

125. The Synthesis of Trimethylcyclopentane-carboxylic Acids

by Bogdan Šolaja¹⁾, Joan Huguet²⁾, Martin Karpf³⁾, and André S. Dreiding*

Organisch-Chemisches Institut der Universität Zürich, Winterthurerstr. 190, CH-8057 Zürich

(20.V.86)

(1*RS*, 5*SR*)-2,2,5-Trimethylcyclopentane-1-carboxylic acid (**17**) and (1*r*, 2*RS*, 5*SR*)-1,2,5-trimethylcyclopentane-1-carboxylic acid (**19**) are the starting materials for the α -alkynone routes to ($\pm$)-capnellene and for similar efforts towards ptychanolide. Since **17** and **19** have, so far, been available only by a branching reaction from the same precursor, the cyanohydrin mixture **2/3**, a modified synthesis for **17** and a new one for **19** was developed (*Scheme 1*). The common precursor **2/3** was treated with POCl₃ which effected normal dehydration to **6** (47%, major path) in competition with Me migration to **8** and **9** (17%). The minor path to **8** and **9** could be reduced to 3% when SOCl₂ was used for the dehydration of **2/3**. This reaction was the basis for an improved synthesis of **17** from **1**, using the steps *b*, *e*, *i*, *r*, and *v* see *Scheme 1* in an overall yield of 35%. The POCl₃ reaction was also studied with the pure cyanohydrins **2** and **3**, the configurations of which were determined by an X-ray analysis of **2**. Me migration did not occur from **2** but only from **3** (25%), which has HO-C(1) and H-C(5) in a *cis* position. With SOCl₂, **3** underwent only 5% Me migration. The new synthesis of **19** started with **4** using the steps *h*, *n*, *p*, and *s* (see *Scheme 1*) in an overall yield of 68%.

1. Introduction. – In our recent syntheses of ($\pm$)-capnellene (**D**) [1] and ($\pm$)-isoptychanolide (**E**) [2], we used the acids **A** and **C**, respectively, for a cyclopentenone anellation sequence **F**→**H** which includes the high temperature α -alkynone cyclization step **G**→**H** (*cf.* [3] [4]). Both acids **A** and **C** were prepared from the same intermediate, the cyanohydrin **B**. The formation of **A** was straightforward, but the formation of **C** involved the unexpected (for our purpose [2] at that time fortunate) migration of a Me group. In view of the low yields (22% **A** and 9% **C**, both from **B**) and since more **C** was required for synthetic efforts towards ptychanolide (differing from **E** in the epoxy configuration), we restudied the transformations of **B**, thereby improving the preparation of **A**, and developed an alternative synthesis of **C**.

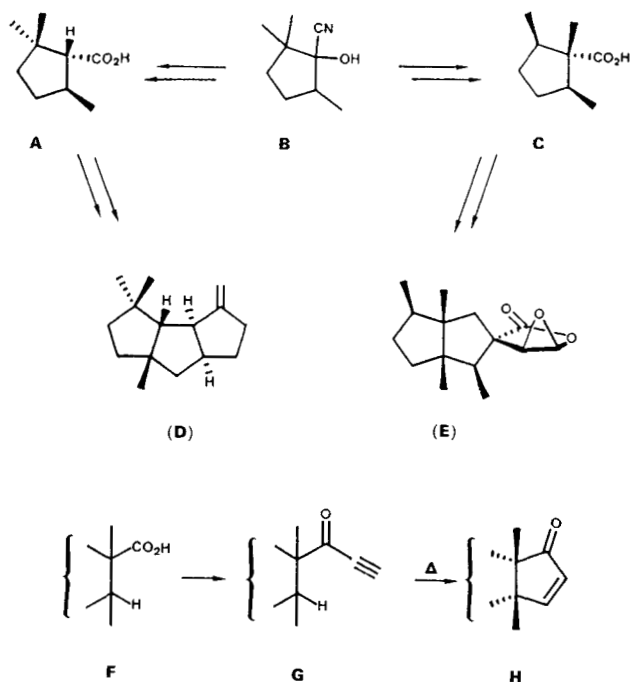
¹⁾ On leave of absence from Faculty of Science, University of Belgrade, 1985–1987.

²⁾ Present address: *Syntex SA*, DIPR 2822, 11000 Mexico City 10, D.F., Mexico.

³⁾ Present address: *F. Hoffmann-La Roche & Co. AG*, Zentrale Forschungseinheiten, CH-4002 Basel.

⁴⁾ The ratio of the nitriles **6**, **8**, and **9** was determined by anal. GC (column *BP-5* at 48°) and confirmed by the ¹H-NMR (400 MHz) signals of the mixture (see *Exper. Part, Exper. 4*). The minor products **8** and **9** had been overlooked in [1] and in our preliminary work (*cf.* [2]), because **8** showed the same retention time as **6** on several other GC columns and because the peak now attributed to **9** appeared to be too small (6%) to be a disturbing factor. This, together with the drastic conditions required for the saponification of **6**, made it appear as if the Me migration had taken place during the latter reaction. We are grateful to Prof. *Paul Bartlett* (University of California, Berkeley) for drawing our attention to the need of reevaluating this situation.

⁵⁾ Shaking the basic solution with Et₂O during workup removed the neutral materials, presumably unreacted **6** and partially saponified **12**.



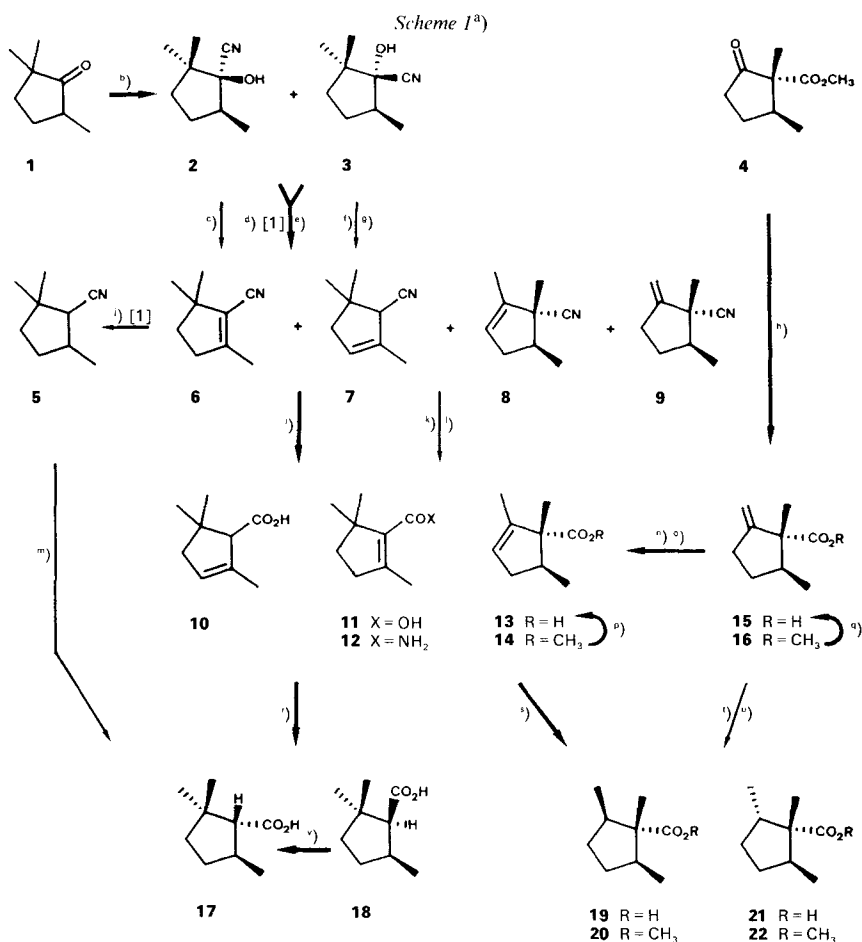
2. Results and Discussion. – All reactions leading to the acids **17** and **19**, which are our starting materials for capnellene and ptychanolide, respectively are given in *Scheme 1*. *Cyanohydrin Formation.* The cyanohydrins **2/3** (98%; 18:82 mixture⁶⁾) were obtained from **1** as in [1] by the general method of [5] and separated by *Lobar* chromatography (*Scheme 1, b*). An X-ray structure determination⁷⁾ (see the *Figure*) of the lower-melting, minor isomer proved the *trans*-configuration of HO–C(1) and H–C(5) on the 5-membered ring in **2** and thus their *cis*-configuration in **3**. The above ratio **3/2** implies a *ca.* 5-times faster attack of $^{\ominus}\text{CN}$ at C(1) *cis* to the Me group at C(5) than *trans*. It looks as if the transition state with pseudoaxial attack at the carbonyl C-atom and with CH_3 –C(5) in pseudoequatorial position (see *Scheme 2, 1*→**3**) is better than another one where one of these aspects is the other way round.

Dehydration of 2 and 3. In a typical *anti*-elimination, the cyanohydrin (OH–C(1) and H–C(5) *trans*) underwent a normal vicinal dehydration to the α , β -unsaturated nitrile **6** (70%) under the influence of POCl_3 in pyridine (*Scheme 1, c*; see *Scheme 2*). There was also 5% cyanohydrin cleavage, reforming the ketone **1**.

Dehydration of cyanohydrin **3** (HO–C(1) and H–C(5) *cis*) with POCl_3 led to only 57% of **6** along with the two β , γ -unsaturated nitriles **8** (24%) and **9** (5%; *Scheme 1, f*)⁴⁾. Evidently, the vicinal dehydration which, in this case, must be a *syn*-elimination is partly suppressed so that a 1,2-migration of one of the geminal Me groups can compete, leading to some **8** and **9** (*Scheme 3*).

⁶⁾ In [1], this ratio was erroneously given as 5:1.

⁷⁾ We thank P. Schönholzer of F. Hoffmann-La Roche & Co. AG, Basel, for this analysis. The X-ray data has been submitted to Cambridge Crystallographic Data Center.



^{a)} Bold-type arrows indicate the preferred synthetic paths to the acids **17** and **19**.

^{b)} (CH₃)₃SiCN, ZnI₂, CH₂Cl₂; HCl: **1**→**2/3** (18:82; 98%).

^{c)} POCl₃, Py, reflux: **2**→**1/6** (6:94; 75%).

^{d)} POCl₃, Py, reflux [1]: **2/3** (18:82)→**6/8/9** (74:20:6; 64%⁴).

^{e)} SOCl₂ (neat), reflux: **2/3** (18:82)→**6/7/8** (86:11:3; 82%).

^{f)} POCl₃, Py, reflux: **3**→**1/6/8/9** (2:68:24:6; 84%).

^{g)} SOCl₂ (neat), reflux: **3**→**6/8** (94:6; 81%).

^{h)} Ph₃PCH₂, THF, r.t.: **4**→**16** (85%).

ⁱ⁾ KOH, diethylene glycol, 200°, 72 h: **6/7/8** (86:11:3)→**10/11/13** (21:64:15; 99% recovery); on chromatography, **10/11/13** (11:79:10; 54%).

^{j)} CuH, THF, r.t. [1]: **6/8/9** (74:20:6)→**5** (69%).

^{k)} KOH, diethylene glycol, 200°, 18 h: **1/6** (6:94)→**6/10/11/12** (18:6:57:19; 77% recovery).

^{l)} KOH, diethylene glycol, 200°, 18 h: **1/6/8/9** (2:68:24:6)→**6/10/11/12/13** (2:3:30:21:44; 108% recovery); **6/8/9** (74:20:6)→**10/11/13** (9:24:67⁵); 69% recovery); on chromatography, **10** (0.4%), **11** (12%), **13** (14%).

^{m)} KOH, diethylene glycol, 200°, 18 h [1]: **5**→**17** (50%).

ⁿ⁾ RhCl₃·H₂O, EtOH, reflux: **16**→**14** (97%).

^{o)} KOH, diethylene glycol, 200°, 18 h: **16**→**13/15** (93:7; 83% recovery); on chromatography, **13** (52%).

^{p)} NaOH, MeOH/H₂O, 50°: **14**→**13** (87%).

^{q)} NaOH, MeOH/H₂O, 50°: **16**→**15** (89%).

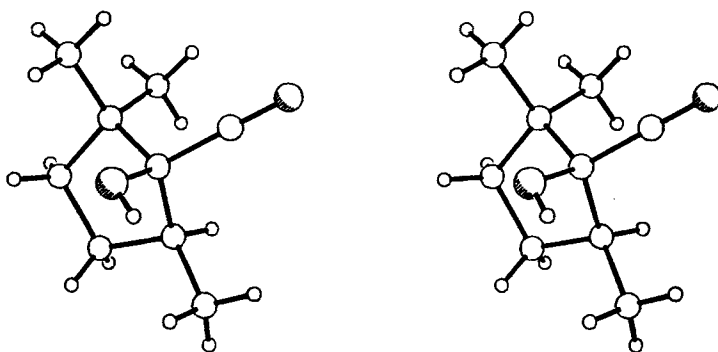
^{r)} H₂, 10% Pt/C, EtOAc: **10/11/13** (11:79:10)→**17/18/19** (23:69:8; 97%).

^{s)} H₂, 10% Pt/C, EtOAc: **13**→**19** (95%).

^{t)} H₂, 10% Pt/C, EtOAc: **16**→**20/22** (39:61; 98%).

^{u)} H₂, 10% Pt/C, EtOAc: **15**→**19/21** (37:63; 85%).

^{v)} NaH, BuLi, Et₂O, r.t.: **17/18/19** (23:69:8)→**17** (83%).

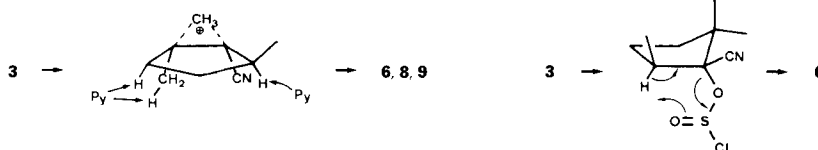
Figure. Stereoscopic view of the X-ray-determined structure of the cyanohydrin **2**

Scheme 2



The isomers **8** and **9** could not be separated from **6**, but their structures were derived from separately visible signals in the difference $^1\text{H-NMR}$ spectrum (**6** subtracted) of the mixture, namely from two signals due to $\text{CH}_3\text{-CHR}_2$ and two due to $\text{CH}_3\text{-CR}_3$ for **8** and **9**, and from two due to $\text{CH}_3\text{-CR=CHR}$ for **8** and two due to $\text{CH}_2\text{=CR}_2$ for **9** (see *Exper. Part, Exper. 4*).

Scheme 3



When neat SOCl_2 was used as dehydrating agent for **3** instead of POCl_3/Py , the product was mostly **6** (only 5% of **8**; *Scheme 1, g*). SOCl_2 dehydration of the 18:82 mixture **2/3** afforded 71% of **6**, 9% of **7** and only 2% of rearranged product **8** (*Scheme 1, e*). The structure of **7** was derived from the separately visible signals in the difference $^1\text{H-NMR}$ spectrum (**6/8** subtracted) of the product mixture. The SOCl_2 reaction seems to involve a concerted *syn*-elimination of the intermediate chlorosulfite [6] of **3** (see *Scheme 3*), so that the Me migration is suppressed. The higher ratio of rearranged (**8** and **9**) to unrearranged (**6**) products with POCl_3/Py (30:68) is attributed to a participation of the Me group (*Scheme 3*). In accord with this mechanism is that the rearranged nitriles **8** and **9** possess the (1*RS*,5*SR*)-configuration, meaning that only the Me group *trans* to the nucleofuge had migrated.

Saponification of the Nitriles. The α,β -unsaturated nitrile **6** was highly resistant to saponification. Even after heating it for 18 h at 200° with KOH in diethylene glycol about 14% of **6** remained unchanged and as much as 15% were saponified only to the amide

stage, *i.e.* to **12** (*Scheme 1, k*). To the extent of *ca.* 5%, the double bond migrated into the β,γ -position to yield **10**, while the rest (44%) was transformed to the expected α,β -unsaturated acid **11**. Evidently, no Me migration had taken place under the alkaline conditions⁴). The saponification of **6/8/9** (68:24:6; containing 2% of **1**) yielded a 2:3:30:21:44 mixture **6/10/11/12/13** (*Scheme 1, l*). From a larger-scale experiment, it was possible to isolate the unsaturated acids **10** (0.4%), **11** (12%)⁸), and **13** (14%), after removal of **6** and **12** during workup.

Synthesis of 19. The β,γ -unsaturated acid **13**, so far available only as a minor by-product (9% in 3 steps from **1**) [1], was synthesized specifically starting with the known keto ester **4** [7] (72% in 3 steps): A Wittig reaction transformed **4** into the unsaturated ester **16** (*Scheme 1, h*). Attempts to hydrogenate the exocyclic double bond in **16** or in the corresponding acid **15** led to *ca.* 2:3 mixtures of the stereoisomeric esters **20** and **22** (98%; *Scheme 1, t*) or acids **19** and **21** (85%; *Scheme 1, u*), respectively. To avoid the formation of the undesired stereoisomers **21** and **22**, the double bond of the ester **16** was first shifted into the ring with a Rh(III) catalyst [4] to give the endocyclic isomer **14** (97%; *Scheme 1, n*) and hence the acid **13**. Hydrogenation of **13** afforded the saturated acid **19** (95%) of *C_s* symmetry, which is the starting material needed for our planned further synthetic work in the ptychanolide field.

Complete saponification of a 86:11:3 mixture **6/7/8** gave a 11:79:10 mixture of the acids **10**, **11**, and **13** (54%; *Scheme 1, i*). Hydrogenation of this mixture over Pt/C led to the saturated acids **17/18/19** (ratio 23:69:8; 97%; *Scheme 1, r*). The structure of **18** was derived from the separately visible signals in the difference ¹H-NMR spectrum (**17** [1]/**19** subtracted). The desired acid **17** was obtained (83%) from the Na salt of **18** by enolization (60 min) with BuLi at r.t. (*Scheme 1, v*); it is identical with the substance used in [1] as the starting material of our capnellene synthesis.

This work was supported by the Swiss National Science Foundation and by Sandoz AG, Basel. We acknowledge the grant of a post-doctoral fellowship from the Legerlotz Fund to B.Š.

Experimental Part

General. See [3]. Compounds which were separated analytically by GC (to obtain product ratios) or preparatively by Lobar chromatography are always listed in the order of their elution sequence. NMR: Varian XL-200 and Bruker AM-400; ¹H-NMR decoupling: irradiated signal in square brackets [δ], followed by the multiplicity and *J* of the ensuing signal.

1. (1RS,5RS)- and (1RS,5SR)-1-Hydroxy-2,2,5-trimethylcyclopentane-1-carbonitriles (**2** and **3**). A stirred soln. of 2,2,5-trimethylcyclopentanone (**1**); 15.28 g, 121 mmol [1] and ZnI₂ 0.86 g, 3 mmol in CH₂Cl₂ (120 ml) was allowed to react with (CH₃)₃SiCN [5] (18 ml, 145 mmol) for 4 h. The solvent was evaporated to leave a mixture of trimethylsilyloxy-nitriles which was dissolved in THF (20 ml) and treated with 10% HCl (50 ml) for 24 h at 40°. The cooled mixture was extracted with Et₂O (5 × 80 ml), and the combined extracts were washed with 10% Na₂S₂O₃ soln. (20 ml), H₂O, and brine, dried over MgSO₄, evaporated, and dried at 22°/14 Torr leaving a colourless solid consisting of **2** and **3** (18.16 g, 98%) in a ratio 18:82^b) (by anal. GC (SE-52, 91°)). Chromatography (Lobar C, hexane/EtOAc 97:3) of 4.35 g of this solid afforded (in two runs) 0.70 g (16%) of **2** and 3.40 g (76%) of **3**. 2:

⁸) A quantum-mechanical study on the geometry [8] and a crystal-structure analysis [9] of 2,5,5-trimethylcyclopent-1-ene-1-carboxylic acid (**11**) have been reported. The sample used in the latter work had been prepared by Prof. H. O. Huisman and collaborators, University of Amsterdam, in connection with the synthesis of the 5-membered ring analog of vitamin A (private communication on unpublished work).

Colourless needles, m.p. 49.5–50.5° from hexane. IR (CHCl₃): 3600*m*, 3600–3130*m* (br.), 2970*s*, 2880*m*, 2235*w* (C≡N), 1465*m*, 1390*m*, 1380*m*, 1370*m*, 1130*m*, 1040*m*, 1000*m*, 990*m*. ¹H-NMR (200 MHz, CDCl₃): 2.68–2.46 (*m*, H–C(5)); 2.28–1.30 (*m*, 2 H–C(3), 2 H–C(4), HO; 1 H exchangeable with D₂O); 1.17 (*s*, 2 CH₃–C(2)); 1.15 (*d*, *J* = 6.2, CH₃–C(5)). MS (70 eV): 153 (2, *M*⁺), 138 (22, *M*⁺ – 15), 126 (25), 111 (10), 97 (17), 84 (100), 70 (29), 55 (62).

3: Colourless needles, m.p. 57.5–58.6° from hexane. IR (CHCl₃): 3590*m*, 3600–3100*m* (br.), 2970*s*, 2880*m*, 2235*w* (C≡N), 1470*m*, 1390*m*, 1375*m*, 1160*m*, 1080*s*. ¹H-NMR (200 MHz, CDCl₃): 2.49 (*s*, HO, exchangeable with D₂O); 2.42–2.18 (*m*, H–C(5)); 2.08–1.85 (*m*, 1 H); 1.78–1.52 (*m*, 2 H); 1.48–1.28 (*m*, 1 H); 1.24 (*d*, *J* = 6.9, CH₃–C(5)); 1.23 (*s*, CH₃–C(2)); 1.03 (*s*, CH₃–C(2)). MS (70 eV): 153 (1, *M*⁺), 138 (64, *M*⁺ – 15), 126 (22), 111 (47), 97 (68), 83 (29), 70 (96), 55 (100).

2. *Dehydration of 2 with POCl₃ in Pyridine*. A stirred soln. of pure **2** (450 mg, 2.9 mmol) in pyridine (3.8 ml) was treated at r.t. with POCl₃ (1.0 ml, *ca.* 11.2 mmol) and then heated to reflux for 5 h. The cooled mixture was poured onto ice (15 g) and conc. HCl (3.5 ml). After separation of the org. layer, the aq. phase was extracted with Et₂O (6 × 5 ml). The combined org. phases were washed with H₂O and brine, dried over MgSO₄, and evaporated to leave 289 mg (*ca.* 75%) of yellow oil consisting of the ketone **1** and 2,5,5-trimethylcyclopent-1-ene-1-carbonitrile (**6**) in a ratio 6:94 (anal. GC (SE-52, 94°)). An anal. sample of **6** was obtained by column chromatography (Lobar A, C₆H₆), followed by bulb-to-bulb distillation at 119°/35 Torr. IR (film): 2960*s*, 2870*s*, 2845*m*, 2220*s* (C≡N), 1645*s*, 1380*m*, 1370*s*, 1330*w*. ¹H-NMR (400 MHz, CDCl₃): 2.42 (*td*, *J* = 7.2, 1.2, 2 H–C(3)); 1.93 (*t*, *J* = 1.2, CH₃–C(2)); 1.78 (*t*, *J* = 7.2, 2 H–C(4)); 1.16 (*s*, 2 CH₃–C(5)); ([1]: ¹H-NMR (90 MHz)). MS (70 eV): 136 (1, *M*⁺ + 1), 135 (9, *M*⁺), 120 (100, *M*⁺ – 15), 93 (32), 77 (13).

3. *Dehydration of 2/3 with SOCl₂*. A soln. of **2/3** (8.48 g, 55 mmol; ratio 18:82, anal. GC (SE-52, 91°)) in SOCl₂ (80 ml) was heated to reflux for 48 h, poured onto ice (400 g), stirred for 1 h, extracted with pentane (3 × 100 ml), washed with sat. NaHCO₃ and brine, dried over MgSO₄, and evaporated. Bulb-to-bulb distillation at 119°/35 Torr yielded 6.12 g (82%) of a colourless oil consisting of 2,5,5-trimethylcyclopent-2-ene-1-carbonitrile (**7**), **8**, and **6** in a ratio of 11:3:86 (anal. GC (BP-5, 48°)). Subtraction of the ¹H-NMR of **8/6** obtained in *Exper. 5* from the ¹H-NMR of **7/8/6** afforded the ¹H-NMR of **7**. ¹H-NMR (400 MHz, CDCl₃): 5.48–5.44 (*m*, H–C(3)); 3.17–3.14 (*m*, H–C(1)); 2.26, 2.21 (*AB* of *ABX*₃, *J*_{AB} = 16.3, *J*_{AX} = 2.2, *J*_{BX} = 2.3, 2 H–C(4)); 1.84–1.82 (*m*, *X*₃ of *ABX*₃, CH₃–C(2)); 1.20 (*s*, CH₃–C(5)); 1.12 (*s*, CH₃–C(5)). GC-MS (70 eV): 136 (4, *M*⁺ + 1), 135 (40, *M*⁺), 120 (100, *M*⁺ – 15), 93 (77), 77 (24).

4. *Dehydration of 3 with POCl₃ in Pyridine*. Using the procedure described in *Exper. 2*, 1.00 g (6.53 mmol) of **3** was transformed (reaction time 45 min) into 744 mg (*ca.* 84%) of a yellow oil consisting of **1**, (1*RS*,5*SR*)-1,2,5-trimethylcyclopent-2-ene-1-carbonitrile (**8**), **6** and (1*RS*,2*SR*)-1,2-dimethyl-5-methylidenecyclopentane-1-carbonitrile (**9**) in a ratio of 2:24:68:6 (anal. GC (BP-5, 48°)). We were not able to separate this mixture with several separation techniques (Lobar, HPLC or semi-prep. GC) under various conditions. The assignment of ¹H-NMR signals to certain protons of **8** and **9** was possible from the difference spectrum after computer subtraction of the ¹H-NMR of **6** from the ¹H-NMR of **6/8/9**. ¹H-NMR (400 MHz, CDCl₃): Signals due to **8**: 5.44–5.41 (*m*, H–C(3)); 2.71–2.61 (*m*, H–C(5)); 1.81–1.78 (*m*, CH₃–C(2)); 1.17 (*s*, CH₃–C(1)); 1.09 (*d*, *J* = 6.6, [2.65] *s*, CH₃–C(5)); signals due to **9**: 5.24, 5.04 (both *t*, *J* = 2.4, CH₂=C(5)); 1.23 (*s*, CH₃–C(1)); 1.04 (*d*, *J* = 6.8, [2.41] *s*, CH₃–C(2)); ¹H-NMR ratio of **6/8/9**, 70:20:7 (*cf.* GC ratio; ratio of **8/9** from olefinic signals (5.42 for **8**, 5.24/5.04 for **9**); ratio of **6** + **8/8** from the CH₃ signals (*ca.* 1.17 for **6** and for **8**) of the mixture spectrum and the difference spectrum (**6** subtracted).

5. *Dehydration of 3 with SOCl₂*. Using the procedure described in *Exper. 3*, 885 mg (5.8 mmol) of **3** were dehydrated (reaction time 15 h) to yield, after bulb-to-bulb distillation at 119°/35 Torr, 634 mg (81%) of a colourless oil consisting of **8/6** in a ratio of 6:94 (anal. GC (BP-5, 48°)).

6. *Methyl (1*RS*,2*SR*)-1,2-dimethyl-5-methylidenecyclopentane-1-carboxylate (16)*. To a stirred soln. of methyl 1,2-dimethyl-5-oxocyclopentane-*r*-1-carboxylate (**4**; 4.99 g, 29.3 mmol) [**7**] in THF (10 ml) under N₂, a *ca.* 0.45 *M* solution of methylidene(triphenyl)phosphorane (73 ml, *ca.* 33 mmol; prepared according to [10]) was added dropwise within 2 h at r.t. After 2 h of stirring at r.t., 1 ml of acetone was added, the soln. was stirred for 1 h and evaporated. The brown semi-solid residue was extracted with pentane (4 × 100 ml) and the combined extract evaporated. After bulb-to-bulb distillation at 85°/14 Torr, 4.21 g (85%) of **16** was obtained as a colourless oil of 92% purity (anal. GC (SE-52, 71°)). An anal. sample of **16** was obtained by chromatography (Lobar A, hexane/EtOAc 99:1) followed by bulb-to-bulb distillation. IR (film): 3080*w* (C=CH₂), 2955*s*, 2880*s*, 2840*w*, 1735*s* (C=O), 1655*m* (C=C), 1260*s*, 1170*s*, 1100*m*, 895*m*. ¹H-NMR (200 MHz, CDCl₃): 4.92, 4.89 (both *t*, *J* = 2.5, CH₂=C(5)); 3.69 (*s*, CH₃OOCC(1)); 2.73–2.52 (*m*, [0.93] *dd*, *J* = 11.0, 6.5, H–C(2)); 2.52–2.38 (*m*, 2 H); 1.94–1.78 (*m*, 1 H);

1.48–1.19 (*m*, 1 H); 1.14 (*s*, CH₃–C(1)); 0.93 (*d*, *J* = 7.0, CH₃–C(2)). MS (70 eV): 153 (2, *M*⁺ – 15), 139 (3), 109 (100), 93 (9), 67 (25). Anal. calc. for C₁₀H₁₆O₂ (168.24): C 71.39, H 9.59; found: C 71.64, H 9.33.

7. *Complete Saponification of 6/7/8*. A soln. of a 86:11:3 mixture 6/7/8 (1.71 g, 12.6 mmol) and KOH (7.10 g, 126.5 mmol) in diethylene glycol (15 ml) was heated at 200° for 72 h. After cooling, H₂O (40 ml) was added and the mixture extracted with Et₂O (5 × 60 ml). The combined Et₂O extracts yielded, after shaking with H₂O and brine, drying over MgSO₄, and evaporation, according to GC, not a trace of 6 or of 12. The aq. alkaline phase was acidified at 0° with conc. HCl and extracted with Et₂O (5 × 60 ml). The combined Et₂O extracts were washed with H₂O and brine, dried over MgSO₄, and evaporated to yield 1.70 g (99% recovery) of a brown solid which, by anal. GC (SE-52, 90°) of an aliquot esterified with an Et₂O soln. of CH₂N₂, contained 10/13/11 in a ratio of 21:15:64. *Lobar* purification (hexane/EtOAc/AcOH 975:25:10) of a 1.21 g sample afforded 0.75 g (54%) of a 11:10:79 mixture 10/13/11 as pale yellow solid.

8. *Incomplete Saponification of 6*. Using the procedure described in *Exper.* 7, 6 (239 mg, 1.8 mmol; containing 6% of 1) was saponified for 18 h and subjected to the same workup. The Et₂O extracts of the alkaline mixture contained 37 mg (15% recovery) of a brown oil which consisted of recovered 6 and 12 in a ratio of 94:5 (anal. GC (SE-52, 100°)). Acidification and subsequent extraction of the aq. alkaline soln. yielded 148 mg (62% recovery) of a brown oil which, by anal. GC (SE-52, 90°) of an aliquot esterified with an Et₂O soln. of CH₂N₂, contained 10/11/12 in a ratio of 7:71:22. ¹H-NMR (200 MHz): signals of all the compounds synthesized in later experiments separately visible.

9. *Saponification of 6/8/9*. Using the procedure described in *Exper.* 8, 271 mg (2.0 mmol) of 1/8/6/9 (ratio 2:24:68:6, anal. GC (BP-5, 48°)) was saponified and subjected to the same workup. The Et₂O extracts of the alkaline mixture contained 41 mg (15% recovery) of brown oil which contained 6/12 in a ratio of 14:51 (anal. GC (SE-52, 100°)). This oil was dissolved in boiling hexane, treated with charcoal, filtered, and the filtrate evaporated to yield 19 mg (6%) of 2,5,5-trimethylcyclopent-1-ene-1-carboxamide (12) as colourless solid with a purity of 93% (anal. GC (BP-5, 150°)). After crystallisation from hexane, 10 mg of 12 were obtained as colourless needles, m.p. 152.8–154.7°. IR (KBr): 3370s, 3175s, 2955m, 2935m, 2860m, 1640s (amide I), 1615s (amide II), 1410s, 1360m, 1115m, 850–620m (br.). ¹H-NMR (400 MHz, CDCl₃): 5.63, 5.40 (2 br. s, NH₂); 2.33 (*q*, *J* = 7.4, 1.1, 2 H–C(3)); 1.88 (*t*, *J* = 1.1, CH₃–C(2)); 1.69 (*t*, *J* = 7.4, 2 H–C(4)); 1.21 (*s*, 2 CH₃–C(5)). MS (70 eV): 153 (43, *M*⁺), 138 (100, *M*⁺ – 15), 121 (89), 109 (17), 95 (70), 84 (26), 77 (21), 67 (39), 55 (21), 44 (20). Anal. calc. for C₉H₁₅NO (153.22): C 70.55, H 9.87, N 9.14; found: C 70.66, H 9.91, N 9.42.

Acidification and subsequent extraction of the aq. soln. yielded 253 mg (93% recovery) of a brown oil which, by anal. GC (SE-52, 90°) of an aliquot esterified with an Et₂O soln. of CH₂N₂, contained 10/13/11/12 in a ratio of 4:46:33:14.

The same saponification was performed with 8/6/9 (10.44 g, 77 mmol; ratio 20:74:6; anal. GC (BP-5, 48°)) and KOH (43.30 g, 773 mmol) in diethylene glycol (100 ml) at 200° for 20 h. After cooling, H₂O (800 ml) was added, the mixture was acidified with conc. HCl and extracted with Et₂O (5 × 100 ml). The combined extracts were washed with H₂O and extracted with 1*N* NaOH (3×). The combined aq. alkaline solns. were acidified to pH 1–2 with conc. HCl and extracted with Et₂O (3 × 100 ml). The combined extracts were washed with brine, dried over MgSO₄, and evaporated to yield 7.14 g (69% recovery) of a brown oil which, by anal. GC (SE-52, 90°) of an aliquot esterified with an Et₂O solution of CH₂N₂, contained 10/13/11 in a ratio of 9:67:24. By repeated column chromatography (300 g of silica gel, hexane/EtOAc/AcOH 975:25:10), 1.45 g (12%) of 2,5,5-trimethylcyclopent-1-ene-1-carboxylic acid (11), 1.65 g (14%) of (1*RS*,5*SR*)-1,2,5-trimethylcyclopent-2-ene-1-carboxylic acid (13), and 50 mg (0.4%) of 2,5,5-trimethylcyclopent-2-ene-1-carboxylic acid (10) were obtained. 11⁸): Colourless solid, m.p. 75–78°. IR (CHCl₃): 3600–2300m (br.), 1675s (C=O), 1625m, 1410w, 1370w, 1360w, 1340w, 1310w, 1280m, 1040w. ¹H-NMR (400 MHz, CDCl₃): 2.42 (*q*, *J* = 7.4, 1.1, 2 H–C(3)); 2.09 (*t*, *J* = 1.1, CH₃–C(2)); 1.70 (*t*, *J* = 7.4, 2 H–C(4)); 1.24 (*s*, 2 CH₃–C(5)). MS (70 eV): 154 (10, *M*⁺), 139 (100, *M*⁺ – 15), 121 (22), 111 (21), 93 (43), 77 (18), 67 (18), 55 (13). Anal. calc. for C₉H₁₄O₂ (154.21): C 70.10, H 9.15; found: C 70.28, H 9.25.

13: Colourless solid, m.p. 48.5–51.0°. IR (KBr): 3600–2700s (br.), 1690s (C=O), 1650m, 1460m, 1450m, 1410m, 1380m, 1285s, 1185m, 1110m, 1020m, 955m, 810m. ¹H-NMR (200 MHz, CDCl₃): 5.44 (br. s, H–C(3)); 2.74 (*sext.*, *J* ca. 7.5, [1.03] *t*, *J* ca. 7.5, H–(5)); 2.58–2.37 (*m*, H–C(4)); 2.03–1.82 (*m*, H–C(4)); 1.76–1.66 (*m*, CH₃–C(2)); 1.10 (*s*, CH₃–C(1)); 1.03 (*d*, *J* = 6.9, CH₃–C(5)). MS (70 eV): 154 (9, *M*⁺), 125 (5), 109 (100), 93 (8), 81 (9), 67 (28), 55 (12). Anal. calc. for C₉H₁₄O₂ (154.21): C 70.10, H 9.15; found: C 69.87, H 8.95.

10: Colourless oil. IR (film): 3650–2200s (br.), 1705 (C=O), 1460w, 1410m, 1370w, 1310w, 1270m, 1220m. ¹H-NMR (200 MHz, CDCl₃): 5.49 (br. s, H–C(3)); 2.98 (br. s, H–C(1)); 2.32 (*dq*, *J* = 16, 2, H–(4)); 2.11 (*dm*, *J* = 16, H–(4)); 1.74 (br. s, CH₃–C(2)); 1.20, 1.11 (both *s*, 2 CH₃–C(5)). MS (70 eV): 154 (31, *M*⁺), 139 (7), 131

(7), 111 (100), 93 (16), 81 (10), 67 (40), 55 (19). Anal. calc. for $C_9H_{14}O_2$ (154.21): C 70.10, H 9.15; found: C 69.80, H 9.25.

10. *Methyl (1RS,5SR)-1,2,5-Trimethylcyclopent-2-ene-1-carboxylate (14)*. A soln. of **16** (2.74 g, 16.3 mmol) and $RhCl_3 \cdot H_2O$ (157 mg; ca. 40% Rh) in 95% EtOH (26 ml) was stirred for 17 h at reflux, cooled, diluted with H_2O (10 ml), and extracted with CH_2Cl_2 (5×10 ml). The combined org. extracts were washed with brine, dried over $MgSO_4$, and evaporated to leave a red oil which was filtered through a short SiO_2 column with hexane/ Et_2O 9:1. The filtrate was evaporated and bulb-to-bulb distilled at 70°/14 Torr to yield 2.67 g (97%) of **14** of 88% purity (anal. GC (SE-52, 71°)) as a yellow oil. An anal. sample resulted from chromatography (*Lobar A*, hexane/ $EtOAc$ 99:1), followed by bulb-to-bulb distillation. IR (film): 3040w, 2960s, 2870m, 1730s (C=O), 1653w (C=C), 1250s, 1125m, 1100s, 800m. 1H -NMR (200 MHz, $CDCl_3$): 5.42 (br. s, H-C(3)); 3.70 (s, $CH_3OOC-C(1)$); 2.78–2.58 (m, H-C(5)); 2.52–2.34 (m, H-C(4)); 2.00–1.80 (m, H-C(4)); 1.70–1.62 (m, $CH_3-C(2)$); 1.08 (s, $CH_3-C(1)$); 1.00 (d, $J = 7.1$, $CH_3-C(5)$). MS (70 eV): 168 (5, M^+), 139 (2), 109 (100), 93 (7), 81 (9), 67 (29). Anal. calc. for $C_{10}H_{16}O_2$ (168.24): C 71.39, H 9.59; found: C 71.65, H 9.85.

11. *Saponification of 16*. Using the procedure described in *Exper. 8*, 168 mg (1 mmol) of **16** was saponified to yield 139 mg (83% recovery) of a brown oil consisting of **15** and the isomeric acid **13** in a ratio of 7:93 (anal. GC (BP-5, 120°)). The oily product was filtered through a short SiO_2 column using hexane/ CH_2Cl_2 1:1 and the filtrate evaporated to leave a colourless solid which, after recrystallisation from pentane at -30° , yielded 80 mg (52%) of **13** as colourless prisms, m.p. 48.2–51.0°.

12. *Saponification of 14*. A soln. of **14** (2.55 g, 15.2 mmol) and NaOH (3.60 g, 90 mmol) in MeOH (20 ml) and H_2O (26 ml) was stirred for 13 h at 45–50°, cooled, acidified with conc. HCl, and extracted with CH_2Cl_2 (5×40 ml). The combined extracts were dried over $MgSO_4$ and evaporated to leave a yellow semi-solid, which was filtered through a short SiO_2 column with C_6H_6 / Et_2O 9:1. The filtrate was evaporated to yield 2.03 g (87%) of **13** of 100% purity (anal. GC (SE-52 101°)) as a colourless solid, after crystallisation from pentane at -30° , as colourless prisms, m.p. 52.9–53.6°. Spectral data: identical to the ones described for **13** in *Exper. 9*.

13. *(1RS,2SR)-1,2-Dimethyl-5-methylenecyclopentane-1-carboxylic acid (15)*. Using the procedure described in *Exper. 12*, 100 mg (0.59 mmol) of **16** was saponified to yield **15** (81 mg, 89%) of 99% purity (anal. GC (BP-5, 150°)) as a colourless solid. Crystallisation from pentane at -30° afforded **15** as colourless needles, m.p. 54.4–55.1°. IR ($CHCl_3$): 3560–2200m (br.), 1695s (C=O), 1650m, 1295m, 1280m, 1185m, 895m. 1H -NMR (200 MHz, $CDCl_3$): 5.02, 4.99 (2 t, $J = 2.4$, $CH_2 = C(5)$); 2.72–2.52 (m, [0.96] dd, $J = 11.0$, 6.5, H-C(2)); 2.52–2.38 (m, 2 H); 1.98–1.78 (m, 1 H); 1.50–1.22 (m, 1 H); 1.16 (s, $CH_3-C(1)$); 0.96 (d, $J = 6.9$, $CH_3-C(2)$). MS (70 eV): 154 (2, M^+), 125 (8), 109 (100), 93 (8), 81 (10), 67 (35), 55 (17).

14. *Hydrogenation of 10/11/13*. A stirred soln. of 730 mg (4.7 mmol) of a 11:79:10 mixture of **10/11/13** (as obtained in *Exper. 7*) in $EtOAc$ (50 ml) was hydrogenated at atmospheric pressure and r.t. for 60 h in the presence of 10% Pt/C (260 mg). The suspension was filtered through *Celite* and the filtrate evaporated to leave 720 mg (97%) of a colourless oil which, by anal. GC (SE-52, 71°) of an aliquot esterified with an Et_2O soln. of CH_2N_2 , contained *(1RS,5RS)-2,2,5-trimethylcyclopentane-1-carboxylic acid (18)*, **17**, and **19** in a ratio of 69:23:8. Subtraction of the 1H -NMR of **17** [1]/**19** from the 1H -NMR of **18**/17/19 enabled the assignment of signals to certain protons of **18**. 1H -NMR (400 MHz, $CDCl_3$): Signals due to **18**: 2.57–2.47 (m, H-C(5)); 2.41 (d, $J = 7.1$, H-C(1)); 1.95–1.81 (m, 2 H); 1.11 (s, $CH_3-C(2)$); 1.08 (s, $CH_3-C(2)$); 1.07 (d, $J = 7.2$, [2.52] s, $CH_3-C(5)$).

15. *(1r,2RS,5SR)-1,2,5-Trimethylcyclopentane-1-carboxylic Acid (19)*. A stirred soln. of **13** (6.14 g, 39.9 mmol) in $EtOAc$ (500 ml) was hydrogenated at atmospheric pressure and r.t. in the presence of 10% Pt/C (2 g) until the expected amount of H_2 (ca. 900 ml) was consumed. The suspension was filtered through *Celite* and the filtrate evaporated to leave **19** (5.89 g, 95%) as colourless solid (purity 99%, anal. GC (SE-52, 90°) of an aliquot esterified with an Et_2O soln. of CH_2N_2). An anal. sample of **19** was obtained after recrystallisation from pentane at -30° as colourless prisms, m.p. 71–73°. IR ($CHCl_3$): 3600–2200s (br.), 1695s (C=O), 1455w, 1410w, 1380w, 1295m, 1190w, 1090w. 1H -NMR (200 MHz, $CDCl_3$): 2.56–2.28 (m, H-C(2), H-C(5)); 2.00–1.76 (m, 2 H); 1.40–1.10 (m, 2 H); 0.93 (d, $J = 7.0$, $CH_3-C(2)$, $CH_3-C(5)$); 0.89 (s, $CH_3-C(1)$). ^{13}C -NMR (25.2 MHz, $CDCl_3$): 184.8 (s, CO_2H); 54.5 (s, C(1)); 43.1 (d, C(2), C(5)); 30.7 (t, C(3), C(4)); 14.8 (q, $CH_3-C(2)$, $CH_3-C(5)$); 9.1 (q, $CH_3-C(1)$). MS (70 eV): 156 (7, M^+), 141 (41, $M^+ - 15$), 114 (35), 101 (100), 95 (39), 83 (24), 69 (52), 55 (82). Anal. calc. for $C_9H_{16}O_2$ (156.23): C 69.19, H 10.32; found: C 69.16, H 10.31.

16. *Hydrogenation of 16*. Using the same conditions as in *Exper. 15*, 200 mg (1.19 mmol) of **16** was hydrogenated affording 196 mg (98%) of a colourless oil consisting of *methyl (2RS,5RS)-1,2,5-trimethylcyclopentane-1-carboxylate (22)* and its diastereoisomer **20** in a ratio of 61:39 (anal. GC (SE-52, 71°)).

17. *Hydrogenation of 15*. Under the same conditions as in *Exper. 15*, 27 mg (0.18 mmol) of **15** was hydrogenated affording 23 mg (85%) of a colourless solid consisting of (2*RS*,5*RS*)-1,2,5-trimethylcyclopentane-1-carboxylic acid (**21**) and its diastereoisomer **19** in a ratio of 63:37 (anal. GC (*SE*-52, 101°)).

18. (1*RS*,5*SR*)-2,2,5-Trimethylcyclopentane-1-carboxylic Acid (**17**). To the stirred suspension of NaH (ca. 4.8 mmol; obtained by washing 212 mg of a 55–60% dispersion of NaH in oil) in Et₂O (10 ml), was added, at r.t., soln. of a 23:69:8 mixture **17/18/19** (712 mg, 4.6 mmol; as obtained in *Exper. 14*) in Et₂O (25 ml) and HMPT (5 ml). After 30 min, BuLi (3 ml; ca. 4.8 mmol; 1.6 M in hexane) was added at r.t., and the pale red mixture was stirred for 60 min, poured onto ice (70 g), acidified with conc. HCl, and extracted with Et₂O (3 × 30 ml). The Et₂O layer was washed with H₂O and brine, dried over MgSO₄, and evaporated to give 876 mg of pale yellow oil. Repeated bulb-to-bulb distillation at 130°/14 Torr yielded 593 mg (83%) of a colourless oil which, by GC (*SE*-52, 100°), was shown to be 87% pure. Spectral data of an anal. sample obtained after *Lobar* purification and bulb-to-bulb distillation: identical to the ones described for **17** in [1].

REFERENCES

- [1] J. Huguet, M. Karpf, A. S. Dreiding, *Helv. Chim. Acta* **1982**, *65*, 2413.
- [2] J. Huguet, M. Karpf, A. S. Dreiding, *Tetrahedron Lett.* **1983**, 4177.
- [3] M. Karpf, A. S. Dreiding, *Helv. Chim. Acta* **1979**, *62*, 852; M. Karpf, J. Huguet, A. S. Dreiding, *ibid.* **1982**, *65*, 13; G. G. G. Manzardo, M. Karpf, A. S. Dreiding, *ibid.* **1983**, *66*, 627; J. Ackroyd, M. Karpf, A. S. Dreiding, *ibid.* **1985**, *68*, 338.
- [4] M. Karpf, A. S. Dreiding, *Helv. Chim. Acta* **1981**, *64*, 1123 and ref. cit. therein.
- [5] P. G. Gassman, J. J. Talley, *Tetrahedron Lett.* **1978**, 3773.
- [6] W. H. Saunders, A. F. Cockerill, 'Mechanisms of Elimination Reactions', Wiley, New York, 1973, pp. 245–251, and ref. cit. therein.
- [7] K. Narasaka, T. Sakakura, T. Uchimarui, D. Guédin-Voung, *J. Am. Chem. Soc.* **1984**, *106*, 2954.
- [8] J. Langlet, H. Van der Meer, *J. Mol. Struct.* **1979**, *10*, 91.
- [9] H. Schenk, Ph.D. Thesis, University of Amsterdam, 1969.
- [10] H. Schmidbaur, H. Stühles, W. Vornberger, *Chem. Ber.* **1972**, *105*, 1084; M. Rey, Ph.D. Thesis, University of Sheffield, 1975.