

Studies on the Constituents of Asclepiadaceae Plants. XL.¹⁾ Absolute Configurations of Tomentogenin and Utendin

HIDEO SETO, KOJI HAYASHI, and HIROSHI MITSUHASHI

Faculty of Pharmaceutical Sciences, Hokkaido University²⁾

(Received January 27, 1976)

The configuration at C-20 of tomentogenin and utendin was determined as s by means of the circular dichroism experiment of 20-O-o-nitrobenzoyl derivatives. From these results, the absolute configurations of tomentogenin and utendin were determined as 5 α -pregnan-3 β ,12 β ,14 β ,17 β ,20 α -pentol and pregn-5-ene-3 β ,12 β ,14 β ,17 β ,20 α -pentol, respectively.

The tentative structure of tomentogenin³⁾ (I), isolated from the stem of *Marsdenia tomentosa* DECNE, was proposed as 5 α -pregnan-3 β ,12 β ,14 β ,17 β ,20-pentol with some ambiguity for the configuration of C-20 hydroxyl group, and the configuration at C-20 of utendin^{3b,c,4)} (II), pregn-5-ene-3 β ,12 β ,14 β ,17 β ,20-pentol, was confirmed to be the same as that of I. Several tomentogenin and utendin derivatives have been isolated from *Caralluma dalzielii* by Tschesche⁵⁾ and from *M. tomentosa* by us.^{1,6)}

Recently, Hayashi and Mitsuhashi proved the configuration at C-20 of sarcostin (III) to be s^{7,8)} on the basis of the optical rotatory dispersion (ORD) examination synchronized with X-ray analysis. In this paper, we report the absolute configurations of tomentogenin (I) and utendin (II).

Partial acetylation of I with 2.5 molar equiv. of acetic anhydride-pyridine afforded 3,12-diacetate (IV), mp 217—220°, $[\alpha]_D^{25} + 31^\circ$ ($c=0.42$, CHCl₃), as a main product, with 3-monoacetate (V), 3,20-diacetate (VI), and 3,12,20-triacetate (VII). The molecular formula of C₂₅H₄₀O₇ was given for IV from its elemental analysis and mass spectrum (M^+ at m/e 452). The infrared (IR) spectrum of IV showed absorptions for hydroxyl groups at 3550, 3490, 3420, and 1030 cm⁻¹, and esters at 1730, 1720, 1260, and 1240 cm⁻¹. The nuclear magnetic resonance (NMR) spectrum of IV showed signals for two tertiary methyl groups at δ 0.84 (s) and 1.22 (s), one secondary methyl group at 1.12 (d, $J=6$ Hz), two acetyl groups at 2.02 (s) and 2.10 (s), and three hydroxy-methines at 3.52 (q, $J=6$ Hz), 4.50 (d.d, $J=6, 11$ Hz) and 4.60 (m). Irradiation of 21-Me group protons (δ 1.12) collapsed the quartet at δ 3.52 assignable for C-20 proton

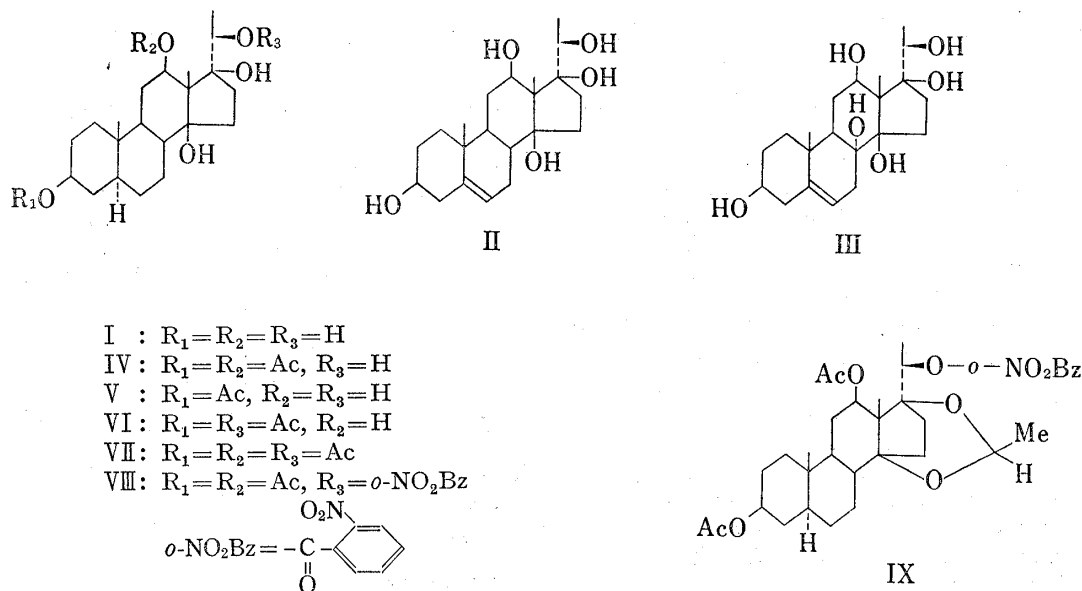
- 1) Part XXXIX: H. Seto, T. Sasaki, K. Hayashi, and H. Mitsuhashi, *Chem. Pharm. Bull.* (Tokyo), **24**, 2185 (1976).
- 2) Location: Nishi-6-chome, Kita-12-jo, Kita-ku, Sapporo, 060, Japan.
- 3) a) H. Mitsuhashi, I. Takemori, Y. Shimizu, T. Nomura, and E. Yamada, *Chem. Pharm. Bull.* (Tokyo), **10**, 804 (1962); b) H. Mitsuhashi, T. Sato, T. Nomura, and I. Takemori, *Chem. Pharm. Bull.* (Tokyo), **12**, 981 (1964); c) H. Mitsuhashi, T. Sato, T. Nomura, and I. Takemori, *Chem. Pharm. Bull.* (Tokyo), **13**, 267 (1965); d) M. Fukuoka and H. Mitsuhashi, *Chem. Pharm. Bull.* (Tokyo), **16**, 1634 (1968).
- 4) E. Abisch, Ch. Tamm, and T. Reichstein, *Helv. Chim. Acta*, **42**, 1014 (1959).
- 5) R. Tschesche and G. Marwede, *Tetrahedron Letters*, **1967**, 1359.
- 6) a) H. Seto, K. Hayashi, and H. Mitsuhashi, *Chem. Pharm. Bull.* (Tokyo), **23**, 1552 (1975); b) H. Seto, K. Hayashi, and H. Mitsuhashi, *Chem. Pharm. Bull.* (Tokyo), **23**, 2397 (1975). In this paper, we gave names of tomentin and dehydrotomentin to 12 β -O,20-O-diacetyltomentogenin and 12 β -O,20-O-diacetyl-utendin, respectively. The name "tomentin" had been used for the coumarin aglycone isolated from *Prunus tomentosa* by Hasegawa [*Bot. Mag.* (Tokyo), **82**, 458 (1969)], and therefore, we discontinue to use these names and change them to tomentinin and dehydrotomentinin; c) H. Seto, K. Hayashi, and H. Mitsuhashi, *Chem. Pharm. Bull.* (Tokyo), **24**, 443 (1976); d) H. Seto, K. Hayashi, and H. Mitsuhashi, *Chem. Pharm. Bull.* (Tokyo), **24**, 1552 (1976).
- 7) K. Hayashi and H. Mitsuhashi, *Chem. Pharm. Bull.* (Tokyo), **23**, 1845 (1975).
- 8) T. Reichstein, *Naturwissenschaften*, **54**, 53 (1967).

to a singlet. 3,20-Diacetate (VI), $C_{25}H_{40}O_7$ (M^+ at m/e 452) showed mp 207—209°, $[\alpha]_D^{25} +13^\circ$ ($c=0.50$, $CHCl_3$). The IR spectrum of VI showed absorptions for hydroxyl groups at 3550, 3400, 1080, 1045, and 1030 cm^{-1} , and esters at 1730, 1710, 1270, and 1250 cm^{-1} . The NMR spectrum of VI showed signals for two tertiary methyl groups at δ 0.84 (s) and 1.12 (s), one secondary methyl group at 1.26 (d, $J=6$ Hz), two acetyl groups at 2.04 (s), and three hydroxy-methines at 3.36 (d.d, $J=6, 11$ Hz), 4.68 (m), and 5.14 (q, $J=6$ Hz). Irradiation of 21-Me group protons (δ 1.26) collapsed the quartet at δ 5.14 assignable for C-20 proton to a singlet. The yield-ratio between IV and VI on this condition was 8:1, which indicates that the acetylation of C-12 β hydroxy group occurred more easily than that of C-20 hydroxyl group in spite of a relatively less hindered position of the latter compared to the former.

o-Nitrobenzoylation of IV with *o*-nitrobenzoyl chloride-pyridine afforded 20-*O*-*o*-nitrobenzoate (VIII), mp 275—278°. The molecular formula of $C_{32}H_{43}O_{10}N$ was given for VIII from its elemental analysis and mass spectrum (M^+ —*o*-nitrobenzoic acid at m/e 434). The IR spectrum of VIII showed absorptions for hydroxyl groups at 3500, 3425, 1070, and 1030 cm^{-1} , acetyl esters at 1730, 1250, and 1240 cm^{-1} , and an *o*-nitrobenzoyl ester at 1710, 1540, 1360, and 1145 cm^{-1} . The NMR spectrum of VIII showed signals for two tertiary methyl groups at δ 0.80 (s) and 1.16 (s), one secondary methyl group at 1.44 (d, $J=6$ Hz), two acetyl groups at 2.04 (s), three hydroxy-methines at 4.60 (m), 4.62 (d.d, $J=6, 11$ Hz), and 4.92 (q, $J=6$ Hz), and four aromatic protons at 7.70 (m). The circular dichroism (CD) spectrum of VIII showed a negative Cotton effect at 330 nm ($[\theta] -2555$).

In order to mask the 17 β -hydroxyl group, VIII was treated in paraldehyde-pyridine with boron trifluoride etherate as a catalyst to afford a cyclic *O*-ethylidene derivative (IX), mp 203—205°. The molecular formula of $C_{34}H_{45}O_{10}N$ was given for IX from its elemental analysis and mass spectrum (M^+ at m/e 627). The IR spectrum of IX showed absorptions for acetyl esters at 1735, 1250, and 1235 cm^{-1} , an *o*-nitrobenzoyl ester at 1725, 1545, 1360, and 1140 cm^{-1} , a cyclic *O*-ethylidene at 1290 and 1120 cm^{-1} , and no hydroxyl group. The NMR spectrum of IX showed signals for two tertiary methyl groups at δ 0.82 (s) and 1.22 (s), two secondary methyl groups at 1.33 (d, $J=6$ Hz) and 1.38 (d, $J=5$ Hz), two acetyl groups at 2.02 (s), three hydroxy-methines at 4.70 (d.d, $J=6, 11$ Hz), 4.72 (m), 4.96 (q, $J=6$ Hz), one proton at 5.18 (q, $J=5$ Hz), and four aromatic protons at 7.70 (m). The CD spectrum of IX showed a positive Cotton effect at 323 nm ($[\theta] +5563$).

Conversion of the Cotton effect of 20-*C*-*o*-nitrobenzoate from a negative to a positive on masking C-17 β hydroxyl group was already established in the experiment on sarcostin⁷⁾ (III).



Chart

Catalytic hydrogenation of utendin (II) with platinum afforded a dihydro derivative, which was identical with tomentogenin^{3b,c)} (I) from the comparison of their spectral data and mixed mp with an authentic sample.

From these facts, the configuration at C-20 of tomentogenin (I) and utendin (II) was determined as *s*, so that the absolute configurations of I and II were finally determined as 5 α -pregnan-3 β ,12 β ,14 β ,17 β ,20 α -pentol and pregn-5-ene-3 β ,12 β ,14 β ,17 β ,20 α -pentol, respectively.

Experimental

Melting points were determined on a Kofler hot stage and are uncorrected. Optical rotations were measured in CHCl_3 solution on a Hitachi S115-4 polarimeter, NMR spectra on a JEOL PS-100 spectrometer operating at 100 MHz with tetramethylsilane (TMS) as an internal standard, mass spectra on a Hitachi RMU-7 mass spectrometer, and IR spectra were taken in Nujol mull on a Hitachi 215 spectrometer. CD spectra were measured in MeOH solution on a JASCO J-20 spectropolarimeter. Thin-layer chromatography (TLC) was performed on silica gel HF₂₅₄ (Merck, Type 60). Tomentogenin and utendin used were obtained from *M. tomentosa*.

Partial Acetylation of Tomentogenin (I)—A solution of 201 mg of tomentogenin (I) in 0.14 ml of Ac_2O and 2 ml of pyridine was allowed to stand for 25 hr at 40° and the reaction mixture was separated by preparative TLC (CHCl_3 : MeOH=97: 3) to afford 25 mg of 3-monoacetate (V), 13 mg of 3,20-diacetate (VI), 108 mg of 3,12-diacetate (IV), and 43 mg of 3,12,20-triacetate (VII). 3-Monoacetate (V) was recrystallized from hexane-acetone to plates, mp 204–208°, $[\alpha]_D^{18} + 40^\circ$ ($c=0.30$, CHCl_3). Mass Spectrum m/e : 410 (M^+), 392 ($\text{M}^+ - \text{H}_2\text{O}$), 374 ($\text{M}^+ - 2\text{H}_2\text{O}$), 365 ($\text{M}^+ - \text{CHOH} \cdot \text{Me}$),⁹⁾ 356 ($\text{M}^+ - 3\text{H}_2\text{O}$), 350 ($\text{M}^+ - \text{AcOH}$), 347 ($\text{M}^+ - \text{CHOH} \cdot \text{Me} - \text{H}_2\text{O}$), 338 ($\text{M}^+ - 4\text{H}_2\text{O}$), 332 ($\text{M}^+ - \text{AcOH} - \text{H}_2\text{O}$), 329 ($\text{M}^+ - \text{CHOH} \cdot \text{Me} - 2\text{H}_2\text{O}$), 314 ($\text{M}^+ - \text{AcOH} - 2\text{H}_2\text{O}$), 305 ($\text{M}^+ - \text{CHOH} \cdot \text{Me} - \text{AcOH}$), 304, 291, 287 ($\text{M}^+ - \text{CHOH} \cdot \text{Me} - \text{AcOH} - \text{H}_2\text{O}$), 286, 269 ($\text{M}^+ - \text{CHOH} \cdot \text{Me} - \text{AcOH} - 2\text{H}_2\text{O}$), 226, 43 (base peak). IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 3350, 1730, 1245, 1030. NMR $\delta_{\text{ppm}}^{\text{CDCl}_3}$: 0.84 (3H, *s*, 19-Me), 1.12 (3H, *s*, 18-Me), 1.20 (3H, *d*, $J=6$ Hz, 21-Me), 2.04 (3H, *s*, OAc), 3.44 (1H, *d,d*, $J=6$, 11 Hz, 12 α -H), 4.00 (1H, *q*, $J=6$ Hz, 20 β -H), 4.66 (1H, *m*, 3 α -H). Anal. Calcd. for $\text{C}_{23}\text{H}_{38}\text{O}_6$: C, 67.29; H, 9.33. Found: C, 67.52; H, 9.59.

3,20-Diacetate (VI) was recrystallized from acetone to needles, mp 207–209°, $[\alpha]_D^{18} + 13^\circ$ ($c=0.50$, CHCl_3). Mass Spectrum m/e : 452 (M^+), 434 ($\text{M}^+ - \text{H}_2\text{O}$), 416 ($\text{M}^+ - 2\text{H}_2\text{O}$), 392 ($\text{M}^+ - \text{AcOH}$), 374 ($\text{M}^+ - \text{AcOH} - \text{H}_2\text{O}$), 365 ($\text{M}^+ - \text{CHOAc} \cdot \text{Me}$),⁹⁾ 356 ($\text{M}^+ - \text{AcOH} - 2\text{H}_2\text{O}$), 347 ($\text{M}^+ - \text{CHOAc} \cdot \text{Me} - \text{H}_2\text{O}$), 338 ($\text{M}^+ - \text{AcOH} - 3\text{H}_2\text{O}$), 332 ($\text{M}^+ - 2\text{AcOH}$), 329 ($\text{M}^+ - \text{CHOAc} \cdot \text{Me} - 2\text{H}_2\text{O}$), 314 ($\text{M}^+ - 2\text{AcOH} - \text{H}_2\text{O}$), 311 ($\text{M}^+ - \text{CHOAc} \cdot \text{Me} - 3\text{H}_2\text{O}$), 304, 291, 286, 226, 43 (base peak). IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 3550, 3400, 1730, 1710, 1270, 1250, 1080, 1045, 1030. NMR $\delta_{\text{ppm}}^{\text{CDCl}_3}$: 0.84 (3H, *s*, 19-Me), 1.12 (3H, *s*, 18-Me), 1.26 (3H, *d*, $J=6$ Hz, 21-Me), 2.04 (6H, *s*, 2 $\times$ OAc), 3.36 (1H, *d,d*, $J=6$, 11 Hz, 12 α -H), 4.68 (1H, *m*, 3 α -H), 5.14 (1H, *q*, $J=6$ Hz, 20 β -H). Anal. Calcd. for $\text{C}_{25}\text{H}_{40}\text{O}_7$: C, 66.34; H, 8.91. Found: C, 66.12; H, 9.02.

3,12-Diacetate (IV) was recrystallized from acetone to needles, mp 217–220°, $[\alpha]_D^{18} + 31^\circ$ ($c=0.42$, CHCl_3). Mass Spectrum m/e : 452 (M^+), 434 ($\text{M}^+ - \text{H}_2\text{O}$), 416 ($\text{M}^+ - 2\text{H}_2\text{O}$), 407 ($\text{M}^+ - \text{CHOH} \cdot \text{Me}$), 392 ($\text{M}^+ - \text{AcOH}$), 389 ($\text{M}^+ - \text{CHOH} \cdot \text{Me} - \text{H}_2\text{O}$), 374 ($\text{M}^+ - \text{AcOH} - \text{H}_2\text{O}$), 371 ($\text{M}^+ - \text{CHOH} \cdot \text{Me} - 2\text{H}_2\text{O}$), 356 ($\text{M}^+ - \text{AcOH} - 2\text{H}_2\text{O}$), 347 ($\text{M}^+ - \text{CHOH} \cdot \text{Me} - \text{AcOH}$), 338 ($\text{M}^+ - \text{AcOH} - 3\text{H}_2\text{O}$), 332 ($\text{M}^+ - 2\text{AcOH}$), 329 ($\text{M}^+ - \text{CHOH} \cdot \text{Me} - \text{AcOH} - \text{H}_2\text{O}$), 314 ($\text{M}^+ - 2\text{AcOH} - \text{H}_2\text{O}$), 311 ($\text{M}^+ - \text{CHOH} \cdot \text{Me} - \text{AcOH} - 2\text{H}_2\text{O}$), 304, 291, 286, 226, 43 (base peak). IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 3550, 3490, 3420, 1730, 1720, 1260, 1240, 1030. NMR $\delta_{\text{ppm}}^{\text{CDCl}_3}$: 0.84 (3H, *s*, 19-Me), 1.12 (3H, *d*, $J=6$ Hz, 21-Me), 1.22 (3H, *s*, 18-Me), 2.02 (3H, *s*, OAc), 2.10 (3H, *s*, OAc), 3.52 (1H, *q*, $J=6$ Hz, 20 β -H), 4.50 (1H, *d,d*, $J=6$, 11 Hz, 12 α -H), 4.60 (1H, *m*, 3 α -H). Anal. Calcd. for $\text{C}_{25}\text{H}_{40}\text{O}_7$: C, 66.34; H, 8.91. Found: C, 66.59; H, 9.04.

3,12,20-Triacetate (VII) was recrystallized from MeOH-acetone to prisms, mp 285–288°, $[\alpha]_D^{11} + 26^\circ$ ($c=0.80$, CHCl_3). Mass Spectrum m/e : 476 (M^+), 434 ($\text{M}^+ - \text{AcOH}$), 416 ($\text{M}^+ - \text{AcOH} - \text{H}_2\text{O}$), 407 ($\text{M}^+ - \text{CHOAc} \cdot \text{Me}$), 398 ($\text{M}^+ - \text{AcOH} - 2\text{H}_2\text{O}$), 389 ($\text{M}^+ - \text{CHOAc} \cdot \text{Me} - \text{H}_2\text{O}$), 374 ($\text{M}^+ - 2 \times \text{AcOH}$), 356 ($\text{M}^+ - 2 \times \text{AcOH} - \text{H}_2\text{O}$), 339 ($\text{M}^+ - \text{CHOAc} \cdot \text{Me} - \text{AcOH}$), 314 ($\text{M}^+ - 3 \times \text{AcOH}$), 304, 291, 286, 226, 43 (base peak). IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 3470, 3400, 1740, 1710, 1275, 1250, 1235, 1050, 1040, 1020. NMR $\delta_{\text{ppm}}^{\text{CDCl}_3}$: 0.82 (3H, *s*, 19-Me), 1.22 (3H, *s*, 18-Me), 1.26 (3H, *d*, $J=6$ Hz, 21-Me), 1.97 (3H, *s*, OAc), 2.02 (3H, *s*, OAc), 2.08 (3H, *s*, OAc), 4.54 (1H, *q*, $J=6$ Hz, 20 β -H), 4.60 (1H, *d,d*, $J=6$, 11 Hz, 12 α -H), 4.62 (1H, *m*, 3 α -H). Anal. Calcd. for $\text{C}_{27}\text{H}_{42}\text{O}_8$: C, 65.56; H, 8.56. Found: C, 65.59; H, 8.64.

***o*-Nitrobenzoylation of Tomentogenin 3,12-Diacetate (IV)**—A solution of 151 mg of tomentogenin 3,12-diacetate (IV) and 232 mg of *o*-nitrobenzoyl chloride in 2 ml of pyridine was stirred for 20 hr at room temperature. The reaction mixture was purified by preparative TLC to afford 65 mg of *o*-nitrobenzoate (VIII), recrystallized from MeOH-acetone-hexane to needles, mp 275–278°. Mass Spectrum m/e : 434 ($\text{M}^+ - o\text{-nitrobenzoic acid}$), 416 ($\text{M}^+ - o\text{-nitrobenzoic acid} - \text{H}_2\text{O}$), 407 ($\text{M}^+ - \text{CHO} \cdot \text{C}_7\text{H}_4\text{O}_3\text{N} \cdot \text{Me}$), 374 ($\text{M}^+ - o\text{-nitrobenzoic acid} - \text{AcOH}$), 356 ($\text{M}^+ - o\text{-nitrobenzoic acid} - \text{AcOH} - \text{H}_2\text{O}$), 346, 341, 314 ($\text{M}^+ - o\text{-nitrobenzoic acid} - 2 \times \text{AcOH}$),

9) M. Fukuoka and H. Mitsuhashi, *Chem. Pharm. Bull.* (Tokyo), **17**, 2448 (1969).

304, 291, 286, 226, 167, 43 (base peak). IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 3500, 3425, 1730, 1710, 1580, 1540, 1360, 1250, 1240, 1145, 1070, 1030. NMR $\delta_{\text{ppm}}^{\text{CDCl}_3}$: 0.80 (3H, s, 19-Me), 1.16 (3H, s, 18-Me), 1.44 (3H, d, $J=6$ Hz, 21-Me), 2.04 (6H, s, $2 \times \text{OAc}$), 4.60 (1H, m, $3\alpha\text{-H}$), 4.62 (1H, d.d, $J=6, 11$ Hz, $12\alpha\text{-H}$), 4.92 (1H, q, $J=6$ Hz, $20\beta\text{-H}$), 7.70 (4H, m, aromatic protons). CD: $[\theta]_{330} -2555$ ($c=3.3 \times 10^{-3}$ M, MeOH). Anal. Calcd. for $\text{C}_{32}\text{H}_{43}\text{O}_{10}\text{N}$: C, 63.88; H, 7.20; N, 2.33. Found: C, 63.74; H, 7.13; N, 2.45.

Reaction of *o*-Nitrobenzoate (VIII) with Paraldehyde—The suspension of 52 mg of *o*-nitrobenzoate (VIII) in 2.5 ml of paraldehyde and three drops of BF_3 -ether was stirred for 1 hr at room temperature. After addition of 100 mg of K_2CO_3 , the reaction mixture was filtered to remove excess K_2CO_3 and concentrated *in vacuo*. The residue was purified by preparative TLC (CHCl_3) to afford 39 mg of cyclic *O*-ethylidene derivative (IX). Cyclic *O*-ethylidene derivative (IX) was recrystallized from acetone-hexane to needles, mp $203-205^\circ$. Mass Spectrum m/e : 627 (M^+), 612 ($\text{M}^+ - \text{Me}$), 597 ($\text{M}^+ - 2 \cdot \text{Me}$), 567 ($\text{M}^+ - \text{AcOH}$), 541, 523, 507 ($\text{M}^+ - 2\text{AcOH}$), 460 ($\text{M}^+ - o\text{-nitrobenzoic acid}$), 304, 291, 286, 226, 150, 120, 43 (base peak). IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 1735, 1725, 1600, 1580, 1545, 1360, 1290, 1250, 1235, 1140, 1120, 1070, 1030. NMR $\delta_{\text{ppm}}^{\text{CDCl}_3}$: 0.82 (3H, s, 19-Me), 1.22 (3H, s, 18-Me), 1.33 (3H, d, $J=6$ Hz, 21-Me), 1.38 (3H, d, $J=5$ Hz), 2.02 (6H, s, $2 \times \text{OAc}$), 4.70 (1H, d.d, $J=6, 11$ Hz, $12\alpha\text{-H}$), 4.72 (1H, m, $3\alpha\text{-H}$), 4.96 (1H, q, $J=6$ Hz, $20\beta\text{-H}$), 5.18 (1H, q, $J=5$ Hz), 7.70 (4H, m, aromatic protons). CD: $[\theta]_{323} +5563$ ($c=3.2 \times 10^{-3}$ M, MeOH). Anal. Calcd. for $\text{C}_{34}\text{H}_{45}\text{O}_{10}\text{N}$: C, 65.05; H, 7.23; N, 2.23. Found: C, 64.94; H, 7.20; N, 2.24.

Hydrogenation of Utendin (II)—A solution of 15 mg of utendin (II) in 10 ml of EtOAc-MeOH (4:1) and one drop of AcOH was hydrogenated over 30 mg of PtO_2 under atmospheric pressure at 20° . It consumed 1 mol. equiv. of H_2 (2 ml) during 3 hr. After the reaction mixture was diluted with H_2O with cooling, the catalyst was removed. The product was purified by preparative TLC (CHCl_3 : MeOH=9:1) to yield 9 mg of dihydroderivative, which showed mp $265-268^\circ$ and mixed mp with I, $263-266^\circ$. Mass Spectrum m/e : 368 (M^+), 350 ($\text{M}^+ - \text{H}_2\text{O}$), 332 ($\text{M}^+ - 2\text{H}_2\text{O}$), 323 ($\text{M}^+ - \text{CHOH} \cdot \text{Me}$), 305 ($\text{M}^+ - \text{CHOH} \cdot \text{Me} - \text{H}_2\text{O}$, base peak), 287 ($\text{M}^+ - \text{CHOH} \cdot \text{Me} - 2\text{H}_2\text{O}$), 269 ($\text{M}^+ - \text{CHOH} \cdot \text{Me} - 3\text{H}_2\text{O}$), 262, 249, 244, 226. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 3400, 1040. NMR $\delta_{\text{ppm}}^{\text{pyridine-}d_5}$: 0.76 (3H, s, 19-Me), 1.54 (3H, d, $J=6$ Hz, 21-Me), 1.64 (3H, s, 18-Me), 3.74 (1H, d.d, $J=6, 11$ Hz, $12\alpha\text{-H}$), 3.80 (1H, m, $3\alpha\text{-H}$), 4.38 (1H, q, $J=6$ Hz, $20\beta\text{-H}$). Anal. Calcd. for $\text{C}_{21}\text{H}_{36}\text{O}_5$: C, 68.44; H, 9.85. Found: C, 68.35; H, 9.99.

Acknowledgement The plant material was kindly collected by Mr. M. Kawaguchi (Owase, Mie Prefecture) and we express our sincere gratitude to him. We also thank Miss M. Takahashi for mass spectral measurement, Miss T. Obara and Mrs. H. Matsumoto for elemental analysis, and Miss T. Okayama for NMR spectral measurement.