A Convenient Synthesis of Alkylated 4-Hydroxy-2-pyrones

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Summary Various alkylated 4-hydroxy-2-pyrones were synthesized from the $\beta\delta$ -diketo-esters (11), (13), and (14), which were easily prepared from the $\gamma\delta$ -acetylenic β keto-esters (3), (4), and (7) via the corresponding enamino- β -keto-esters (8)—(10).

The present study was undertaken to establish a convenient method for the preparation of 3,6- and 5,6-dialkyl- and 3,5,6-trialkyl-4-hydroxy-2-pyrones corresponding to the pyrone units in luteoreticulin, 1 citreoviridin, 2 and aureothin. 3 Syntheses of 3,6-4 and 5,6-dialkyl-4-hydroxy-2-pyrones⁵ have been carried out extensively in recent years, but no 3,5,6-trialkyl derivatives have been synthesized.

The Grignard reagents (1) and (2) reacted with a solution of ethyl methylmalonyl chloride in tetrahydrofuran at -20° to give the $\gamma\delta$ -acetylenic β -keto-esters (3) (65%) and (4) (85%), respectively, as viscous oils. The imidazolide of tetrolic acid reacted with the malonate magnesium enolates (5) and (6) in tetrahydrofuran to give the esters (3) (50%) and (7) (50%). Addition of diethylamine to (3), (4), and (7) in ether afforded the corresponding enamino- β -ketoesters, (8) (74%), (9) (48%), and (10) (41%). Hydrolysis of the enamine (9) with N-hydrochloric acid at 50° gave ethyl 2-methyl-3,5-dioxoheptanoate (11) (81%). Ring closure with polyphosphoric acid at 100° for 2 h then furnished 6-ethyl-4-hydroxy-3-methyl-2-pyrone (12) (51%), m.p. 182-183°. When the enamines (8) and (10) were heated with 3 equiv. of methyl iodide at 100° for 12 h, predominant C-alkylation was observed on the carbon atom of the enamine function. Without purification, the alkylated enamines were hydrolysed with boiling water for 1 h to give ethyl 2,4-dimethyl-3,5-dioxohexanoate (13) (54%) and ethyl 4-methyl-3,5-dioxohexanoate (14) (74%). Treatment of (13) with polyphosphoric acid at 100° for 1 h afforded 3,5,6-trimethyl-4-hydroxy-2-pyrone (15) (54%), m.p. 153—

RC
$$\equiv$$
 CMg Br R^1 C \equiv C \cdot CO \cdot CH \cdot CO $_2$ Et
(1) R = Me (3) R^1 = R^2 = Me (2) R = Et (4) R^1 = Et, R^2 = Me R^3 R² (7) R^1 = Me, R^2 = H R^1 CO \cdot CH \cdot CO \cdot CH \cdot CO $_2$ Et (11) R^1 = Et, R^2 = Me, R^3 = H (13) R^1 = R^2 = R^3 = Me (14) R^1 = R^3 = Me, R^2 = H (5) R = Me (6) R = H R^2 R¹C = CH \cdot CO \cdot CH \cdot CO $_2$ Et (5) R = Me (6) R = H (10) R^1 = Et, R^2 = Me (12) R^1 = Et, R^2 = Me, R^3 = H (15) R^1 = R^2 = R^3 = Me (16) R^1 = R^3 = Me, R^2 = H

154°. 5,6-Dimethyl-4-hydroxy-2-pyrone (16) was prepared (50%) from (14) in the same way. The physical properties (m.p., i.r., and n.m.r.) of (16) were in good agreement with those cited in the literature.5a

All new compounds gave satisfactory spectral and analytical data.

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- Y. Koyama, Y. Fukakusa, N. Kyomura, and S. Yamagishi, Tetrahedron Letters, 1969, 355.
 N. Sakabe, T. Goto, and Y. Hirata, Tetrahedron Letters, 1964, 1825.
 Y. Hirata, H. Nakata, K. Yamada, K. Okuhara, and T. Naito, Tetrahedron, 1961, 14, 252.

- (a) K. Boltz and K. Heidenbluth, Chem. Ber., 1958, 91, 2849; (b) T. E. Acker, P. E. Brenneisen, and S. W. Tanenbaum, J. Amer. Chem. Soc., 1966, 88, 834.
- ⁵ (a) T. M. Harris, C. M. Harris, and M. P. Wachter, Tetrahedron, 1968, 24, 6897; (b) T. Kappe and H. Schmidt, Tetrahedron Letters, 1970, 5105.
 - ⁶ G. Bram, Tetrahedron Letters, 1967, 4069.