## A Regioselective Synthesis of 8,10-Dideoxycarminomycinone

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Summary A carminomycinone derivative has been synthesised from 5-hydroxyquinizarin and 2-methoxy-5-methyl-4-nitromethyltetrahydrofuran.

Over the past few years there has been extensive synthetic work directed towards the synthesis of anthracyclinones and a number of elegant solutions to the regiochemical problem have been produced.¹ We chose the readily available 5-hydroxyquinizarin (1) as starting point, which required the development of methods for selective substitution at C-2 or C-3. We have shown that Marshalk (aldol) condensation of leuco-5-hydroxyquinizarin can be made selective for alkylation at C-2 or C-3 under differing conditions.² More recently we have established³ that nitronate carbanions add with >95% selectivity to C-2 of compound (1). We have now used this selectivity for the synthesis of an anthracyclinone derivative.

An appropriate nitro-compound (9) was synthesised from ethyl levulinate which was brominated  $(Br_2-CCl_4)$  and dehydrobrominated  $(Et_3N-CH_2Cl_2)$  to give (5). Reduction of (5) with NaBH<sub>4</sub>-MeOH gave the alcohol (6) (82%)

which underwent Michael addition4 (MeNO<sub>2</sub>-Triton B-ButOH) and lactonisation in situ to form (7)† (79%)  $[\nu_{\rm max}~1770,~1545~{\rm cm^{-1}},~\tau~5\cdot4~(1{\rm H,~m}),~5\cdot50~(2{\rm H,~d},~J~6~{\rm Hz}),$ and 8.60 (3H, pair d, I 6 and 7 Hz)]. Bui, AlH-PhMe at -78 °C reduced the lactones (7) to the lactols (8)† which were converted (MeOH- $H_2SO_4$ ) into the ethers (9)† (55%) overall)  $[v_{\text{max}} 1550 \text{ cm}^{-1}, \tau 5.02 \text{ (1H, m)}, 5.3 - 5.8 \text{ (2H, m)},$ 6.72 (3H, br.s), and 8.6—8.9 (3H, m)]. Refluxing a mixture of the quinone (1), ethers (9), and NaOMe (1:3:1.4) in MeOH gave the substitution products (2)  $\uparrow$  (65%) [ $\tau$  6.62(s), 6.64(s), 6.66(s), and 6.69(s) (1:3:2.5:3); ratio ArH: OCH<sub>3</sub> Hydrolysis of (2) (1n HCl-dimethoxyethane) yielded the lactols; (3)† (85%) which were reduced and the leuco-derivatives were cyclised (Na<sub>2</sub>S<sub>2</sub>O<sub>4</sub>-1% NaOH, 90 °C) to the alcohols (10)  $\dagger$  (65%) [ $\tau$  2·12 (1H, dd, J 7 and 2 Hz), 2.35 (1H, t, J 7 Hz), 2.75 (1H, dd, J 7 and 2 Hz), and 8.67 (3H, d, J 6 Hz)]. Pyridinium chlorochromate<sup>5</sup> oxidized (10) to the ketone<sup>1c</sup> (11) (67%) [ $\nu_{max}$  1705 cm<sup>-1</sup>;  $\tau$  2·12 (1H, dd, J 7 and 2 Hz), 2·35 (1H, t, J 7 Hz), 2·70 (1H,dd, J 7 and 2 Hz), and 7·70 (3H,s)]. The structure of (11) was shown to be 8,10-dideoxycarminomycinone by conversion into the daunomycinone derivative (12) with

<sup>†</sup> Mixture of diastereoisomers.

<sup>‡</sup> This compound was characterised by oxidation to the  $\gamma$ -lactones (4).

O HO
O HO
$$X$$

O HO
 $X$ 

(3) 
$$R = -CH_2$$

(4) 
$$R = -CH_2$$

$$X$$

$$CO_2Et$$
(5)  $X = 0$ 
(6)  $X = H_1OH$ 

(11) 
$$R = H, X = =0$$
  
(12)  $R = Me, X = =0$ 

(13)

 $CH_2N_2$ - $CH_2Cl_2^6$  (32% yield on 52% conversion)§ or by trimethylation (MeI-K2CO3-Me2CO) followed by silver(II) oxide cleavage7 (24% overall). The ether (12) was shown to be identical to an authentic sample by usual criteria; in particular we were unable to detect any of the regioisomer which has an n.m.r. spectrum distinctive from that of (12).

When the same sequence of reactions is carried out using quinizarin as starting material the deoxy-compound (13) can be prepared efficiently (27% overall yield).

Reaction of the nitrolactones (7) with (1) and NaOMe-MeOH gave the lactones (4) $\dagger$  (8%) [ $\nu_{max}$  1770 cm<sup>-1</sup>;  $\tau$  2·17 (1H,dd, J 7 and 2 Hz), 2·29 (1H,t, J 7 Hz), 2·65 (1H,dd, J7 and 2 Hz), 2.85 (1H,s), 5.35 (1H,m), 8.60  $(\frac{3}{2}$ H,d, J 7 Hz), and 8.72  $(\frac{3}{2}$ H,d, J 7 Hz)]; a control experiment established that the low yield was due to destruction of (7) under the reaction conditions. In principle the nitrolactols (8) could react with (1) by the sequence Michael addition, aldol cyclisation of the leuco-adduct, and elimination of nitrite to give the 7-hydroxy-derivatives of (10). In practice reaction of (1) and (8) in NaOMe-MeOH gave the lactols (3) (7%), the ethers (2) (2%), and an ether (2%) which on reduction  $(Na_2S_2O_4)$  yields the alcohol (10). ¶ From this it would appear that, under our conditions, elimination of nitrite is faster than aldol condensation. Compounds (3) and (8) gave satisfactory accurate mass measurements. All other new compounds gave satisfactory combustion analyses.

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§ Another monomethyl ether (16%) was isolated. Since it shows only hydrogen-bonded quinone carbonyl absorption in the i.r. it is assumed to be the 6-methoxy-compound.

¶ This result suggests that the other ether terminus is at C-7 or C-10.

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