

## The Absolute Stereochemistry at C-2 of Thiazolidines derived from *R*-Penicillamine and Aldehydes

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IN connection with the synthesis of some penicillin analogues we became interested in the factors which control the stereochemistry at C-2 of thiazolidines formed from *R*-penicillamine and aldehydes. Here we describe a convenient method for determining the absolute stereochemistry at C-2.

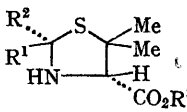
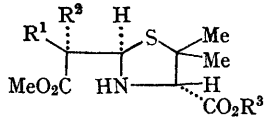
Foster and his co-workers<sup>1</sup> have successfully determined the absolute configuration at the acetal centre of 1,3-dioxolans by n.m.r. spectroscopy. The acetal proton was shown to be deshielded by *cis*-alkyl substituents at C-3 or C-4.<sup>2</sup> Consequently, it seemed likely that the chemical shift of the C-4 proton of 4-carboxythiazolidines would reflect the stereochemistry at C-2. Support for this expectation was found in the chemical shifts of the C-4 protons of 4*R*-carboxy-5,5-dimethylthiazolidine (I)<sup>3</sup> and 4*R*-carboxy-2,2,5,5-tetramethylthiazolidine (II)<sup>4</sup> which occurred at  $\tau$  6.18 and 5.78, respectively. The model compounds indicated that a methyl group at C-2 *cis* to the hydrogen at C-4 deshielded the latter by 0.40 p.p.m.

2*S*,4*R*-Dicarboxy-5,5-dimethylthiazolidine (III), m.p. 188—190° (decomp),  $[\alpha]_D - 13^\circ$  (pyridine), was synthesized from *R*-penicillamine and glyoxylic acid.<sup>5</sup> The stereochemistry of (III) was confirmed

by conversion to its *N*-ethoxycarbonyl derivative, m.p. 148—150°,  $[\alpha]_D - 79^\circ$  (chloroform), which gave a syrupy anhydride ( $\nu_{\max}$  1820, 1770, and 1715 cm.<sup>-1</sup>;  $M^+$  259.0536) with dicyclohexylcarbodi-imide. Diazomethane esterification of (III) gave (IV), m.p. 63—64°,  $[\alpha]_D + 4^\circ$  (chloroform), which was equilibrated with 2*R*,4*R*-dimethoxycarbonyl-5,5-dimethylthiazolidine (V), m.p. 96°,  $[\alpha]_D + 162^\circ$  (chloroform) in methanolic hydrogen chloride. 4*R*-Methoxycarbonyl-2*S*-carboxy-5,5-dimethylthiazolidine (VI), m.p. 82—84°,  $[\alpha]_D + 73^\circ$  (chloroform) and 4*R*-methoxycarbonyl-2*R*-carboxy-5,5-dimethylthiazolidine (VII), m.p. 112—114°,  $[\alpha]_D + 197^\circ$  (chloroform), were obtained by monosaponification of (IV) and (V), respectively. 2*S*-Methoxycarbonyl-4*R*-carboxy-5,5-dimethylthiazolidine (VIII), m.p. 108—110°,  $[\alpha]_D + 58^\circ$  (chloroform) was obtained from *R*-penicillamine and methyl glyoxylate. Its stereochemistry was confirmed by conversion to (IV) with diazomethane.

The chemical shifts of the C-4 protons of these thiazolidines are shown in the Table. In all cases the C-4 proton resonated at lower field when *cis* to an alkyl substituent at C-2, compared to compounds in which the C-4 proton was *cis* to a hydrogen at C-2.

TABLE  
Chemical shifts<sup>a</sup> of C-4 protons in thiazolidine derivatives

|  |                    |                    |                |        |  |                         |                |                |        |
|---|--------------------|--------------------|----------------|--------|--|-------------------------|----------------|----------------|--------|
| Compound  | R <sup>1</sup>     | R <sup>2</sup>     | R <sup>3</sup> | $\tau$ | Compound   | R <sup>1</sup>          | R <sup>2</sup> | R <sup>3</sup> | $\tau$ |
| (I) <sup>3</sup>  | H                  | H                  | H              | 6.18   | (X) <sup>8</sup>   | PhCH <sub>2</sub> ·CONH | H              | H              | 5.80   |
| (II) <sup>4</sup>   | Me                 | Me                 | H              | 5.78   | (XI) <sup>9</sup>  | PhCH <sub>2</sub> ·CONH | H              | Me             | 5.86   |
| (III)   | H                  | CO <sub>2</sub> H  | H              | 5.89   | (XII) <sup>10</sup>  | H                       | Cl             | Me             | 5.91   |
| (IV)  | H                  | CO <sub>2</sub> Me | Me             | 6.05   | (XIII) <sup>11</sup>   | Ph(CO) <sub>2</sub> N   | H              | H              | 5.73   |
| (V)   | CO <sub>2</sub> Me | H                  | Me             | 5.62   | (XIV) <sup>11</sup>  | Ph(CO) <sub>2</sub> N   | H              | Me             | 5.78   |
| (VI)  | H                  | CO <sub>2</sub> H  | Me             | 6.02   |  |                         |                |                |        |
| (VII)   | CO <sub>2</sub> H  | H                  | Me             | 5.52   |  |                         |                |                |        |
| (VIII)  | H                  | CO <sub>2</sub> Me | H              | 6.01   |  |                         |                |                |        |
| (IX) <sup>b</sup>   | CO <sub>2</sub> H  | H                  | H              | 5.26   |  |                         |                |                |        |

<sup>a</sup> N.m.r. spectra were measured in pyridine at 60 Mc./sec. with tetramethylsilane as an internal standard; <sup>b</sup> This compound has not been fully characterized; its presence in the mother liquor after crystallization of (III) was implied by n.m.r. spectroscopy.

4*R*-Carboxy-2,5,5-trimethylthiazolidine, m.p. 149—150°, [ $\alpha$ ]<sub>D</sub> + 133° (pyridine), was prepared from *R*-penicillamine and acetaldehyde and was shown to be a mixture of stereoisomers by n.m.r. spectroscopy. The tertiary protons, which appeared at  $\tau$  6.07 and 5.88 in the ratio of 4:1, suggested that the predominant stereoisomer was 4*R*-carboxy-2*S*-5,5-trimethylthiazolidine. Similarly 4*R*-carboxy-5,5-dimethyl-2-phenylthiazolidine<sup>6</sup> was found to be a mixture of stereoisomers. The tertiary protons, which resonated at  $\tau$  5.93 and 5.87, were present in the ratio of 3:1 implying that the major isomer was 4*R*-carboxy-5,5-dimethyl-2*S*-phenylthiazolidine.

A number of penicilloic acid derivatives and related compounds were also examined and the results are shown in the Table. In these cases, in which the substituent at C-2 of the thiazolidine is known to possess the *R*-configuration, the chemical shift of the C-4 protons fell between  $\tau$  5.73 and

5.91. Consequently, it was expected that if a penicilloate possessed the 2*S*-stereochemistry then the C-4 proton would resonate at higher field. The  $\gamma$ -isomer of methyl 4*R*-carboxy-5,5-dimethyl- $\alpha$ -phthalimido-2-thiazolidine acetate, m.p. 180—182°, [ $\alpha$ ]<sub>D</sub> - 8° (dioxan), was prepared by Sheehan's method.<sup>7</sup> The C-4 proton resonated at  $\tau$  6.12 suggesting that this isomer possessed the 2*S*-configuration.

The results described indicate that, in the case of the 4-carboxythiazolidine derivatives, the chemical shift of the C-4 proton is a valuable guide to the relative stereochemistry at C-2. In particular if both stereoisomers at C-2 are available the isomer in which the C-4 proton appears at lowest field possessed the *trans*-configuration.

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<sup>1</sup> N. Baggett, K. W. Buck, A. B. Foster, and J. M. Webber, *J. Chem. Soc.*, 1965, 3401; N. Baggett, K. W. Buck, A. B. Foster, M. H. Randall, and J. M. Webber, *ibid.*, 1965, 3394; N. Baggett, J. M. Duxbury, A. B. Foster, and J. M. Webber, *J. Chem. Soc. (C)*, 1966, 208.

<sup>2</sup> M. Anteunis and F. Alderweireldt (*Bull. Soc. chim. belg.*, 1964, **73**, 889) have reached the opposite conclusion but their interpretation is based upon a misassignment of stereochemistry. We thank Dr. J. G. Buchanan for pointing out this paper to us.

<sup>3</sup> H. T. Clarke, J. R. Johnson, and R. Robinson, "The Chemistry of Penicillin", Princeton University Press, Princeton, 1949, p. 958.

<sup>4</sup> Ref. 3, p. 26.

<sup>5</sup> A similar condensation utilizing the methyl ester of penicillamine has been described (R. Bentley, A. H. Cook, J. A. Elvidge, and G. Shaw, *J. Chem. Soc.*, 1949, 2351; see also ref. 3, p. 964). No stereochemical assignments were made.

<sup>6</sup> Ref. 3, p. 946.

<sup>7</sup> J. C. Sheehan and D. A. Johnson, *J. Amer. Chem. Soc.*, 1954, **76**, 158.

<sup>8</sup> Ref. 3, p. 582.

<sup>9</sup> Ref. 3, p. 613.

<sup>10</sup> I. Mcmillan and R. J. Stoodley, *Tetrahedron Letters*, 1966, 1205. The absolute stereochemistry at C-2 and the  $\beta$ -position has now been established by a chemical method.

<sup>11</sup> J. C. Sheehan and P. A. Cruickshank, *J. Amer. Chem. Soc.*, 1956, **78**, 3677.