

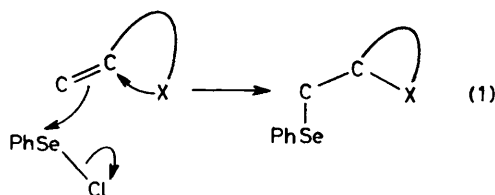
## A New Use of Cyclofunctionalisation With Selenenyl Reagents: An Example of Carbon–Carbon Bond Formation

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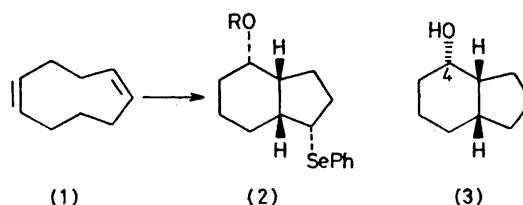
**Summary** (*Z,Z*)-Cyclonona-1,5-diene reacts stereoselectively with PhSeCl in AcOH to give (**2**; R = Ac).

CYCLOFUNCTIONALISATION<sup>1</sup> is an efficient route to heterocycles<sup>1,2</sup> containing the synthetically useful benzeneseleno-group<sup>3</sup> [equation (1)]. We have found a transannular example of this process that demonstrates how it can be used to make a C–C bond [equation (1), X = C].<sup>†</sup>



Under appropriate conditions the readily available<sup>4</sup> diene (**1**) reacts with PhSeCl to give the substituted hydrindan (**2**; R = Ac).<sup>5</sup> Thus, when an AcOH solution (0.04 M) of PhSeCl (1.02 equiv.) is added (0.3 equiv. min<sup>-1</sup>) to the diene in AcOH (0.04 M) containing anhydrous NaOAc (1 equiv.) the liquid hydrindanyl acetate (**2**; R = Ac) is produced.<sup>‡</sup> This compound, isolated 10 min after the end of the addition is treated with MeOH–H<sub>2</sub>O–K<sub>2</sub>CO<sub>3</sub> (5 ml, 0.85 ml, 84 mg, respectively, per mmol of acetate) at room temperature for 16 h and the crystalline alcohol (**2**; R = H) can be isolated as an analytically pure solid (m.p. 85–86 °C) in 68% yield [based on (**1**)]. Acetylation (Ac<sub>2</sub>O, pyridine, 3.5 h, room

temp.) of (**2**; R = H) regenerates (70%) the original reaction product (**2**; R = Ac). Treatment of the alcohol (**2**; R = H) with Ph<sub>3</sub>SnH<sup>6</sup> in refluxing PhMe gives the alcohol (**3**) (90%) shown, by comparison with an authentic sample<sup>7</sup> and by comparison of the derived sharp-melting acid phthalates,<sup>7</sup> to have the structure and stereochemistry defined by (**3**).<sup>§</sup>



On the basis of the firmly established<sup>8</sup> *trans* mode of addition of selenenyl reagents to double bonds the transannular reactions<sup>9</sup> of (**1**) are expected to proceed as in (**4**) or (**5**)<sup>¶</sup> to place the PhSe- and AcO- groups in a *cis*-1,4-relationship and *trans* to the bridgehead hydrogens. The validity of these predictions, and the preferential formation of (**6**) [= (**2**)] rather than (**7**), is established by the reduction to (**3**) together with the following observations.

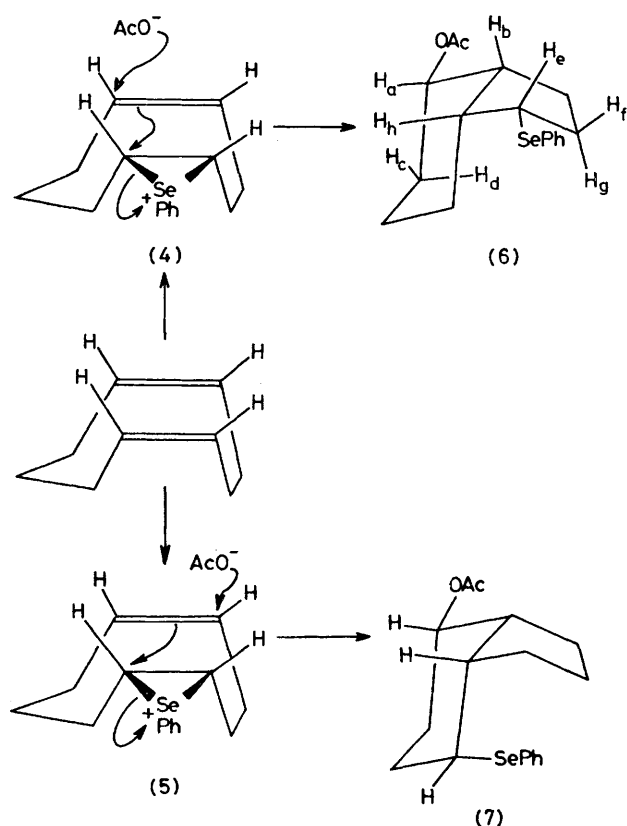
The n.m.r. signal (CDCl<sub>3</sub>; 100 MHz) centred at δ 5.0 due to H<sub>a</sub> [see (**6**)] is a 1:2:2:2:1 quintet corresponding to a pair of overlapping triplets (*J*<sub>ab</sub> = *J*<sub>ac</sub> = 5.2; *J*<sub>ad</sub> = 12.1 Hz). The signal centred at δ 3.58 due to H<sub>e</sub> is a sextet (*J*<sub>ef</sub> = *ca.* *J*<sub>eh</sub> = 9.9; *J*<sub>eg</sub> = 5.8 Hz<sup>10</sup>). Dreiding models

<sup>†</sup> For a previous attempt to make a C–C bond using PhSeCl, see ref. 1.

<sup>‡</sup> With EtOAc as solvent (and no NaOAc), a mixture of (**2**; R = Ac) and (**2**; R = H) is obtained.

<sup>§</sup> Compound (**3**), and not the C-4 epimer, is the main product obtained by oxymercuration–reduction of (**1**) (*cf.* G. Nagendrappa and D. Devaprabhakara, *Tetrahedron Letters*, 1970, 4687; S. N. Moorthy and D. Devaprabhakara, *ibid.*, 1975, 257).

<sup>¶</sup> For simplicity, other conformations of (**7**) are not shown. The pathway *via* (**5**) involves a boat transition state and has not been observed in other (ref. 9 and the ref. in footnote §) transannular reactions of (**1**).



reveal that the observed pair of splitting patterns due to  $\text{H}_a$  and  $\text{H}_e$  would not arise from (7). Therefore, as anticipated on mechanistic grounds,<sup>9</sup> the  $\text{PhSe-}$  group is attached to the  $\text{C}_5$ -ring in the orientation shown. This assignment is corroborated by mass spectral measurements. A diagnostically significant peak (21% of base peak) occurs at  $m/e$  97.0652 in the spectrum of (3) and corresponds to  $[\text{C}_6\text{H}_9\text{O}]^+$ . Such a species is expected in terms of one of the standard fragmentation pathways of cyclohexanols<sup>11</sup> [see (8);  $\text{R} = \text{H}$ ]. When (2) is reduced with  $\text{Ph}_3\text{SnD}$ <sup>6</sup> the alcohol produced should, and does, give a corresponding peak for  $[\text{C}_6\text{H}_8\text{DO}]^+$  (8;  $\text{R} = \text{D}$ ) (13.5% of base peak,  $m/e$  found 98.0714, calc. 98.0716).

Correct mass measurements and satisfactory ( $\pm 0.3$ ) combustion analytical data were obtained for all new compounds.

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- <sup>9</sup> J. K. Sutherland, *Tetrahedron*, 1974, **30**, 1651.
- <sup>10</sup> Cf. Footnote 13 of ref. 5b.
- <sup>11</sup> F. W. McLafferty, 'Interpretation of Mass Spectra,' 2nd edn., Benjamin, New York, 1973, p. 117.