

A PSEUDOGUAIANOLIDE AND A HYDROXYGERANYLNEROL FROM *KINGIANTHUS PARADOXUS**

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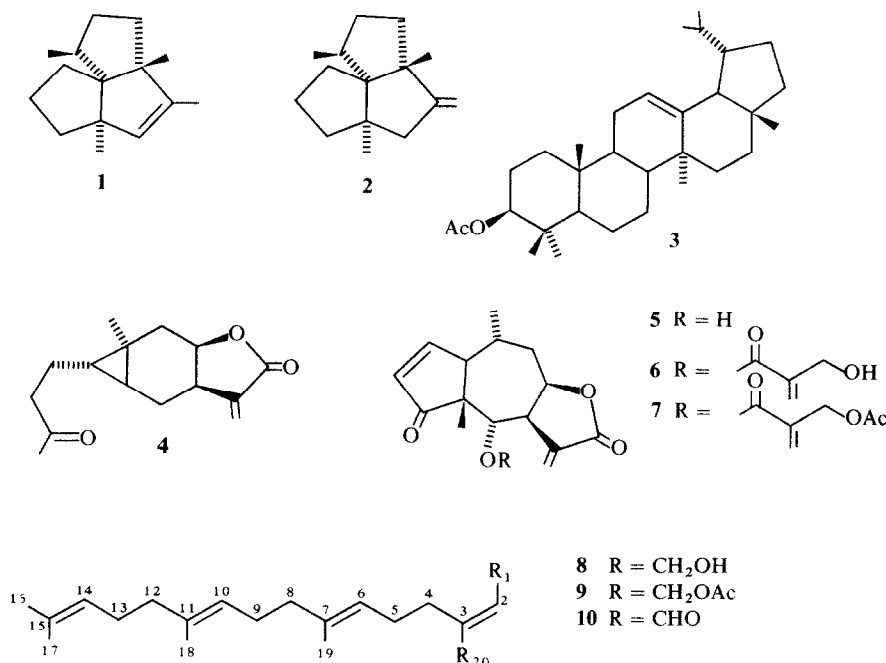
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Key Word Index—*Kingianthus paradoxus*; Compositae; Heliantheae; sesquiterpene lactones; pseudoguaianolide; kingiolide; 20-hydroxygeranylnerol.

Abstract—Examination of *Kingianthus paradoxus* yielded kingiolide, a new pseudoguaianolide, and 20-hydroxygeranylnerol.

So far nothing is known about the chemistry of the new genus *Kingianthus* [1, 2] which is placed in the tribe Heliantheae (Compositae). To see whether the chemistry gives any indication of relationships to other parts of this large tribe, we have investigated *K. paradoxus* H. Robins. The aerial parts, collected in Ecuador, afforded α -pinene, germacrene D, germacrene B, α -humulene, phytol, isocomene (1), β -isocomene (2), lupenone, lupeol, lupeyl acetate and its isomer 3, carabrone (4) [3] as the main constituent, helenalin (5) [4] and two further compounds, namely 6, the hydroxymethacrylate of 5 and the diterpene diol 8. The structure of 6 was deduced from the ^1H NMR

data and those of the corresponding acetate 7, obtained by heating with acetic anhydride (Table 1). The presence of a pseudoguaianolide structure, closely related to helenalin (5), was clear from the similarity of the ^1H NMR signals, except for the expected downfield shift of H-6. The similarity of the couplings of H-6 through H-9 further indicated the presence of an 8,12-lactone. In the ^1H NMR spectrum of 7 first-order interpretations of all signals were possible. We have named compound 6 kingiolide. The structure of the diterpene 8 also followed from the ^1H NMR data and those of the corresponding diacetate 9, obtained by mild acetylation, as well as from those of the



*Part 327 in series "Naturally Occurring Terpene Derivatives". For Part 326 see Bohlmann, F., Dhar, A. K., King, R. M. and Robinson, H. (1981) *Phytochemistry* 20, 1144.

Table 1. ^1H NMR spectral data of compounds **6** and **7** (270 MHz, TMS as internal standard, CDCl_3)

	6	7
H-1	3.04 <i>ddd</i>	3.02 <i>ddd</i>
H-2	6.08 <i>dd</i>	6.08 <i>dd</i>
H-3	7.69 <i>dd</i>	7.68 <i>dd</i>
H-6	5.46 <i>br. d</i>	5.46 <i>br. d</i>
H-7	3.59 <i>dddd</i>	3.59 <i>dddd</i>
H-8	4.90 <i>ddd</i>	4.90 <i>ddd</i>
H-9	2.41 <i>ddd</i>	2.42 <i>ddd</i>
H-9'	1.75 <i>ddd</i>	1.74 <i>ddd</i>
H-10	2.19 <i>m</i>	2.19 <i>dddq</i>
H-13	6.42 <i>d</i>	6.48 <i>d</i>
H-13'	6.16 <i>d</i>	6.18 <i>d</i>
H-14	1.29 <i>d</i>	1.28 <i>d</i>
H-15	1.04 <i>s</i>	1.05 <i>s</i>
OCOR	6.05 <i>dt</i> 5.84 <i>q</i> 4.27 <i>br. d</i>	6.25 <i>dt</i> 5.85 <i>q</i> 4.70 <i>br. s</i>
OAc	—	2.08 <i>s</i>

J (Hz): 1, 2 = 2.7; 1, 3 = 1.8; 1, 10 = 11.5; 2, 3 = 6; 6, 7 = 1; 7, 8 = 7.5; 7, 13 = 3; 7, 13' = 2.7; 8, 9 = 7.5; 8, 9' = 2.5; 9, 9' = 15; 9, 10 = 3.5; 9', 10 = 8; 10, 14 = 7; 3', 3' = 3', 4' ~ 1; 4', OH = 5.

dialdehyde **10**, obtained by manganese dioxide oxidation (Table 2). The stereochemistry of the 2,3-double bond was deduced from the shift differences of the signals for H-2 and H-4 in the spectra of **8** and **10** and the chemical shifts of H-1 and H-20 in the spectrum of **10**. Compound **8** has also been isolated from a *Disynaphia* species [5].

Carabrone (**4**) has been isolated from a *Monactis* species (F. Bohlmann *et al.*, unpublished results)

indicating a relationship of this genus to *Kingianthus*. These groups are also taxonomically very close to each other [2]. However, the pseudoguaianolides are typical for the genera *Helenium*, *Gaillardia*, *Ambrosia*, *Arnica* and *Parthenium* [6], while these lactones have not so far been isolated from members of the subtribe Verbesininae. Relationships to *Zaluzania*, where *Kingianthus* was placed earlier [1], are not apparent.

EXPERIMENTAL

The air-dried plant material (voucher RMK 7813) was extracted with Et_2O -petrol (1:2) and the resulting extract was first separated by CC (Si gel) and further by repeated TLC (Si gel). Known compounds were identified by comparing the IR and ^1H NMR spectra with those of authentic material.

The hydrocarbons were identified by GC/MS. Compounds obtained were 1 mg α -pinene, 20 mg germacrene D, 30 mg germacrene B, 10 mg α -humulene, 50 mg lupenone, 2 mg lupeol, 50 mg lupeyl acetate, 20 mg phytol, 1 mg **1**, 1 mg **2**, 50 mg **3**, 250 mg **4**, 1 mg **5**, 65 mg **6** and 190 mg **8**.

Kingiolide (**6**). Colourless gum, IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm^{-1} : 3600 (OH), 1760 (γ -lactone), 1715 ($\text{C}=\text{CCO}_2\text{R}$); MS m/z (rel. int.): 346.142 (M^+ , 1), 328 ($\text{M} - \text{H}_2\text{O}$, 0.5), 262 ($\text{M} - \text{C}_3\text{H}_4\text{O}_2$, 32), 244 ($\text{M} - \text{RCO}_2\text{H}$, 38), 85 (RCO^+ , 100).

$$[\alpha]_{\text{D}}^{24} = \frac{589}{-70.1} \frac{578}{-78.8} \frac{546 \text{ nm}}{-87.4} (c = 6.5).$$

Compound **6** (10 mg) in 0.1 ml Ac_2O was heated for 1 hr at 70° . TLC (Et_2O -petrol, 1:1) afforded 8 mg **7**, colourless gum; IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm^{-1} : 1785 (γ -lactone), 1750, 1235 (OAc), 1720 ($\text{C}=\text{CCO}_2\text{R}$); MS m/z (rel. int.): 388 (M^+ , 3), 346 ($\text{M} - \text{ketene}$, 1), 329 ($\text{M} - \text{OAc}$, 1), 244 ($\text{M} - \text{RCO}_2\text{H}$, 24), 125 ($\text{C}_6\text{H}_5\text{O}_3$, 100).

20-Hydroxygeranylnlerol (**8**). Colourless gum, IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm^{-1} : 3580 (OH), 1630 ($\text{C}=\text{C}$); MS m/z (rel. int.): 306 (M^+ , 0.5), 238.193 ($\text{M} - \text{isoprene}$, 6) ($\text{C}_{15}\text{H}_{26}\text{O}_2$), 220 ($238 - \text{H}_2\text{O}$, 15), 69 (C_5H_9^+ , 100). Compound **8** (5 mg) in 0.1 ml Ac_2O was heated for

Table 2. ^1H NMR spectral data of compounds **8–10** (270 MHz, TMS as internal standard, CDCl_3)

	8	9	9 (C_6D_6)	10
H-1	4.22 <i>br. d</i>	4.64 <i>br. d</i>	4.67 <i>br. d</i>	10.22 <i>d</i>
H-2	5.70 <i>br. t</i>	5.63 <i>br. t</i>	5.67 <i>br. t</i>	6.51 <i>d</i>
H-4, 5	2.07 <i>m</i>	2.15 <i>m</i>	2.22 <i>m</i>	$\left\{ \begin{array}{l} 2.73 \text{ br. } t \\ 2.22 \text{ dt} \end{array} \right.$
H-6	5.12 <i>br. t</i>	5.11 <i>br. t</i>	5.32 <i>tq</i>	5.10 <i>br. t</i>
H-8, 9	2.07 <i>m</i>	2.00 <i>m</i>	2.11 <i>m</i>	2.02 <i>m</i>
H-10	5.10 <i>br. t</i>	5.11 <i>br. t</i>	5.29 <i>tq</i>	5.10 <i>br. t</i>
H-12, 13	2.07 <i>m</i>	2.00 <i>m</i>	2.11 <i>m</i>	2.02 <i>m</i>
H-14	5.10 <i>br. t</i>	5.11 <i>br. t</i>	5.20 <i>tq</i>	5.10 <i>br. t</i>
H-16*	1.68 <i>br. s</i>	1.69 <i>br. s</i>	1.73 <i>br. s</i>	1.69 <i>br. s</i>
H-17 to H-19*	1.60 <i>br. s</i>	1.61 <i>br. s</i>	$\left\{ \begin{array}{l} 1.67 \text{ br. } s \\ 1.63 \text{ br. } s \\ 1.61 \text{ br. } s \end{array} \right.$	$\left\{ \begin{array}{l} 1.61 \text{ br. } s \\ 1.61 \text{ br. } s \\ 1.54 \text{ br. } s \end{array} \right.$
H-20	4.10 <i>br. s</i>	4.55 <i>br. s</i>	4.56 <i>br. s</i>	9.67 <i>s</i>
OAc	—	2.10 <i>s</i>	1.72 <i>s</i>	—
	—	2.08 <i>s</i>	1.70 <i>s</i>	—

* In C_6D_6 : 1.73 *br. s*, 1.65 *br. s*, 1.61 *br. s*, 1.59 *br. s*.

J (Hz): 1, 2 = 7; (9: 7.5); 5, 6 = 9, 10 = 13, 14 ~ 6.5.

1 hr at 70°. TLC (Et₂O–petrol, 3:1) afforded 5 mg **9**, colourless gum; IR $\nu_{\text{max}}^{\text{CCl}_4}$: 1740 and 1240 (OAc); MS (CI, isobutane) m/z (rel. int.): 391 ($M + 1$, 1), 331 (391 – HOAc, 15), 271 (331 – HOAc, 100). Compound **8** (5 mg) in 2 ml Et₂O was stirred for 2 hr with 50 mg MnO₂. TLC (Et₂O–petrol, 3:1) afforded 2 mg **10**, colourless gum; IR $\nu_{\text{max}}^{\text{CCl}_4}$: 1695 and 1685 (C=CCHO); MS m/z (rel. int.): 302 (M^+ , 4), 136 (C₁₀H₁₆⁺, 17), 81 (C₆H₉⁺, 40), 69 (C₅H₉⁺, 100).

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