## THE MASS SPECTRA OF *CYCLO*-TRIPEPTIDES OF B-ALANINE AND 3-AMINO-3-METHYL BUTANOIC ACID

C. N. C. DREY and J. LOWBRIDGE

Department of Chemistry and Polymer Technology, The Polytechnic of the South Bank, Borough Road, London SE1 OAA, England

(Received 3 November 1972; accepted (revised) 4 December 1972)

Abstract—The cyclo-tripeptides of B-alanine and of 3-amino-3-methyl butanoic acid have been synthesised, the latter for the first time, and structures were confirmed by elemental analysis and mass spectrometry; accurate mass measurements showed that carbon monoxide was lost from the molecular ions; prominent peaks in the mass spectrum of both cyclic peptides are otherwise consistent with those reported for  $\alpha$ -peptides.

BOTH cyclo-(tri-B-alanyl)<sup>1</sup> (I) and cyclo-(tri-3-amino-3-methyl butyryl) (II) were prepared by cyclisation of the corresponding linear tripeptides,  $(III)^{2,*}$  and  $(IV)^{3,\dagger}$ ; (I) was also obtained via the active ester (V) and the hydrazide (VI). All the appropriate intermediates and final products analysed correctly.

$$\label{eq:constraint} \begin{split} \underbrace{ \begin{bmatrix} NH \cdot C(R^1)_2 \cdot CH_2 \cdot CONH \cdot C(R^1)_2 \cdot CH_2 \cdot CO) \\ (I) & R^1 = H \\ (II) & R^1 = Me \\ NH_2 \cdot C(R^1)_2 \cdot CH_2 \cdot CONH \cdot C(R^1)_2 \cdot CH_2 \cdot CO) \\ H^1 = H, & R^2 = OH \\ (II) & R^1 = H, & R^2 = OH \\ (IV) & R^1 = H, & R^2 = OH \\ (V) & R^1 = H, & R^2 = OTcp^a \\ (VI) & R^1 = H, & R^2 = N_2H_3 \\ \end{split}$$

<sup>a</sup> Tcp = 2,4,5-Trichlorophenol

It is known that cyclo-dimerisation of linear peptides can readily occur.<sup>4</sup> The low resolution mass spectra of the *cyclo*-peptides (I and II) were examined to ascertain whether this in fact had happened. Molecular ions m/e 213 and 297 corresponded to the expected values for I and II, respectively. There was no evidence for higher molecular weight derivatives. In both cases, however, prominent ions corresponding to  $[M - 28]^{+}$  were observed. Metastable peaks m/e 160·7 and m/e 243·5 provided evidence for decomposition of the molecular ions of I and II to peaks corresponding to m/e 185 and 269, respectively.

The  $[M - 28]^{+}$  ion derived from the B-alanyl derivative (I) may be ascribed to loss of ethylene or carbon dioxide from the molecular ion; in the case of the *cyclo*-peptide (II), the  $[M - 28]^{+}$  ion cannot be due to the loss of ethylene and may be rationalised as loss of carbon monoxide.

Cleavage of  $\alpha$ -peptide residues at the CO—N bond gives rise to acylium ions, which readily lose carbon monoxide leaving a mesomerically stabilised acyliminium ion;<sup>5</sup> the absence of such [M - 28]<sup>+</sup> peaks from the spectra of B-alanyl, B-aspartyl,  $\gamma$ -glutaminyl and B-lysyl peptides has been recommended as a criterion for distinguishing  $\alpha$  from  $\omega$ -linked peptides.<sup>5</sup>

Accurate mass determinations for the ions m/e 213, 185 from the peptides I and

\* Previously isolated as the hemi-hydrate, obtained as the mono-hydrate.

† From the benzoyl tripeptide<sup>3</sup> by electrolytic reduction.

1

III and m/e 297, 269 from the cyclo-peptide (II), respectively, have shown that the loss of the 28 fragment corresponds to the loss of carbon monoxide in each case.

It seems therefore that the absence of  $[M - 28]^+$  peaks as a criterion for the recognition of non  $\alpha$ -peptidic bonds is equivocal.

Prominent peaks in the mass spectrum of both cyclic peptides are otherwise consistent with fragmentation pathways reported for  $\alpha$ -peptides.<sup>6</sup>

The mass spectrum of the cyclo-aminobutyryl derivative (II) gives rise to strong peaks at m/e 182 and 83 (-1 series).<sup>6</sup> Metastable peaks at m/e 210.9 and 147.2 indicate that the fragment of m/e 182 arises via fragments m/e 240 and 225—a feasible mechanism (Scheme 1) depends on the ring having opened with loss of an amine fragment; additionally, analogous precursor peaks of the m/e 83 fragment also appear. A similar fragmentation pathway can be discerned for the B-alanyl derivative (I).



## SCHEME 1

## EXPERIMENTAL

Mass spectra were obtained with 70 eV on an AEI MS-902; the samples were introduced at  $225^{\circ}$ .

The measured masses were within 5 ppm of the calculated values.

Acknowledgement—One of us (J. L.) thanks the I.L.E.A. for a studentship. We would also like to thank Dr B. J. Millard and Dr J. F. J. Todd for helpful discussion.

## REFERENCES

- 1. M. Rothe, Acta. Chim. Acad. Sci. Hung. 18, 449 (1959); Angew. Chem. 71, 700 (1959).
- 2. E. Adams, N. C. Davis and E. L. Smith, J. Biol. Chem. 199, 845 (1952).
- 3. J. Lowbridge and C. N. C. Drey, Chem. Commun. 791 (1970).
- 4. R. Schwyzer and P. Sieber, Helv. Chim. Acta 41, 2186, 2190, 2199 (1958).
- 5. J. Van Heijenoort, E. Bricas, B. C. Das, W. A. Wolstenholme and E. Lederer, *Tetrahedron* 23, 3403 (1967); L. I. Rostovtseva and A. A. Kiryushkin, *Org. Mass. Spectrom.* 6, 1 (1972).
- 6. B. J. Millard, Tetrahedron Letters 3041 (1965).