

# Synthetic Approach to Medium-sized Cycloalkanones. A One-pot Three-carbon Ring Expansion of Carbocyclic $\beta$ -Keto Esters

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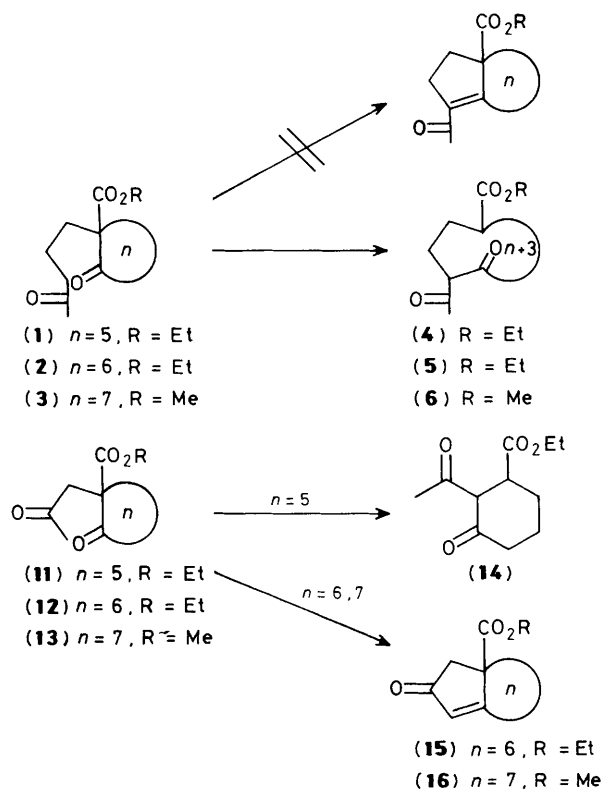
By treatment with Bu<sup>t</sup>OK in Me<sub>2</sub>SO, carbocyclic  $\beta$ -keto-esters (5-, 6-, and 7-membered rings) with a 4-oxopentyl function at the  $\alpha$ -position afforded three-carbon ring expansion products (8-, 9-, and 10-membered rings, respectively).

New syntheses of medium-sized rings<sup>1</sup> are attractive to organic chemists because, at present, there are only limited synthetic procedures available. Recent approaches have involved the fragmentation of a cross-piece bond in ring systems composed of two or more small rings, e.g. Grob fragmentation.<sup>2</sup>

Here we report a one-pot three-carbon ring expansion by treatment of carbocyclic  $\beta$ -keto esters, having a 4-oxopentyl moiety at the  $\alpha$ -position, with Bu<sup>t</sup>OK in Me<sub>2</sub>SO at room temperature. This carbon zip reaction<sup>3</sup> involves aldol and retro-aldol condensation.<sup>4</sup> Treatment of (3) (7-membered ring)<sup>†</sup> with Bu<sup>t</sup>OK (1.2 equiv.) in Me<sub>2</sub>SO at room temperature afforded a three-carbon ring expansion product (6)

(10-membered ring) in 78% yield.<sup>‡</sup> This one-pot reaction was applied to the 5-membered ring  $\beta$ -keto esters (1) and (2), and the ring expansion products (4) and (5) were obtained in 62 and 54% yields respectively (Scheme 1). The cyclized dehydration products were not obtained in contrast to the result obtained by Nakashita and Hesse<sup>5</sup> which concerned the reaction of  $\alpha$ -nitro-cyclopentanone and -cyclohexanone with methyl 3-oxopent-4-enoate.

Treatment of (3) with Bu<sup>t</sup>OK (1.2 equiv.) in tetrahydrofuran (THF) at  $-78^\circ\text{C}$  gave the normal aldol condensation product (7), with a small amount of (6). Compound (7) could be converted to (6) by treatment with Bu<sup>t</sup>OK. This result suggests that the three-carbon ring expansion process is thermodynamically preferable to the dehydration process. The stereochemistry of (7) was established by Baeyer–Villiger oxidation with CF<sub>3</sub>CO<sub>2</sub>H and subsequent methanolysis with K<sub>2</sub>CO<sub>3</sub>/MeOH to convert it to the five membered lactone (9). Facile formation of (9) indicates that the methyl ketone in (7) should be *cis* relative to the angular methyl ester. By treatment with 2,2-dimethoxypropane/*p*-MeC<sub>6</sub>H<sub>4</sub>SO<sub>3</sub>H, (9) was converted to the acetonide (10), indicating the methyl ketone in (7) to be *cis* relative to the angular OH. Thus, the above aldol condensation at  $-78^\circ\text{C}$  provides a simple method for the construction of *cis*-fused carbocyclic ring systems with the methyl ketone in the *cis*-position (Scheme 2). It is noteworthy that (7) was easily converted to the bicyclo-[4.4.1]undecanone derivative (17) under reductive conditions



**Scheme 1.** Reagents and conditions: Bu<sup>t</sup>OK/Me<sub>2</sub>SO, room temp. Yields: (4) (54%), (5) (62%), (6) (78%), (14) (41%), (15) (46%), (16) (41%).

<sup>†</sup> Compounds (1)–(3) were prepared *via* alkylation of the corresponding carbocyclic  $\beta$ -keto ester with 5-chloropentan-2-one ethylene acetal in the presence of Bu<sup>t</sup>OK in Me<sub>2</sub>SO and subsequent treatment with 10% HCl in MeOH. Compounds (11)–(13) were obtained by similar alkylation with allyl chloride followed by Wacker oxidation.

<sup>‡</sup> All yields refer to isolated and purified compounds. An FeCl<sub>3</sub> test on (4), (5), (6), and (14) was positive, indicating the presence of a 1,3-diketone.

**Selected spectroscopic data for (4):** i.r. (neat) 1740, 1710, 1440 cm<sup>-1</sup>; <sup>1</sup>H n.m.r. (CDCl<sub>3</sub>)  $\delta$  1.28 (3H, t, *J* 7.2 Hz), 2.24 (3H, s), 3.40 (1H, t, *J* 6.9 Hz), 4.18 (2H, q, *J* 7.2 Hz); <sup>13</sup>C n.m.r. (CDCl<sub>3</sub>)  $\delta$  14.2 (q), 20.7 (t), 26.1 (t), 27.4 (t), 29.0 (q), 29.4 (t), 38.0 (t), 48.9 (d), 59.8 (d), 61.5 (t), 169.6 (s), 202.9 (s), 213.5 (s); mass spec. *m/z* 240 (*M*<sup>+</sup>), 194, 166.

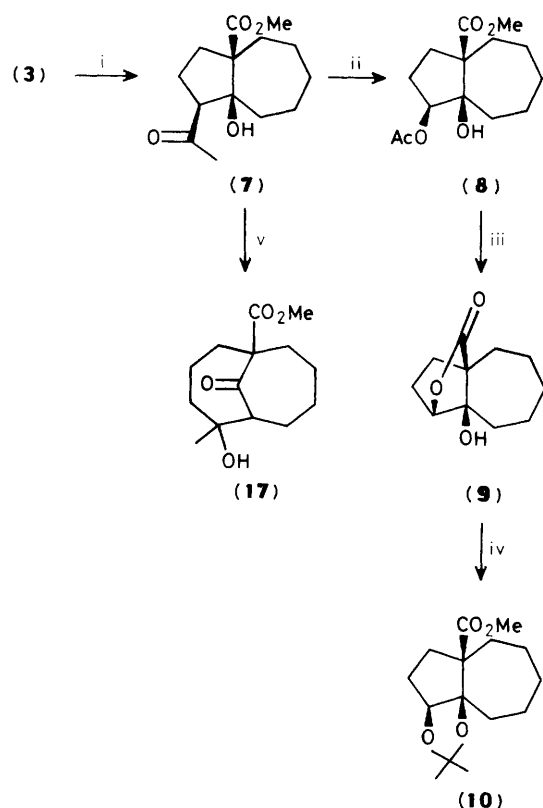
**For (5):** i.r. (neat) 1710 (br), 1640 cm<sup>-1</sup>; <sup>1</sup>H n.m.r. (CDCl<sub>3</sub>)  $\delta$  1.28 (3H, t, *J* 7.2 Hz), 2.24 (3H, s), 3.40 (1H, t, *J* 7.1 Hz), 4.18 (2H, q, *J* 7.2 Hz); <sup>13</sup>C n.m.r. (CDCl<sub>3</sub>)  $\delta$  14.1 (q), 25.0 (t), 25.3 (t), 25.9 (t), 27.2 (t), 28.8 (q), 34.1 (t), 42.1 (t), 50.5 (d), 59.9 (d), 61.3 (t), 169.7 (s), 203.2 (s), 212.7 (s); mass spec. *m/z* 254 (*M*<sup>+</sup>), 209.

**For (6):** i.r. (neat) 1740–1690 (br), 1635 cm<sup>-1</sup>; <sup>1</sup>H n.m.r. (CDCl<sub>3</sub>)  $\delta$  2.24 (3H, s), 3.42 (1H, dd, *J* 7.4, 14.6 Hz), 3.74 (3H, s); <sup>13</sup>C n.m.r. (CDCl<sub>3</sub>)  $\delta$  24.2 (t), 25.1 (t), 25.9 (t), 28.7 (q), 29.4 (t), 29.8 (t), 31.4 (t), 43.0 (t), 51.8 (q), 52.4 (d), 59.6 (d), 170.1 (s), 202.9 (s), 215.5 (s); mass spec. *m/z* 254 (*M*<sup>+</sup>), 223.

**For (7):** i.r. (neat) 3460, 1710, 1690 cm<sup>-1</sup>; <sup>1</sup>H n.m.r. (CDCl<sub>3</sub>)  $\delta$  2.22 (3H, s), 2.97 (1H, dd, *J* 7.6, 10.7 Hz), 3.71 (3H, s); mass spec. *m/z* 254 (*M*<sup>+</sup>), 236.

**For (14):** i.r. (neat) 1710 (br), 1600 cm<sup>-1</sup>; <sup>1</sup>H n.m.r. (CDCl<sub>3</sub>)  $\delta$  1.26 (3H, t, *J* 7.3 Hz), 3.46 (1/2 H, br), 4.14 (2H, q, *J* 7.3 Hz), 16.00 (1/2 H, s); mass spec. *m/z* 212 (*M*<sup>+</sup>), 170.

**For (17):** i.r. (neat) 3460, 1690 cm<sup>-1</sup>; <sup>1</sup>H n.m.r. (CDCl<sub>3</sub>)  $\delta$  1.22 (3H, s), 3.71 (3H, s), 3.77 (1H, s, OH); <sup>13</sup>C n.m.r. (CDCl<sub>3</sub>)  $\delta$  22.7 (q), 25.4 (t), 26.1 (t), 27.2 (t), 30.5 (t), 30.8 (t), 34.0 (t), 43.9 (t), 51.2 (d), 51.8 (q), 66.5 (s), 81.5 (s), 176.4 (s), 218.5 (s); mass spec. *m/z* 254 (*M*<sup>+</sup>), 236.



**Scheme 2.** Reagents and conditions: i, Bu<sup>t</sup>OK/THF, -78 °C, 64%; ii, CF<sub>3</sub>CO<sub>3</sub>H/CH<sub>2</sub>Cl<sub>2</sub>, 61%; iii, K<sub>2</sub>CO<sub>3</sub>/MeOH, 85%; iv, Me<sub>2</sub>C(OMe)<sub>2</sub>/*p*-MeC<sub>6</sub>H<sub>4</sub>SO<sub>3</sub>H/dimethylformamide, 81%; v, L-Selectride/THF, 80% or NaBH<sub>4</sub>/MeOH, -78 °C, 25%.

using L-Selectride or NaBH<sub>4</sub> at -78 °C. § Ring expansions in (1)–(3) may proceed *via* aldol condensation [such as (7)] followed by a retro-aldol reaction, which may be caused by the stereoelectronic repulsion between the ester, OH, and methyl ketone in a *cis*-configuration.

For further application of this zip reaction, carbocyclic β-keto esters (11)–(13), having a 2-oxopropyl moiety were subjected to the ring expansion conditions. In the case of compound (11), the three-carbon ring expansion product was not obtained, but a one-carbon expansion product (14) was obtained in 41% yield. Interestingly, (12) and (13) afforded only the cyclized dehydration products (15) and (16) without any formation of one- or three-carbon ring expansion products.

Received, 27th October 1987; Com. 1564

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§ The structure of (17) was suggested by a referee. Compound (17) was obtained as the major product, in addition to a small amount of reduction products. Reaction of (7) with lithium acetylide also afforded (17), and the addition product was not obtained. Treatment of (17) with Bu<sup>t</sup>OK afforded (6).