 117.18, 134.04, 140.65, 142.05, 166.62, 215.94; EIMS m/z : 223 (17, $M^+ + H$), 222 (100, M^+), 221 (16, $M^+ - H$); HRMS m/z (M^+): calcd. for $C_{13}H_{18}O_3$, 222.1256; found, 222.1239.

Ethyl (2E)-4-ethyl-2-(3-oxocyclopent-1-yl)-2,4-pentadienoate (8c). Colorless oil (420 mg). The spectral data for **8c** have been described elsewhere.^{5b)}

Ethyl (2E)-4-butyl-2-(3-oxocyclopent-1-yl)-2,4-pentadienoate (8d). Colorless oil (674 mg); IR ν_{max} (film) cm^{-1} : 3082, 2959, 1747, 1714, 1623, 1464, 1404, 1367, 1337, 1294, 1243, 1156, 1094, 1026, 903, 776; 1H -NMR (C_6D_6) δ : 0.88 (3H, t, $J=7.3$ Hz, $CH_2CH_2CH_2Me$), 1.01 (3H, t, $J=7.3$ Hz, CO_2CH_2Me), 1.15–1.37 (4H, m, $CH_2CH_2CH_2Me$), 1.70–1.95 (2H, m, C_4 -H, C_5 -H), 2.00 (2H, t, $J=7.3$ Hz, $CH_2CH_2CH_2Me$), 2.21 (1H, dd, $J=8.9$, 18.1 Hz, C_2 -H), 2.20–2.36 (2H, m, C_4 -H, C_5 -H), 2.84 (1H, dd, $J=8.9$, 18.1 Hz, C_2 -H), 3.52 (1H, quint., $J=8.9$ Hz, C_7 -H), 4.02 (2H, q, $J=7.3$ Hz, CO_2CH_2Me), 4.84 (1H, br.s, C_5 -H), 5.00 (1H, br.s, C_5 -H), 7.30 (1H, s, C_3 -H); ^{13}C -NMR (C_6D_6) δ : 13.93, 14.08, 22.45, 28.42, 30.48, 35.71, 36.54, 38.21, 43.04, 60.47, 114.85, 134.90, 141.91, 145.21, 166.44, 215.76; EIMS m/z : 264 (100, M^+); HRMS m/z (M^+): calcd. for $C_{16}H_{24}O_4$, 264.1725; found, 264.1724.

Typical procedure for intramolecular 1,6-conjugate addition with pyrrolidine and subsequent isomerization with DBU. A solution of **8c** (2.10 g, 8.92 mmol) and pyrrolidine (0.07 ml, 0.89 mmol) in benzene (40 ml)/*tert*-BuOH (2.6 ml) was stirred at room temperature for 10 h. The reaction mixture was acidified with 2 N HCl and extracted with EtOAc (3 times). The combined extracts were successively washed with sat. aq. $NaHCO_3$ and brine, dried over anhydrous Na_2SO_4 , and concentrated under reduced pressure. The crude residue was purified by column chromatography (hexane:EtOAc=9:1) to give a mixture of β,γ -unsaturated esters **9c** (1.97 g) as a colorless oil which was used for the next reaction.

A solution of β,γ -unsaturated esters (39 mg, 0.165 mmol) and DBU (0.05 ml, 0.33 mmol) in benzene (2.0 ml) was refluxed for 64 h. The same work-up and purification as that just described gave a mixture of α,β -unsaturated esters (**10c** and **11c**, 34 mg, 1:1 ratio) as a colorless oil, which was further separated by medium-pressure column chromatography (hexane:EtOAc=96:4).

Ethyl (3aS,7aS*)-2,3,3a,6,7,7a-hexahydro-1-oxo-1H-indene-4-carboxylate (10a)*. According to the typical procedure, **10a** (360 mg) was obtained as a colorless oil; IR ν_{max} (film) cm^{-1} : 2981, 1739, 1715, 1645, 1465, 1447, 1428, 1381, 1253, 1202, 1147, 1089, 1058, 1023, 901, 750, 725; 1H -NMR ($CDCl_3$) δ : 1.31 (3H, t, $J=7.3$ Hz, CH_2Me), 1.17 (2H, q, $J=6.3$ Hz, C_7 -H₂), 1.83 (1H, dq, $J=12.5$, 7.9 Hz, C_3 -H), 2.15–2.24 (2H, m, C_6 -H₂), 2.26 (2H, t, $J=7.9$ Hz, C_2 -H₂), 2.33–2.47 (2H, m, C_3 -H, C_{7a} -H), 3.22 (1H, dq, $J=1.3$, 7.9 Hz, C_{3a} -H), 4.23 (2H, m, CH_2Me), 7.07 (1H, dt, $J=1.3$, 4.3 Hz, C_5 -H); ^{13}C -NMR ($CDCl_3$) δ : 14.22, 19.50, 23.85, 27.35, 35.71, 37.07, 46.60, 60.38, 131.81, 140.20, 166.68, 220.59; EIMS m/z : 209 (28, $M^+ + H$), 208 (100, M^+); HRMS m/z (M^+): calcd. for $C_{12}H_{16}O_3$, 208.1099; found, 208.1106.

Ethyl (3aS,6R*,7aS*)-2,3,3a,6,7,7a-hexahydro-6-methyl-1-oxo-1H-indene-4-carboxylate and ethyl (3aS*,6S*,7aS*)-2,3,3a,6,7,7a-hexahydro-6-methyl-1-oxo-1H-indene-4-carboxylate (10b and 10b)*. According to the typical procedure, a mixture of **10b** and **11b** (120 mg, 1:1 ratio) was obtained as a colorless oil, which was further separated by medium-pressure column chromatography (hexane:EtOAc=96:4).

10b: IR ν_{max} (film) cm^{-1} : 2961, 1739, 1713, 1643, 1456, 1376, 1251, 1095, 1057, 905, 752; 1H -NMR ($CDCl_3$) δ : 1.06 (1H, q, $J=13.2$ Hz, C_7 -H), 1.11 (3H, d, $J=7.3$ Hz, *Me*), 1.31 (3H, t, $J=7.3$ Hz, CO_2CH_2Me), 1.57 (1H, dt, $J=8.5$, 11.6 Hz, C_3 -H), 1.82 (1H, dt, $J=13.2$, 4.8 Hz, C_7 -H), 2.20–2.46 (4H, m, C_2 -H₂, C_6 -H, C_{7a} -H), 2.56 (1H, dt, $J=11.6$, 6.9 Hz, C_3 -H), 3.07 (1H, dt, $J=11.6$, 6.9 Hz, C_{3a} -H), 4.17–4.26 (2H, m, CO_2CH_2Me), 6.84 (1H, s, C_5 -H); ^{13}C -NMR ($CDCl_3$) δ : 14.25, 20.43, 28.14, 28.36, 31.09, 35.87, 38.12, 46.79, 60.45, 131.28, 145.01, 166.77, 220.32; EIMS m/z : 223 (20, $M^+ + H$), 222 (94, M^+), 105 (100); HRMS m/z (M^+): calcd. for $C_{13}H_{18}O_3$, 222.1256; found, 222.1256.

11b: IR ν_{max} (film) cm^{-1} : 2961, 1739, 1715, 1645, 1456, 1373, 1249, 1105, 1082, 885, 758; 1H -NMR ($CDCl_3$) δ : 1.03 (3H, d, $J=7.3$ Hz, *Me*), 1.29 (4H, t, $J=7.3$ Hz, C_7 -H, CO_2CH_2Me), 1.85–2.04 (2H, m, C_3 -H, C_7 -H), 2.07–2.33 (4H, m, C_2 -H₂, C_3 -H, C_6 -H), 2.46 (1H, dt, $J=5.0$, 7.3 Hz, C_{7a} -H), 3.23 (1H, q, $J=7.3$ Hz, C_{3a} -H), 4.20 (2H, q, $J=7.3$ Hz, CH_2Me), 6.84 (1H, d, $J=3.3$ Hz, C_5 -H); ^{13}C -NMR ($CDCl_3$) δ : 14.20, 20.17, 26.88, 27.78, 28.02, 35.60, 36.73, 45.64, 60.41, 130.85, 146.00, 166.86, 221.15; EIMS m/z : 223 (22, $M^+ + H$), 222 (91, M^+), 105 (100); HRMS m/z (M^+): calcd. for $C_{13}H_{18}O_3$, 222.1256; found, 222.1263.

Ethyl (3aS,6R*,7aS*)-6-ethyl-2,3,3a,6,7,7a-hexahydro-1-oxo-1H-indene-4-carboxylate and ethyl (3aS*,*

6*S**, 7*aS**)-6-ethyl-2,3,3*a*,6,7,7*a*-hexahydro-1-oxo-1*H*-indene-4-carboxylate (**10c** and **11c**). The spectral data for **10c** and **11c** have been described elsewhere.^{5b)}

Ethyl (3*aS**,6*R**,7*aS**)-6-butyl-2,3,3*a*,6,7,7*a*-hexahydro-1-oxo-1*H*-indene-4-carboxylate and ethyl (3*aS**,6*S**,7*aS**)-6-butyl-2,3,3*a*,6,7,7*a*-hexahydro-1-oxo-1*H*-indene-4-carboxylate (**10d** and **11d**). According to the typical procedure, a mixture of **10d** and **11d** (579 mg, 1:1 ratio) was obtained as a colorless oil, which was further separated by medium-pressure column chromatography (hexane:EtOAc=96:4).

10d: IR ν_{\max} (film) cm^{-1} : 2930, 1744, 1715, 1645, 1467, 1381, 1301, 1255, 1143, 1102, 1075, 1029, 752; $^1\text{H-NMR}$ (CDCl_3) δ : 0.90 (3H, t, $J=7.3$ Hz, $\text{CH}_2\text{CH}_2\text{CH}_2\text{Me}$), 1.06 (1H, dt, $J=11.2$, 13.2 Hz, $\text{C}_7\text{-H}$), 1.31 (3H, t, $J=7.3$ Hz, $\text{CO}_2\text{CH}_2\text{Me}$), 1.28–1.55 (6H, m, $\text{CH}_2\text{CH}_2\text{CH}_2\text{Me}$), 1.59 (1H, dt, $J=8.6$, 12.5 Hz, $\text{C}_3\text{-H}$), 1.84 (1H, dt, $J=13.2$, 4.6 Hz, $\text{C}_7\text{-H}$), 2.24 (1H, m, $\text{C}_6\text{-H}$), 2.28–2.47 (3H, m, $\text{C}_2\text{-H}_2$, $\text{C}_{7a}\text{-H}$), 2.56 (1H, dt, $J=12.5$, 7.3 Hz, $\text{C}_3\text{-H}$), 3.07 (1H, dt, $J=7.3$, 12.5 Hz, $\text{C}_{3a}\text{-H}$), 4.13–4.29 (2H, m, $\text{CO}_2\text{CH}_2\text{Me}$), 6.90 (1H, s, $\text{C}_5\text{-H}$); $^{13}\text{C-NMR}$ (CDCl_3) δ : 13.93, 14.27, 22.64, 26.26, 28.14, 28.77, 34.68, 36.03, 36.19, 38.11, 46.70, 60.45, 131.36, 144.21, 166.83, 220.54; EI-MS m/z : 265 ($\text{M}^+ + \text{H}$), 264 (100, M^+), 147 (85); HRMS m/z (M^+): calcd. for $\text{C}_{16}\text{H}_{24}\text{O}_4$, 264.1725; found, 264.1741.

11d: IR ν_{\max} (film) cm^{-1} : 2930, 1744, 1716, 1645, 1464, 1381, 1251, 1144, 1097, 888, 756; $^1\text{H-NMR}$ (CDCl_3) δ : 0.90 (3H, t, $J=7.3$ Hz, $\text{CH}_2\text{CH}_2\text{CH}_2\text{Me}$), 1.18–1.42 (7H, m, $\text{C}_7\text{-H}$, $\text{CH}_2\text{CH}_2\text{CH}_2\text{Me}$), 1.32 (3H, t, $J=7.3$ Hz, $\text{CO}_2\text{CH}_2\text{Me}$), 1.89–2.05 (2H, m, $\text{C}_3\text{-H}$, $\text{C}_7\text{-H}$), 2.10 (1H, m, $\text{C}_6\text{-H}$), 2.15–2.35 (3H, m, $\text{C}_2\text{-H}_2$, $\text{C}_3\text{-H}$), 2.47 (1H, q, $J=6.0$ Hz, $\text{C}_{7a}\text{-H}$), 3.27 (1H, q, $J=6.0$ Hz, $\text{C}_{3a}\text{-H}$), 4.19–4.28 (2H, m, $\text{CO}_2\text{CH}_2\text{Me}$), 6.96 (1H, d, $J=3.0$ Hz, $\text{C}_5\text{-H}$); $^{13}\text{C-NMR}$ (CDCl_3) δ : 14.06, 14.37, 22.80, 25.87, 27.11, 29.18, 33.20, 34.57, 35.99, 36.94, 45.82, 60.57, 131.19, 145.26, 167.09, 221.32; EIMS m/z : 265 (26, $\text{M}^+ + \text{H}$), 264 (100, M^+), 147 (49); HRMS m/z (M^+): calcd. for $\text{C}_{16}\text{H}_{24}\text{O}_4$, 264.1725; found, 264.1724.

(3*aS**, 7*aS**)-2,3,3*a*,6,7,7*a*-hexahydro-1-oxo-1*H*-indene-4-carboxylic acid (**12a**). By the same method as that described for the synthesis of (\pm)-**1**,^{5b)} **12a** (209 mg) was obtained as colorless crystals [crystallized from acetone- H_2O (6:4)]; mp 109–110°C; IR ν_{\max} (KBr) cm^{-1} : 2943, 1737, 1673, 1429, 1284, 1246, 1201, 1148, 1093, 930, 722; $^1\text{H-NMR}$ (CDCl_3) δ : 1.60–1.74 (2H, m, $\text{C}_7\text{-H}_2$), 1.83 (1H, dt, $J=6.0$, 12.2 Hz, $\text{C}_3\text{-H}$), 2.21–2.29 (4H, m, $\text{C}_2\text{-H}_2$, $\text{C}_6\text{-H}_2$), 2.31–2.46 (2H, m, $\text{C}_3\text{-H}$, $\text{C}_{7a}\text{-H}$), 3.14 (1H, q, $J=6.0$ Hz, $\text{C}_{3a}\text{-H}$), 7.19 (1H, t, $J=4.0$ Hz, $\text{C}_5\text{-H}$), 11.0 (1H, br., COOH); $^{13}\text{C-NMR}$ (CDCl_3) δ : 19.36, 24.15, 27.32, 35.46, 37.09, 46.51, 131.05, 143.36, 172.13, 220.68; EIMS m/z : 181 (12, $\text{M}^+ + \text{H}$), 180 (100, M^+); HRMS m/z (M^+): calcd. for $\text{C}_{10}\text{H}_{12}\text{O}_3$, 180.0786; found, 180.0786.

(3*aS**,6*R**,7*aS**)-2,3,3*a*,6,7,7*a*-hexahydro-6-methyl-1-oxo-1*H*-indene-4-carboxylic acid (**12b**). By the same method,^{5b)} **12b** (60 mg) was obtained as colorless crystals [crystallized from acetone- H_2O (6:4)]; mp 129–130°C;

IR ν_{\max} (KBr) cm^{-1} : 2948, 1733, 1683, 1628, 1470, 1425, 1290, 1140, 1058, 946, 727; $^1\text{H-NMR}$ (CDCl_3) δ : 1.08 (1H, q, $J=13.2$ Hz, $\text{C}_7\text{-H}$), 1.14 (3H, d, $J=7.3$ Hz, Me), 1.57 (1H, ddd, $J=8.5$, 11.8, 13.2 Hz, $\text{C}_3\text{-H}$), 1.87 (1H, dt, $J=13.2$, 4.9 Hz, $\text{C}_7\text{-H}$), 2.23–2.47 (4H, m, $\text{C}_2\text{-H}_2$, $\text{C}_6\text{-H}$, $\text{C}_{7a}\text{-H}$), 2.56 (1H, dt, $J=11.8$, 6.6 Hz, $\text{C}_3\text{-H}$), 3.07 (1H, dt, $J=11.8$, 7.3 Hz, $\text{C}_{3a}\text{-H}$), 7.02 (1H, s, $\text{C}_5\text{-H}$), 11.0 (1H, br., COOH); $^{13}\text{C-NMR}$ (CDCl_3) δ : 20.29, 28.07, 28.21, 31.36, 35.64, 38.13, 46.70, 130.55, 148.09, 171.97, 220.09; EIMS m/z : 195 (13, $\text{M}^+ + \text{H}$), 194 (100, M^+); HRMS m/z (M^+): calcd. for $\text{C}_{11}\text{H}_{14}\text{O}_3$, 194.0943; found, 194.0958.

(3*aS**,6*R**,7*aS**)-6-ethyl-2,3,3*a*,6,7,7*a*-1-oxo-1*H*-indene-4-carboxylic acid [(\pm)-coronafacic acid; **12c**]. Colorless crystals (140 mg). The synthesis and spectral data for **12c** have been described elsewhere.^{5b)}

(3*aS**, 6*R**, 7*aS**)-6-butyl-2,3,3*a*,6,7,7*a*-hexahydro-1-oxo-1*H*-indene-4-carboxylic acid (**12d**). By the same method,^{5b)} **12d** (153 mg) was obtained as colorless crystals [crystallized from acetone- H_2O (6:4)]; mp 145–146°C; IR ν_{\max} (KBr) cm^{-1} : 2955, 1736, 1683, 1627, 1468, 1430, 1284, 1141, 1075, 950, 728; $^1\text{H-NMR}$ (CDCl_3) δ : 0.90 (3H, t, $J=7.3$ Hz, $\text{CH}_2\text{CH}_2\text{CH}_2\text{Me}$), 1.09 (1H, dt, $J=11.2$, 12.9 Hz, $\text{C}_7\text{-H}$), 1.24–1.68 (7H, m, $\text{C}_3\text{-H}$, $\text{CH}_2\text{CH}_2\text{CH}_2\text{Me}$), 1.88 (1H, dt, $J=12.9$, 5.0 Hz, $\text{C}_7\text{-H}$), 2.22–2.47 (4H, m, $\text{C}_2\text{-H}_2$, $\text{C}_6\text{-H}$, $\text{C}_{7a}\text{-H}$), 2.60 (1H, dt, $J=13.1$, 6.3 Hz, $\text{C}_3\text{-H}$), 3.07 (1H, dt, $J=12.2$, 6.3 Hz, $\text{C}_{3a}\text{-H}$), 7.06 (1H, s, $\text{C}_5\text{-H}$), 11.0 (1H, br., COOH); $^{13}\text{C-NMR}$ (CDCl_3) δ : 13.95, 22.64, 26.09, 28.05, 28.75, 34.50, 35.94, 36.26, 38.12, 46.60, 130.69, 147.15, 172.11, 220.34; EIMS m/z : 237 (18, $\text{M}^+ + \text{H}$), 236 (100, M^+); HRMS m/z (M^+): calcd. for $\text{C}_{14}\text{H}_{20}\text{O}_3$, 236.1412; found, 236.1380.

(3*aS**,6*S**,7*aS**)-2,3,3*a*,6,7,7*a*-hexahydro-6-methyl-1-oxo-1*H*-indene-4-carboxylic acid (**13b**). By the same method,^{5b)} **13b** (33 mg) was obtained as a colorless oil; IR ν_{\max} (film) cm^{-1} : 2961, 1733, 1682, 1635, 1456, 1404, 1274, 1146, 1036, 885, 760; $^1\text{H-NMR}$ (270 MHz) δ : 1.06 (3H, d, $J=6.9$ Hz, Me), 1.34 (1H, ddd, $J=5.0$, 7.9, 13.2 Hz, $\text{C}_7\text{-H}$), 1.96–2.10 (2H, m, $\text{C}_3\text{-H}$, $\text{C}_7\text{-H}$), 2.12–2.35 (4H, m, $\text{C}_2\text{-H}_2$, $\text{C}_3\text{-H}$, $\text{C}_6\text{-H}$), 2.51 (1H, dt, $J=5.0$, 7.9 Hz, $\text{C}_{7a}\text{-H}$), 3.25 (1H, q, $J=5.0$ Hz, $\text{C}_{3a}\text{-H}$), 7.07 (1H, d, $J=2.6$ Hz, $\text{C}_5\text{-H}$), 11.0 (1H, br., COOH); $^{13}\text{C-NMR}$ (CDCl_3) δ : 20.04, 26.88, 27.64, 28.34, 35.38, 36.77, 45.59, 130.15, 149.07, 172.31, 221.17; EIMS m/z : 195 (12, $\text{M}^+ + \text{H}$), 194 (100, M^+); HRMS m/z (M^+): calcd. for $\text{C}_{11}\text{H}_{14}\text{O}_3$, 194.0943; found, 194.0938.

(3*aS**,6*S**,7*aS**)-6-ethyl-2,3,3*a*,6,7,7*a*-1-oxo-1*H*-indene-4-carboxylic acid [(\pm)-6-*epi*-coronafacic acid; **13c**]. By the same method,^{5b)} **13c** (100 mg) was obtained as a colorless oil; IR ν_{\max} (film) cm^{-1} : 2963, 2633, 1733, 1683, 1464, 1418, 1277, 1146, 1054, 929, 891, 758; $^1\text{H-NMR}$ (CDCl_3) δ : 0.91 (3H, t, $J=7.3$ Hz, CH_2Me), 1.25–1.44 (3H, m, $\text{C}_7\text{-H}$, CH_2Me), 1.89–2.30 (6H, m, $\text{C}_2\text{-H}_2$, $\text{C}_3\text{-H}_2$, $\text{C}_6\text{-H}$, $\text{C}_{7a}\text{-H}$), 2.44 (1H, q, $J=6.0$ Hz, $\text{C}_{7a}\text{-H}$), 3.20 (1H, q, $J=6.0$ Hz, $\text{C}_{3a}\text{-H}$), 7.08 (1H, d, $J=2.3$ Hz, $\text{C}_5\text{-H}$), 11.0 (1H, br., COOH); $^{13}\text{C-NMR}$ (CDCl_3) δ : 11.38,

25.27, 26.87, 27.48, 34.86, 35.62, 36.73, 45.73, 130.49, 147.89, 172.25, 221.2; EIMS m/z : 209 (14, $M^+ + H$), 208 (100, M^+); HRMS m/z (M^+): calcd. for $C_{12}H_{16}O_3$, 208.1099; found, 208.1083.

(3*aS**, 6*S**, 7*aS**)-6-butyl-2,3,3*a*,6,7,7*a*-hexahydro-1-oxo-1*H*-indene-4-carboxylic acid (**13d**). By the same method,^{5b} **13d** (40 mg) was obtained as a colorless oil; IR ν_{\max} (film) cm^{-1} : 2930, 2632, 1740, 1683, 1635, 1458, 1419, 1275, 1145, 1066, 887; 1H -NMR ($CDCl_3$) δ : 0.88 (3H, t, $J=7.3$ Hz, $CH_2CH_2CH_2Me$), 1.24–1.43 (7H, m, C_7 -H, $CH_2CH_2CH_2Me$), 1.92–2.04 (2H, m, C_3 -H, C_7 -H), 2.09–2.37 (4H, m, C_2 -H₂, C_3 -H, C_6 -H), 2.49 (1H, dt, $J=5.6, 7.3$ Hz, C_{7a} -H), 3.24 (1H, q, $J=5.6$ Hz, C_{3a} -H), 7.12 (1H, d, $J=2.0$ Hz, C_5 -H), 11.0 (1H, br., COOH); ^{13}C -NMR ($CDCl_3$) δ : 13.93, 22.64, 25.55, 26.90, 29.00, 33.30, 34.25, 35.60, 36.77, 45.61, 130.33, 148.12, 172.34, 221.22; EIMS m/z : 237 (22, $M^+ + H$), 236 (100, M^+); HRMS m/z (M^+): calcd. for $C_{14}H_{20}O_3$, 236.1412; found, 236.1418.

(3*aS*, 6*R*, 7*aR*)-6-ethyl-2,3,3*a*,6,7,7*a*-1-oxo-1*H*-indene-4-carboxylic acid (C_{7a} -*epi*-coronafacic acid). 1H -NMR (500 MHz, $CDCl_3$) δ : 1.02 (3H, t, $J=7.4$ Hz, CH_2Me), 1.36 (1H, m, CH_2Me), 1.49–1.65 (3H, m, CH_2Me , C_3 -H, C_7 -H), 1.94 (1H, dt, $J=3.0, 13.0$ Hz, C_2 -H), 2.05 (1H, dd, $J=3.8, 13.0$ Hz, C_2 -H), 2.25 (1H, ddd, $J=8.3, 9.5, 18.0$ Hz, C_{7a} -H), 2.39–2.46 (2H, m, C_3 -H, C_6 -H), 2.49 (1H, dd, $J=9.5, 18.0$ Hz, C_7 -H), 2.75 (1H, ddd, $J=5.0, 8.3, 13.5$ Hz, C_{3a} -H), 7.05 (1H, br.s, C_5 -H), 11.0 (1H, br., COOH).

Bioassays. Cell expansion-inducing activity was examined with cultures of tissue discs of potato tubers (*Solanum tuberosum* L. cv. Irish Cobbler) *in vitro*, as reported previously.⁷ In the range from 10^{-6} to 10^{-4} M concentration, all analogues of **1**, as well as **12c** (each as a racemate), exhibited almost no activity.

Tuber-inducing activity was examined with cultures of single-node segments of potato stems *in vitro*, as reported previously.⁸ The positive standard, a mixture of minor (\pm)-**3a** (ca. 7%) and major (\pm)-**4a**, was pre-

pared from a commercially available mixture of minor (\pm)-**3b** and major (\pm)-**4b**.^{2e}

Acknowledgments

We are grateful to Dr. H. Matsuura and Miss Y. Fujino in our laboratory for their assistance in the tuber-inducing assay, and also to Mr. K. Watanabe and Dr. E. Fukushima in our faculty for measuring the MS data.

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