

## KAURENE DERIVATIVES FROM *LASIANTHEA FRUTICOSA*, REVISION OF STEREOCHEMISTRY OF RELATED COMPOUNDS

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**Key Word Index**—*Lasianthea fruticosa*; Compositae; kaurene derivatives.

**Abstract**—Two new kaurene diterpenes, 12 $\alpha$ -angeloyloxy and 12 $\alpha$ -seneciyoxy-9,11-dehydro-*ent*-kaur-16-en-19-oic acids, have been isolated from *Lasianthea fruticosa*. The structures and stereochemistry of the compounds were determined on the basis of spectroscopic techniques and minor chemical transformation. The stereochemistry of a few similar compounds has been revised.

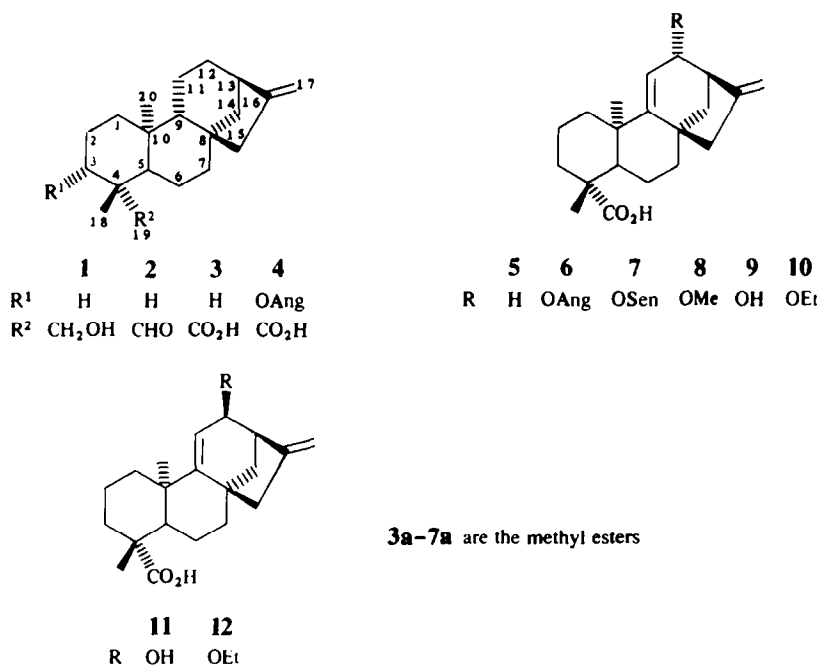
### INTRODUCTION

*Lasianthea fruticosa* R. Becker is a common shrub native in Central America [1]. A sesquiterpene named lasidiol angelate was reported earlier from this plant by Wiemer *et al.* [2]. More recent investigation on this plant by Khan *et al.* [3] afforded five kaur-16-ene diterpenes.

### RESULTS AND DISCUSSION

The extract of the aerial parts of *L. fruticosa* yielded in addition to germacrene D, 1 [4], 2 [4], 3 [5], 4 [6] and 5 [7], the new diterpenes 6 and 7. Surprisingly we could not detect the presence of the sesquiterpene lasidiol angelate.

The structure of compounds 6 and 7 followed from the <sup>1</sup>H NMR spectral data (Table 1) of the methyl esters 6a and 7a obtained upon addition of diazomethane. The spectral data of 6a differed from those of 7a only by the signals of the angelate and senecioate ester groups, respectively. Otherwise the spectra of both compounds were similar to that of 5a. In the case of 6a, a double doublet at  $\delta$ 5.08 ( $J$  = 2.5, 4.5 Hz) which coupled with the olefinic signal at  $\delta$ 5.36 (H-11,  $J$  = 1, 4.5 Hz) indicated that the ester group is at C-12. In compounds 11 and 12 [8, 9] a  $\beta$ -configuration for the oxygen function was assigned at C-12, based mainly on the downfield shift of H-17 [8]. On the other hand the epimeric 12-oxykaur-16-enes occurred



3a-7a are the methyl esters

Table 1.  $^1\text{H}$  NMR spectral data of compounds **6a**, **7a** and **8** (400 MHz,  $\text{CDCl}_3$ , TMS as int. standard)

H	<b>6a</b>	<b>7a</b>	<b>8</b>
1 $\alpha$	1.91 m	*	1.96 br d
1 $\beta$	1.23 br ddd	1.22 m	1.22 br ddd
2 $\alpha$	1.83 dddd	*	1.86 m
2 $\beta$	1.50 m	†	1.51 br d
3 $\alpha$	2.15 br d	2.15 br d	2.15 br d
3 $\beta$	1.01 ddd	1.00 ddd	1.02 ddd
5	1.61 dd	1.61 dd	1.62 dd
6 $\alpha$	2.50 m	2.49 m	2.50 dddd
6 $\beta$	1.87 m	*	1.87 m
7 $\alpha$	1.52 m	†	1.46 br dd
7 $\beta$	2.01 m	*	2.03 ddd
11	5.36 dd	5.32 dd	5.37 dd
12	5.08 dd	5.02 dd	3.44 dd
13	2.91 br s	2.89 br s	2.95 br s
14 $\alpha$	1.70 dd	1.68 dd	1.66 dd
14 $\beta$	1.45 dd	1.42 dd	1.37 dd
15 $\alpha$	2.15 br d	2.15 br d	2.12 br ddd
15 $\beta$	2.43 br d	2.43 br d	2.41 br d
17	4.96 br s	4.96 br s	4.91 br s
17'	5.16 br s	5.16 br s	5.01 br s
18	1.18 s	1.19 s	1.24 s
20	0.96 s	0.98 s	1.08 s
$\text{CO}_2\text{Me}$	3.68 s	3.66 s	3.41 s (OMe)
OR			
3'	6.01 qq	5.68 qq	
4'	1.95 dq	1.88 d	
5'	1.87 dq	2.17 d	

\*1.75–2.05 m.

†1.45–1.55 m.

$J$  [Hz]: 1 $\alpha$ ,1 $\beta$ =1 $\beta$ ,2 $\alpha$ =2 $\alpha$ ,2 $\beta$ =2 $\alpha$ ,3 $\beta$ =3 $\alpha$ ,3 $\beta$ =13; 1 $\alpha$ ,2 $\alpha$ =1 $\beta$ ,2 $\beta$ =2 $\alpha$ ,3 $\alpha$ =2 $\beta$ ,3 $\beta$ =4; 5,6 $\alpha$ =11.5; 5,6 $\beta$ =8.5; 6 $\alpha$ ,6 $\beta$ =7 $\alpha$ ,7 $\beta$ =14; 6 $\alpha$ ,7 $\beta$ =6 $\beta$ ,7 $\alpha$ =6 $\beta$ ,7 $\beta$ =9.5; 11,12=4.5; 11,13=1; 12,13=2.5; 14 $\alpha$ ,14 $\beta$ =11; 14 $\alpha$ ,15 $\beta$ =3; 15 $\alpha$ ,15 $\beta$ =16; 15 $\alpha$ ,17=15 $\alpha$ ,17'=2.5; OAng: 3',4'=7; 3',5'=4',5'=1.5; OSen: 2',4'=2',5'=1.5.

[9]. For this reason we carefully checked the stereochemistry of the compounds by a NOE experiment. As from the NOE effects of compound **6a** no clear decision was possible, it was hydrolysed by KOH in methanol solution to produce the 12-hydroxy derivative. After usual work-up we obtained a mixture of the corresponding hydroxy (**9**) and methoxy (**8**) compounds (1:1). The multiplicity and the couplings of H-12 signal in the  $^1\text{H}$  NMR spectrum of **8** indicated the same stereochemistry as in compounds **6a** and **7a**. Compound **8** was probably formed during acidic work-up of the reaction mixture by addition of methanol from the  $\alpha$ -side to the allylic cation. The NOE experiment was performed on methoxy compound **8**. Irradiation of H-12 gave a NOE with one of the exomethylene proton (3%) indicating that the ester group at C-12 is  $\alpha$ -oriented. Conversely, irradiation of the same exomethylene proton showed an effect with H-12 (4%) and H-13 (4%) and the remaining exomethylene proton (12%). As the splitting of H-12 in the spectra of our compounds and in those of **11** and **12** were identical, the stereochemistry at C-12 in the latter compounds should be revised to compounds **9** and **10**, respectively.

Table 2.  $^{13}\text{C}$  NMR spectral data of **6a** and **8** (100 MHz,  $\text{CDCl}_3$ ,  $\delta$ -values)

C	<b>6a</b>	<b>8</b>
1	38.23 t	38.17 t
2	18.44 t	18.35 t
3	29.12 t	29.03 t
4	43.08 s	43.43 s
5	44.85 t	43.81 d
6	20.08 t	20.07 t
7	40.52 t	40.60 t
8	44.84 s	44.66 s
9	161.81 s	160.28 s
10	38.93 s	38.94 s
11	114.13 d	115.42 d
12	74.06 d	81.79 d
13	46.09 d	46.17 d
14	46.17 t	40.60 t
15	46.94 t	47.17 t
16	151.36 s	153.00 s
17	109.59 t	108.12 t
18	28.1 q	28.22 q
19	177.76 s	182.98 s
20	23.1 q	23.41 q
$\text{CO}_2\text{Me}$	51.3 q	56.53 q (OMe)
OR		
1'	167.79 s	
2'	128.54 s	
3'	136.64 d	
4'	20.63 q	
5'	15.70 q	

## EXPERIMENTAL

The aerial parts of *L. fruticosa* (1 kg, collected in March 1989 in Rincon de la Vieja, Costa Rica, voucher deposited at the National Herbarium of Costa Rica) were extracted with  $\text{MeOH-Et}_2\text{O}$ -petrol (1:2:1) at room temp. for 24 hr. After removing the fatty materials by  $\text{MeOH}$  treatment, the extract was first separated by CC (silica gel) using solvent mixts comprising petrol,  $\text{Et}_2\text{O}$  and  $\text{MeOH}$  with increasing polarities. The compounds were further sep'd and purified by repeated TLC. The extract yielded 300 mg germacrene D, 500 mg **1**, 500 mg **2**, 6 g **3**, 200 mg **4**, 2 g **5**, 2 g **6** and 200 mg **7**. The structures of the known compounds were determined by comparing their spectral data with those of reference compounds.

12 $\alpha$ -Angeloyloxy-9,11-dehydro-ent-kaur-16-en-19-oic acid (**6**). (Isolated as its methyl ester **6a**); oil,  $R_f$  0.70 (E-PE, 1:4); IR  $\nu_{\text{max}}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1720 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 412 [ $\text{M}^+$ ] (8) (calc. for  $\text{C}_{26}\text{H}_{36}\text{O}_4$ : 412.262), 297 [ $\text{M}-\text{Me}^+$ ] (3), 353 [ $\text{M}-\text{CO}_2\text{Me}^+$ ] (4), 330 [ $\text{M}-\text{O}=\text{C}=\text{C}(\text{Me})-\text{CH}=\text{CH}_2^+$ ] (54), 329 [ $\text{M}-\text{C}_5\text{H}_7\text{O}^+$ ] (7), 313 [ $\text{M}-\text{C}_5\text{H}_7\text{O}_2^+$ ] (12), 312 [ $\text{M}-\text{C}_5\text{H}_8\text{O}_2^+$ ] (7), 297 [312-Me] $^+$  (10), 83 [ $\text{C}_5\text{H}_7\text{O}^+$ ] (100); [ $\alpha$ ] $_D^{25} + 123$  ( $\text{CHCl}_3$ ;  $c$  2.5).

12 $\alpha$ -Seneciolyoxy-9,11-dehydro-ent-kaur-16-en-19-oic acid (**7**). (Isolated as its methyl ester **7a**); oil,  $R_f$  0.63 (E-PE, 1:4); IR  $\nu_{\text{max}}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1720 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 412 [ $\text{M}^+$ ] (3) (calc. for  $\text{C}_{26}\text{H}_{36}\text{O}_4$ : 412.262), 353 [ $\text{M}-\text{CO}_2\text{Me}^+$ ] (1), 330 [ $\text{M}-\text{O}=\text{C}=\text{C}(\text{Me})_2^+$ ] (56), 329 [ $\text{M}-\text{C}_5\text{H}_7\text{O}^+$ ] (2), 313 [ $\text{M}-\text{C}_5\text{H}_7\text{O}_2^+$ ] (5), 312 [ $\text{M}-\text{C}_5\text{H}_8\text{O}_2^+$ ] (4), 297 [312-Me] $^+$  (4), 83 [ $\text{C}_5\text{H}_7\text{O}^+$ ] (100); [ $\alpha$ ] $_D^{25} + 76.6$  ( $\text{CHCl}_3$ ;  $c$  0.3).

Compound **6** was hydrolysed with KOH in  $\text{MeOH}$  soln at 60°

for 1 hr. Usual work-up afforded a mixt. of corresponding 12-hydroxy and methoxy compounds (1:1). On TLC separation we obtained compounds **8** and **9**. Our spectral data of compound **9** were identical to those of the 12-hydroxy compound reported from *Montanoa pteropoda* [8].

12 $\alpha$ -Methoxy-9,11-dehydro-ent-kaur-16-en-19-oic acid (**8**). Oil; IR  $\nu_{\text{max}}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1695; MS  $m/z$  (rel. int.): 330 [M]<sup>+</sup> (100) (calc. for C<sub>21</sub>H<sub>30</sub>O<sub>3</sub>: 330.219), 315 [M-Me]<sup>+</sup> (28), 298 [M-HOMe]<sup>+</sup> (12), 283 [298-Me]<sup>+</sup> (12).

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## SESQUITERPENE LACTONES FROM *ARTEMISIA RUTIFOLIA*

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**Key Word Index**—*Artemisia rutifolia*; Compositae; sesquiterpene lactones; guaianolides; germacranolide; eudesmane derivative.

**Abstract**—The polar fractions of *Artemisia rutifolia* afforded three new guaianolides, a germacranolide and a trihydroxycostic acid methyl ester. Structures were elucidated by high field NMR techniques.

#### INTRODUCTION

In our previous study of *Artemisia rutifolia* Steph. ex Spreng [1], we obtained from this plant a very polar mixture of lactones, the structures of which were not elucidated. We have re-investigated this fraction from plant material collected in Bulgan-Aimack in Mongolia.

#### RESULTS AND DISCUSSION

In addition to compounds isolated previously [1], we isolated the guaianolides **1**–**3**, the germacranolide **4** and the eudesmane derivative **5** from the polar column chromatography fractions.

The <sup>1</sup>H NMR spectrum of **1** (Table 1) indicated the presence of a guaianolide with a 2,3-double bond by a pair of doublets at  $\delta$ 5.61 and 5.94. As these protons showed no further vicinal couplings, C-1 and C-4 are

quarternary. All data agreed with the proposed structure. The stereochemistry was determined by NOED. Clear effects were observed between H-6 and H-15 (5%) but not between H-6 and OH-1, between H-7 and H-5 (6%) as well as between H-15, H-6 (10%) and H-3 (6%).

The <sup>1</sup>H NMR spectrum of **2** (Table 1) was similar to that of **1**. However, the exomethylene proton signals were replaced by a singlet at  $\delta$ 1.45 indicating a 10-hydroxyguaianolide. The configuration at C-10 was proposed by comparison with the spectra of similar guaianolides and from biogenetic considerations as several 10 $\alpha$ -hydroxyguaianolides were isolated from this species [1].

The <sup>1</sup>H NMR spectra of **3** and its acetate **3a** (Table 1) showed that a guaianolide was again present. Spin decoupling indicated a 12,6 $\alpha$ -olide with oxygen functions at C-2, C-5 and C-10. Comparison of the chemical shifts of **3** and **3a** with those of related lactones led to the assignment of the configurations at the chiral centres. The