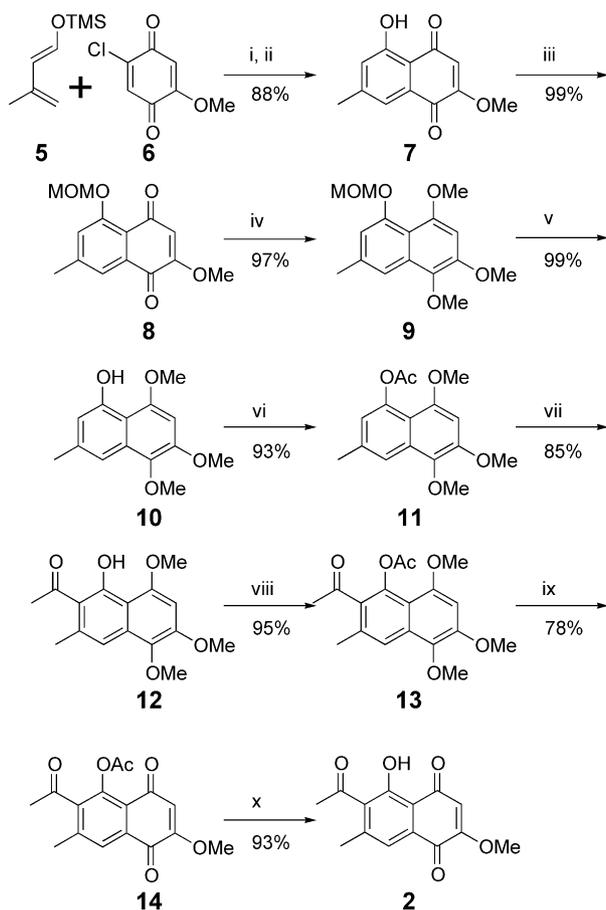


Synthesis of 2-Methoxystyandrone

The quantity of 2-methoxystyandrone isolated from the plant extract was very small and resupply of the compound for further biological evaluations had to be accomplished by a total synthesis. A 14-step total 9% overall yielding synthesis of this compound was reported by Hughes and Sargent.^{6b} However, we decided to design a new synthesis that would be more amenable for analogue synthesis for SAR. In our synthetic approach we took advantage of the regio-selective Diels–Alder reactions⁷ of suitably substituted chloro-benzoquinone and appropriately substituted *O*-silyloxy butadiene followed by oxidation for the construction of the naphthoquinone. A 10-step 45% overall yielding synthesis (Scheme 1) of **2** follows.

Thus, Diels–Alder reaction of 1-trimethylsilyloxy-1,3-butadiene⁸ (**5**) with 2-methoxy-5-chloro-benzoquinone (**6**) in refluxing benzene followed by Jones oxidation afforded naphthoquinone (**7**) in 88% yield. MOM protection of the phenolic group produced compound **8** in 99% yield, which was reduced with sodium dithionite and alkylated with dimethyl sulfate to give trimethyl ether (**9**) in 97% yield. Deprotection of the MOM ether

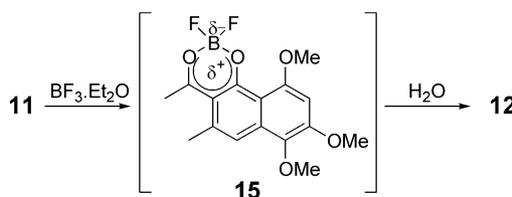


Scheme 1. Reagents and conditions: (i) Benzene, reflux, 24 h; (ii) Jones oxidation; rt, 20 min; (iii) MOMCl, DIPEA, CH₂Cl₂, 0 °C to rt, 16 h; (iv) (a) Sodium dithionite, dioxane–H₂O, 1:1, rt, 15 min; (b) KOH, dimethyl sulfate, rt, 2 h; (v) dioxane–MeOH–H₂O–H₂SO₄, 3:6:2:0.8, rt, 16 h; (vi) Ac₂O, C₅H₅N, rt, 20 h; (vii) BF₃·Et₂O, 60 °C, 35 min; (viii) Ac₂O, C₅H₅N, rt, 20 h; (ix) CAN (excess), CH₃CN–H₂O, 3:1, 30 min; (x) KOH, MeOH–H₂O, 6:1, 0.5 min, 6 N HCl.

turned out to be challenging but was efficiently accomplished by using a mixture of dioxane–MeOH–H₂O–H₂SO₄ (3:6:2:0.8), which gave deprotected product **10** in 99% yield after chromatography. Standard acetylation of compound **10** furnished compound **11** in 93% yield. Fries rearrangement of compound **11** with BF₃·Et₂O at 60 °C exclusively produced *ortho*-acetylated product **12** in 85% yield (vide infra), which was acetylated to give acetate **13** in 95% yield. Oxidation of compound **13** with ceric ammonium nitrate (CAN) in a mixture of CH₃CN–H₂O yielded⁹ 78% of *para*-quinone **14**, which upon saponification furnished **2** which was identical (TLC, HPLC, ¹H NMR) to natural product 2-methoxystyandrone.

Fries Rearrangement

BF₃·Et₂O catalyzed Fries rearrangement of compound **11** produced stable BF₂-adduct **15** (Scheme 2) which was purified by silica gel chromatography and was characterized by spectral analysis.¹⁰ It showed a molecular ion at *m/z* 338 (with *m/z* 337, which was 20% of *m/z* 338 peak). The ¹¹B NMR spectrum showed one signal for the boron nucleus at δ 0.20 (relative to external reference of BF₃·Et₂O at δ 0.0). The ¹⁹F NMR spectrum of **15** showed two signals for equivalent fluorine atoms at δ –147.52 and δ –147.59 (relative to internal C₆F₆ at δ –163.0) in a 1:4 ratio consistent with the natural abundance ratios of ¹⁰B and ¹¹B, respectively. In addition, the broad band decoupled ¹³C NMR spectrum of **15** showed two triplets at δ 193.9 (*J* = 1.7 Hz) and δ 171.1 (*J* = 2.2 Hz) due to covalent bonding of the boron atom to CH₃CO and the aromatic oxygen. The hydrolysis of the adduct in water was slower (45% conversion in 30 min and 100% after overnight as observed by ¹H NMR analysis in CD₂Cl₂ and addition of D₂O) and was fast in MeOH.



Scheme 2.

Sources of Analogues

Compound **16** (Table 1) was prepared by standard acetylation of compound **7**. Compounds **17** and **18** were prepared by a method described by Jung et al.¹¹ Compound **19** was isolated as an occasional by-product of the Jones oxidation reaction during preparation of **7**. 5-Hydroxy-7-methyl-1,4-naphthoquinone (**20**) was prepared by analogous cycloaddition reactions of **5** with 2-chloroquinone followed by oxidation. Compounds **21**–**26** were acquired from Aldrich Chemical Company. Xanthomegnin (**27**) was obtained from our internal collections.

Table 1. List of compounds and biological activities

Compd	Structure	HRV3C (IC ₅₀ , μM)	Papain (IC ₅₀ , μM)	Compd	Structure	HRV3C (IC ₅₀ , μM)	Papain (IC ₅₀ , μM)
1		47	NT	19		10.5	NT
2		4.6	23	20		6.4	NT
3		> 150	NT	21		26	NT
4		> 150	NT	22		> 150	NT
7		2.4	NT	23		21	76% (29 μM)
14		8.1	33	24		8.2	67% (120 μM)
16		40	NT	25		1.4	24% (91 μM)
17		124	NT	26		46.5	45% (18 μM)
18		> 150	NT	27		6	NT

NT, not tested.

Structure–Activity Relationship

2-Methoxystypanone (**2**) was evaluated in HRV 3C-protease HPLC assay.^{2a} It showed an IC₅₀ value of 4.6 μM and the inhibition was time dependent. In order to measure the selectivity of this compound against other cysteine proteases, it was evaluated against papain. It showed 5-fold selectivity for HRV 3C-protease over papain (Table 1). Naphthoquinone without acetyl substitution (cf. **7**) was 2-fold more potent than the corresponding acetylated compound **2**. Acetylation of the phenolic group caused 2-fold reduction (cf. **14**) compared to the parent compound **2** and over 16-fold reduction (cf. **16**) compared to compound **7**. Elimination of either phenolic (**19**) or methoxy (cf. **20**) groups from **7** caused 5- or 3-fold diminution of the HRV 3C-protease activity. Elimination of both the methyl and the hydroxy groups of **7** caused more significant reduction in HRV 3C-protease activity (cf. **21**) but the elimination of methoxy group (cf. **23**), the third substituent, did not have any deleterious effect on the potency (26 vs 21 μM). However, it appears that the elimination of all

three substituents (cf. **23**) led to the elimination of the selectivity for HRV 3C-protease. The transposition of the methoxy group from C-2 to C-3 caused significant reduction in the activity (**14** vs **17**). Complete loss of the activity was observed when the methoxy group was replaced with a hydroxy group (**21** vs **22**) and (**14**, **17** vs **18**). The *ortho*-quinones were generally more potent than corresponding *para*-quinones (**23** vs **24**). The size of the quinones has little effect on the protease activity. For example, complex tricyclic naphthoquinones such as thysanone (**1**) and xanthomegnin (**27**) showed IC₅₀ values of 47 and 6 μM, respectively. Substituted 1,4-antraquinones such as emodin (**3**) and methylemodin (**4**) were inactive. In contrast the simple 9,10-phenanthraquinone was the most active compound of the series and showed an IC₅₀ value of 1.4 μM with a high degree of selectivity. Chrysenequinone (**26**) showed a modest activity (IC₅₀ 46.5 μM). The activity profiles of 1,4-quinones (cf. **2**) and 1,2-quinones (cf. **25**) were different. The former compound showed a time dependent inhibition which was not affected by pre-incubation reaction with DTT whereas the latter is

a time dependent inhibitor but inhibition is potentiated by reaction with DTT.

2-Methoxystypandrone and other active compounds reported in this letter contain reactive sites, which is consistent with most of the inhibitors of HRV 3C-protease reported in the literature that also possess the Michael addition site including clinical candidate AG-7088.^{3,4} The mode of action of these inhibitors are through irreversible inactivation of the enzyme.

In summary, we have described the discovery, efficient synthesis and HRV 3C-protease activity of 2-methoxystypandrone and analogues. In addition, we also described and compared the activities of *ortho*- versus *para*-quinones. It appears that the mode of action of the two classes of compounds is different.

Acknowledgements

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- Experimental details. Aqueous solution of CAN (300 mg in 0.4 mL H₂O) was added drop wise (in 1 h) to a CH₃CN solution of **13** (50 mg in 0.4 mL CH₃CN). After 3 h stirring at rt the reaction mixture was diluted with 10 mL H₂O, extracted with CH₂Cl₂ and chromatographed on silica gel column. Elution with 49:1 CH₂Cl₂-MeOH afforded 35 mg of **14** as a chromatographically homogeneous yellow powder.
- 15**: HREIMS (*m/z*) 338.1143 (calcd for C₁₆H₁₇O₅BF₂:338.1136), ¹H NMR (400 MHz, CD₂Cl₂) δ, 2.60 (3H, d, *J*=1.9 Hz), 2.79 (3H, s), 3.82 (3H, s), 4.02 (3H, s), 4.03 (3H, s), 6.58 (1H, s), 7.26 (1H, m), ¹³C NMR (100 MHz, CD₂Cl₂) 25.1, 28.3, 56.7, 57.1, 61.6, 95.6, 109.3, 114.4, 117.3, 134.4, 136.1, 136.7, 158.2, 160.7, 171.1 (t, *J*=2.2 Hz), 193.9 (t, *J*=1.8 Hz). **12**: ¹H NMR (250 MHz, CD₂Cl₂) δ, 2.34 (3H, d, *J*=0.8 Hz), 2.56 (3H, s), 3.84 (3H, s), 3.97 (3H, s), 4.05 (3H, s), 6.60 (1H, s), 7.31 (1H, t, *J*=0.8 Hz), 9.78 (1H, br s).
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