# Preparation and X-Ray Structure of 4-BrC<sub>6</sub>H<sub>4</sub>CNSC(Cl)N

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Crystal Structure, 1,2,4-Thiadiazole, Intramolecular Cyclization

The title compound was obtained in 82% yield by the intramolecular cyclization of 4-BrC<sub>6</sub>H<sub>4</sub>C(NSCCl<sub>3</sub>)[N(SiMe<sub>3</sub>)<sub>2</sub>] in CH<sub>2</sub>Cl<sub>2</sub> at 23°C. It crystallizes in the triclinic system, space group P1, a = 7.957(3) Å, b = 10.864(5) Å, c = 5.625(1) Å,  $\alpha = 95.94(3)^{\circ}$ ,  $\beta = 97.79(2)^{\circ}$ ,  $\gamma = 100.72(3)^{\circ}$ , V = 469.2(3) Å<sup>3</sup>, and Z = 2. The bond lengths of the planar C<sub>2</sub>N<sub>2</sub>S ring indicate partial  $\pi$ -delocalization.

### Introduction

A variety of synthetic approaches to the 1,2,4thiadiazole ring system is available [1]. For example, the cyclocondensation of carbamidines  $RC(NH_2)(NH_2)^+Cl^-$  with Cl<sub>3</sub>CSCl produces RCNSC(Cl)N [1]. In this note we report the synthesis and X-ray structure of 3-*p* -bromophenyl-5-chloro-1,2,4-thiadiazole (2), which was obtained by the spontaneous intramolecular cyclization of 4-BrC<sub>6</sub>H<sub>4</sub>C(NSCCl<sub>3</sub>)[N(SiMe<sub>3</sub>)<sub>2</sub>] (1), according to eq. (1).

### **Results and Discussion**

The monothiolated benzamidine **1** was prepared from 4-BrC<sub>6</sub>H<sub>4</sub>CN<sub>2</sub>(SiMe<sub>3</sub>)<sub>3</sub> [2,3] and CCl<sub>3</sub>SCl (1:1 molar ratio) *cf.* synthesis of PhC(SCCl<sub>3</sub>)[N(SiMe<sub>3</sub>)<sub>2</sub>] [4]. The <sup>1</sup>H NMR spectrum of **1** showed resonances for 4-BrC<sub>6</sub>H<sub>4</sub> (an AA'XX' pattern centred at  $\delta$  8.80 and 7.56) and SiMe<sub>3</sub> groups ( $\delta$  0.30) in the intensity ratio 4:18. In addition, a second weak AA'XX' pattern ( $\delta$ 8.15 and 7.65) attributed to **2** was evident. A solution of **1** in CH<sub>2</sub>Cl<sub>2</sub> was kept for 72 h at 23 °C and, after subsequent work-up, the heterocycle **2** was obtained in 82% yield. A small amount of 4-BrC<sub>6</sub>H<sub>4</sub>C(NH)(NH<sub>2</sub>), the hydrolysis product of **1**, was also isolated.

The structure of 2 was determined by X-ray crystallography (see Fig. 1). The pertinent bond lengths, bond angles and torsion angles are summarized in Table I. In common with other 1,2,4-

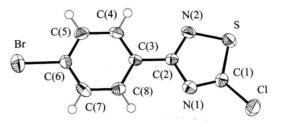


Fig. 1. ORTEP drawing and atomic numbering scheme for **2**.

thiadiazoles [5-10], the heterocyclic ring in **2** is essentially planar and the bond lengths indicate some  $\pi$ -delocalization. Thus the S-N distance of 1.650(6) Å is significantly shorter than the predicted single bond value of 1.73 Å [11] and the sequence of C-N bond lengths is 1.331(7), 1.380(8) and 1.302(7) Å (*cf.* single and double bond values of *ca.* 1.29 and 1.47 Å, respectively [12]). The 4-BrC<sub>6</sub>H<sub>4</sub> substituent is coplanar with the heterocyclic ring and there are no significant intermolecular interactions.

The synthetic route represented by eq. (1) represents an alternative to the use of carbamidines [2] for the synthesis of 1,2,4-thiadiazoles containing a 5-Cl substituent that can be subsequently functionalized.

## **Experimental Section**

# Preparation of 4-BrC<sub>6</sub>H<sub>4</sub>CNSC(Cl)N (2)

A solution of Cl<sub>3</sub>CSCl (0.60 g, 3.22 mmol) in 25 ml of CH<sub>2</sub>Cl<sub>2</sub> was added dropwise to 4-BrC<sub>6</sub>H<sub>4</sub>CN<sub>2</sub>(SiMe<sub>3</sub>)<sub>3</sub>, (1.37 g, 3.30 mmol) in 25 ml of CH<sub>2</sub>Cl<sub>2</sub> at 23°C under an atmosphere of N<sub>2</sub>. The reaction mixture was stirred for 72h and then solvent and Me<sub>3</sub>SiCl were removed by vacuum transfer to give a viscous yellow oil containing a small amount of white powder. The oil was dissolved

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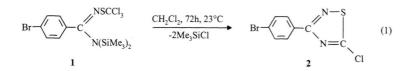


Table I. Selected bond lengths (Å) and bond angles (deg) for **2**.

Bond lengths		Bond angles		
S-N(2)	1.650(6)	N(2)-S-C(1	) 91.8(3)	
S-C(1)	1.709(7)	C(1)-N(1)-C	C(2) = 107.9(5)	
N(1)-C(1)	1.302(7)	S-N(2)-C(2	) 108.4(5)	
C(2)-N(1)	1.380(8)	Cl-C(1)-S	122.8(4)	
N(2)-C(2)	1.331(7)	Cl-C(1)-N(	1) 123.5(5)	
Cl-C(1)	1.706(7)	S-C(1)-N(1	) 113.6(5)	
C(2)-C(3)	1.467(8)	N(1)-C(2)-N	N(2) 118.3(6)	
	Tors	ion angles		
	~			
	Cl-C(1)-S-N	· · /		
	Cl-C(1)-N(1)	/ / /		
	S-N(2)-C(2)	)-N(1) 0.	0(7)	
	S-N(2)-C(2)	)-C(3) 179.	8(4)	
	S-C(1)-N(1)	)-C(2) -0.	6(6)	
	C(1)-S-N(2)	)-C(2) $-0.$	3(5)	

in CH<sub>2</sub>Cl<sub>2</sub>:diethyl ether and the insoluble white solid was removed by filtration. Solvent was removed from the filtrate. Slow evaporation of a diethyl ether solution of the residue gave pale yellow rectangular crystals of 4-BrC<sub>6</sub>H<sub>4</sub>CNSC(Cl)N (0.73 g, 82%). M.p. 85°C.

### Analysis for $C_8H_4BrClN_2S$

Calcd C 34.87 H 1.46 N 10.17%, Found C 35.39 H 1.50 N 9.33%.

<sup>1</sup>H NMR (in CDCl<sub>3</sub>):  $\delta$  8.15 and 7.65 (4-BrC<sub>6</sub>H<sub>4</sub>, AA'XX' pattern). <sup>13</sup>C NMR (in CDCl<sub>3</sub>):  $\delta$  173.3 and 171.2 (NC(S)Cl and CN<sub>2</sub>), 132.1, 130.8, 129.6 and 125.6 (C<sub>6</sub>H<sub>4</sub>). The white solid was identified as 4-BrC<sub>6</sub>H<sub>4</sub>C(NH)(NH<sub>2</sub>) (0.10 g, 16%) by EI-MS (*m/z* = 198 and 200, M<sup>+</sup>).

### X-ray analysis

The crystal structure of 2 was determined by using a Rigaku AFC6S diffractometer. Experimental details are summarized in Table II<sup>\*</sup>. The structure was solved by the heavy atom method [13] and expanded using Fourier techniques [14]. The non-hydrogen atoms were refined

Tabl	e II. Cryst	tal data for 4-1	$BrC_6H_4CNS$	SC(CI)N 2.

Formula	$C_8H_4N_2SBrCl$	
Fw.	275.55	
Crystal System	triclinic	
Space Group	P1 (No. 2)	
a [Å]	7.957(3)	
b [Å]	10.864(5)	
<i>c</i> [Å]	5.625(1)	
$\alpha$ [deg]	95.94(3)	
$\beta$ [deg]	97.79(2)	
$\gamma [deg]$	100.72(3)	
$V[Å^3]$	469.2(3)	
Z	2	
$d_{calc} [g cm^{-3}]$	1.950	
Crystallogr. Dim. [mm]	$0.50 \times 0.40 \times 0.20$	
Cell detn, refls	25	
Cell detn, $2\theta$ range, [deg]	18.50 - 29.97	
Radiation, MoK <sub><math>\alpha</math></sub> [Å]	$\lambda = 0.71069$	
Temperature [K]	170(1)	
Scan type	$\omega$ - $2\dot{ heta}$	
Scan rate [deg min <sup><math>-1</math></sup> ]	16.0	
Scan width [deg]	$1.57 + 0.34 \tan \theta$	
$2\theta \max [deg]$	50.1	
$\mu [\mathrm{cm}^{-1}]$	48.48	
Absorption correction	empirical	
Absorption, range	0.32 - 1.00	
Total reflections	1797	
Unique reflections	1666	
Data with $I > 3\sigma(I)$	1149	
Parameters refined	118	
R	0.042	
$R_w$	0.041	
GOF	2.54	
Largest $\Delta/\sigma$	0.00	
Final diff. map [e Å <sup><math>-3</math></sup> ]	0.53, -1.22	
F(000)	268	

anisotropically. Hydrogen atoms were included at geometrically idealized positions. Scattering factors were taken from Cromer and Waber [15] and allowance was made for anomalous dispersion [16]. All calculations were performed using teXsan [17].

#### Acknowledgments

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<sup>\*</sup>Further crystal structure data may be obtained from the Fachinformationszentrum Karlsruhe, D-76344 Eggenstein-Leopoldshafen, by quoting the Registry No. CSD-406259.

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