## Synthesis of (±)-Nanaimoal and (±)-Nanaimool from N,N-Diethylgeranylamine

## Takashi YAMADA and Kunihiko TAKABE\*

Department of Applied Chemistry, Faculty of Engineering, Shizuoka University, Hamamatsu 432

(±)-Nanaimoal and (±)-nanaimool were synthesized via Diels-Alder Reaction of 1,1-dimethyl-2,3-dimethylenecyclohexane, a novel building block for cyclic terpenoids, easily prepared from N,N-diethylgeranylamine.

Cyclic sesquiterpene aldehydes such as polygodial<sup>1)</sup>and olepuane<sup>2)</sup> are fish antifeedants which are thought to act as chemical defense substances. In 1984 Andersen<sup>3)</sup> et al. have isolated Nanaimoal (1), a fragrant sesquiterpene aldehyde having a new carbon skeleton, from *Acanthodoris nanaimoensis*, and synthesized its derivative in a very low yield for the confirmation of the structure of 1. In our continuous studies<sup>4)</sup> on the synthesis of terpenoids using N,N-dialkyl geranylamine (3),5,6) we wish to report a simple synthetic method of 1 from 1,1-dimethyl-2,3-dimethylenecyclohexane (5), a novel building block for cyclic terpenoids, easily prepared from 3.

Cyclization<sup>7)</sup> of 3 with BF3·Et2O afforded a 81:19 mixture of N,N-diethyl  $\gamma$ - and  $\alpha$ -cyclogeranylamines (4) (87% yield). Oxidation of 4 with 30% H2O2, and subsequent thermal elimination gave selectively<sup>8)</sup>1,1-dimethyl-3,4-dimethylenecyclohexane (5) in 70% yield. Diels-Alder reaction of 5 with methyl methacrylate (MMA) at 170 °C for 20 h afforded adducts 6 in 91% isolated yield. NMR analysis of 6 (R=Me) showed slightly splitted two singlet peaks at  $\delta$  3.61 and 3.63 (-CO2CH3), suggesting that 6 is a

NEt<sub>2</sub>

$$\begin{array}{c}
X \\
1: X = CHO \\
2: X = CH2OH
\end{array}$$
NEt<sub>2</sub>

$$\begin{array}{c}
X \\
8 \\
7a
\end{array}$$
NEt<sub>2</sub>

$$\begin{array}{c}
CO_{2}R \\
7a
\end{array}$$
OTS

mixture of the regioisomers. However, the regioisomers could not be separated by chromatographic methods. Reduction of the mixture (6) with LiAlH4, followed by tosylation gave the tosylates (7) in 98% yield. Hplc analysis of 7 showed clearly separated two peaks, and the mixture was fractionated to 7a and 7b [ 7a: 7b = 56: 44].9,10)

The desired **7a** was treated with NaCN in dimethylsulfoxide (DMSO) at 110-115 °C for 3 h to give the nitrile **8**<sup>11</sup> in 85% yield, and the subsequent reduction of diisobutyl aluminum hydride (DIBALH) afforded (±)-nanaimoal (1) (80% yield), which was converted into (±)-nanaimool (2) by NaBH4 in a quantitative yield. <sup>12,13</sup>)

The authors would like to thank Professor T. Katagiri for his encouragement, and also thank Professor R. J. Andersen, University of British Columbia, for kindly sending the spectra of nanaimool and its derivative.

## References

- 1) M. D'Ischia, G. Prota, and G. Sodano, Tetrahedron Lett., 23, 3295 (1982).
- 2) R. K. Okuda, P. J. Scheuer, J. E. Hochlowski, R. P. Walker, and D. J. Falkner, J. Org. Chem., 48, 1866 (1983).
- 3) S. W. Ayer, J. Hellou, N. Tischer, and R. J. Andersen, *Tetrahedron Lett.*, **25**, 141 (1984); S. W. Ayer, R. J. Andersen, H. Chun-heng, and J. Clardy, *J. Org. Chem.*, **49**, 2654 (1984).
- 4) K. Takabe, T. Yamada, and T. Katagiri, Chem. Lett., 1982, 1981.
- 5) K. Takabe, T. Katagiri, J. Tanaka, T. Fujita, S. Watanabe, and K. Suga, Org. Synth., 67, 44 (1988).
- 6) S. Otsuka and K. Tani, Synthesis, 1991, 665; K. Takabe, The Koryo, 139, 41 (1983), and references cited therein.
- 7) K. Takabe, T. Yamada, T. Sato, and T. Katagiri, Nippon Kagaku Kaishi, 1980, 776.
- 8) The reaction products were 5 and  $\alpha$ -4, and 5 could be separated readily by ether extraction under acidic conditions (dil. HCl). The purity of 5 was more than 99% by GLC analysis (OV-101, 0.2 mm 25 m, He).
- 9) Hplc conditions: Shim-pack PREP-ODS, CH3CN:H2O = 88:12, 9.0 ml/min.
- 10) In order to increase the yield of the desired isomer (7a), Diels-Alder reactions of 5 with methacrylates were examined. The use of AlCl<sub>3</sub> as a catalyst in this reaction (25 °C, 6 h) decreased the ratio of the desired isomer [Yield of 6 (R=Me): 70%, 7a:7b=34:66], but the thermal reaction (170 °C, 17 h) of 5 with t-butyl methacrylate slightly increased the ratio of the desired isomer [Yield of 6 (R=t-Bu): 72%, 7a:7b=59:41].
- 11) IR: 2230 cm<sup>-1</sup>;  ${}^{1}$ H-NMR(CDCl<sub>3</sub>):  $\delta$  0.97(6H, s), 1.06(3H, s), 1.2-2.2(12H, m), 2.23(2H, s); HRMS Found: m/z 217.1820. Calcd for C<sub>15</sub>H<sub>23</sub>N: M, 217.1829.
- 12) The spectral data of **1** and **2** agreed with those of natural nanaimoal and nanaimool. **1**: IR: 1722 cm<sup>-1</sup>; <sup>1</sup>H-NMR(CDCl<sub>3</sub>): δ 0.97(6H, s), 1.04(3H, s), 1.3-2.2(12H, m), 2.25(2H, d, J=3.1Hz), 9.86(1H, t, J=3.1Hz); HRMS Found: m/z 220.1807. Calcd for C<sub>1</sub>5H<sub>2</sub>4O: M, 220.1825. **2**: IR: 3320, 1020 cm<sup>-1</sup>; <sup>1</sup>H-NMR(CDCl<sub>3</sub>): δ 0.88(3H, s), 0.99(6H, s), 1.1-2.1(15H, m), 3.73(2H, d, J=7.6Hz); HRMS Found: m/z 222.1985. Calcd for C<sub>1</sub>5H<sub>2</sub>6O: M, 222.1983.
- 13) Nanaimoal (1) could be also synthesized from myrcene (9) via the scheme described below.

(Received September 1, 1992)