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## Studies on the Constituents of Medicinal and Related Plants in Sri Lanka. I. New Triterpenes from *Hedyotis lawsoniae*

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Three new triterpene acids were isolated from the twigs and leaves of *Hedyotis lawsoniae*, together with sixteen known ones. The new compounds were determined to be  $3\beta,23$ -dihydroxyurs-12-en-28-oic acid,  $3\beta,24$ -dihydroxyurs-12-en-28-oic acid, and  $2\alpha,3\beta,24$ -trihydroxyurs-12-en-28-oic acid on the basis of spectroscopic evidence.

**Keywords**—*Hedyotis lawsoniae*; Rubiaceae; triterpene;  $3\beta,23$ -dihydroxyurs-12-en-28-oic acid;  $3\beta,24$ -dihydroxyurs-12-en-28-oic acid;  $2\alpha,3\beta,24$ -trihydroxyurs-12-en-28-oic acid

Some of the *Hedyotis* plants (Rubiaceae) are used as folk medicines in India and Sri Lanka. For instance, *H. auricularia* L. was used for the treatment of diarrhea and dysentery in India, while it is taken as a vegetable effective to reduce high blood pressure in Sri Lanka.<sup>1)</sup> In the course of our studies on the biologically active constituents of medicinal plants and related plants in Sri Lanka, we obtained three new triterpenes from *Hedyotis lawsoniae* (DC.) WIGHT et ARN., along with sixteen known ones. This paper describes the structure assignments of these triterpenes.

The methanolic extract of dried twigs and leaves of *Hedyotis lawsoniae* was separated into the ethyl acetate-soluble part and the water-soluble one. The former was roughly separated by silica gel column chromatography, and the fractions were further separated by preparative layer chromatography (PLC), or high-performance liquid chromatography (HPLC), giving new triterpenes (**1a**—**3a**) along with known ones (**4a**—**19a**) (see Experimental). Among these, ursolic acid (**4a**),<sup>2)</sup> 3-epiursolic acid (**5a**),<sup>2)</sup> ursonic acid (**6a**), asiatic acid (**8a**),<sup>3)</sup> euscaphic acid (**12a**),<sup>4)</sup> oleanolic acid (**13a**), oleanonic acid (**14a**), sumaresinolic acid (**15a**),<sup>5)</sup> hederagenin (**16a**), arjunolic acid (**17a**),<sup>3)</sup> betulinic acid (**18a**), and betulin (**19a**) were identified by direct comparisons of the proton nuclear magnetic resonance (<sup>1</sup>H-NMR) spectra, gas chromatographic (GC) retention times, and gas chromatography-mass spectrometry (GC-MS) data with those of corresponding authentic samples. On the other hand, identification of  $2\alpha$ -hydroxyursolic acid (**7a**),<sup>6)</sup>  $2\alpha$ -hydroxy-3-epiursolic acid (**9a**),<sup>7)</sup>  $2\alpha,3\alpha,23$ -trihydroxyurs-12-en-28-oic acid (**10a**),<sup>8)</sup> and benthamic acid (**11a**)<sup>9)</sup> was performed by comparisons of the MS and <sup>1</sup>H-NMR spectra with those described in the literature.

Compounds **1a** and **2a** were isolated as the methyl ester diacetates: **1b**, mp 180—181.5 °C, C<sub>35</sub>H<sub>54</sub>O<sub>6</sub>, [ $\alpha$ ]<sub>D</sub> + 64.1 ° (CHCl<sub>3</sub>), and **2b**, mp 180—182 °C, C<sub>35</sub>H<sub>54</sub>O<sub>6</sub>, [ $\alpha$ ]<sub>D</sub> + 50.8 ° (CHCl<sub>3</sub>), after methylation and acetylation. The purification of **1b** was difficult, but it was achieved by preparative HPLC.

In the MS, both compounds showed the base peak at  $m/z$  262, ascribable to the

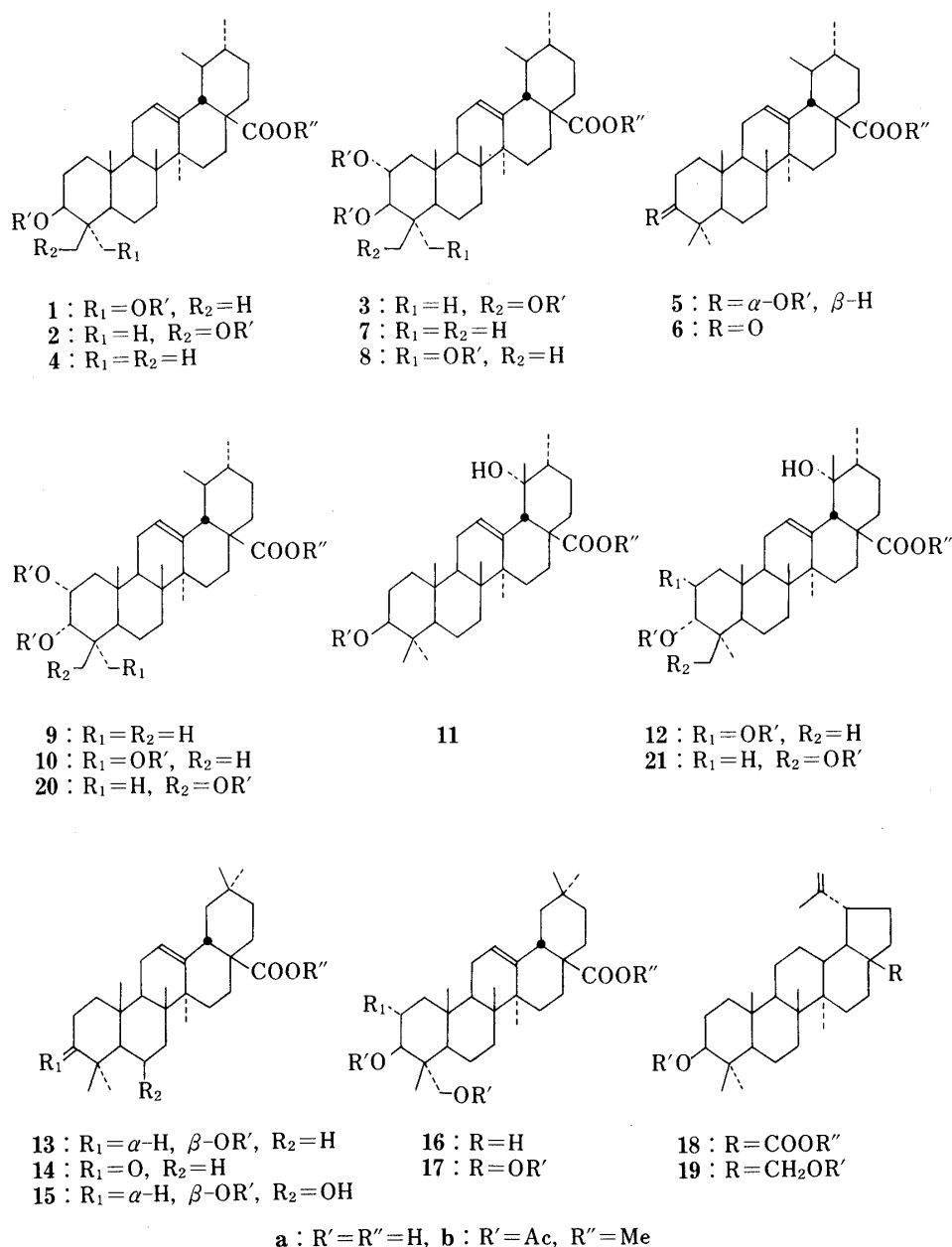


Chart 1

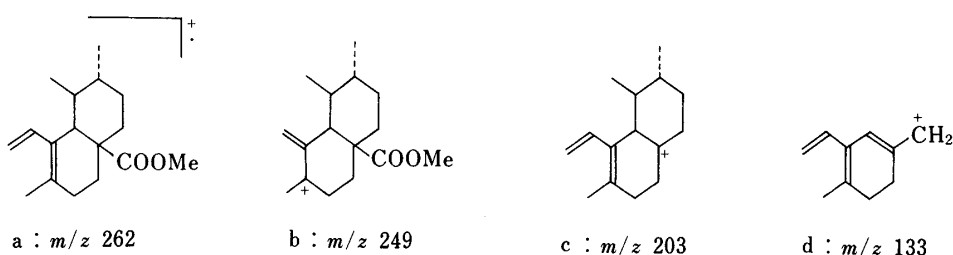


Chart 2

retro-Diels-Alder fragment ion (a), along with peaks at  $m/z$  249 (b), 203 (c), 189 (c-14), and 133 (d), which are characteristic of  $\Delta^{12}$ -oleanene or  $\Delta^{12}$ -ursene type triterpenes (Chart 2).<sup>10)</sup> The  $^1H$ -NMR spectra of **1b** and **2b** showed significant signals assignable to the 18-proton at  $\delta$  2.25 (d,  $J=11$  Hz) and the 12-olefinic proton at  $\delta$  5.25 (br t,  $J=3.5$  Hz), together with two

*sec*-methyl and four *tert*-methyl signals, whose signal pattern is suggestive of the structural features of urs-12-en-28-oic acid.<sup>11)</sup> In view of these data, **1b** and **2b** were presumed to be methyl urs-12-en-28-oate derivatives having two acetoxyl groups. These two acetoxyls may be present at ring A rather than ring B, since the acetylation of **1a** and **2a** leading to **1b** and **2b**, respectively, proceeded smoothly under very mild conditions. The positions and configuration of these acetoxyls in each compound were determined by <sup>1</sup>H-NMR analyses as described below.

The <sup>1</sup>H-NMR spectrum of **1b** showed signals due to a methine proton at  $\delta$  4.81 (dd,  $J=10, 5.5$  Hz, CH<sub>2</sub>OAc) and methylene protons at  $\delta$  3.71 and 3.88 (each 1H, d,  $J=12$  Hz, CH<sub>2</sub>OAc), along with two acetyl methyl signals ( $\delta$  2.02 and 2.07). This spectral pattern closely resembles that of authentic **16b**. Thus, the compound was determined to be methyl 3 $\beta$ ,23-diacetoxyurs-12-en-28-oate (**1b**).

On the other hand, compound **2b** showed a <sup>1</sup>H-NMR spectrum very similar to that of **1b**, except for signals arising from a CH<sub>2</sub>OAc group and a *tert*-methyl group. The signals due to the acetoxyl-bearing methylene ( $\delta$  4.13 and 4.36, each 1H, d,  $J=12$  Hz) were shifted downfield relative to the 23-methylene signals of **1b**, and these chemical shifts are virtually the same as those of the 24-methylene protons of methyl 2 $\alpha$ ,3 $\alpha$ ,24-triacetoxyurs-12-en-28-oate (**20b**)<sup>12)</sup> and methyl barbinervate diacetate (**21b**).<sup>13)</sup> In addition, the methyl group at the C-4 position resonated downfield (*ca.* 0.08 ppm) relative to the 23-methyl signal of **4b**. This behavior was also parallel with those of **20b** and **21b**, whose 23-methyl signals appeared downfield relative to those of **9b** and **5b**, as shown in Table I. On the other hand, in the case of 23-acetoxytriterpenes, the signal of the 24-methyl group was reported to appear at about the same region as the corresponding signal of non-substituted compounds.<sup>11a)</sup> Therefore, these data indicated that the primary acetoxyl group of **2b** is located at the 24-position. Since the signal due to the 3-proton was observed at  $\delta$  4.60 (dd,  $J=9, 7$  Hz), this compound was established to be methyl 3 $\beta$ ,24-diacetoxyurs-12-en-28-oate (**2b**).

Compound **3a** was also obtained as the methyl ester triacetate (**3b**), amorphous powder, C<sub>37</sub>H<sub>56</sub>O<sub>8</sub>,  $[\alpha]_D +51^\circ$  (CHCl<sub>3</sub>). The gross structure of this compound was suggested by the MS base peak at  $m/z$  262 (a) and the <sup>1</sup>H-NMR pattern, which is similar to that of **2b**. Furthermore, the <sup>1</sup>H-NMR spectrum showed signals due to the methine protons geminal to acetoxyl groups at  $\delta$  4.86 (d,  $J=10.5$  Hz) and 5.18 (td,  $J=10.5$  and 4.5 Hz), indicating the presence of a 2 $\alpha$ ,3 $\beta$ -glycol diacetate system. The signals arising from the methylene protons at  $\delta$  4.21 (s) and a methyl group at  $\delta$  1.02 (Table I) suggested that the third acetoxyl group should be located at the 24-position. In addition, the 3 $\alpha$ -proton of **2b** and **3b** resonated downfield by 0.09 and 0.08 ppm, respectively, relative to the corresponding proton of **4b** and **7b**,

TABLE I. Chemical Shifts of Methyl Signals in Several 24-Acetoxytriterpenes ( $\delta$  in CDCl<sub>3</sub>)

Compound (Position of AcO group)	Methyl signals			
	23	24	25	26
<b>4b</b> (3 $\beta$ )	0.87	0.88	0.94	0.74
<b>2b</b> (3 $\beta$ , 24)	0.95 <sup>a)</sup>	—	1.01 <sup>a)</sup>	0.74
<b>5b</b> (3 $\alpha$ )	0.86	0.90	0.95	0.76
<b>21b</b> (3 $\alpha$ , 24)	0.95 <sup>a)</sup>	—	0.93 <sup>a)</sup>	0.68
<b>7b</b> (2 $\alpha$ , 3 $\beta$ )	0.90	0.90	1.05	0.74
<b>3b</b> (2 $\alpha$ , 3 $\beta$ , 24)	1.02	—	1.06	0.73
<b>9b</b> (2 $\alpha$ , 3 $\alpha$ )	0.87	0.98	1.04	0.75
<b>20b</b> (2 $\alpha$ , 3 $\alpha$ , 24)	0.98	—	1.05	0.75

a) Assignments may be interchanged as indicated in each compound.

TABLE II. Chemical Shifts of Methine and Methylene Protons of Carbons Carrying Acetoxyl Groups ( $\delta$  in  $\text{CDCl}_3$ )

Compound (Position of AcO group)	Chemical shifts		
	$2\beta\text{-H}$	$3\alpha\text{-H}$	$23\text{-H}_2$ or $24\text{-H}_2$
<b>4b</b> ( $3\beta$ )	—	4.51 (dd, $J=9$ , 7 Hz)	—
<b>1b</b> ( $3\beta$ , 23)	—	4.81 (dd, $J=10$ , 5.5 Hz)	3.71, 3.88 (each d, $J=12$ Hz)
<b>2b</b> ( $3\beta$ , 24)	—	4.60 (dd, $J=9$ , 7 Hz)	4.13, 4.36 (each d, $J=12$ Hz)
<b>7b</b> ( $2\alpha$ , $3\beta$ )	5.14 (td, $J=10$ , 4.5 Hz)	4.78 (d, $J=10$ Hz)	—
<b>8b</b> ( $2\alpha$ , $3\beta$ , 23)	5.2 (td, $J=10$ , 4.5 Hz)	5.12 (d, $J=10$ Hz)	3.62, 3.89 (each d, $J=12$ Hz)
<b>3b</b> ( $2\alpha$ , $3\beta$ , 24)	5.18 (td, $J=10.5$ , 4.5 Hz)	4.86 (d, $J=10.5$ Hz)	4.20 (s)

whereas that of **1b** and **8b** appeared further downfield by 0.30 and 0.34 ppm, respectively, in comparison with that of **4b** and **7b**, as shown in Table II. These results supported the *trans* relationship between the acetoxymethyl group and the  $3\alpha$ -proton in **3b**. From the above findings, compound **3b** was determined to be  $2\alpha,3\beta,24$ -triacetoxyurs-12-en-28-oate (**3b**).

### Experimental

Melting points were determined with a Kofler-type apparatus and are uncorrected. Optical rotations were measured with a JASCO DIP-4 automatic polarimeter.  $^1\text{H-NMR}$  spectra were recorded on a Varian Associates XL-200 spectrometer and tetramethylsilane (TMS) in  $\text{CDCl}_3$  was used as an internal standard. MS and high resolution MS were obtained with a JEOL JMS-D 300 spectrometer. Column chromatography was carried out using Mallinckrodt silica gel and the columns were eluted with  $\text{MeOH-CHCl}_3$  mixtures of increasing MeOH contents, unless otherwise stated. The eluted solutions were concentrated *in vacuo*. For thin layer chromatography (TLC), Kieselgel 60  $\text{F}_{254}$  (Merck) was used and spots were detected by spraying a  $\text{Ce}(\text{SO}_4)_2\text{-aq. H}_2\text{SO}_4$  reagent. For PLC, Kieselgel  $\text{PF}_{254}$  or  $\text{F}_{254}$  (Merck) was employed and the plates were examined after exposure to  $\text{I}_2$  vapor or under ultraviolet (UV) light. Extraction of substances from the silica gel was done with  $\text{MeOH-CH}_2\text{Cl}_2$  mixture. The organic solutions were dried over anhydrous  $\text{MgSO}_4$ . Gas chromatography (GC) was done on a GC-6AM instrument (Shimadzu) with a 3% SE-300 or a 2% OV-17 column ( $2\text{m} \times 3\text{mm}$  i.d. glass tube; injection temp.  $300^\circ\text{C}$ , column temp.  $280^\circ\text{C}$ , carrier gas  $\text{N}_2$ ).

**Isolation of Triterpenes from *H. lawsoniae***—Dried twigs and leaves (1.69 kg) of *H. lawsoniae*, collected at Horton Plains in Sri Lanka, were extracted with boiling MeOH ( $5\text{ l} \times 3$ ) and the extracts were combined and concentrated *in vacuo*. The residue was diluted with water (1 l) and extracted with AcOEt ( $500\text{ ml} \times 3$ ). The combined AcOEt extract was washed with water, dried, and concentrated *in vacuo* to give a syrupy residue (20 g). This was suspended in  $\text{MeOH-CH}_2\text{Cl}_2$  (1:9) (200 ml) and the insoluble material (F-I) (4.58 g) was separated by filtration. The filtrate was concentrated *in vacuo* and the residue (15 g) was subjected to silica gel (450 g) column chromatography, giving four fractions containing triterpenes: viz. the  $\text{MeOH-CHCl}_3$  (2:98) eluate (F-II, 2.57 g), the earlier part of the  $\text{MeOH-CHCl}_3$  (5:95) eluate (F-III, 1.81 g), the later part of the  $\text{MeOH-CHCl}_3$  (5:95) eluate (F-IV, 1.45 g), and the  $\text{MeOH-CHCl}_3$  (1:9) eluate (F-V, 1.12 g).

F-I was rechromatographed on a silica gel (200 g) column. The eluate with  $\text{MeOH-CHCl}_3$  (1:99) was concentrated to give ursolic acid (**4a**) (1.45 g) as a colorless powder. The subsequent eluate with  $\text{MeOH-CHCl}_3$  (5:95) (1.27 g) was methylated with  $\text{CH}_2\text{N}_2$  and subjected to PLC with acetone- $\text{CHCl}_3$  (5:95) as the eluent. The less polar band gave methyl ursolate (0.97 g), which was identified by GC, MS, and  $^1\text{H-NMR}$  comparisons with an authentic sample. The more polar band afforded the methyl ester of benthamic acid (**11a**) (60 mg), amorphous powder,  $[\alpha]_D^{25} +29.8^\circ$  ( $c=0.55$ ,  $\text{CHCl}_3$ ). MS  $m/z$ : 486 ( $\text{M}^+$ ), 468, 278, 260, 201, 179. High resolution MS: Found 486.3699, Calcd for  $\text{C}_{31}\text{H}_{50}\text{O}_4$  ( $\text{M}^+$ ) 486.3708. This ester was acetylated with  $\text{Ac}_2\text{O-pyridine}$  and recrystallized from MeOH to give the monoacetate (**11b**) (31 mg), colorless needles, mp  $197\text{--}199^\circ\text{C}$ .  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : 0.69 (3H, s,

26-H<sub>3</sub>), 0.87 (3H, s, 23-H<sub>3</sub>), 0.88 (3H, s, 24-H<sub>3</sub>), 0.94 (3H, s, 25-H<sub>3</sub>), 0.96 (3H, d,  $J=6$  Hz, 30-H<sub>3</sub>), 1.23 (3H, s, 27-H<sub>3</sub>), 1.25 (3H, s, 29-H<sub>3</sub>), 2.07 (3H, s, OAc), 2.62 (1H, s, 18-H), 3.63 (3H, s, COOCH<sub>3</sub>), 4.54 (1H, dd,  $J=9, 7$  Hz, 3-H), and 5.39 (1H, t,  $J=3.5$  Hz, 12-H). The above data are practically identical with those of methyl benthamate monoacetate (**11b**) reported in the literature.<sup>9)</sup>

F-II and F-III were each suspended in CH<sub>2</sub>Cl<sub>2</sub> and the insoluble substances (450 mg and 710 mg, respectively) were collected by filtration and methylated with CH<sub>2</sub>N<sub>2</sub> to give an additional crop of methyl ursolate. The filtrate from F-II was concentrated and the residue (2.1 g) was diluted with dil. NaOH (100 ml), then extracted with CH<sub>2</sub>Cl<sub>2</sub> (50 ml  $\times$  3). The CH<sub>2</sub>Cl<sub>2</sub> layer was washed with water, dried, and concentrated to give a residue (1.5 g). This residue was roughly separated by silica gel (80 g) column chromatography and the fractions were further purified by repeated PLC using acetone-CHCl<sub>3</sub> (5:95) as the eluent to give oleanolic acid (**13a**) (130 mg), 3-epiursolic acid (**5a**) (4.2 mg), a mixture of ursonic acid (**6a**) and oleanonic acid (**14a**) (3.9 mg), betulinic acid (**18a**) (13 mg), and betulin (**19a**) (2 mg). These compounds except for **5a** were methylated with CH<sub>2</sub>N<sub>2</sub> and identified by GC, MS, and <sup>1</sup>H-NMR comparisons with corresponding authentic samples. Compound **5a** (1.6 mg) was methylated with CH<sub>2</sub>N<sub>2</sub> and oxidized with CrO<sub>3</sub>-AcOH at room temperature for 2 h. The product was purified by PLC (developed with acetone-CHCl<sub>3</sub>, 5:95) to give **6b** (1 mg), which was identified by GC and <sup>1</sup>H-NMR comparison with an authentic sample.

F-IV was acetylated with Ac<sub>2</sub>O-pyridine at room temperature for 4 h and the product was subjected to silica gel (70 g) column chromatography with AcOEt-hexane as the eluent. The AcOEt-hexane (2:8) eluate (178 mg) was methylated with CH<sub>2</sub>N<sub>2</sub> and the product was subjected to PLC with CHCl<sub>3</sub> to afford the following compounds.

1) Methyl 2 $\alpha$ -hydroxyursolate diacetate (**7b**) (18 mg), amorphous powder,  $[\alpha]_D^{22} + 17^\circ$  ( $c=1.7$ , CHCl<sub>3</sub>). MS  $m/z$ : 570 (M<sup>+</sup>), 510, 450, 262, 249, 203, 189, 133. High resolution MS: Found 570.3912, Calcd for C<sub>35</sub>H<sub>54</sub>O<sub>6</sub> (M<sup>+</sup>) 570.3919. <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.74 (3H, s, 26-H<sub>3</sub>), 0.83–0.90 (6H, 29- and 30-H<sub>3</sub>), 0.90 (6H, s, 23- and 24-H<sub>3</sub>), 1.05 (3H, s, 25-H<sub>3</sub>), 1.07 (3H, s, 27-H<sub>3</sub>), 1.98 and 2.07 (each 3H, s, OAc), 2.25 (1H, d,  $J=11$  Hz, 18-H), 3.62 (3H, s, COOCH<sub>3</sub>), 4.78 (1H, d,  $J=10$  Hz, 3-H), 5.14 (1H, td,  $J=10, 4.5$  Hz, 2-H), and 5.23 (1H, m, 12-H). These spectral data are compatible with those of **7b** reported in the literature.<sup>6)</sup>

2) A mixture of methyl 3 $\beta$ ,23-diacetoxyurs-12-en-28-oate (**1b**) and methyl hederagenin diacetate (**16b**) (12 mg, approximately 7:3 on GC analysis). Purification of this mixture by HPLC is described below.

3) Methyl 3 $\beta$ ,24-diacetoxyurs-12-en-28-oate (**2b**) (2 mg), colorless needles (from MeOH), mp 180–182 °C,  $[\alpha]_D^{22} + 50.8^\circ$  ( $c=0.12$ , CHCl<sub>3</sub>). MS  $m/z$ : 570 (M<sup>+</sup>), 510, 450, 262, 249, 203, 189, 133. High resolution MS: Found 570.3882, Calcd for C<sub>35</sub>H<sub>54</sub>O<sub>6</sub> (M<sup>+</sup>) 570.3919; Found 262.1922, Calcd for C<sub>17</sub>H<sub>26</sub>O<sub>2</sub> (a) 262.1932; Found 249.1815, Calcd for C<sub>16</sub>H<sub>25</sub>O<sub>2</sub> (b) 249.1854; Found 203.1790, Calcd for C<sub>15</sub>H<sub>23</sub> (c) 203.1800; Found 189.1627, Calcd for C<sub>14</sub>H<sub>21</sub> (c-14) 189.1642; Found 133.0991, Calcd for C<sub>10</sub>H<sub>13</sub> (d) 133.1016. IR  $\nu_{\max}^{\text{KBr}}$  cm<sup>-1</sup>: 1730, 1230. <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.74 (3H, s, 26-H<sub>3</sub>), 0.85 and 0.94 (each 3H, d,  $J=6$  Hz, 29- and 30-H<sub>3</sub>), 0.95 (3H, s, 23-H<sub>3</sub>), 1.01 (3H, s, 25-H<sub>3</sub>), 1.07 (3H, s, 27-H<sub>3</sub>), 2.05 and 2.07 (each 3H, s, OAc), 2.25 (1H, d,  $J=11$  Hz, 18-H), 3.61 (3H, s, COOCH<sub>3</sub>), 4.13 and 4.36 (each 1H, d,  $J=12$  Hz, 24-H<sub>2</sub>), 4.60 (1H, dd,  $J=9, 7$  Hz, 3-H), and 5.26 (1H, br t,  $J=3.5$  Hz, 12-H).

4) Methyl 2 $\alpha$ -hydroxy-3-epiursolate diacetate (**9b**) (3 mg), amorphous powder,  $[\alpha]_D^{22} + 15^\circ$  ( $c=0.25$ , CHCl<sub>3</sub>). MS  $m/z$ : 570 (M<sup>+</sup>), 510, 450, 262, 249, 203, 189, 133. High resolution MS: Found 570.3876, Calcd for C<sub>35</sub>H<sub>54</sub>O<sub>6</sub> (M<sup>+</sup>) 570.3919. <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.75 (3H, s, 26-H<sub>3</sub>), 0.85 and 0.93 (each 3H, d,  $J=6$  Hz, 29- and 30-H<sub>3</sub>), 0.87 (3H, s, 23-H<sub>3</sub>), 0.98 (3H, s, 24-H<sub>3</sub>), 1.04 (3H, s, 25-H<sub>3</sub>), 1.11 (3H, s, 27-H<sub>3</sub>), 1.97 and 2.12 (each 3H, s, OAc), 2.25 (1H, d,  $J=11$  Hz, 18-H), 3.62 (3H, s, COOCH<sub>3</sub>), 4.99 (1H, d,  $J=2.5$  Hz, 3-H), and 5.20–5.32 (2H, m, 2- and 12-H). These spectral data are compatible with the reported values.<sup>7)</sup>

5) Methyl euscaphate diacetate (**12b**) (4 mg), colorless needles, mp 115–118 °C,  $[\alpha]_D^{22} + 17^\circ$  ( $c=0.35$ , CHCl<sub>3</sub>). This compound was identified by direct comparison of the TLC behavior, <sup>1</sup>H-NMR, and GC with those of an authentic sample.

6) Methyl sumaresinolate monoacetate (**15b**) (4 mg), colorless needles (from EtOH), mp 220–222 °C. This was identified by GC and <sup>1</sup>H-NMR comparisons with an authentic sample.

F-V (1.0 g) was also acetylated in the same manner as described above and the product was chromatographed on a silica gel (50 g) column. The eluate with MeOH-CHCl<sub>3</sub> (2:98) (120 mg) was treated with CH<sub>2</sub>N<sub>2</sub> and the resulting methyl ester was separated by PLC with AcOEt-benzene (2:8) as the eluent. The least polar band afforded methyl 2 $\alpha$ ,3 $\alpha$ ,23-triacetoxyurs-12-en-28-oate (**10b**) (10 mg), amorphous powder,  $[\alpha]_D^{22} + 37^\circ$  ( $c=0.35$ , CHCl<sub>3</sub>). MS  $m/z$ : 628 (M<sup>+</sup>), 568, 508, 448, 262, 249, 203, 189, 133. High resolution MS: Found 628.4022, Calcd for C<sub>37</sub>H<sub>56</sub>O<sub>8</sub> 628.3975. <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.75 (3H, s, 26-H<sub>3</sub>), 0.85 and 0.93 (each 3H, d,  $J=6$  Hz, 29- and 30-H<sub>3</sub>), 1.07 (3H, s, 24-H<sub>3</sub>), 1.11 (6H, s, 25- and 27-H<sub>3</sub>), 1.95, 2.00, and 2.06 (each 3H, s, OAc), 2.25 (1H, d,  $J=11$  Hz, 18-H), 3.61 (3H, s, COOCH<sub>3</sub>), 3.73 and 4.11 (each 1H, d,  $J=12$  Hz, 23-H<sub>2</sub>), and 5.18–5.37 (3H, m, 2-, 3-, and 12-H).<sup>8)</sup> The middle band gave a mixture of methyl asiatic triacetate (**8b**) and methyl arjunolate triacetate (**17b**) (40 mg) (approximate ratio of 62:38 by GC analysis). Both compounds were identified by GC and GC-MS comparisons with corresponding authentic samples. The most polar band gave impure methyl 2 $\alpha$ ,3 $\beta$ ,24-triacetoxyurs-12-en-28-oate (**3b**) (6 mg), purification of which is described below.

**Separation of the Mixture of Methyl 3 $\beta$ ,23-Triacetoxyurs-12-en-28-oate (1b) and Methyl Hederagenin Diacetate (16b) by HPLC**—Preparative HPLC was performed on a Waters Associates ALC/GPC 201D compact type liquid chromatograph using a TSK-GEL ODS-120A column (column size, 4.6 mm i.d.  $\times$  25 cm; detector setting, UV

215 nm) with acetonitrile as the eluent (flow rate 0.5 ml/min). The mixture (**1b** and **16b**) (12 mg) gave **16b** (1.0 mg) and **1b** (3.7 mg). The former was identified by GC, MS, and  $^1\text{H-NMR}$  comparisons with an authentic sample. Methyl  $3\beta,23$ -diacetoxys-12-en-28-oate (**1b**) was obtained as colorless needles (from EtOH), mp  $180\text{--}181.5^\circ\text{C}$ ,  $[\alpha]_{\text{D}}^{21} + 64.1^\circ$  ( $c=0.22$ ,  $\text{CHCl}_3$ ). MS  $m/z$ : 570 ( $\text{M}^+$ ), 510, 450, 262, 249, 203, 189, 133. High resolution MS: Found 570.3913, Calcd for  $\text{C}_{35}\text{H}_{54}\text{O}_6$  ( $\text{M}^+$ ) 570.3919; Found 262.1937, Calcd for  $\text{C}_{17}\text{H}_{26}\text{O}_2$  (a) 262.1933; Found 249.1842, Calcd for  $\text{C}_{16}\text{H}_{25}\text{O}_2$  (b) 249.1854; Found 203.1789, Calcd for  $\text{C}_{15}\text{H}_{23}$  (c) 203.1799. IR  $\nu_{\text{max}}^{\text{KBr}} \text{cm}^{-1}$ : 1730, 1230.  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : 0.75 (3H, s, 26- $\text{H}_3$ ), 0.83 (3H, s, 24- $\text{H}_3$ ), 0.87 and 0.94 (each 3H, d,  $J=6$  Hz, 29- and 30- $\text{H}_3$ ), 0.98 (3H, s, 25- $\text{H}_3$ ), 1.07 (3H, s, 27- $\text{H}_3$ ), 2.02 and 2.07 (each 3H, s, OAc), 2.25 (1H, d,  $J=11$  Hz, 18-H), 3.62 (3H, s,  $\text{COOCH}_3$ ), 3.71 and 3.88 (each 1H, d,  $J=12$  Hz, 23- $\text{H}_2$ ), 4.81 (1H, dd,  $J=10, 5.5$  Hz, 3-H), and 5.23 (1H, br t,  $J=3.5$  Hz, 12-H).

**Purification of Methyl  $2\alpha,3\beta,24$ -Triacetoxys-12-en-28-oate (**3b**) by HPLC**—Preparative HPLC of the above-mentioned impure **3b** under the same conditions as used for **1b**, gave a pure sample (**3b**) (1.4 mg), amorphous powder,  $[\alpha]_{\text{D}}^{21} + 51^\circ$  ( $c=0.1$ ,  $\text{CHCl}_3$ ). MS  $m/z$ : 628 ( $\text{M}^+$ ), 568, 508, 262, 249, 203, 189, 133. High resolution MS: Found 628.3954, Calcd for  $\text{C}_{37}\text{H}_{56}\text{O}_8$  ( $\text{M}^+$ ) 628.3974; Found 262.1918, Calcd for  $\text{C}_{17}\text{H}_{26}\text{O}_2$  (a) 262.1932; Found 249.1820, Calcd for  $\text{C}_{16}\text{H}_{25}\text{O}_2$  (b) 249.1854; Found 203.1785, Calcd for  $\text{C}_{15}\text{H}_{23}$  (c) 203.1799; Found 189.1629, Calcd for  $\text{C}_{14}\text{H}_{21}$  (c-14) 189.1642; Found 133.1000, Calcd for  $\text{C}_{10}\text{H}_{13}$  (d) 133.1016. IR  $\nu_{\text{max}}^{\text{KBr}} \text{cm}^{-1}$ : 1740, 1235.  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : 0.73 (3H, s, 26- $\text{H}_3$ ), 0.83 and 0.94 (each 3H, d,  $J=6$  Hz, 29- and 30- $\text{H}_3$ ), 1.02 (3H, s, 23- $\text{H}_3$ ), 1.06 (3H, s, 25- $\text{H}_3$ ), 1.07 (3H, s, 27- $\text{H}_3$ ), 1.97, 2.05, and 2.06 (each 3H, s, OAc), 2.25 (1H, d,  $J=11$  Hz, 18-H), 3.60 (3H, s,  $\text{COOCH}_3$ ), 4.20 (2H, s, 24- $\text{H}_2$ ), 4.86 (1H, d,  $J=10.5$  Hz, 3-H), 5.18 (1H, td,  $J=10.5, 4.5$  Hz, 2-H), and 5.25 (1H, br t,  $J=3.5$  Hz, 12-H).

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