

## Pd<sup>II</sup>-catalysed Isomerisations of 3-Acetoxy-1,4-dienes to 1-Acetoxy-2,4-dienes: Stereochemical and Preparative Aspects

By BERNARD T. GOLDING\* and COLIN PIERPOINT

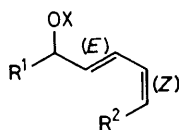
(Department of Chemistry and Molecular Sciences, University of Warwick, Coventry CV4 7AL)

and RAJINDRA ANEJA

(Unilever Research, Colworth Laboratory, Colworth House, Sharnbrook, Bedford MK44 1LO)

**Summary** The rapid Pd<sup>II</sup>-catalysed rearrangement of 3-acetoxy-1,4-dienes to 1-acetoxy-2,4-dienes [*e.g.* (2*E*, 5*Z*)-4-acetoxydeca-2,5-diene → (3*E*, 5*Z*)-2-acetoxydeca-3,5-diene (*ca.* 80%)] occurs in a stereoselective and regioselective manner.

THE structural fragment (1) occurs in several natural substances or their products of degradation.<sup>1</sup> We have found that this unit can be efficiently generated [*cf.* Table] by the Pd<sup>II</sup>-catalysed rearrangement of 3-acetoxy-1,4-dienes.<sup>2</sup>



(1) X = H, OH, or COR

The scope of this type of reaction was studied with substrates that possess either a vinyl group and a disubstituted double-bond (*E* or *Z*) or two disubstituted double-bonds (all combinations of *E* and *Z*) (*cf.* Table). For such transforma-

tions monitored by <sup>1</sup>H n.m.r. spectroscopy in [<sup>2</sup>H<sub>6</sub>]benzene we used as catalyst (PhCN)<sub>2</sub>PdCl<sub>2</sub> (5 mol %); for preparative experiments, (MeCN)<sub>2</sub>PdCl<sub>2</sub> (5 mol %) in tetrahydrofuran (THF) was employed. In a typical experiment, to (*E*, *Z*)-4-acetoxyhepta-2,5-diene (0.5 g) in dry THF (5 cm<sup>3</sup>) was added (MeCN)<sub>2</sub>PdCl<sub>2</sub> (42 mg) with stirring. After 5 min at room temperature the solution was evaporated, pentane was added, and the resulting suspension was filtered. The filtrate was concentrated and fractionally distilled (Kugelröhr; b.p. 85–87 °C/2 mmHg) to give an oil (0.46 g, 92%) containing (3*E*, 5*Z*)-2-acetoxyhepta-3,5-diene (*ca.* 80%) [ $\delta$  (CCl<sub>4</sub>) 1.30 (d, *J* 6 Hz, MeCHOAc), 1.75 (d, *J* 6.5 Hz, MeCH=), 1.9 (s, OCOMe), 5.3–5.7 (m, 2-, 5-, and 6-H), 5.9 (dd, *J* 10.5 and 10–11 Hz, 3-H), and 6.45 (dd, *J* 10.5 and 15 Hz, 4-H); *m/z* 154 (*M*<sup>+</sup>, 16), 112 (24), 95 (39), 79 (100), and 43 (83);  $\lambda_{\text{max}}$  (hexane) 228 nm ( $\epsilon$  23000)] and the (3*E*, 5*E*)-isomer (*ca.* 20%) (principally detected by the resonance for 4-H at  $\delta$  6.1) [assignments aided by additions of Eu(fod)<sub>3</sub> (fod = 1,1,1,2,2,3,3-heptafluoro-7,7-dimethyloctane-4,6-dionato) and *cf.* ref. 1b].

The results presented show that the Pd<sup>II</sup>-catalysed isomerisations of 3-acetoxy-1,4-dienes occur preferentially at (*E*)-disubstituted double-bonds. This circumstance is

TABLE. Pd<sup>II</sup>-catalysed rearrangement of 3-acetoxy-1,4-dienes.<sup>a</sup>

Substrate	Product(s) <sup>b</sup>
( <i>E,E</i> )-MeCH=CHCH(OAc)CH=CHMe	$\xrightarrow{10 \text{ min}}$ ( <i>E,E</i> )-MeCH=CHCH=CHCH(OAc)Me <sup>c</sup> (ca. 100%)
( <i>Z,Z</i> )-MeCH=CHCH(OAc)CH=CHMe	$\xrightarrow{30 \text{ min}}$ ( <i>Z,E</i> )-MeCH=CHCH=CHCH(OAc)Me (ca. 100%)
( <i>Z,E</i> )-R <sup>1</sup> CH=CHCH(OAc)CH=CHR <sup>2</sup> (R <sup>1</sup> , R <sup>2</sup> = Me, Me <sup>d</sup> ; Me, Bu; or Bu, Me)	$\xrightarrow{20 \text{ min}}$ ( <i>Z,E</i> )-R <sup>1</sup> CH=CHCH=CHCH(OAc)R <sup>2</sup> (80%) + 20% ( <i>E,E</i> )-isomer(s)
( <i>E</i> )-CH <sub>2</sub> =CHCH(OAc)CH=CHMe	$\xrightarrow{30 \text{ min}}$ ( <i>E</i> )-CH <sub>2</sub> =CHCH=CHCH(OAc)Me ( $\geq 95\%$ )
( <i>Z</i> )-CH <sub>2</sub> =CHCH(OAc)CH=CHMe	$\xrightarrow{48 \text{ h}}$ ( <i>E</i> )-CH <sub>2</sub> =CHCH=CHCH(OAc)Me (60%) + AcOCH <sub>2</sub> CH=CHCH=CHMe [30% ( <i>E,E</i> ) + 10% ( <i>E,Z</i> )-isomer]

<sup>a</sup> All transformations at room temperature, both in n.m.r. tubes and on a preparative scale (isolated yields > 90%). <sup>b</sup> All new compounds gave <sup>1</sup>H n.m.r., i.r., u.v., and electron impact mass spectra in accord with their assigned structures. <sup>c</sup> Identical with a sample prepared from sorbic aldehyde. <sup>d</sup> Increasing % of (*E,E*)-product with longer reaction times.

favourable for synthetic applications because (i) (*E,Z*)-3-acetoxy-1,4-dienes are easy to prepare from (*E*)- $\alpha,\beta$ -unsaturated aldehydes *via* (*E,Z*)-3-hydroxy-1,4-dienes;<sup>3</sup> they are converted into predominantly (*2E,4Z*)-1-acetoxy-2,4-dienes. (ii) The preferred direction of allylic rearrangement can be predicted [(*E,Z*)-R<sup>1</sup>CH=CHCH(OAc)CH=CHR<sup>2</sup>  $\rightarrow$  (*E,Z*)-R<sup>1</sup>CH(OAc)CH=CHCH=CHR<sup>2</sup>].

These points were illustrated with the substrates (*2E,5Z*)-4-acetoxydeca-2,5-diene and (*2Z,5E*)-4-acetoxydeca-2,5-diene. The (*2E,5Z*)-diene [in THF, (MeCN)<sub>4</sub>PdCl<sub>2</sub> (5 mol %), 10 min] gave an oil (92%) containing (*3E,5Z*)-2-acetoxydeca-3,5-diene (ca. 80%) and other isomer(s) (ca. 20%)<sup>†</sup> (analysis by <sup>1</sup>H n.m.r. spectroscopy). Hydrogenation (H<sub>2</sub>-Pt-THF) of this product gave 2-acetoxydecane (ca. 95%) and 5-acetoxydecane (5%) (analysis by g.l.c.). Similarly, (*2Z,5E*)-4-acetoxydeca-2,5-diene gave (*2Z,4E*)-6-acetoxydeca-2,4-diene (ca. 80%) and other isomer(s) (ca. 20%).<sup>†</sup> Hydrogenation of this product gave 5-acetoxydecane (ca. 88%) and 2-acetoxydecane (ca. 12%).

The reactions described are likely to be mechanistically related to Pd<sup>II</sup>-catalysed isomerisations of allylic acetates which probably take place *via* an intermediate acetoxonium ion.<sup>4-7</sup> The configuration of the main product is *E*, irrespective of whether the starting material is an (*E*)- or (*Z*)-allylic acetate.<sup>7</sup> It is significant that Pd<sup>0</sup>-catalysed rearrangements of 3-acetoxy-1,4-dienes take a quite different stereochemical course from the Pd<sup>II</sup>-catalysed reactions described. Thus, treatment of either (*E,E*)- or (*Z,Z*)-4-acetoxyhepta-2,5-diene with (Ph<sub>3</sub>P)<sub>4</sub>Pd (5 mol %)<sup>8</sup> in benzene gave, within a few minutes at room temperature, (*E,E*)-2-acetoxyhepta-3,5-diene (ca. 100%). Both (*E*)- and (*Z*)-3-acetoxyhexa-1,4-diene with (Ph<sub>3</sub>P)<sub>4</sub>Pd (5 mol %) in benzene gave predominantly (> 80%) (*E,E*)-1-acetoxyhexa-2,4-diene.

(Received, 19th June 1981; Com. 718.)

<sup>†</sup> The principal by-product appears to be the corresponding (*E,E*)-isomer (dd at  $\delta$  6.1 for 4-H). *N.B.* (*E,E*)-6-acetoxydeca-2,4-diene did not equilibrate with (*E,E*)-2-acetoxydeca-3,4-diene when incubated for 30 min at room temperature with (PhCN)<sub>2</sub>PdCl<sub>2</sub> (5 mol %) in benzene.

<sup>1</sup> (a) P. Wlodawer and B. Samuelsson, *J. Biol. Chem.*, 1973, **248**, 5673; (b) R. Tabacchi, J. Garnerio, and P. Buil, *Helv. Chim. Acta*, 1975, **58**, 1184; (c) N. A. Porter, B. A. Weber, H. Weenen, and J. A. Khan, *J. Am. Chem. Soc.*, 1980, **102**, 5597; N. A. Porter, D. H. Roberts, and C. B. Ziegler, *ibid.*, 5912.

<sup>2</sup> For some recent methods for preparing (*E,Z*)-conjugated dienes see: (a) H. Bosshardt and M. Schlosser, *Helv. Chim. Acta*, 1980, **63**, 2393; (b) G. Decodts, G. Dressaire, and Y. Langlois, *Synthesis*, 1979, 510; (c) G. Cassani, P. Massardo, and P. Piccardi, *Tetrahedron Lett.*, 1979, 633; (d) G. R. Knox and I. C. Thom, *J. Chem. Soc., Chem. Commun.*, 1981, 373; (e) J. E. Baldwin, N. V. Reed, and F. J. Thomas, *Tetrahedron*, 1981, **37**, 263; and ref. 1c.

<sup>3</sup> E. A. Braude and J. A. Coles, *J. Chem. Soc.*, 1951, 2078, 2085.

<sup>4</sup> P. M. Henry, *J. Am. Chem. Soc.*, 1972, **94**, 5200.

<sup>5</sup> L. E. Overman and F. M. Knoll, *Tetrahedron Lett.*, 1979, 321.

<sup>6</sup> J. Tsuji, K. Tsuruoka, and K. Yamamoto, *Bull. Chem. Soc. Jpn.*, 1976, **49**, 1701.

<sup>7</sup> P. A. Grieco, T. Takigawa, S. L. Bongers, and H. Tanaka, *J. Am. Chem. Soc.*, 1980, **102**, 7588.

<sup>8</sup> Cf. B. M. Trost, *Acc. Chem. Res.*, 1980, **13**, 385, and references cited therein.