ISSN 1070-3632, Russian Journal of General Chemistry, 2007, Vol. 77, No. 10, pp. 1821–1822. © Pleiades Publishing, Ltd., 2007. Original Russian Text © O.S. Attaryan, G.A. Akopyan, K.S. Badalyan, G.G. Minasyan, G.V. Asratyan, 2007, published in Zhurnal Obshchei Khimii, 2007, Vol. 77, No. 10, pp. 1757–1758.

> LETTERS TO THE EDITOR

Bromination of 1,3-Dimethyland 1,5-Dimethyl-1*H*-pyrazole-4-carbaldehydes

O. S. Attaryan^a, G. A. Akopyan^b, K. S. Badalyan^b, G. G. Minasyan^b, G. V. Asratyan^a

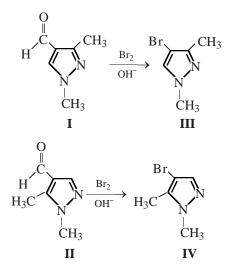
^aARIAK Institute of Applied Chemistry of the Republic of Armenia, Artashatskoe shosse 5/2, Yerevan, 375053 Armenia

^b Institute of Organic Chemistry, National Academy of Sciences of Armenia, Yerevan, Armenia

Received July 13, 2007

DOI: 10.1134/S1070363207100295

Proceeding with the search for synthetic approaches to 5-bromo-1,3-dimethyl- and 3-bromo-1,5-dimethylpyrazoles [1] we have studied bromination of 1,3-dimethyl- and 1,5-dimethyl-1*H*-pyrazole-4-carbaldehydes (**I**, **II**). It was found that the reaction in aqueous alkali involves substitution of the formyl group by bromine by the following scheme:



The structure of compounds **III** and **IV** was established by means of IR and ¹H NMR spectroscopy and elemental analysis. The ¹H NMR spectrum contains no aldehyde proton signal at δ 9.78 ppm, and the IR spectrum shows no bands at 1680 cm⁻¹ related to the formyl group. The integral intensities of the rest protons are completely consistent with the suggested structures.

Starting pyrazoles I and I were prepared by known procedures [2, 3], mp 50 and 60°C, respectively.

4-Bromo-1,3-dimethyl-1*H***-pyrazole** (**III**) To a solution of 12.5 g of 1,3-dimethyl-1*H*-pyrazole-4carbaldehyde (**I**) in 100 ml of water, 12 g of NaOH was added and then 16 g of bromine was added dropwise at room temperature. The reaction mixture was then extracted with chloroform (3×50 ml), the extract was dried over MgSO₄, the solvent was removed, and the residue was distilled in a vacuum to obtain 12 g (69%) of compound **III**, bp 45°C (1 mm Hg), n_D^{20} 1.521, d_4^{20} 1.4059 [1, 4]. IR spectrum, v, cm⁻¹: 1510 (pyrazole ring). ¹H NMR spectrum, δ, ppm: 2.15 s (3H, 3-CH₃), 3.81 s (3H, N-CH₃), 7.51 s (1H, H⁴). Found, %: C 34.38. H 4.31, Br 45.88, N 16.43. C₅H₇BrN₂. Calculated, %: C 34.31, H 4.02, Br 45.65, N 16.00.

4-Bromo-1,5-dimethyl-1*H***-pyrazole** (**IV**) was obtained in a similar way from 12.4 g of 1,5-dimethyl-1*H*-pyrazole-4-carbaldehyde (**II**) and 16 g of bromine. Yield 11 g (63%), bp 54°C (1 mm Hg), mp 43°C [1, 4]. IR spectrum, v, cm⁻¹: 1530 (pyrazole ring). ¹H NMR spectrum, δ , ppm: 2.25 s (3H, 3-CH₃), 3.80 s (3H, N-CH₃), 7.21 s (1H, H⁴). Found, %: C 34.62, H 4.42, Br 45.89, N 16.58. C₅H₇BrN₂. Calculated, %: C 34.31, H 4.02, Br 45.65, N 16.00.

The ¹H NMR spectra were taken on a Varian Mercury-300 spectrometer in DMSO- d_6 . The IR spectra were obtained on a Specord IR-75 spectrometer in KBr pellets and in thin layer.

REFERENCES

 Attaryan, O.S., Akopyan, G.A., Badalyan, K.S., and Asratyan, G.V., *Zh. Obshch. Khim.*, 2007, vol. 77, no. 2, p. 335.

- Mal'tseva, S.P., Borodulina, Z.A., and Stepanov, B.I., *Zh. Org. Khim.*, 1973, vol. 9, no. 4, p. 815.
- Antanosyan, S.K., *Khim. Zh. Arm.*, 2006, vol. 59, no. 2, p. 99.
- Andreeva, M.A., Bolonov, M.I., Isaev, Sh.G., Mushiy, R.Ya., Gerevalov, V.P., Seraya, V.I., and Stepanov, B.I., *Zh. Obshch. Khim.*, 1980, vol. 50, no. 9, p. 2116.