

DITERPENOIDS FROM *RABDOSIA KUNMINGENSIS*

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Key Word Index—*Rabdosia kunmingensis*; rabdokunmin A, B, C, D, E; *ent*-kaurene diterpenoid.

Abstract—The investigation of the dried leaves of *Rabdosia kunmingensis* afforded rabdoloxin B, 4-*epi*-isopimaric acid, callitrisic acid and five new *ent*-kaurene diterpenoids. The structures were elucidated by spectroscopic and chemical means.

INTRODUCTION

The genus *Rabdosia* (Labiateae) is known to produce *ent*-kaurene diterpenoids [1]. This paper reports our examination of the chemical constituents of *Rabdosia kunmingensis* C. Y. Wu et H. W. Li, a perennial herb of south-eastern China which has not been investigated previously for its chemical constituents. From the ethereal extract of dried leaves of *R. kunmingensis*, eight diterpenoids (1–8), including five new *ent*-kaurene diterpenoids (1–5) were isolated.

RESULTS AND DISCUSSION

The leaves gave 4-*epi*-isopimaric acid (7) [2], callitrisic acid (8) [3], rabdoloxin B (6) [4], as well as five new diterpenoids, named rabdokunmin A (1), B (2), C (3), D (4), and E (5).

Rabdokunmin A (1), $C_{22}H_{32}O_6$ ($[M]^+$ at m/z 392), was shown to be a monoacetate from the NMR spectra (Tables 1 and 2) (δ 2.10, 3H, s; δ 21.9 q, 171.4 s). The very similar mass spectrum to that of rabdoloxin B (6) suggested that 1 was a monoacetate of rabdoloxin B (6). This was confirmed by treating 1 with 5% sodium carbonate to give rabdoloxin B (6). The appearance of the 14 α -H signal of 1 at a lower field (δ 7.25, 1H, *br s*) comparing with that of rabdoloxin B (δ 5.94, 1H, *br s*) revealed that the acetoxy group took the place of the secondary hydroxyl group at C-14 of 1. Structure 1 is therefore proposed for rabdokunmin A.

Rabdokunmin C (3), $C_{20}H_{30}O_5$ ($[M]^+$ at m/z 350), has a five-membered ketone conjugated with an *exo*-methylene group [λ_{max} 231 nm (log ϵ 3.81); ν_{max} 1720, 1650 cm^{-1} ; δ 5.42, 6.29 (each 1H, *br s*); δ 117.0 (t), 147.7 (s), 209.0 (s)] and two tertiary methyl group (δ 0.87, 1.67, each 3H, s; δ 17.0 q, 18.0 q). The ^{13}C NMR spectrum (Table 2) also showed the presence of six methylenes, six methines and three quaternary carbons. Based on these facts and the consideration of the structures of diterpenoids from the genus *Rabdosia*, together with the negative Cotton effect of dihydrorabdokunmin C (9) [5], we presume compound 3 has an *ent*-kaurene skeleton. Rabdokunmin C (3) has four hydroxy groups (ν_{max} 3340 cm^{-1} ;

δ 6.15, 6.93, 7.41, 7.88, each 1H, *br s*, disappeared after D_2O), which could be confirmed by acetylation of 3 with pyridine–acetic anhydride giving a tetraacetate (10) ($C_{28}H_{38}O_9$ [$M-MeCOOH$] $^+$ at m/z 518; ν_{max} 1735, 1238 cm^{-1} ; δ 1.79, 2.11, 2.15, 2.16, each 3H, s; δ 20.8 q, 21.1 q, 21.2 q, 21.3 q, 169.6 s, 169.7 s, 170.7 s, 170.9 s). The four hydroxy groups were presumably assigned as 18-OH, 14 β -OH, 12 α -OH, 7 α -OH of 3, respectively, by observing the 1H and ^{13}C NMR spectral signals. The 1H and ^{13}C NMR spectra of 3 showed the presence of a hydroxymethyl group (δ 3.37, 3.69, each 1H, AB, d , J = 10.0 Hz; δ 71.6 t), which was assigned to C-18 according to the following observation [6]. The location of the hydroxymethyl group at C-18 resulted in a striking γ -effect on C-3 ($\Delta\delta$ – 6.2) and β -effect on C-4 ($\Delta\delta$ + 5.4) compared with compound 6. The signal for C-5 showed nearly the same shielding γ -effects ($\Delta\delta$ – 6.7) (Table 2). The presence of 14 β -OH (δ 5.82, 1H, *br s*, 14 α -H) and 7 α -OH (δ 4.98, *dd*, J = 3.7, 10.0 Hz, 7 β -H) groups was confirmed by formation of the acetone (11) owing to their co-planar relationship [7] ($C_{23}H_{34}O_5$, [M] $^+$ at m/z 390), ν_{max} 3430, 1728, 1641, 1109 cm^{-1} ; δ 1.43, 1.60, each 3H, s, 2 \times Me on dimethoxypropyl group; 4.82, 1H, *dd*, J = 3.7, 10.0 Hz, 7 β -H; 5.44, 1H, *br s*, 14 α -H; δ 25.4 q, 31.4 q, 67.4 d, 71.2 d, 97.4 s) which proved to be the 7 α , 14 β -acetone of compound 3. The downfield shift of the ^{13}C NMR signal for C-13 (δ 55.7 d) and the downfield shifts of 20-H (1.63, 3H, s) and 14 α -H (5.59 1H, *br s*) indicated the presence of a 12 α -OH (4.35, 1H, *m*, 12 β -H), which was confirmed by observing the γ -*gauche* shielding effect between 12 α -OH and 14 α -H (Table 2). Compound 3 was oxidized with Beckmann's reagent to give 12, the 1H and ^{13}C NMR spectral data of which were identical with the known compound macrocalyxin C (12) [8]. Thus, structure 3 was assigned to rabdokunmin C.

Rabdokunmin B (2), D (4), and E (5) have similar structures to rabdokunmin C (3). By observing the spectra of these three compounds, we noticed that their NMR spectra were similar to those of 3, which suggested that 2, 4 and 5 had the same skeleton as 3. The 1H and ^{13}C NMR spectra showed the presence of 18-CH₂OH and 14 β -OH in 2, 4 and 5 (Tables 1 and 2).

Rabdokunmin B (2), $C_{20}H_{30}O_4$ ($[M]^+$ at m/z 334), λ_{max} 231 nm (log ϵ 3.82), was shown to contain three hydroxy groups including 18-CH₂OH and 14 β -OH (δ 5.13, 6.62, 7.07, each 1H, *br s*, disappeared after D_2O).

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Table 1. ^1H NMR spectra of compounds 1–6 (ppm from int. TMS)

	1	2	3	4	5	6
7 β -H	4.89 <i>dd</i> (4.2, 12.0)		4.98 <i>dd</i> (3.7, 10.0)	5.02 <i>dd</i> (3.6, 11.7)	5.12 <i>dd</i> (3.3, 11.8)	4.99 <i>dd</i> (4.5, 11.8)
11 α -H	4.52 <i>br s</i>			4.25 <i>d</i> (3.3)	4.45 <i>br s</i>	4.43 <i>br s</i>
12 β -H	4.81 <i>d</i> (3.1)	4.46 <i>m</i>	4.35 <i>m</i>		4.73 <i>m</i>	4.73 <i>m</i>
13 α -H	3.69 <i>m</i>	3.68 <i>m</i>	3.58 <i>m</i>	3.32 <i>m</i>	3.74 <i>m</i>	3.75 <i>m</i>
14 α -H	7.23 <i>br s</i>	5.59 <i>br s</i>	5.82 <i>br s</i>	5.41 <i>br s</i>	5.97 <i>br s</i>	5.94 <i>br s</i>
17-H _a	5.55 <i>br s</i>	5.38 <i>br s</i>	5.42 <i>br s</i>	5.26 <i>br s</i>	5.48 <i>br s</i>	5.52 <i>br s</i>
17-H _b	6.35 <i>br s</i>	6.28 <i>br s</i>	6.29 <i>br s</i>	6.27 <i>br s</i>	6.35 <i>br s</i>	6.39 <i>br s</i>
18-H _a		3.37 <i>d</i> (10.6)	3.32 <i>d</i> (10.6)	3.30 <i>d</i> (10.6)	3.34 <i>d</i> (10.5)	
18-H _b		3.69 <i>d</i> (10.6)	3.65 <i>d</i> (10.6)	3.65 <i>d</i> (10.6)	3.66 <i>d</i> (10.5)	
18-Me	0.85 <i>s</i>					0.83 <i>s</i>
19-Me	0.80 <i>s</i>	0.84 <i>s</i>	0.87 <i>s</i>	0.88 <i>s</i>	0.91 <i>s</i>	0.82 <i>s</i>
20-Me	1.94 <i>s</i>	1.63 <i>s</i>	1.67 <i>s</i>	1.12 <i>s</i>	1.68 <i>s</i>	1.58 <i>s</i>
OAc	2.10					
OH*	6.29, 6.43 7.62	6.10, 6.82 7.07	6.15, 6.93 7.88	5.80, 6.05 7.47, 8.12	6.04, 7.15 7.45, 7.88	6.24, 7.35 7.48, 8.06

*These signals disappeared on addition of D₂O.

Table 2. ^{13}C NMR spectra of compounds 1–6 (ppm from int. TMS)

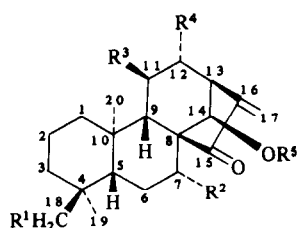
C	1	2	3	4	5	6
1	39.4	39.7	39.5	39.2	39.1	39.4
2	18.8	18.4	18.3	18.3	18.5	18.7
3	41.9	35.9	35.6	35.4	35.8	41.8
4	33.4	39.0	38.7	40.4	39.1	33.3
5	52.5	48.7	46.5	46.9	46.9	53.2
6	29.0	18.6	29.7	29.3	29.9	30.5
7	73.3	27.1	74.5	76.0	75.0	74.8
8	61.4	59.4	61.7	60.0	60.3	60.1
9	68.9	57.4	57.2	64.3	68.0	67.6
10	39.2	38.2	38.0	38.6	38.2	39.0
11	70.8	26.0	26.5	66.3	71.2	71.0
12	79.8	73.6	72.5	37.8	79.4	79.2
13	53.9	55.7	55.6	45.9	54.8	54.7
14	72.9	68.6	71.3	74.5	71.9	71.7
15	206.7	210.9	209.0	207.6	208.1	208.0
16	146.3	147.5	147.7	150.3	147.8	147.7
17	115.4	115.9	117.0	114.7	116.0	115.9
18	33.6	71.6	71.3	71.1	71.6	33.5
19	21.9	17.9	18.0	18.5	18.2	21.8
20	17.2	16.9	17.0	17.9	18.1	17.4
Ac ₂ O	171.4 21.9					

This compound was identified, as the 7-dehydroxy derivative (2) of rabdokunmin C (3) (Table 1 and 2).

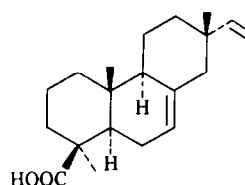
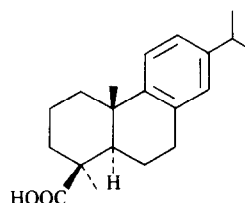
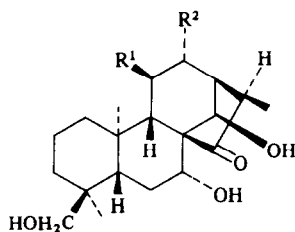
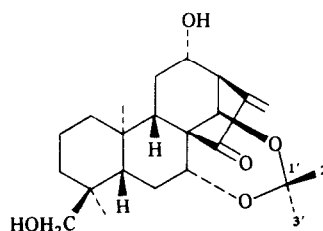
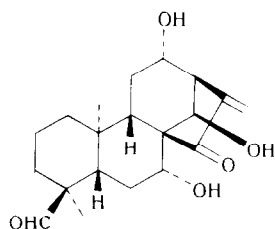
Rabdokunmin D (4), C₂₀H₃₀O₅ ([M]⁺ at *m/z* 350), λ_{max} 237 nm (log ϵ 3.83), was shown to have two hydroxy groups in addition to those at 18-CH₂OH and 14 β -OH. The signal at δ 5.02 (*dd*, *J* = 3.6, 11.7 Hz) was attributed to 7 β -H compared with that of 3. The signal at δ 4.25 due to

the 11 α -proton appeared as a doublet (*J* = 3.3 Hz). Rabdokunmin D was therefore elucidated to have structure 4.

Rabdokunmin E (5), C₂₀H₃₀O₆ ([M]⁺ at *m/z* 366), λ_{max} 237 nm (log ϵ 3.75), was shown to have three hydroxy groups in addition to 18-CH₂OH and 14 β -OH. The comparison of the ^1H and ^{13}C NMR spectra of 5 with those of



- 1** $R^1 = H, R^2 = R^3 = R^4 = OH, R^5 = Ac$
2 $R^1 = R^4 = OH, R^2 = R^3 = R^5 = H$
3 $R^1 = R^2 = R^4 = OH, R^3 = R^5 = H$
4 $R^1 = R^2 = R^3 = OH, R^4 = R^5 = H$
5 $R^1 = R^2 = R^3 = R^4 = OH, R^5 = H$
6 $R^1 = R^5 = H, R^2 = R^3 = R^4 = OH$
10 $R^1 = R^2 = R^4 = OAc, R^3 = H, R^5 = Ac$

**7****8****9** $R^1 = H, R^2 = OH$ **13** $R^1 = OH, R^2 = H$ **11****12**

rabdoloxin B (**6**) led to the assignment of these three hydroxy groups (Table 1 and 2). Thus, structure **5** was assigned to the fifth new diterpenoid.

EXPERIMENTAL

General. Mps: uncorr. CD: MeOH. IR: KBr. UV: EtOH. $^1\text{H NMR}$ (400.13 MHz) and $^{13}\text{C NMR}$ (100.61 MHz, broad band and DEPT): pyridine- d_5 , TMS as int. standard; EIMS: 70 eV.

The dried leaves (4.5 kg) of *R. kunmingensis*, which were collected in Kunming, Yunnan, China in Nov. 1987, were extracted with Et_2O to give a green residue after evapn of the Et_2O . The residue was refluxed in MeOH with charcoal (50 g) for 1 hr then filtered ($\times 3$) to give a yellow soln. The yellow soln was

evapd to 1/3 volume of the mother liquid and filtered to give 50 g yellow powder. Evaporation of the MeOH left 100 g residue, which was chromatographed on a silica gel (1 kg) column. Elution with $\text{Me}_2\text{CO}-\text{CHCl}_3$ gave rabdokunmin A (**1**) (70 mg), B (**2**) (30 mg), C (**3**) (1.5 g), D (**4**) (500 mg), E (**5**) (80 mg), rabdoloxin B (**6**) (1.3 g) and crystals (11 g). The crystalline material was chromatographed on a RP-8 silica gel column to give 4-*epi*-isopimaric acid and callitrisic acid.

Rabdokunmin A (1). Colourless rods (from MeOH), mp $212.0-214.0^\circ$, $[\alpha]_D^{21} -51.0^\circ$ (Me $_2$ CO; c 0.51); UV λ_{max} nm (log ϵ): 242 (3.77); IR ν_{max} cm^{-1} : 3410, 1722, 1706, 1645, 1260; EIMS m/z : 392 $[\text{M}]^+$, 350 $[\text{M}-\text{CH}_2=\text{C}=\text{O}]^+$, 332 $[\text{350}-\text{H}_2\text{O}]^+$, 314 $[\text{350}-2\text{H}_2\text{O}]^+$. (Found: C, 64.00; H, 8.33. $\text{C}_{22}\text{H}_{32}\text{O}_6 \cdot \text{H}_2\text{O}$ requires: C, 64.40; H, 8.35%).

Rabdokunmin B (2). Colourless needles (from MeOH), mp $259.5-261.5^\circ$, $[\alpha]_D^{21} -46.2^\circ$ (MeOH; c 0.52); UV λ_{max} nm (log ϵ): 231 (3.83); IR ν_{max} cm^{-1} : 3340, 1701, 1641, 1078; EIMS m/z : 334 $[\text{M}]^+$, 316 $[\text{M}-\text{H}_2\text{O}]^+$, 298 $[\text{M}-2\text{H}_2\text{O}]^+$. (Found: C, 71.60; H, 9.01. $\text{C}_{20}\text{H}_{30}\text{O}_4$ requires: C, 71.80; H, 9.04%).

Rabdokunmin C (3). Colourless needles (from MeOH), mp $145.0-146.0^\circ$, $[\alpha]_D^{21} -85.7^\circ$ (MeOH; c 0.54); UV λ_{max} nm (log ϵ): 231 (3.81); IR ν_{max} cm^{-1} : 3340, 1720, 1650, 1091; EIMS m/z : 350 $[\text{M}]^+$, 332 $[\text{M}-\text{H}_2\text{O}]^+$, 314 $[\text{M}-2\text{H}_2\text{O}]^+$, 296 $[\text{M}-3\text{H}_2\text{O}]^+$. (Found: C, 65.30; H, 8.70. $\text{C}_{20}\text{H}_{30}\text{O}_5 \cdot \text{H}_2\text{O}$ requires: C, 65.30; H, 8.75%).

Rabdokunmin D (4). Colourless needles (from Me₂CO), mp 254.0–257.0°, [α]_D²¹ –113.3 (MeOH; c 0.57); UV λ_{\max} nm (log ϵ): 237 (3.83); IR ν_{\max} cm^{–1}: 3370, 3270, 1719, 1647; EIMS m/z : 332 [M–H₂O]⁺, 314 [M–2H₂O]⁺. (Found: C, 65.50; H, 8.75. C₂₀H₃₀O₅·H₂O requires: C, 65.20; H, 8.75%).

Rabdokunmin E (5). Colourless needles (from Me₂CO), mp 286.0–288.0°, [α]_D²¹ –110.5° (MeOH; c 0.51); UV λ_{\max} nm (log ϵ): 237 (3.75); IR ν_{\max} cm^{–1}: 3320, 1715, 1645; EIMS m/z : 348 [M–H₂O]⁺, 330 [M–2H₂O]⁺, 312 [M–3H₂O]⁺. (Found: C, 62.20; H, 7.90. C₂₀H₃₀O₆·H₂O requires: C, 62.50; H, 7.87%).

Hydrogenation of 3. A mixture of Pd/C (8 mg), MeOH, and **3** (70 mg) was stirred under H₂ at room temp. for 4 hr. After removal of the catalyst the soln was evapd affording a residue which was crystallized from MeOH to give **9** (68 mg), mp 284.5–286.0° (decomp.); CD: [θ]₃₀₆ –48.9; IR ν_{\max} cm^{–1}: 3390, 1716, 1050, 1030; ¹H NMR: δ 0.87 (3H, s, 19-Me), 1.18 (3H, d, J = 7.3 Hz, 17-Me), 1.67 (3H, s, 20-Me), 3.26 (1H, d, J = 7.2 Hz, 13 α -H), 3.31, 3.64 (each 1H, AB d, J = 10.5 Hz, 18-H₂), 4.38 (1H, m, 12 β -H), 4.81 (1H, br d, J = 11.2 Hz, 7 β -H), 5.88 (1H, br s, 14 α -H); ¹³C NMR: δ 9.6 (C-17), 17.0 (C-20), 17.9 (C-19), 18.2 (C-2), 26.5 (C-11), 29.6 (C-6), 35.6 (C-3), 37.9 (C-10), 38.4 (C-4), 39.2 (C-1), 43.3 (C-16), 46.7 (C-5), 50.9 (C-13), 57.0 (C-9), 60.6 (C-8), 66.6 (C-12), 71.3 (C-18), 71.5 (C-14), 75.0 (C-7), 221.7 (C-15); EIMS m/z : 334 [M–H₂O]⁺, 316 [M–2H₂O]⁺, 298 [M–3H₂O]⁺.

Acetylation of 3. Acetylation of **3** (50 mg) with 10 ml Ac₂O–pyridine (1:1) at room temp. overnight gave **10** (55 mg) after work-up in the usual manner. Mp 121.0–122.5°, IR ν_{\max} cm^{–1}: 1735, 1367, 1235, 1030; ¹H NMR: δ 0.82 (3H, s, 19-Me), 1.52 (3H, s, 20-Me), 1.86, 2.11, 2.15, 2.16 (each 3H, s, 4 \times COMe), 3.48 (1H, d, J = 3.5 Hz, 13 α -H), 3.65, 4.02 (each 1H, AB d, J = 11.1 Hz, 18-H₂), 5.22 (1H, t, J = 3.9, 12.0 Hz, 7 β -H), 6.54 (1H, br s, 14 α -H); ¹³C NMR: δ 16.3 (C-20), 17.7 (C-19), 17.9 (C-2), 20.8, 21.1, 21.2, 21.3 (4 \times COMe), 23.0 (C-11), 24.6 (C-6), 35.8 (C-3), 36.5 (C-4), 38.6 (C-1), 39.2 (C-10), 47.3 (C-5), 49.0 (C-13), 56.5 (C-9), 61.2 (C-8), 71.5 (C-14), 72.4 (C-18), 74.7 (C-7), 74.7 (C-12), 120.1 (C-17), 143.2 (C-16), 169.6, 169.9, 170.7, 170.9 (4 \times COMe), 204.1 (C-15); EIMS m/z : 458 [M–MeCOOH]⁺, 430 [458–CO]⁺, 416 [458–H₂C=C=O]⁺, 398 [458–MeCOOH]⁺, 356 [416–MeCOOH]⁺, 328 [356–CO]⁺.

7 α ,14 β -Acetonide of 3. Compound **3** (50 mg) was refluxed in Me₂CO in the presence of dry CuSO₄ for 24 hr. The reaction product was treated in the usual manner to yield **11** (49 mg), mp 127.0–128.0°, IR ν_{\max} cm^{–1}: 3430, 1730, 1640, 1109; ¹H NMR: δ 0.91 (3H, s, 19-Me), 1.40, 1.60 (each 3H, s, 2 \times Me on dimethoxypropyl group), 1.66 (3H, s, 20-Me), 3.35, 3.68 (each 1H, AB d, J = 10.6 Hz, 18-H₂), 3.52 (1H, d, J = 3.3 Hz, 13 α -H), 4.41 (1H, m, 12 β -H), 4.82 (1H, dd, J = 3.7, 12.0 Hz, 7 β -H), 5.38, 6.29 (each 1H, br s, 17-H₂), 5.44 (1H, br s, 14 α -H), 6.30, 7.15 (each 1H, br s, disappeared after D₂O, 2 \times OH); ¹³C NMR: δ 16.8 (C-20), 17.9 (C-19), 18.2 (C-2), 25.4 (C-2'), 31.4 (C-3'), 26.6 (C-11), 27.9 (C-6), 35.8 (C-3), 38.0 (C-4), 38.0 (C-10), 39.4 (C-1), 45.1 (C-5), 52.1 (C-13), 54.6 (C-9), 55.0 (C-8), 67.4 (C-14), 71.2 (C-18), 71.2 (C-12), 71.8 (C-7), 97.2 (C-1'), 116.7 (C-17), 146.5 (C-16), 207.2 (C-15); EIMS m/z : 375 [M–Me]⁺.

Oxidation of 3. To a soln of **3** (30 mg) in Me₂CO (5 ml), Beckmann's reagent (10 drops) was added at 0°, and the mixture was stirred for 5 min then MeOH (1 ml) was added. After neutralization with aq. Na₂CO₃, the solvent was distilled off. The mixture after extraction with Me₂CO was chromatographed over silica gel. The prep. TLC yielded a white powder (15 mg) which proved to be **12**, IR ν_{\max} cm^{–1}: 3450, 3300, 2790, 1726, 1705, 1640, 1225, 940; ¹H NMR: δ 1.12 (3H, s, 19-Me), 1.61 (3H, s, 20-Me), 3.63 (1H, d, J = 3.2 Hz, 13 α -H), 4.38 (1H, t, J = 3.8 Hz, 12 β -H), 5.01 (1H, dd, J = 3.6, 12.0 Hz, 7 β -H), 5.41, 6.32 (each 1H, br s, 17-H₂), 5.79 (1H, br s, 14 α -H), 6.12, 7.20, 7.95

(each 1H, disappeared after D₂O, br s, 3 \times OH), 9.29 (1H, s, 18-CHO); ¹³C NMR: δ 14.2 (C-19), 16.6 (C-20), 17.1 (C-2), 26.3 (C-11), 32.3 (C-6), 32.2 (C-3), 37.6 (C-10), 38.6 (C-1), 45.3 (C-5), 49.7 (C-4), 55.6 (C-13), 56.6 (C-9), 61. (C-8), 71.1 (C-14), 72.2 (C-12), 73.7 (C-7), 117.2 (C-17), 147.5 (C-16), 205.9 (C-18), 208.4 (C-15); EIMS m/z : 348 [M]⁺, 330 [M–H₂O]⁺, 319 [M–CHO]⁺, 315 [M–H₂O–Me]⁺, 304 [M–CHO–Me]⁺.

Saponification of 1. To a soln of **1** (10 mg) in Me₂CO, 5% NaOH aq. soln (10 drops) was added, and the mixture was refluxed for 1 hr. Treatment of the reaction product as usual gave a white powder, which proved to be rabdoloxin B (**6**) by TLC.

Hydrogenation of 4. Compound **4** (70 mg) was hydrogenated in MeOH with Pd/C (10 mg) at room temp for 5 hr, and then the reaction product was treated in the usual manner to afford **13** (70 mg), mp 253.0–255.0°; CD: [θ]₃₀₅ –50.0; IR ν_{\max} cm^{–1}: 3395, 3280, 1722, 1028; ¹H NMR: δ 0.85 (3H, s, 19-Me), 1.10 (3H, s, 20-Me), 1.67 (3H, d, J = 7.1 Hz, 17-Me), 3.29, 3.64 (each 1H, AB d, J = 10.3 Hz, 18-H₂), 3.42 (1H, t, J = 7.0 Hz, 13 α -H), 4.14 (1H, d, J = 5.3 Hz, 11 α -H), 4.91 (1H, dd, J = 3.7, 11.2 Hz, 7 β -H), 5.32 (1H, brs, 14 α -H), 5.95, 6.22, 7.44, 8.00 (each 1H, br s, disappeared after D₂O, 4 \times OH); ¹³C NMR: δ 11.2 (C-17), 18.1 (C-20), 18.4 (C-2), 18.6 (C-19), 29.5 (C-6), 34.5 (C-12), 35.6 (C-3), 38.0 (C-10), 38.4 (C-4), 39.2 (C-1), 43.2 (C-16), 44.8 (C-13), 47.1 (C-5), 59.4 (C-8), 63.6 (C-9), 65.5 (C-11), 71.2 (C-18), 75.2 (C-14), 76.0 (C-7), 218.9 (C-15); EIMS m/z : 334 [M–H₂O]⁺, 316 [M–2H₂O]⁺, 298 [M–3H₂O]⁺.

Rabdoloxin B (6). C₂₀H₃₀O₅, colourless needles, mp 256.0–258.0°, [α]_D²¹ –92.5° (Me₂CO; c 0.51); UV λ_{\max} nm (log ϵ): 231 (3.82); IR ν_{\max} cm^{–1}: 3380, 1720, 1642.

4-epi-isoPimaric acid (7). Colourless needles, mp 145.5–146.0°, [α]_D²¹ 0° (Me₂CO; c 0.50); UV λ_{\max} nm (log ϵ): 265.5 (3.63); IR ν_{\max} cm^{–1}: 3300–2500, 1690, 1632, 1268, 1259; ¹H NMR (CDCl₃): δ 0.79 (3H, s, 17-Me), 0.86 (3H, s, 18-Me), 1.25 (3H, s, 20-Me), 4.87 (1H, dd, J = 1.0, 17.5 Hz, 16-H₂), 4.93 (1H, dd, J = 1.0, 17.5 Hz, 16-H₂), 5.38 (1H, d, J = 4.1 Hz, 14-H), 5.81 (1H, dd, J = 10.6, 17.5 Hz, 15-H); ¹³C NMR (CDCl₃): δ 14.3 (C-20), 19.6 (C-2), 21.0 (C-11), 21.6 (C-17), 24.4 (C-6), 29.2 (C-18), 36.0 (C-10), 36.5 (C-12), 36.9 (C-13), 38.1 (C-3), 39.9 (C-1), 43.9 (C-4), 46.3 (C-14), 51.4 (C-5), 51.8 (C-9), 109.2 (C-16), 121.5 (C-7), 134.8 (C-8), 150.4 (C-15), 183.9 (C-19); EIMS m/z : 302 [M]⁺, 285 [M–OH]⁺, 273 [M–CHO]⁺, 256 [M–CH₂=C=O]⁺, 242 [M–MeCOOH]⁺.

Callitricic acid. C₂₀H₂₈O₂, colourless crystals, mp 97.0–100°, [α]_D²¹ +127.6° (MeOH; c 0.52); UV λ_{\max} nm (log ϵ): 216 (4.00); IR ν_{\max} cm^{–1}: 3500–2500, 1690, 1495, 1265, 820; ¹H NMR (CDCl₃): δ 1.20 (6H, d, J = 6.9 Hz, 16,17-Me), 1.40 (6H, s, 18, 20-Me), 7.00 (1H, br s, 14-H), 7.10, 7.33 (each 1H, AB d, J = 8.2 Hz, 11, 12-H₂); ¹³C NMR (CDCl₃): δ 21.1 (C-2), 22.0 (C-6), 24.2 (C-16), 24.2 (C-17), 29.8 (C-18), 32.7 (C-7), 33.8 (C-15), 38.8 (C-10), 39.0 (C-3), 40.4 (C-1), 44.6 (C-4), 53.4 (C-5), 123.5 (C-8), 124.2 (C-12), 125.9 (C-11), 127.2 (C-14), 145.6 (C-18), 147.1 (C-9), 181.0 (C-18); EIMS m/z : 300 [M]⁺, 285 [M–Me]⁺, 267 [285–H₂O]⁺, 257 [285–CO]⁺, 249 [285–2H₂O]⁺, 240 [M–MeCOOH]⁺.

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REFERENCES

- Fujita, E. and Node, M. (1984) *Progr. Chem. Org. Nat. Prod.* **46**, 138.
- Kitajima, J., Komori, T. and Kawasaki, T. (1982) *Chem. Pharm. Bull.* **30**, 3912.

3. Welch, S. C., Hagan, C. P., Kim, J. H. and Chu, D. S. (1977) *J. Org. Chem.* **42**, 2879.
4. Sun Handong, Lin Zhongwen, Takada, Y. and Fujita, T. (1985) International Symposium on Organic Chemistry of Medicinal Natural Products, Abstracts B-172. Shanghai, China
5. MacMillan, J. and Walker, E. R. H. (1972) *J. Chem. Soc. Perkin I*, 986.
6. Gonzalez, Antonio G., Fraga, Braulio M., Hernandez, Melcheor C. and Hanson, J. R. (1981) *Phytochemistry*. **20**, 846.
7. Hirotsu, K., Kamikawa, T., Kubota, T. and Shimada, A. (1973) *Chem. Letters (Japan)* 255.
8. Wang Xianrong, Wan Zhaoquan, Dong Jinguang and Xue Zhaowen. (1984) *Acta Bot. Sinica* **26**, 425.