ALKALOID CONSTITUENTS FROM ERYTHRINA XBIDWILLII FLOWERS

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Key Word Index-Erythrina xbidwillii; Leguminosae; flowers; alkaloids; erythristemine-N-oxide.

Abstract—The alkaloids present in the flowers of *Erythrina xbidwillii* have been screened by GC-MS. A novel alkaloid, erythristemine-N-oxide has been isolated and its structure established by spectroscopic methods.

INTRODUCTION

A series of studies on the alkaloid content of different parts of some 70 species of *Erythrina* have been undertaken in our laboratories and in that of Rinehart at Illinois [1]. These have been screened for alkaloids using GC/MS as the primary analytical tool to facilitate chemotaxonomic studies. Erythrinine [2] and erybidine [3] have been earlier isolated from the leaves of *E. xbidwillii*. We now report our studies on the flowers of *E. xbidwillii* which has led to the characterization of a new natural product, erythristemine-*N*-oxide (1), in addition to other known alkaloids.

RESULTS AND DISCUSSION

The GC and GC-MS examination of the petrol-soluble and methanol-soluble free alkaloid fractions of E. *xbidwillii* showed the presence of erysotrine (57 and 21%) and erythrartine (43 and 70%), respectively. The latter fraction also indicated the presence of erythristemine (9%). The liberated fraction afforded only erythrartine (16%).

Preparative scale isolation of the petrol-soluble alkaloid fraction by chromatography over alumina afforded two fractions identified as erysotrine and erythrartine. The structural assignment of erythrartine (3) was achieved by high field ¹H NMR spectroscopy (360 MHz); the chemical shifts of all resonances and coupling constants are given in Table 1. The ¹³C NMR data is also consistent with the structure for erythrartine. Sarragiotto et al. [4] have assigned C-1 and C-2 resonances at δ 125.3 ppm and 131.2, respectively. However, a twodimensional ¹³C-¹H NMR chemical shift correlation experiment revealed these resonances at δ 131.4 and 125.5, respectively [5]. The EI mass spectrum of the base showed significant peaks at m/z 329 [M]⁺ 314 [M-Me]⁺ (33), 311 [M-H₂O]⁺ (20) [M-OMe]⁺ (100), 296 (30) and 280 (24). (69). (20), 298

A new natural product, erythristemine-N-oxide (1) (0.008%) as well as the known erysotrine-N-oxide (2) (0.01%) [4] were isolated in addition to erysotrine, erythrartine and erysotramidine from the methanol-soluble free alkaloid fraction. This is the first reported occurrence of 1 from a natural source although it is known as a synthetic compound [4]. A complete study of their spectral characteristics and comparison with their parent compounds is summarized in the Experimental and in Table 1.

The ¹HNMR spectrum of erythristemine-N-oxide yielded a H_e-4 resonance at a high field ($\delta 2.02$), whereas the H_a-4 was shifted downfield (δ 3.23). The H-8 (δ 5.10, 4.34) and H-10 (δ 4.34, 3.84) resonances were 1 ppm downfield as compared to those of the parent compound. The H-11 proton also resonated downfield (0.5 ppm). The presence of a $[M-16]^+$ peak in the EI mass spectrum, characteristic of N-oxides [6] identified 1 as erythristemine N-oxide. Similar features have also been observed for erysotrine-N-oxide. The identity of both the N-oxide alkaloids was confirmed by treating the parent alkaloids with m-chloroperbenzoic acid. The resulting N-oxides were identical in all respects to the natural products. Earlier erysotrine-N-oxide and erythrartine-N-oxide have been isolated from E. mulungu flowers [4]; the investigators also emphasized that the N-oxides are in fact natural products and not artifacts.

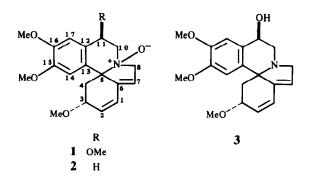
EXPERIMENTAL

Flowers of *E. xbidwillii* Lindley were collected from Chandigarh (India) and their authenticity was certified by Dr Rupert C. Barneby (Curator, New York Botanical Garden, U.S.A.). A voucher specimen is deposited in the herbarium, Department of Pharmaceutical Sciences, Panjab University, Chandigarh. The dried flowers were reduced to a moderately coarse powder before extraction.

Analysis of alkaloids. Alkaloids were extracted from a small sample of flowers (10 g) by the same method as used previously [7] and MS were determined with a GC via a two-stage Watson Biemann separator. The ion source was maintained at 220°, and the accelerating and ionising voltages were set at 3 and 70 eV, respectively.

Isolation of alkaloids. Prep. scale isolation of the petrol-sol alkaloid fr. (1.4 g) obtained from flowers (800 g) [7] on

H Deceased.



chromatography over neutral alumina yielded erysotrine (0.43 g) and another alkaloid fr., mp 159–160°, characterized as erythrartine (3, 0.34 g). $v_{max}^{CHC1_3}$ cm⁻¹: 3220 (O–H), 1600 (C=C). ¹H NMR (360 MHz, CDCl₃): Table 1. ¹³C NMR (CDCl₃): δ 131.4 (d, C-1), 125.5 (d, C-2), 75.8 (d, C-3), 40.7 (t, C-4), 66.2 (s, C-5), 142.0 (s, C-6), 123.7 (d, C-7), 58.8 (t, C-8), 50.8 (t, C-10), 64.5 (d, C-11), 129.3 (s, C-12), 129.5 (s, C-13), 108.1 (d, C-14), 146.2 (s, C-15)^a, 145.4 (s, C-16)^a, 113.1 (d, C-17), 57.5, 55.8 (q, OMe). ^aSignals of C-15 and C-16 may be reversed. MS see text. Found: C, 69.32; H, 7.00; N, 4.61. C₁₉H₂₃O₄N requires: C, 69.28; H, 7.04; N, 4.25.

The MeOH-sol. free alkaloid fr. (11.1 g) on repeated chromatography over neutral alumina afforded five alkaloids; three were identified as erysotrine (0.22 g), erythratine (3.1 g) and erysotramidine (0.38 g) by comparison with authentic samples. The other two were characterized as erythristemine-N-oxide (1, 0.06 g), and erysotrine-N-oxide (2, 0.09 g).

Erythristemine-N-oxide. $[\alpha]_{D}^{25}-3.8^{\circ}$ (CHCl₃; c0.01). UV (MeOH) λ_{max} nm (log ε) 255 (3.52), IR ν_{max}^{finax} cm⁻¹: 1610 (C=C). ¹H NMR: Table 1. MS m/z (rel. int.): 359 (3) [M]⁺, 343 (17) [M-16]⁺, 328 (9) [M-31]⁺, 313 (14) [M-46]⁺ and 312 (49) [M-47]⁺.

Preparation of erythristemine N-oxide. NaH (20 mg) was added in small portions to a cold, stirred soln of 3 (50 mg) (5°) in THF (10 ml) followed by addition of MeI ($\simeq 0.05$ ml) and the mixt. left at room temp. for 24 hr. MeOH was added to destroy excess NaH. The contents were then poured into ice-cold H₂O (50 ml), extracted with CHCl₃ (5 × 10 ml), washed, dried and solvent removed to leave an oily residue (30 mg). To the stirred soln (0°) of this residue in CHCl₃ (5 ml) was added *m*-chloroperbenzoic acid (40 mg). The reaction was carried out at room temp. for 4 hr. The mixt. was then poured into cold H₂O and processed as usual to give an oily residue. This was chromatographed over alumina (10 g). Elution with CHCl₃ gave the desired pure *N*oxide (20 mg) which was found to be identical with the natural product 1.

Erysotrine-N-oxide. IR v_{max}^{film} cm⁻¹: 1605 (C=C). ¹H NMR: Table 1. MS m/z (rel. int.): 329 (3) [M]⁺, 313 (48) [M-16]⁺, 311 (55) [M-18]⁺, 298 (52) [M-31]⁺, 296 (52) [M-33]⁺ and 282 (100) [M-47]⁺.

Preparation of erysotrine-N-oxide. m-Chloroperbenzoic acid (70 mg) was added to a stirred mixt. $(0-2^{\circ})$ of erysotrine (100 mg) in CHCl₃ (5 ml) and the mixt. left at room temp. for 4 hr. The CHCl₃ soln was evapd and the residue purified by prep. TLC in CHCl₃-MeOH (1:1), affording 48 mg of erysotrine-N-oxide (2) which was identical (IR, ¹H NMR) with the natural product 2. The liberated alkaloid fr. (0.9 g) on chromatographic resolution over neutral alumina yielded erythrartine (70 mg) and an intractable mixt. (10 mg). The remaining aq layer afforded hypaphorine as the H₂O-sol base.

	OMe	392 (s, OMe-16), 3.77 (s, OMe-15), 3.62 (s, OMe-11), 3.36 (s, OMe-3)	3.88 (s, OMe-16), 3.76 (s, OMe-15), 3.36 (s, OMe-3)	3.90 (s, OMe-16), 3.78 (s, OMe-15) 3.32 (s, OMe-3)					
Table 1. ¹ H NMR spectral data of <i>Erythrina</i> alkaloids (δ values: in CDCl ₃)	H-17	6.80 s	6.68 <i>s</i>	s 99.3		10e. 11e	2.5		2.5
	H-14	6.70 s	6.62 s	6.85 s		10a. 11e	、 		4.0
	H-11	4.48 <i>d</i>	3.66 m	4.70 t		10a. 10e	15.0	1	14.5
	H _e -10	3.84 <i>dd</i>	3.92 m	3.10 <i>dd</i>		8a, 8 <i>b</i>			2
	H ₁ -10	4.34 m	4.07 m	3.59 dd	its (Hz)			1	14.5
	8-H	5.10 d, 4.34 m	4.43 br s	3.97 d, 3.87 dd	Coupling constants (Hz)	7, 8α, 8 <i>β</i>	2.0		3.5
	Н-7	5.78 brs	5.80 brs	5.75 brs	Co	4a, 4e	11.0	11.0	11.0
	H _e -4	2.02 <i>dd</i>	2.14 dd	2.42 dd		3a, 4e	6.0	5.5	5.5
	H _a -4	3.23 t	3.24 t	1.81 <i>t</i>		3a, 4a	0	0	0
	Н-3	4.34 m	4.24 m	4.06 m				11.	11.
	H-2	6.68 <i>dd</i>	6.78 <i>dd</i>	6.61 <i>dd</i>		2, 3	2.5	2.5	2.5
	H-1	6.17 <i>d</i>	6.18 <i>d</i>	6.02 <i>d</i>		1, 2	10.0	10.0	10.0
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