SESQUITERPENE LACTONES AND DITERPENOIDS FROM HELIANTHUS ARGOPHYLLUS*

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Abstract—Three germacranolide sesquitepene lactones (argophyllin-A and -B and eupatolide), three diterpenoids (ciliaric acid, $(-)-16-\alpha$ -hydroxy-kaur-11-en-19-oic acid and $(-)-16-\alpha$ -hydroxykaurane) and one flavonoid (nevadensin) were isolated and characterized from a chloroform extract of *Helianthus argophyllus*. Argophyllin-A and -B are both described here for the first time. Their structures were deduced by ¹H NMR and ¹³C NMR. Argophyllin-A and -B were found to show anti-auxin effects while eupatolide exhibited weak insecticidal activity.

INTRODUCTION

As a part of continuing biochemical systematic investigation of Helianthus (tribe Heliantheae) [1-4], we analysed a CHCl₃ extract of *H. argophyllus* T. and G. (the silverleaf sunflower), an annual species native to south Texas and Florida. This species is closely related to the commercially important H. annuus. In accord with previous work on the terpenoid constituents of this genus, which has been reviewed elsewhere [5] (see also refs. [6-8] for even more recent results), sesquiterpene lactones and diterpene carboxylic acids were found to be abundant secondary metabolites in this species. In this paper, we report the isolation and structural elucidation of two new sesquiterpene lactones, argophyllin-A and -B, and five previously described compounds including another germacranolide sesquiterpene lactone, eupatolide [9], a trachylobane-type diterpene carboxylic acid, ciliaric acid [10], two kauranoid diterpenes, (-) - 16 - α - hydroxykaur - 12 - en - 19 - oic acid [4], and $(-) - 16 - \alpha$ - hydroxykaurane [11], and a flavonoid, nevadensin [12], from H. argophyllus. Some biological activities of the isolated compounds are also described.

RESULTS AND DISCUSSION

Air-dried and ground leaves and stems of *H. argo-phyllus* were extracted with $CHCl_3$. The extract was purified by standard procedures [13] and then subjected to Si gel column chromatography with a toluene-EtOAc gradient solvent system initiated with toluene. The most abundant compound **4a**, eluted first (toluene-EtOAc, 5:1) and crystallized upon removal

of the solvent. The ¹H and ¹³C NMR and IR data indicated that it was eupatolide (8- β -hydroxycostunolide). The spectral data and mps for 4a and its acetate 4b were identical to those previously published [9].

Argophyllin-A (1a) was eluted with toluene-EtOAc (1:3). On evaporation of the solvent, 1a, $C_{20}H_{28}O_7$ (high resolution), was obtained and recrystallized from MeOH (mp 190-192°). The presence of an α methylene- γ -lactone was indicated by the IR (1755 1650 cm⁻¹) and the ¹H NMR (a pair of one-proton doublets at δ 6.33 and 5.74, J = 2.0, 2.5 Hz) spectra. A five carbon α,β -unsaturated ester side-chain attached to the skeleton of the sesquiterpene lactone was indicated by the IR spectra (1720 cm^{-1}) and the MS (m/z 83, base peak). The side-chain was shown to be an angelic ester by typical ¹H NMR signals at δ 6.09 [1H, q(br), $J = \sim 2$, 7.5 Hz), 1.95 [3H, d(br), $J = \sim 1$, 7.5 Hz] and 1.83 [3H, s(br), $J = \sim 1$, 2 Hz]. Two sharp three-proton singlets at δ 1.42 and 1.39 showed that the methyl groups of the basic skeleton, C-14 and C-15, both had geminal oxygen functions and no vicinal protons, indicating that positions 4 and 10 were both fully substituted. The presence of hydroxyl groups was clear from the IR spectrum (3400 cm^{-1}) .



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Extensive decoupling experiments were conducted to determine the location of these different functional groups. When a multiplet at δ 2.86 was irradiated, two doublets for H-13a and H-13b changed into sharp singlets, suggesting that this signal should be assigned to H-7; at the same time two one-proton multiplets at δ 5.20 and 4.78 were sharpened. In view of these chemical shifts and the decoupling data, the signal at δ 5.20 was assigned to a proton with a geminal acyloxy group and the signal at δ 4.78 was assigned to the proton attached to the carbon bearing the lactone oxygen function. At this point lactone 1a was con-sidered to be a C-6, C-7 *trans*-fused lactone, since almost all of the sesquiterpene lactones isolated from this genus so far are lactonized in this direction. The co-occurrence of eupatolide (4a) in this species also supported this assumption (which was later confirmed by X-ray analysis). Accordingly, the two multiplets at δ 5.20 and 4.78 were assigned to H-8 and H-6, respectively. The complexity of the H-6 signal requires a methylene group adjacent to C-6. On irradiation at δ 5.20 (H-8), double doublets at δ 2.81 and 1.38 became doublets (J = 15 Hz). Therefore, these two signals are assignable to H-9a and H-9b. Moreover, position 10 must be fully substituted in accord with the presence of a sharp three-proton singlet for the C-14 methyl group. The values of the coupling constants between H-9a, b and H-8 as well as between H-7 and H-8 (J = 4.0 Hz) indicated an α -orientation for H-8 [14].

The ¹³C NMR spectrum of 1a exhibited only four signals in the carbon-carbon double bond region, which accounted for the carbon-carbon double bonds in the angelic side-chain ester and the α -methylene- γ -lactone (Table 2). The above data suggested partial structure A, which accounted for six of the seven oxygen functions and six of the seven degrees of unsaturation. The nature of the remaining oxygen function in **1a** was established as follows. Lactone **1a** gave a monoacetate, 1b, on acetylation with Ac₂O-pyridine. This monoacetate still showed the presence of a hydroxyl group in its IR spectrum (3400 cm⁻¹). Since this hydroxyl group was not acetylated under the usual reaction conditions, it had to be tertiary and therefore was located at C-4 or C-10. In the monoacetate 1b, the double doublet at δ 3.73 (J = 10.0, 4.0 Hz) in 1a was shifted downfield to δ 5.07. This signal was spin-coupled to a two-proton multiplet at δ 2.10 in 1a, suggesting the presence of a secondary hydroxyl group at C-2 or C-3 with a methylene group at the other position. A C-1,10epoxide accounted for the last unsaturated function and was supported by ¹³C NMR signals at δ 60.6 (d) and 59.2 (s), which are typical for 1,10-epoxy ring carbons [14]. A ¹H NMR signal at δ 2.86 (m) also indicated the presence of an epoxy ring and was assigned to H-1. Since this signal overlapped those for H-7 and H-9a, the magnitude of the spin coupling between H-1 and H-2a, b could not be fully analysed making it difficult to ascertain the configurations at C-1 and C-10. (The ¹H NMR data for 2a, however, discussed below, strongly suggested a $1-\beta$, $10-\alpha$ orientation for the epoxide.) The complexity of the H-1 signal, even without exact information on the coupling constants, indicated an adjacent methylene group at C-2, thus placing the secondary hydroxyl at C-3. The tertiary alcohol must then be at C-4. These

data completed the structure of argophyllin-A (1a) except for the configuration of the 1,10-epoxy ring and the orientation of the C-3 and C-4 hydroxyl groups. An X-ray diffraction analysis carried out by V. Zabel and W. H. Watson indicated that 1a is a $1-\beta,10-\alpha$ -epoxygermacranolide and the C-3 hydroxyl group is on the β -side of the molecule and the C-4 hydroxyl group on the α -side. (The X-ray analysis will be the subject of a separate paper.)

Argophyllin-B (2a) proved to be an isomer of argophyllin-A, $C_{20}H_{28}O_7$ (high resolution MS), and its ¹H and ¹³C NMR spectra showed close relationships to those of 1a. A double doublet at δ 2.98 (J = 10.0, 5.0 Hz) was assigned to H-1, and corresponded to the signal at δ 2.86 (m) in 1a. The coupling constants between H-2a, b and H-1, which were very similar to those of heliangine (3), whose structure was established by an X-ray diffraction analysis [15, 16] indicated an α -orientation of H-1 and a β -orientation of the C-14 methyl group. Thus, the stereochemistry of 2a at C-1 and C-10 was equivalent to that of 1a. A one-proton multiplet appeared at δ 4.23, instead of the double doublet for H-3 at δ 3.73 in 1a, indicating that C-4 was not fully substituted. Further evidence for this was the lack of a signal corresponding to the sharp three-proton singlet observed at δ 1.39 in 1a. Instead, two double doublets appeared at δ 3.98 and 3.90, the AB part of an ABX system.

Lactone 2a gave the diacetate 2b on acetylation with Ac₂O-pyridine. The IR spectrum of 2b indicated the absence of a hydroxyl group, and in the 'H NMR spectrum the multiplet at δ 4.23 and the double doublets centered at δ 3.94 in 2a shifted downfield to δ 5.20 and 4.15, respectively. The former signal was assigned to H-3, while the latter were assigned to the two protons of a hydroxymethyl group attached to C-4. These data provided the structure of 2a, except for the configuration at C-4. Since the 'H NMR spectrum of 2a in CDCl₃ (90 MHz) was poorly resolved in the region of the signals for H-3 and H-4, it was difficult to analyse the stereochemistry at C-4. In a 200 MHz ¹H NMR spectrum of 2a in C_6D_6 , however, the signals were resolved well and the spin coupling constant between H-3 and H-4 appeared to be ca 7 Hz (see Table 1). The configuration for H-4 should therefore be α on the basis of the dihedral angle between H-3 and H-4 predicted by the Karplus equation. Hence the most probable structure for argophyllin-B must be as depicted in formula 2a. The structure of argophyllin-B is of rare occurrence, since it lacks any unsaturated function at C-4. Almost all the germacranolides found so far have either a C-C



	12	1b	(CDCI)	2a	2b	4a (Duriding d.)
			(CDCl ₃)	$(200 \text{ MHZ}, C_6 D_0)$	s)	(Pynume-a ₅)
H-1	2.86 m	*	2.98 dd	2.55 dd	3.27 dd	4.75 dd
2	2.10 m	*	*	*	*	*
3	3.73 dd†	5.07 dd	4.23 m	3.84 ddd	5.20 m	*
6	4.78 m	5.30 m	4.88 m	4.64 m	4.95 m	5.65 t
7	2.86 m	2.90 m	2.91 m	2.10 m	3.37 m	*
8	5.20 m	5.30 m	5.28 m	5.02 m	5.30 m	5.25
9a	2.81 dd	2.83 dd	2.75 dd	2.59 dd	2.60 dd	*
9b	1.38 dd	*	1.40 dd	*	*	*
13a	6.33 d	6.10 d	6.40 d	6.33 d	6.12 d	6.40 d
13b	5.84 d	5.08 d	5.80 d	5.75 d	5.10 d	5.87 d
14	1.42 s‡	1.46 d‡	1.47 s‡	1.22 s‡	1.55 s‡	1.65 s(br)‡
15	1.39 s‡	1.42 s‡	a 3.98 dd	3.60 dd	4.23 dd	$1.80 \ s(br)^{\ddagger}$
			b 3.90 dd	3.50 dd	4.07 dd	
18	$6.09 \ q(br)$	6.16 q(br)	$6.09 \ q(br)$	5.64 $q(br)$	6.10 q(br)	
19	1.95 d(br)‡	1.97 d(br)	1.95 d(br)‡	$1.86 \ d(br)$ ‡	1.95 d(br)	
20	1.83 m‡	1.86 m‡	1.86 m‡	1.75 m‡	1.85 m‡	
0		2.10 s‡			2.00 s‡	
11					2.05 s‡	
—Ü—Me					·	

Table 1. ¹H NMR data for compounds 1a-2b and 4a (90 MHz, CDCl₃, TMS as internal standard)

*Could not be observed because of overlap of signals.

*Coupling constants were virtually identical for 1a-2b. Those for 1a are given as representatives; values for those which are characteristic for 2a and 2b, are also given; J values in Hz for compound 1a: 2a,3 = 10.0: 2b,3 = 4.0; 7,8 = 4.0; 8,9a = 5.0; 8,9b = 3.0; 9a,b, = 15.0; 7,13a = 2.5; 7,13b = 2.0; 18,19 = 7.5; 18,20 = -2; 19,20 = -1; compound 2a: 1,2a = 10.0; 1,2b = 5.0; 3,4 = 7.0.

[‡]Three protons.

double bond or an oxygen function at this position. Argophyllin-B belongs to the rare class of C-15-oxygenated 4,5-dihydrogermacranolides represented by melnerines A, B [17] and 9-acetoxymelnerins A, B [18].

Four previously known compounds were also isolated from this plant: a kauranoid diterpene alcohol (5), a hydroxykaurenoic acid (6), a trachylobanetype diterpene acid, ciliaric acid (7) and the flavonoid nevadensin (8). The spectral data and mps of these compounds were identical to those previously reported.

A number of reports have been published concerning the anti-auxin activity of heliangine (3), isolated from *H. tuberosus* [19, 20]. Since 1a and 2a are closely related to 3, the inhibitory effect of 1a and 2b on the IAA-induced elongation of Azuki (Azukia angularis) hypocotyl sections was examined (Table 3). The potencies of 1a and 2a were almost comparable to that reported for 3 [21]. The biochemical or ecological significance of this kind of activity, however, is currently unclear.

Observation of a number of wild *Helianthus* species has indicated that they are resistant to some of the serious insect pests that feed on the cultivated varieties [22, 23]. Since it is thought that chemical factors might be important in this resistance, and several diterpene carboxylic acids from *Helianthus* have been previously demonstrated to inhibit insect growth [22], we tested two of the isolated compounds, a diterpene carboxylic acid and a sesquiterpene lactone, for insecticidal activity. Eupatolide (4a), showed weak activity against tobacco cutworm larvae

Carbon	1a	2a	4a (pyridine-d ₅)
1‡	60.6 d§	60.0 d	124.0 d
2	31.0 t	36.6 t*	26.5 t
3	74.8 d	70.8 d	39.6 t
4	74.2 s	43.9 d†	141.8 s
5	42.5 t	36.9 t*	123.8 d
6	78.1 d	74.9 d	75.7 d
7	50.5 d	49.1 d	54.3 d
8	77.5 d	75.3 d	71.4 d
9	49.2 t	43.3 t†	48.4 t
10	59.2 s	58.1 s	140.3 s
11	136.6 s	137.1 s	136.8 s
12	169.7 s	169.1 s	170.9 s
13	125.1 t	124.9 t	120.0 <i>t</i>
14	19.7 q	19.5 q	19.9 q
15	18.0 q	66.0 t	17.3 q
16	166.3 s	166.4 s	-
17	126.6 s	127.0 s	
18	141.0 <i>d</i>	140.2 d	
19	15.8 q	15.7 q	
20	20.3 q	20.2 q	

Table 2. ¹³C NMR data for compounds 1a, 2a and 4a (22.6 MHz, CDCl₃)

*,†Assignments are interchangeable.

‡Signals were assigned by means of partially decoupled off-resonance spectra and comparison with those reported for heliangine **3** and eupalinin-A [25].

§Indicates multiplicities on partially decoupled spectra.

Table 3. Inhibitory effect of argophyllin-A and -B on the IAA-induced elongation of (A. angularis) hypocotyl sections. The concentration of 1a and 2a was 10^{-4} M (38 ppm)

Conc of IAA	Average elongation of (mm) of 12 sections					
(M)	Without 1a or 2a	With 1a	With 2a			
0	13.0	12.0	11.9			
10-5	15.5	12.6	12.2			
10-4	18.2	12.0	12.1			



3 Heliangine







5 (-)-16- α -Hydroxykaurane







(Spodoptera litura, 63% kill after 48 hr, 2000 ppm in an artificial diet) and mosquito larvae (*Culex pipens*, 60% kill after 24 hr in 10 ppm solution). Ciliaric acid (7), however, did not show either of these activities. Eupatolide could thus be involved in the natural resistance of *H. argophyllus* to insect predation.

EXPERIMENTAL

¹H and ¹³C NMR spectra were measured at 90 and 22.6 MHz, respectively, with TMS as an int. standard, except for one ¹H NMR spectrum of argophyllin-B which was measured at 200 MHz. Si gel 60 (Merck, 70–230 mesh) was used for column chromatography separations. Analyt. TLC and prep. TLC (0.5 and 2.0 mm) were done on precoated Si gel 60 GF₂₅₄. MS were recorded by direct inlet at 70 eV ionization.

Extraction and isolation. Air-dried and ground leaves and stems (1.63 kg) of H. argophyllus (collected along Texas Highway 80, 7.4 miles south of U.S. Highway 90, along the edges of a cultivated field, Gonzalez Co., TX, 25 Aug. 1979, voucher deposited in the Herbarium at the University of Texas at Austin), were extracted with CHCl₃ ($\times 2$, 4.0 and 6.01.) at room temp. The extract was filtered and concd in vacuo to give 256 g of a dark green syrup. The syrup was purified by standard procedures [13] to give 88.0 g of yellow syrup which was charged on the top of a Si gel column (1.76 kg). The column was eluted with a toluene-EtOAc gradient solvent system initiated with toluene; 11. of eluent was collected for each fraction. All the fractions were monitored by TLC with a toluene-EtOAc, solvent system. Fractions 23 and 24 (eluted with toluene-EtOAc, 15:1) yielded 0.4 g (-) - 16 - α - hydroxykaurane (5) as white needles. Fractions 51-55, which were eluted with toluene-EtOAc (5:1), gave 1.5 g of crystalline eupatolide (4a) (0.09% yield from the dried plant). Colourless prisms were obtained by recrystallization from MeOH. Fractions 61-63 yielded 30 mg of yellow needles of nevadensin (8). Fraction 64 yielded 20 mg of crystalline (-) - 16 - α - hydroxy - kaur -11 - en - 19 - oic acid (6). Fractions 65-75 (eluted with toluene-EtOAc, 3: 1-1: 1) gave 2.0 g of white needles of ciliaric acid (7). Fractions 96-102 (eluted with toluene-EtOAc, 1:3) showed several spots of TLC and were combined; the material (9.0 g) from these fractions was purified on a smaller Si gel column (300 g) eluted with CHCl₃-Me₂CO (1:1). The material from fractions 77-89 (1.30 g) was purified $\times 2$ by prep. TLC (CHCl₃-Me₂CO, 1:2, 2 mm; CHCl₃-iso-PrOH, 10:1, 0.5 mm) to give 400 mg of crystalline argophyllin-A (1a). Colourless prisms were obtained by recrystallization from MeOH. These crystals were used for X-ray diffraction analysis. Fractions 110–114 (eluted with EtOAc) gave 4.5 g yellow syrup. The syrup was purified through a smaller Si gel column (250 g) which was developed with CHCl₃-MeOH (10:1). Fractions 38–43 were combined and gave 300 mg syrup which showed one main spot. The material from these fractions was purified by prep. TLC (2 mm; CHCl₃-iso-PrOH, 10:1 and 0.5 mm; CHCl₃-iso-PrOH, 8:1) to give 250 mg argophyllin-B (2a) as white needles.

Argophyllin-A (1a). Mp 190–192° (MeOH), $[\alpha]_{D}^{22} - 157.0°$ (CHCl₃; c 0.10) C₂₀H₂₈O₇, for M⁺ found m/z 380.1826 calc. m/z 380.1835, MS m/z (rel. int.): 380 (1), 362 (0.5), 178 (11), 95 (12), 83 (100), 55 (64), 43 (45); IR $\nu_{\rm Max}^{\rm Nujol}$ cm⁻¹: 3400, 3100, 1755, 1720, 1680, 1650, 1230, 1150, 1040, 1000.

Argophyllin-B (2a). Mp 63–69° (CHCl₃), $[\alpha]_{D}^{22} - 137.5°$ (CHCl₃; c 0.14) C₂₀H₂₈O₇, for M⁺ found m/z 380.1837 calc. 380.1835; MS m/z (rel. int.): 380 (0.3), 362 (0.7), 336 (0.7), 305 (1), 83 (100), 55 (75), 43 (32). IR ν_{max}^{Nujol} cm⁻¹: 3400, 1760, 1725, 1650, 1220, 1140, 1040.

Eupatolide (4a). Mp 185–188° (MeOH) (lit. 182–188° [9]), IR and ¹H NMR spectra and mp were identical to the previously published data. 4b, which was prepared with Ac_2O -pyridine, exhibited IR and ¹H NMR spectra identical to those of an authentic specimen.

(-)-16- α -Hydroxykaurane (5). Mp 211-213° (lit. 212-214°), $[\alpha]_{D}^{22}$ -40.0° (CHCl₃; c 0.10) (lit. -45.0°). The IR and NMR were identical to those previously reported [11].

(-)-16- α -Hydroxy-kaur-11-en-19-oic acid (6). Mp 242–244° (lit. 244–246° [4]). The IR and NMR were identical to those of an authentic specimen.

Ciliaric acid (7). Mp (methyl ester) $132-133^{\circ}$ (lit. $136-137^{\circ}$ [10]). IR and ¹H NMR spectra were identical to those of an authentic specimen. The methyl ester which was prepared by the usual method (CH₂N₂) exhibited IR and ¹H NMR spectra identical to those of previously reported [10].

Nevadensin (8). Mp $190-192^{\circ}$ (lit. $193-195^{\circ}$ [12]). ¹H NMR, UV data and mp were identical to the published data.

Acetylation of argophyllin-A (1a). 30 mg 1a was acetylated with Ac₂O (1 ml) and pyridine (0.5 ml) for 12 hr at room temp. After the usual work-up, the crude product was purified by prep. TLC (EtOAc) to give 12 mg of the monoacetate 1b, colourless oil, $[\alpha]_{D}^{22} - 79.8^{\circ}$ (CHCl₃; c 0.14) MS m/z (rel. int.): 422 (0.3), 404 (0.7), 374 (0.7), 83 (100), 55 (57), 43 (90). IR $\nu_{max}^{CHCl_3}$ cm⁻¹: 3450, 2940, 1760, 1730, 1720, 1680, 1640, 1230 and 1140.

Acetylation of argophyllin-B (2a). 56 mg 2a was acetylated with Ac₂O (2 ml) and pyridine (1 ml) for 12 hr at room temp. After prep. TLC (EtOAc) of the crude product, 47 mg of the diacetate 2b was obtained as colourless oil, $[\alpha]_{22}^{22}$ – 70.2° (CHCl₃; c 0.05) MS m/z: 464 (0.5), 422 (0.4), 245 (12), 83 (100), 55 (58), 43 (66); IR $\nu_{mc}^{CHCl_3}$ cm⁻¹: 2940, 1760, 1740, 1720, 1700, 1690, 1650, 1230, 1140, 1040.

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REFERENCES

- 1. Ohno, N. and Mabry, T. J. (1979) Phytochemistry 18, 1003.
- 2. Ohno, N., Mabry, T. J., Zabel, V. and Watson, W. H. (1979) Phytochemistry 18, 1687.
- 3. Ohno, N. and Mabry, T. J. (1980) Phytochemistry 19, 609.
- 4. Ohno, N., Gershenzon, J., Neuman, P. and Mabry, T. J. (1981) Phytochemistry 20, 2393.
- 5. Gershenzon, J., Ohno, N. and Mabry, T. J. (1981) Rev. Latinoam. Quim. 12, 53.
- Bohlmann, F., Jakupovic, J., King, R. M. and Robinson, H. (1980) Phytochemistry 19, 863.
- 7. Herz, W. and Kumar, N. (1981) Phytochemistry 20, 93.
- 8. Herz, W. and Kumar, N. (1981) Phytochemistry 20, 99.
- 9. Dolejs, L. and Herout, V. (1962) Collect. Czech. Chem. Commun. 27, 2654.
- 10. Bjeldanes, L. F. and Geissman, T. A. (1972) Phytochemistry 11, 427.
- 11. Cross, B. E., Galt, R. H. B., Hanson, J. R., Curtis, P. J., Grove, J. F. and Morrison, A. (1963) J. Chem. Soc. 2937.
- 12. Farkas, L., Nogracli, M., Sudarsanam, V. and Herz, W. (1966) J. Org. Chem. 31, 3228.
- 13. Mabry, T. J., Miller, H. E., Kagan, H. B. and Renold, W. (1966) *Tetrahedron* 22, 1139.
- 14. Doskotch, R. W. and El-Feraly, F. A. (1970) J. Org. Chem. 35, 1928.
- Nishikawa, M., Kamiya, K., Takabatake, A., Oshio, H., Tomiie, Y. and Nitta, I. (1966) *Tetrahedron* 22, 3601.
- 16. Neidle, S., and Rogers, D. (1972) J. Chem. Soc. Chem. Commun. 140.
- Watkins, S. F., Korp, J. D., Bernal, I., Perry, D. L., Bhacca, N. S. and Fischer, N. H. (1978) J. Chem. Soc. Perkin Trans. 2, 599.
- Olivier, E. J., Perry, D. L., and Fischer, N. H. (1980) J. Org. Chem. 45, 4028.
- Iriuchijima, S., Kuyama, S., Takahashi, N. and Tamura, S. (1966) Agric. Biol. Chem. 30, 1152.
- Morimoto, H., Sanno, Y. and Oshio, H. (1966) Tetrahedron 22, 3173.
- 21. Shibaoka, H. (1961) Plant Cell Physiol. 2, 175.
- 22. Rogers, C. E. and Thompson, T. E. (1978) J. Econ. Entomol. 71, 221.
- 23. Rogers, C. E. and Thompson, T. E. (1978) J. Econ. Entomol. 71, 622.
- Elliger, C. A., Zinkel, D. R., Chan, B. G. and Waiss, A. C., Jr. (1976) *Experientia* 32, 1364.
- 25. Ito, K., Sakakibara, Y., Haruna, M. and Lee, K. H. (1979) Chem. Letters 1469.