## A New Synthetic Route to Mesoionic Thiazoles

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Summary Mesoionic thiazoles are obtained in good yield by the reaction of gem-dicyano epoxides with thioamides in a neutral medium.

ANHYDRO-5-HYDROXYTHIAZOLIUM HYDROXIDES (3) are masked 1,3 dipoles and are useful synthetically in heterocyclic chemistry, 1 but only a few synthetic routes leading to them are known. 2 We report here a new synthetic route to the mesoionic compounds (3).

TABLE. Preparation of the mesoionic thiazoles (3)a

x	$R^1$	R²	M.p. (t/°C)	Yield (%)	$ \nu_{\rm C=O}/{\rm cm^{-1}} $ (CCl <sub>4</sub> )
H	Ph	$\operatorname{Ph}$	270 <sup>b</sup>	72	1630
C1	Ph	$\mathbf{P}\mathbf{h}$	300	94	1630
$NO_2$	Ph	$\mathbf{P}\mathbf{h}$	273	70	1654
Cl	${ m Ph}$	$PhCH_2$	168	65	1625
$NO_2$	Ph	PhCH <sub>2</sub>	210	71	1638
Cl	Me	$\operatorname{Ph}$	180	30	1716, 1 <b>62</b> 8
$NO_2$	Me	$\operatorname{Ph}$	280	60	1644

 $^{\rm a}$  Combustion analyses and mass spectra of the compounds herein are in agreement with this structure.  $^{\rm b}$  Ref. 5.

The gem-dicyano-epoxides<sup>3</sup> (1) react in solution in dioxan or acetone with stoicheiometric quantities of the

thioamides (2) (Scheme 1).4 In most cases the reaction is complete after 24 h at room temperature and the meso-

ionic thiazoles (3) are obtained after evaporation. They are usually deep red, with the ring carbonyl i.r. band in the

Scheme 2. For (3),  $\delta$ (Me) 2·48(s); for (4);  $\delta$ (CH) 5·18(s),  $\delta$ (CH<sub>2</sub>) 4·34 (ABq,  $J_{AB}$  2·5 Hz). All signals disappear on addition of CD<sub>3</sub>CO<sub>2</sub>D.

range<sup>2</sup> 1620—1650 cm<sup>-1</sup> (Table). It is interesting that compound (3; X = Cl,  $R^1 = Me$ ,  $R^2 = Ph$ ) shows two carbonyl bands in solution in CCl4, whereas the solid (Nujol mull) shows only one band, at 1623 cm<sup>-1</sup>. Its n.m.r. spectrum (CHCl<sub>3</sub>) shows the existence of a tautomeric equilibrium (3) ⇒ (4) (Scheme 2).

This result suggests that the compound obtained by Ohta

et al., 5 is not a mesoionic compound (3; X = H,  $R^1 = Me$ ,  $R^2 = Ph$ ). The unusually high value of the carbonyl band  $1710 \, \mathrm{cm^{-1}}$  (KBr) observed for this compound, is in best agreement with a tautomeric form similar to (4).

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