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SPONTANEOUS DEHYDROGENATION OF TETRAHYDRO-2,3-TRIMETHYLENEQUINAZOLINES AND -QUINAZOLONES

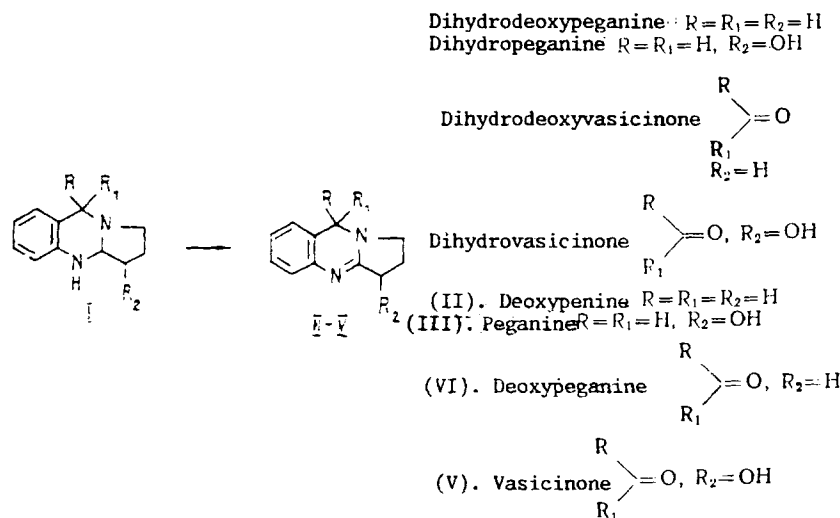
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In a study of the mass spectra of dihydro derivatives of quinazoline and quinazalone alkaloids (I) and their deuterio analogues [1, 2] we detected traces of dehydrogenation products in all the specimens (II), deoxypeganine; $R = R_1 + R_2 = H$; (III), peganine:

$R = R_1 = H, R_2 = OH$; (IV), deoxyvasicinone, $R_1 = O, R_2 = H$, (V), vasicinone: $R_1 = O, R_2 = OH$.

$R_2 = OH$. To find whether this process takes place spontaneously or under the action of electron impact, a sample of (I) was heated to the melting point. Heating was also carried out in amyl alcohol and in methanol in a sealed tube at 120°C. In all cases the formation of (II)-(V) was established with the aid of TLC and mass spectrometry. After samples of (I) in the form of bases had been stored for 2-5 years their mass spectra were recorded again.



The spectra showed the conversion of dihydropeganine (M^+ 190 into (III) by approximately 50%; of dihydrodeoxyvasicinone (M^+ 188) into (IV) completely; and of dihydrovasicinone (M^+ 204) into (IV) and (V) in equal proportions.

Among quinazoline derivatives similar transformations have been described previously only with the aid of oxidants (chromium trioxide [3] and potassium ferricyanide [4]) and only in the synthesis of the β -carbolinoquinazoline alkaloid evodiamine has the spontaneous dehydration of the tetrahydroquinazoline moiety of the molecule to the dihydroquinazalone analogue been observed [5].

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