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Absolute Configuration of 3,3'-Dihydroxy-4,4'-biphenanthryl as Determined by the Stereochemistry of Cyclic Diester Formation with 1,1'-Binaphthyl-2,2'-dicarboxylic Acid

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Abstract: Condensation of racemic 3,3'-dihydroxy-4,4'-biphenanthryl [(\pm)-2] with (R)-1,1'-binaphthyl-2,2'-dicarboxylic acid dichloride (5) allows cyclization of only (-)-2 to give the 12-membered cyclic diester (7), the absolute configuration of which can be assigned (R,R) from steric reasons. Thus, absolute stereochemistry of (-)-2 is determined to be (R).

Previously we reported that in order to assemble a 12-membered cyclic diester containing two sets of biaryl units joined by ester-CO₂-linkages between the *ortho,ortho'*-positions, *e. g.* a compound 1, the directions of axial twist of the two biaryl units must be same for steric reasons as the CPK molecular models clearly indicate. Thus, this provides a convenient method for recognizing axial chirality of biaryls because by knowing one of the configurations of the two biaryl units of cyclic diesters such as 1 one can assign the configuration of the other.^{2,3}

Yamamoto et al. applied this method for the determination of the configuration of 3,3'-dihydroxy-4,4'-biphenanthryl (2) via the intramolecular Ullmann coupling of a diester of 1-bromo-2-naphthoic acid derived from (-)-2 to give the cyclic diester 7.4 They claimed that hydrolysis of the diester 7 afforded (S)-(-)-1,1'-binaphthyl-2,2'-dicarboxylic acid (S)-4, and thus assigned (-)-2 to be (S). On the contrary, Hayashi et al.⁵ have recently pointed out that (-)-2 should be (R) based on CD spectrum reported by Yamamura et al., who also had done an X-ray crystallographic analysis on a related compound 3.6

These contradictory reports prompted us to make sure whether the axial chirality recognition method can be applied to biphenanthrol 2 or it is an exceptional compound where the method can not be applied. Herein we report that the method still holds for 2, giving the result contrary to that of Yamamoto et al.⁴ and in agreement with Hayashi et al.⁵

To a well stirred, refluxing solution of benzene (160 ml), pyridine (5.0 ml) and 4-(dimethylamino)-pyridine (379 mg, 3.10 mmol) were added slowly (1 h) a solution of $(\pm)-2^7$ (600 mg, 1.55 mmol) in benzene

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(70 ml) and also a solution of (R)-5 (588 mg, 1.55 mmol) in benzene (70 ml) at a same rate, and then, the mixture was stirred under reflux for another 12 h. After the usual workup which included aqueous washings, extractions with diethyl ether and drying over MgSO₄, volatiles were removed under reduced pressure to leave a pale yellow residue. This material was chromatographed on a silica-gel column with hexane - dichloromethane (1:1) to give the cyclic diester 7 as the first eluted component [153 mg, 14% based on (R)-5]. Diastereomeric homogeneity of 7 was confirmed by the fact that reductive cleavage of the ester linkages by treatment with LiAlH₄ in THF gave enantiomerically pure (-)-29 as judged by chiral HPLC analyses. It can be said that upon treatment with $(\pm)-2$, (R)-5 serves as the template to pick up only (-)-2 to complete a cycle of the diester 7, the configuration of which should be (R,R). Thus, the absolute stereochemistry of (-)-2 is determined to be (R) as suggested by Hayashi et al.5

In conclusion, we have shown here an additional example where the axial chirality recognition method can be applied, correctly assigning the configuration of 3,3'-dihydroxy-4,4'-biphenanthryl (2). The intramolecular Ullmann coupling route by Yamamoto et al.4 should also have reached to the same conclusion but seemingly the rotation of 2 used for the Ullmann reaction after having been separated by a chiral HPLC might have been inadvertently exchanged.

References and Notes

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- 8. (R,R)-7: m.p. 290 °C (decomp.); $[\alpha]_D^{20}$ +622 (c 1.53, CHCl₃); IR (KBr) 1755 cm⁻¹; ¹H NMR (250 MHz; $CDCl_3$) δ 6.83 – 7.90 (m, ArH); FD-MS m/z (rel intensity) 694 (13), 693 (54), 692 (M⁺; 100), and 346 $(M^{2+}; 10)$. Found: C, 86.7; H, 4.35%. Calcd for $C_{50}H_{28}O_4$: C, 86.7; H, 4.1%. 9. (-)-2: m.p. 144 – 145 °C; $[\alpha]_D^{20}$ –68 (c 0.48, CHCl₃) [lit., 5 $[\alpha]_D^{20}$ –67 (c 0.13, CHCl₃)].

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