

## The Synthesis of De-ethylasycarpidone

By A. JACKSON and J. A. JOULE\*

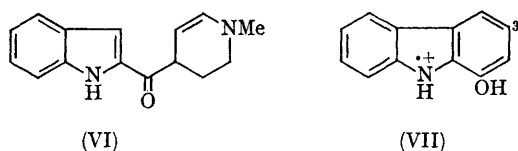
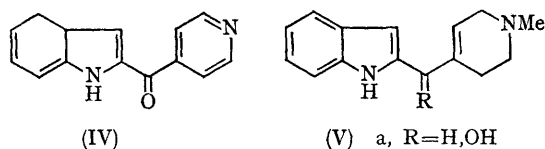
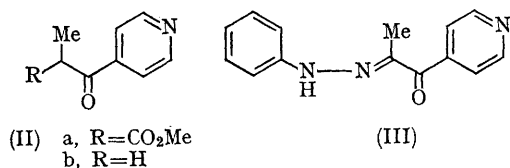
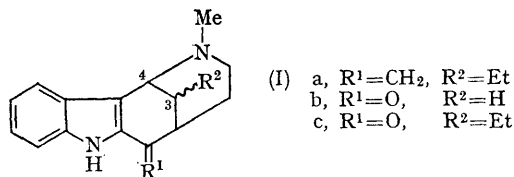
(*Chemistry Department, University of Manchester, Manchester 13*)

SINCE the elucidation<sup>1</sup> of the structure of uleine (Ia) by Büchi, six other alkaloids have been discovered<sup>2</sup> which are based on the same carbon skeleton. We present here the first synthesis of the basic four-ring system possessed by these alkaloids.

The Claisen condensation of methyl isonicotinate with methyl propionate, catalysed by sodium hydride, gave methyl 2-isonicotinylpropionate (IIa) as an oil, (b.p. 128—130°/1.5 mm.) in 65% yield. The  $\beta$ -keto-ester (IIa) was hydrolysed and decarboxylated in 80% yield by boiling with a mixture of acetic and sulphuric acids. The resulting ketone (IIb) was converted into an enamine with pyrrolidine in benzene and the

enamine coupled,<sup>3</sup> without purification, with benzenediazonium chloride. The phenylhydrazone (III) was thus obtained in 50% yield, (m.p. 223—225°). Indolisation of the phenylhydrazone, using phosphoric acid, gave 2-isonicotinylindole (IV) in 33% yield, (m.p. 172—174°,  $\lambda_{\text{max}}$  220, 265, and 329 m $\mu$ ;  $\log \epsilon$  4.31, 3.89, 4.24,  $\nu_{\text{max}}$  3300 and 1635 cm.<sup>-1</sup>). The methiodide of (IV) was reduced with potassium borohydride, giving an unsaturated alcohol (Va) in 90% yield, (m.p. 160—162°,  $\lambda_{\text{max}}$  220, 273, 282, 290 m $\mu$ ;  $\log \epsilon$  4.46, 3.90, 3.90, and 3.76,  $\nu_{\text{max}}$  3280 cm.<sup>-1</sup>). Manganese dioxide oxidation converted the alcohol (Va) into the corresponding conjugated ketone (Vb) in 67% yield, (m.p. 161—162°,  $\lambda_{\text{max}}$  241, 319 m $\mu$ ;  $\log \epsilon$  4.00,

4.23,  $\nu_{\max}$  3140, 1640, 1615  $\text{cm}^{-1}$ ). Deconjugation of the ketone (Vb) was achieved by successive



treatment with dimethyl sodium and ammonium chloride and the enamine (VI) (m.p. 152—155°) obtained in 55% yield. Ring closure of the enamine to de-ethyl-dasycarpidone (Ib) was achieved in 50% yield by treatment with 50% aqueous acetic acid at 100°. This process is envisaged as involving initial protonation at the  $\beta$ -carbon of the enamine to produce an electrophilic immonium group. Mannich condensation of the immonium grouping at the indole  $\beta$ -position then gives de-ethyl-dasycarpidone (Ib), (m.p. 215—216°,  $\lambda_{\max}$  237, 314  $\text{m}\mu$ ;  $\log \epsilon$  4.02, 4.21,  $\nu_{\max}$  3450, 1665  $\text{cm}^{-1}$ ). The mass spectrum of de-ethyl-dasycarpidone (Ib) showed a fragmentation pattern completely in accord with expectation, based upon the fragmentation of dasycarpidone (Ic) itself. For example, the base peak in the spectrum of (Ib) is at  $m/e$  183, corresponding to an ion (VII). The base peak in the spectrum of dasycarpidone (Ic) occurs<sup>4</sup> at  $m/e$  211, an ion which is the 3-ethyl homologue of (VII). The n.m.r. spectrum of de-ethyl-dasycarpidone (Ib) is also completely consistent with the structure assigned. The most relevant features are (a) the presence of only four aromatic hydrogen signals and the absence of an indole  $\beta$ -proton signal; (b) a one-proton multiplet at  $\tau$  5.55, which is attributed to the hydrogen at C-4; and (c) a one-proton signal at  $\tau$  0.41 for the indole  $N$ -hydrogen. In the n.m.r. spectrum of dasycarpidone,<sup>2b</sup> the hydrogen at C-4 resonates at  $\tau$  5.65 and is a doublet due to coupling with the single hydrogen at C-3.

This method is being extended to the synthesis of dasycarpidone and uleine.

(Received, March 28th, 1967; Com. 287.)

<sup>1</sup> G. Büchi and E. W. Warnhoff, *J. Amer. Chem. Soc.*, 1959, **81**, 4433.

<sup>2</sup> (a) M. Ohashi, J. A. Joule, B. Gilbert, and C. Djerassi, *Experientia*, 1964, **20**, 263; (b) J. A. Joule, M. Ohashi, B. Gilbert, and C. Djerassi, *Tetrahedron*, 1965, **21**, 1717.

<sup>3</sup> V. I. Shvedov, L. B. Altukhova, and A. N. Grinev, *Zhur. Org. Khim.*, 1965, **1**, 879. (*Chem. Abs.*, 1965, **63**, 6893h.)

<sup>4</sup> J. A. Joule and C. Djerassi, *J. Chem. Soc.*, 1964, 2777.