

107.7 (C-5', *d*), 120.8 (C-6', *d*), 39.0 (C- α , *t*), 134.2 (C- β , *d*), 117.6 (C- γ , *t*), 101.0 (O₂CH₂, *t*). MS *m/z* (rel. int.): 328 (7), 174 (15), 173 (100), 163 (25), 162 (59), 161 (28), 151 (12), 149 (29), 135 (68), 131 (21), 122 (12), 121 (18), 115 (21), 105 (14), 104 (20), 103 (28). *Dihydro-3b*. To **3a** (70 mg) in THF (40 ml), LiAlH₄ (15 mg) was added. The mixture was stirred (4 hr, room temp). Excess reagent was destroyed with NH₄Cl. The mixture was extracted with CHCl₃. The CHCl₃ soln was washed, dried and evapd. The residue (64 mg) was purified by TLC (silica gel, C₆H₆-EtOAc, 4:1) to dihydro-**3b** (40 mg). ¹H NMR (60 MHz, CDCl₃): δ 4.40 (H-2, *d*, *J* = 10 Hz), 2.9–1.9 (H-3, 2H-4, 2H- α , *m*), 6.2–5.3 (H-5, H-6, H-7, H- β , *m*), 7.2–6.7 (3ArH, *m*), 5.97 (O₂CH₂, *s*), 5.37–4.87 (2H- γ , *m*). ¹³C NMR (20 MHz, CDCl₃): δ 85.9 (C-2, *d*), 49.7 (C-3, *d*), 9.5 (Me-3, *q*) 49.5 (C-3a, *s*), 28.5 (C-4, *t*), 127.1 (C-5, *d*), 127.3 (C-6, *d*), 72.5 (C-7, *d*), 100.4 (C-7a, *s*), 135.4 (C-1', *s*), 107.8 (C-2', *d*), 147.8 (C-3', *s*), 147.2 (C-4', *s*), 107.6 (C-5', *d*), 121.0 (C-6', *d*), 39.4 (C- α , *t*), 134.5 (C- β , *d*), 117.5 (C- γ , *t*), 101.0 (O₂CH₂, *t*).

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TWO ANTHRAQUINONES FROM *VENTILAGO CALYCVLATA*

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Abstract—Two new anthraquinones have been isolated from the root bark of *Ventilago calyculata* and their structures shown to be 2,4,8-trihydroxy-1-methoxy-3-methylanthraquinone and 2,4,8-trihydroxy-1,6-dimethoxy-3-methylanthraquinone.

INTRODUCTION

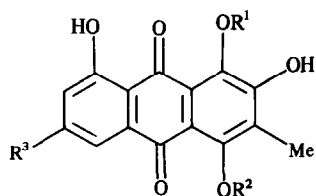
The genus *Ventilago* is a rich source of anthraquinones [1–4]. Further investigation of the root bark of *Ventilago calyculata* afforded two more new anthraquinones, **2** and **4**, in addition to the eleven anthraquinones reported previously [2].

RESULTS AND DISCUSSION

Pigment **2**, C₁₆H₁₂O₆, gives a positive reduction test with Na₂S₂O₄, indicating its anthraquinone nature, and forms a trimethyl ether. It is soluble in aqueous sodium carbonate suggesting the presence of a β -hydroxyl group, and as there is only one IR carbonyl band, at 1630 cm⁻¹, the remaining two hydroxyl groups are in α -positions, and from the visible spectrum (λ_{\max} 443 nm) it must be a 1,5 or 1,8-dihydroxyanthraquinone. The ¹H NMR spectrum confirms the presence of a β -hydroxyl and two *peri*-

hydroxyl groups, and shows signals for a methoxyl and a β -methyl group. In addition an ABC pattern for three aromatic protons at δ 7.26 (*d*, *J* = 8 Hz), 7.63 (*t*, *J* = 8 Hz) and 7.79 (*d*, *J* = 8 Hz) indicated that one ring carries only a *peri*-substituent. Hydrolysis of the methoxyl group with 80% H₂SO₄ indicates that it is located at a *peri* position, and as the product was 2-hydroxyislandicin **1** the new pigment must be either **2** or **3**. The latter was excluded on the basis of a negative zirconium nitrate test, and the absence of a shift in the visible spectrum on addition of NaOAc–H₃BO₃ (for alizarin the shift is 22 nm) thus demonstrating the absence of a vicinal dihydroxy system. Hence the pigment has structure **2**. The trimethyl ether of **2** was identical with the tetramethyl ether of 2-hydroxyislandicin (**1**).

The second new pigment **4**, C₁₇H₁₄O₇, shows all the features of **2** except that the ¹H NMR spectrum reveals the presence of a second methoxyl group and the ABC system is replaced by two *meta* protons. Demethylation of



- 1 $R^1 = R^2 = R^3 = H$
- 2 $R^1 = Me, R^2 = R^3 = H$
- 3 $R^1 = R^3 = H, R^2 = Me$
- 4 $R^1 = Me, R^2 = H, R^3 = OMe$
- 5 $R^1 = R^2 = H, R^3 = OMe$

4 with 80% H_2SO_4 afforded calyculatone (5), and the trimethyl ether of 4 was found to be identical with calyculatone tetramethyl ether. These reactions show that 4 must be the 1- or 8-monomethyl ether of 5. A negative zirconium nitrate test and the absence of a bathochromic shift in the visible spectrum in the presence of $NaOAc-H_3BO_3$ indicate the absence of a vicinal dihydroxy system in the quinone. Hence the new pigment is calyculatone 1-methyl ether (4).

EXPERIMENTAL

For the isolation of the pigments, and other details, see ref. [2]. Fractions 57–96 afforded 2 and 4 after repeated CC (C_6H_6) and prep. TLC (C_6H_6).

Compound 2 was obtained as orange-red needles, (C_6H_6 -petrol) mp 210° (Found: C, 64.10; H, 4.24. $C_{16}H_{12}O_6$ requires C, 64.00; H, 4.03%); λ_{max}^{MeOH} nm (log ϵ): 225 (4.58), 255 (4.39), 280 (4.12), 443 (3.67); $\lambda_{max}^{MeOH/OH^-}$ nm: 226, 263, 325, 480; IR ν_{CHCl_3} cm^{-1} : 1630, 3400; 1H NMR (220 MHz, $CDCl_3$): δ 2.27 (3H, s, Me), 3.99 (3H, s, OMe), 6.95 (1H, br s, exchangeable with

D_2O , OH-2), 7.26 (1H, br d, $J = 8$ Hz, H-7), 7.63 (1H, br t, $J = 8$ Hz, H-6), 7.79 (1H, br d, $J = 8$ Hz, H-5), 12.90 and 13.99 (each 1H, s, exchangeable with D_2O , peri-OH); MS (Found: M^+ 300.0634; $C_{16}H_{12}O_6$ requires M , 300.0634) m/z (rel. int.): 300 (100), 285 (8), 283 (11), 282 (44), 272 (7), 271 (11), 257 (37), 254 (8), 253 (11), 244 (7), 226 (12). Methyl ether ($Me_2SO_4-K_2CO_3-Me_2CO$) yellow needles (petrol), mp $149-150^\circ$ (lit. [2] 150°). On demethylation (80% H_2SO_4 for 90 min at 100°) 2 afforded 2-hydroxyislandicin (1).

Compound 4 was obtained as orange-red micro needles (C_6H_6), mp 253° (Found C, 61.72, H, 4.38. $C_{17}H_{14}O_7$ requires C, 61.82; H, 4.27%); λ_{max}^{MeOH} nm (log ϵ): 227 (4.56), 270 sh (4.13), 285 (4.38), 422 (4.05), 440 (3.78); $\lambda_{max}^{MeOH/OH^-}$ nm: 228, 267, 302, 490; IR ν_{CHCl_3} cm^{-1} : 1630, 3450; 1H NMR (220 MHz, $CDCl_3$): δ 2.28 (3H, s, Me), 3.95 (3H, s, OMe), 4.00 (3H, s, OMe), 6.68 (1H, d, $J = 2$ Hz, H-7), 6.97 (1H, br s, exchangeable with D_2O , OH-2), 7.36 (1H, d, $J = 2$ Hz, H-5), 13.20 and 13.96 (each 1H, s, exchangeable with D_2O , peri-OH); MS m/z (rel. int.): 330 (100), 315 (5), 313 (16), 312 (72), 302 (7), 301 (16), 300 (7), 287 (24), 284 (8), 283 (6), 268 (8), 259 (5), 256 (8). Methyl ether ($Me_2SO_4-K_2CO_3-Me_2CO$) yellow needles (petrol) mp $145-146^\circ$ (lit. [3] 146°). On partial demethylation (80% H_2SO_4 for 90 min at 100°) 4 afforded calyculatone (5).

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