## SYNTHESIS OF N-SUBSTITUTED 4-AMINO-2,3-PENTAMETHYLENEQUINOLINES

## É. S. Abramochkin and M. E. Konshin

4-Amino-2,3-pentamethylenequinolines are of interest as physiologically active compounds [1,2]. The present work is devoted to the synthesis of the N-substituted 4-amino-2,3-pentamethylenequinolines (V - XXIX) with the aim of testing their biological activity and of establishing a possible relationship between the activity and the structure. The synthesis was carried out by the interaction of 4-chloro-2,3-pentamethylenequinolines I-IV with primary or secondary amines in phenol medium:



The reaction was carried out with difficulty, requiring prolonged heating and involving, in the majority of cases, the formation of significant quantities of the corresponding 4-phenoxy-2,3-pentamethylenequinolines. To obtain analytically pure compounds it was necessary to resort to multiple crystallizations, which led to reduced yields. Compound I reacted more smoothly and at a lower temperature (140°C), whereas compounds II-IV only reacted at 170-175°C.

N-substituted 4-amino-2,3-pentamethylenequinolines (see Table 1) are colorless crystalline compounds having a basic character which react with mineral acids to form salts dissolved in water.

The UV spectra of V-XXII have three maxima: 232-236, 306-312, and 320-326 nm.\* On introduction of halogen or methyl groups at position 6 of 2,3-pentamethylenequinolines, there is observed a clearly defined bathochromic shift of the absorption band to longer wavelengths The nature of the amine entering position 4 has no effect on the position of the maxima.

Compounds V-XI, XIII, XIV, XVI-XXIX were subjected to biological tests.<sup>†</sup> The toxicity  $(LD_{50})$ , the analgesic activity (by the heat method of MacDonald and Wolf), and the anticurare activity (the ability to elimiate or reduce the myorelaxant effect of diplacinum) were investigated. The greatest toxicity (68-148 mg/kg) is possessed by XXIII and XXIX. Morpholino and arylpiperazino derivatives showed significantly lower toxicity (205-1050 mg/kg), decreasing in the following order: XVI > V, VI > XIII, XIV > XI > IX > X > VIII. Compounds VII, IX, XIII, XIV, and XVI raise the threshold of pain perceptibility 2-2 1/2 times; the analgesia is retained for 60-90 min. Weak anticurare activity was found in VI, VII, and XXIV.

## EXPERIMENTAL

<u>4-Chloro-6-bromo-2,3-pentamethylenequinoline</u> (IV). To 0.2 mole of phosphorus oxychloride was gradually added 0.05 mole of 4-hydroxy-6-bromo-2,3-pentamethylenequinoline. The mixture was heated on a water bath for 30 min and then poured into ice water; the solution formed after decomposition of the

\*UV spectra were run in alcohol on an SF-4 spectrophotometer.

†The tests were carried out by Prof. A. S. Zakos and medical student L. G. Zil'bermintsii, to whom the authors express gratitude.

Perm Pharmaceutical Institute. Translated from Khimiko-Farmatsevticheskii Zhurnal, Vol. 4, No. 7, pp. 10-13, July, 1970. Original article submitted July 17, 1969.

© 1971 Consultants Bureau, a division of Plenum Publishing Corporation, 227 West 17th Street, New York, N. Y. 10011. All rights reserved. This article cannot be reproduced for any purpose whatsoever without permission of the publisher. A copy of this article is available from the publisher for \$15.00.

Compound	R	N <sup>/R'</sup>	Yield, %	Мр, °С		Found	Empirical	Calculated
				base	hydro- chloride	N, %	formul a	N, %
V	Н	C6H5N.N	60,5	178—80	222-4	11,81	$C_{24}H_{27}N_3$	11,72
VI	Н	p – CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub> N N	49,7	146—7	221-2	[[,6]	$C_{25}H_{29}N_3$	11,32
VII	Н	o-CH3C6H4NN	47,4	139-40	215—7	11,59	C <sub>25</sub> H <sub>29</sub> N <sub>3</sub>	FT,32
VIII	CH <sub>3</sub>	C6H5N_11	58,6	1879	227—9	11,22	$C_{25}H_{20}N_3$	II, <b>3</b> 2
IX	CH3	p - CH3C6114N	48,3	l 49—50	262-4	11,08	$C_{26}H_{31}N_3$	10,92
Х	СН <sub>3</sub>	0 - CH3C6H4N_11	46,7	161-2	207—9	10,71	$C_{26}H_{31}N_3$	10,92
XI	СН <sub>з</sub>	ООИ	44,5	193—4	2524	9,49	$C_{13}H_{24}N_2O$	9,44
XII	CH3	CH3N N	43,8	115—6	2324	13,73	$C_{20}H_{27}N_3$	13,59
XIII	CI	C <sub>6</sub> H <sub>5</sub> N N	61,2	1889	2424	10,95	$C_{24}H_{26}CIN_3$	10,71
XIV	Cl	p – CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub> N	60,8	163—4	261—3	10,62	C <sub>25</sub> H <sub>28</sub> ClN <sub>3</sub>	10,34
XV	Cl	0 – CH3C6H4N	59,1	153—4	230—3	10,11	$C_{25}H_{28}ClN_{3}$	10,34
XVI	Cl	оДи	50,8	160—1	230—2	9,04	C18H21CIN2O	8,83
XVII	CI	CH3N N	50,ľ	130—1	265—7	12,65	$C_{19}H_{24}CIN_3$	[2,72
XVIII	Br	C6H5N	53,4	195—6	245—7	9,73	C <sub>24</sub> H <sub>26</sub> BrN <sub>3</sub>	9,63
XIX	Br	p – CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub> N	53,2	146—8	265—7	9,59	C <sub>25</sub> H <sub>28</sub> BrN <sub>3</sub>	9,33
XX	Br	0 - CH3C6H4N	50,8	136—7	223—5	9,18	C <sub>25</sub> H <sub>28</sub> BrN <sub>3</sub>	9,33
XXI	Br	о	47,4	1668	246—8	7,92	$C_{18}H_{21}BrN_2O$	7,75
XXII	Br	CH3N N	51,4	131—2	267—9	11,37	C <sub>19</sub> H <sub>24</sub> BrN <sub>3</sub>	[1,23
XXIII·HCl	н	C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> NH	57,7		24925	7,95	$C_{21}H_{22}N_2HCl$	8,27
X X IV HCl	н	n — C4H9NH	43,2	—	231-4	9,0	C18H24N2HCl	9,19
X X V · HCl	Н	cyclo-C <sub>6</sub> H <sub>11</sub> NH	36,1		260—1	8,10	C <sub>20</sub> H <sub>26</sub> N <sub>2</sub> HCl	8,48
XXVI HCl	н	(C2H5)2N	31,8	<sup>1</sup>	229—30	9,43	C <sub>18</sub> H <sub>24</sub> N <sub>2</sub> HCl	9,20
XXVII·HCl	CH3	C6H5CH2NH	39,7	_	255—6	8,27	$C_{22}H_{24}N_2HCl$	7,95
X VIII · HCl	CH3	n — C4H9NH	45,8		167 <b>—9</b>	9,13	$C_{19}H_{26}N_{2}HCl$	8,81
XXIX HCI	Cl	C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> NH	44,1	-	253—4	7,63	C <sub>21</sub> H <sub>21</sub> ClN <sub>2</sub> HCl	7,50

TABLE 1. N-Substituted 4-Amino-2,3-pentamethylenequinolines

reaction mixture was filtered. The filtrate was neutralized with ammonia and the precipitate which separated was filtered. Yield was 79%, mp 110-111°C (from acetone). Found %: N 4.59.  $C_{14}H_{13}BRC1N$ . Calculated %: N 4.50.

 $\frac{4-Chloro-6-methyl-2,3-pentamethylenequinoline (II)}{(from acetone). Found \%: N 6.00. C_{15}H_{16}ClN. Calculated \%: N 5.71.$ 

<u>N-Substituted 4-Amino-2,3-pentamethylenequinolines (V - XXIX).</u> To 0.01 mole of I-IV in 0.04 mole of molten phenol was added 0.02 mole of amine, and the mixture was heated at 140-175°C for 20-24 h. The reaction mixture was cooled, then ether and 10% caustic were added. The ether layer was separated and evaporated. The volatile components were removed from the residue by steam distillation, the residue was acidified with dilute hydrochloric acid and filtered, the filtrate was made alkaline. The precipitated base was crystallized from a suitable solvent.

## LITERATURE CITED

- 1. G. K. Patnaik, M. M. Vohra, I. Bindra, et al., J. Med. Chem., 9, 483 (1966).
- M. V. Sigal, B. I. Brent, and P. Marchini, U. S. Patent No. 3,232,945 (1966); Ref. Zhur. Khimiya, No. 4, No. 4H42OP (1968).