

ALTERPORRIOL C: A MODIFIED BIANTHRAQUINONE FROM *ALTER-NARIA PORRI**

RIKISAKU SUEMITSU, TOSHIBUMI UESHIMA, TOSHINARI YAMAMOTO and SATOSHI YANAGAWASE

Department of Applied Chemistry, Faculty of Engineering, Doshisha University, Kamikyo-ku, Kyoto 602, Japan

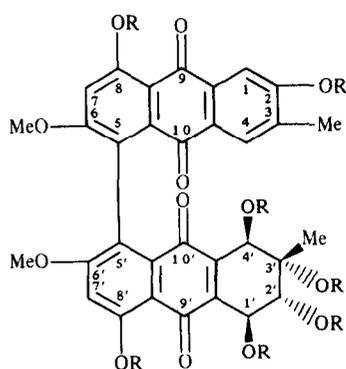
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Key Word Index—*Alternaria porri*; α,β' -bianthraquinone; alterporriol C.

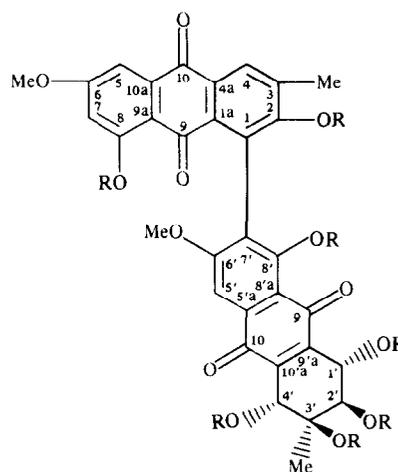
Abstract—The spent growth medium and mycelia of *Alternaria porri* afforded a novel anthraquinone named alterporriol C, whose structure was determined by spectroscopic methods as well as by degradation to known compounds. Alterporriol C was found to be an α,β' -bianthraquinone derived from macrosporin and altersolanol A.

INTRODUCTION

In the course of our investigation on the pigments of *Alternaria porri* (Ellis) Ciferri, the causal fungus of black spot disease of stone-leek (Japanese name: negi), we isolated three dark red pigments from the culture liquid and mycelia [1]. The structures of two of these pigments, alterporriol A (Ap-A, **1**) and B (Ap-B, **1**), have already been determined and found to be atropisomers of each other [1, 2]. This paper deals with the structural elucidation of the third pigment (Ap-C, **2**). Cladofulvin [3] and asphodelin, [4] have been reported as naturally occurring α,β' -bianthraquinone. The former is a product of *Cladosporium fulvum* and the latter is a component of *Aloe saponaria* HAW. Ap-C is the second example of an α,β' -bianthraquinone of fungal origin.



R
1 H
1a Ac



R
2 H
2a Ac

RESULTS AND DISCUSSION

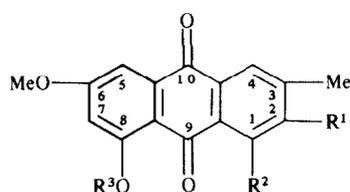
When the $^1\text{H NMR}$ spectrum of Ap-C was compared with those of macrosporin (**3**) and altersolanol A (**4**), both are metabolic pigments of *Alternaria porri* [5, 6], it was obvious that the spectrum of Ap-C was the sum of those of **3** and **4** (Table 1). This suggested Ap-C was a modified bianthraquinone formed from **3** and **4**. To confirm the identities of the structural fragments of Ap-C, the reductive cleavage products formed on treatment of Ap-C with alkaline sodium dithionite were examined. Compounds **3**, **4** and 1,2,8-trihydroxy-6-methoxy-3-methylantraquinone (**5**) were identified by comparing their R_s with those of authentic samples [1]. The combined results showed that Ap-C is another modified bianthraquinone derived from the monomeric units **3** and **4**.

The manner of the attachment of **4** to **3** followed from the $^1\text{H NMR}$ spectrum of Ap-C. As shown in Table 1, the

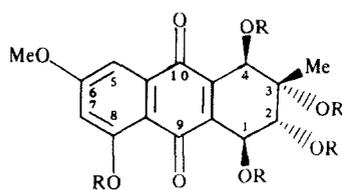
*Studies of the Metabolic Products of *Alternaria porri*, part 14. For Part 13 see ref. [2].

Table 1. ^1H NMR spectral data of macrosporin (**3**), altersolanol A (**4**) and alterporriol C (**2**) (400 MHz, d_8 -THF with TMS)

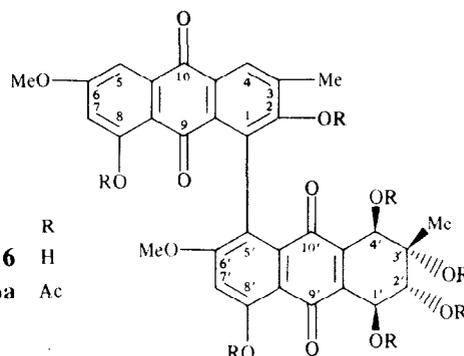
H	3	H	4	H	2
4	8.10 s	4	8.10 s	4	8.10 s
1	7.51 s	1	---	1	---
7	6.70 d $J = 2.2$ Hz	7	6.61 d $J = 2.2$ Hz	7	6.61 d $J = 2.2$ Hz
5	7.27 d $J = 2.2$ Hz	5	7.25 d $J = 2.2$ Hz	5	7.25 d $J = 2.2$ Hz
Me-3	2.32 s	Me-3	2.32 s	Me-3	2.32 s
OMe-6	3.91 s	OMe-6	3.88 s	OMe-6	3.88 s
OH-2	9.70 s	R-OH $\times 4$	2.58 br, s	OH-5 and 8'	{ 12.59 13.08
OH-8	12.86 s	OH-8	12.21 s	4'	4.30 s
		4	4.41 s	2'	4.15 d $J = 6.5$ Hz
		2	3.77 d $J = 6.6$ Hz	1'	4.96 d $J = 6.5$ Hz
		1	4.64 d $J = 6.6$ Hz	7'	---
		7	6.71 d $J = 2.2$ Hz	5'	6.85 s
		5	7.11 d $J = 2.2$ Hz	Me-3'	1.27 s
		3	1.37 s	OMe-6'	3.71 s
		OMe-6	3.90 s		



	R ¹	R ²	R ³
3	OH	H	H
3a	OAc	H	Ac
5	OH	OH	H



	R
4	H
4a	Ac



	R
6	H
6a	Ac

^1H NMR spectra of **3** and **4** exhibit the signals of H-5 and H-7 as a pair of doublets because of *meta*-coupling. In the spectrum of Ap-C, a pair of doublets appear at δ 6.61 and 7.25 corresponding to H-7 (δ 6.70) and H-5 (δ 7.27) in **3**, respectively. In addition, an aromatic proton signal at δ 8.10 in Ap-C can be attributed to H-4 (δ 8.10) in **3**. Moreover, the 2D ^1H NMR (NOESY) spectrum of Ap-C heptaacetate (**2a**) shows that the signal due to the aromatic proton at δ 8.21 (H-4) is correlated to the aromatic methyl protons at δ 2.26 (Me-3). These results suggest that the **3** moiety of Ap-C is linked at C-1 to the **4** moiety, and it follows that the internuclear C-C link must be C-1 to C-7' (**2**) or C-5' (**6**).

Ogihara *et al.* have reported [7] that in the ^1H NMR spectra of acetoxyated bianthraquinones the signals due to the protons of acetate groups *ortho* to the internuclear bond resonate at higher field than usual on account of anisotropy caused by the aromatic ring of the other 'half' of the molecule. In the ^1H NMR data of Ap-A previously reported [2], the signals due to methoxyl protons *ortho* to the internuclear C-C link resonate at higher field than those of its constituents, **3** and **4**. In the case of **2a**, a methoxyl group (OMe-6') and two acetoxy groups (OAc-2 and OAc-8') resonate at unusually high field as shown in Table 2. Namely, signals of OMe-6' (δ 3.70; **4a**: 3.96), OAc-2 (δ 2.14; **3a**: 2.33) and OAc-8' (δ 2.10; **4a**: 2.38) are observed. Hence, the structure of **2** and **2a** are compatible with those of Ap-C and Ap-C heptaacetate. The structure **6** must be excluded, because it has a methoxyl and an acetoxy group located *ortho* to the internuclear C-C linkage.

EXPERIMENTAL

Extraction and isolation of alterporriol C. The cultural condition using the stone-leek decoction as a medium and the method of isolation have already been reported [1]. Crude Ap-C obtained by prep. TLC was purified by reversed phase HPLC: (YMC-S343 ODS-type, solvent MeCN-H₂O 3:7). Yields of Ap-C were 81 mg from 30 l culture medium.

Table 2. Assignment of acetyl and methoxyl protons (400 MHz, CDCl₃ with TMS)

Ap-C heptaacetate (6a)	Macrosporin diacetate (3a)	As-A pentaacetate (4a)*
1.91 (C-4)†		1.98 (C-4)
1.94 (C-1')		2.06 (C-1)
2.02 (C-2')		2.13 (C-2)
2.08 (C-3')		2.15 (C-3)
2.10 (C-8')		2.38 (C-8)
2.14 (C-2)	2.33 (C-2)	
2.27 (C-8)	2.31 (C-8)	
3.70 (C-6')		3.96 (C-6)
3.96 (C-6)	3.93 (C-6)	

*Values documented in ref. [8].

†Carbon number attached to OAc and OMe group.

Table 3. ¹³C NMR spectral data of alterporriol C heptaacetate (2a) (INEPT method) (400 MHz, CDCl₃ with TMS)

Me	C-3	16.83	quat. -C	C-3'	80.54	OCOMe	166.64
	C-3'	17.01		C-9'a	116.74		167.16
OAc		19.67		C-7'	116.74		168.04
		20.36		C-9a	118.57		168.17
		20.57		C-1	125.13		168.41
		20.59		C-4a	129.81		168.81
		20.92		C-1a	130.74		169.67
		21.08		C-10'a	132.40	C=O	C-9 179.67
	21.85		C-10a	135.69		C-9' 180.16	
OMe	C-6	55.89		C-3	136.13		C-10 180.91*
	C-6'	56.53		C-5'a	137.97		C-10' 181.27†
-CH	C-4'	64.00		C-8'a	141.87		
	C-1'	66.89		C-8'	150.05		
	C-2'	72.85		C-8	150.77		
	C-5	108.65		C-2	151.30		
	C-5'	109.60		C-6	161.39		
	C-7	115.30		C-6'	163.37		
	C-4	129.15					

Assignments of * and † may be reversed, respectively.

Alterporriol C. Dark-red amorphous, mp > 300° (dec.), [α]_D²⁵ -268° (EtOH; c 0.05). IR ν_{\max}^{KBr} cm⁻¹: 3600–3000(OH), 1670, 1650 (free C=O), 1640 (chelated C=O), 1210 (Ar-OH), 1100 (sec OH); UV $\lambda_{\max}^{\text{EtOH}}$ nm (log. ϵ) 228 (5.91), 267 (5.68), 320 (5.33), 360 (4.51), 422 (4.69), 450 (4.69); MS(FD) (rel. int.) m/z : 618 [M]⁺ (24) 582 [M - 2H₂O]⁺ (100), 566; MS(SIMS) m/z (rel. int.): 643 [M + Na + 2H]⁺ (4), 621 [M + 3H]⁺ (44), 567.1306 (100), Calc. for C₃₂H₂₃O₁₆, m/z 567.1285.

Alterporriol C heptaacetate (2a). On acetylation with Ac₂O containing a few drops of 70% perchloric acid, followed by purification using HPLC (YMC-S343 ODS-type, solvent MeCN-H₂O 11:9), Ap-C (20 mg) gave a yellow acetate (8 mg), mp 270–271°. MS(FD) m/z : 912. [M]⁺, corresponding to C₃₂H₁₉O₁₃Ac₇ (912.78); NMR $\delta_{\text{Me}_4\text{Si}}^{\text{CDCl}_3}$: 1.53 (3H, s, Me-3'), 2.41 (3H, s, Me-3), 1.91, 1.94, 2.02, 2.08, 2.10, 2.14, 2.27 (3H, each, s, ROCOMe), 3.70, 3.96, (3H, each, s, Ar-OMe), 5.53 (1H, d, J = 7.3 Hz, H-2'), 6.21 (1H, d, J = 7.3 Hz, H-1'), 6.84 (1H, s, H-4'), 6.79 (1H, s, H-5'), 6.80 (1H, d, J = 2.6 Hz, H-7), 7.70 (1H, d, J = 2.6 Hz, H-5) 8.20 (1H, s, H-4); ¹³C NMR: Table 3. The assign-

ments were based on comparison with the chemical shifts of 3a and 4a [8].

Reductive cleavage of Ap-C. To a sol of Ap-C (2 mg) in 0.5 M NaOH (5 ml), an aq. sol of Na₂S₂O₄ (5 mg) in 5 ml H₂O was added. After heating at 70° for 30 min, the reaction mixture was cautiously neutralized with cooling and extd with EtOAc. After evapn of the dried extract (Na₂SO₄), the residue was analysed by HPLC. For analysis, a YMC-A314 ODS-type column was used with a mobile phase of MeOH-H₂O (7:3) for 30 min, which was then increased to 100% MeOH during the next 40 min. Compounds 3, 4 and 5 were identified by comparing their R_s with those of authentic samples.

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