

## SESQUITERPENE LACTONES FROM *CARPESIUM ABROTANOIDES*

MASAO MARUYAMA, AKIO KARUBE\* and KIYOKO SATO†

Department of Chemistry, Miyagi University of Education, Sendai, Japan 980; \*Department of Chemistry, Akita Technical College, Akita, Japan 011; †Department of Chemical Engineering, Ichinoseki Technical College, Ichinoseki, Japan 021

(Received 3 May 1983)

**Key Word Index**—*Carpesium abrotanoides*; Compositae; carabrol; ivaxillin; 11(13)-dehydroivaxillin; eriolin.

**Abstract**—Carabrol, ivaxillin and 11(13)-dehydroivaxillin were obtained from *Carpesium abrotanoides*, and the structures of carabrol and 11(13)-dehydroivaxillin were elucidated. The structure of eriolin was also elucidated by interconversion from 11(13)-dehydroivaxillin.

### INTRODUCTION

In the course of a continuing search for antifungal and antibacterial material from *Carpesium abrotanoides* L. we have reported the isolation and stereochemistry of granilin [1], the isolation of ivalin [2] and carabrone [3], and the isolation and structure of carpesiolin [3]. This paper reports the isolation of ivaxillin (1), the isolation and structure elucidation of carabrol (2) and 11(13)-dehydroivaxillin (3), and the structure of eriolin (4).

### RESULTS AND DISCUSSION

One of the active fractions obtained from the chromatography of the active extract afforded a non-crystalline, viscous substance, carabrol (2), bp 110° (bath temperature)/0.2 mm Hg,  $[\alpha]_D^{25} + 74.9^\circ$ ,  $C_{15}H_{22}O_3$ . The  $^1H$  NMR spectrum of 2 was similar to that of carabrone (5) [4]; their only difference was that a methyl singlet at  $\delta 2.13$  in the spectrum of 5 was replaced by a 3H doublet at 1.17, and a 1H multiplet was present at  $\delta 3.7$  in the spectrum of 2 which shifted to 5.3 in the spectrum of its 3,5-dinitrobenzoate (6), mp 123–126°,  $C_{22}H_{24}O_8N_2$ . The IR spectrum of 2 lacked a carbonyl absorption which was present at  $1714\text{ cm}^{-1}$  in 5, and instead showed a hydroxyl absorption at  $3610\text{ cm}^{-1}$ . These facts suggested that 2 had a secondary hydroxyl group in place of the carbonyl group of 5. This was proved by oxidation of 2 which afforded 5.

This fraction also yielded an inactive crystalline substance, mp 177–178°,  $[\alpha]_D^{25} - 130^\circ$ ,  $C_{15}H_{22}O_4$ . The  $^1H$  NMR and  $^{13}C$  NMR spectral data of this substance were very similar to those of ivaxillin (1) [5], and their IR spectra were superimposable. Therefore, this substance was identified as compound 1.

Another active fraction afforded an antibacterial crystalline compound 3, mp 166–167°,  $[\alpha]_D^{25} - 57.0^\circ$ ,  $C_{15}H_{20}O_4$ . Compound 3 had an  $\alpha$ -methylene- $\gamma$ -lactone [ $\nu_{\max}$   $1770\text{ cm}^{-1}$ ;  $\lambda_{\max}$  214 nm ( $\epsilon$  17600);  $\delta$  5.71 ( $d$ ,  $J = 3\text{ Hz}$ ) and 6.36 ( $d$ ,  $J = 3\text{ Hz}$ )] instead of the secondary methyl group ( $\delta$  1.25,  $d$ ,  $J = 9\text{ Hz}$ ) and  $\gamma$ -lactone ( $\nu_{\max}$   $1779\text{ cm}^{-1}$ ) found in compound 1. The rest of its  $^1H$  NMR spectrum was very similar to that of compound 1. Hydrogenation of 3 over platinum catalyst afforded two products, one of which was identified as compound 1. Therefore, compound 3 was established as 11(13)-dehydroivaxillin. The other product 4, mp 218–220°,  $[\alpha]_D^{25} - 46.5^\circ$ ,  $C_{15}H_{22}O_4$ , was a C-11 epimer of ivaxillin.

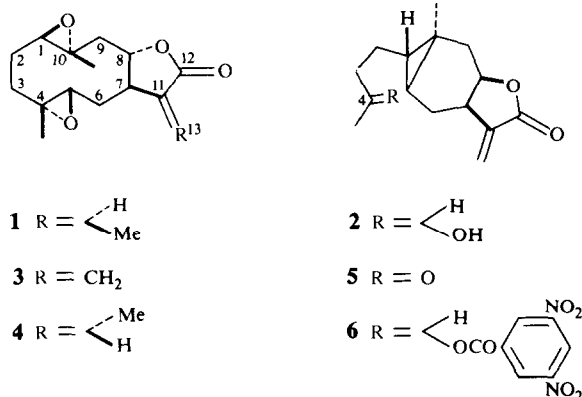
Eriolin [6] has been reported to have the same planar structure as compounds 1 and 4, and Herz and his co-workers [5] have mentioned that eriolin may be the C-11 epimer of compound 1. Although neither a sample nor the IR spectrum of eriolin was obtained and compound 4 and eriolin were not compared directly, their  $^1H$  NMR spectra [7] were superimposable. Therefore, compound 4 was identical with eriolin and its structure was determined as the C-11 epimer of compound 1.

Furthermore, the diepoxygermacranolide obtained by Romo de Vivar from *Schkuhria pinnata* and mentioned by Herz [5] is identical with compound 3.

### EXPERIMENTAL

Mps are uncorr.  $^1H$  NMR spectra were measured in  $CDCl_3$  with TMS as int. standard.

**Isolation of carabrol (2) and ivaxillin (1).** Fraction G (4.98 g) obtained from the silica gel CC of the partitioned extract of *C. abrotanoides* [1] was repeatedly chromatographed over silica gel in  $C_6H_6$ -EtOAc (5:1) followed by  $CHCl_3$  to give an oily fraction (1.10 g) and a crystalline fraction (0.10 g). The oily fraction was chromatographically homogeneous and a small portion was distilled to give a viscous, colourless oil, carabrol (2), bp 110° (bath temp.)/0.2 mm Hg;  $[\alpha]_D^{18} + 74.9^\circ$ ,  $c$  0.41 ( $CHCl_3$ ). (Found: C, 71.35; H, 8.94. Calc. for  $C_{15}H_{22}O_3$ : C, 71.97; H, 8.86%). MS:



$m/z$  232  $[M - H_2O]^+$ ; IR  $\nu_{\max}^{\text{CHCl}_3}$   $\text{cm}^{-1}$ : 3610, 1755, 1655; UV  $\lambda_{\max}^{\text{MeOH}}$  nm ( $\epsilon$ ): 213 (8300);  $^1\text{H}$  NMR (60 MHz):  $\delta$  0.4 (2H, *m*, H-1 and H-5), 1.03 (3H, *s*, C-10 Me), 1.17 (3H, *d*,  $J = 3$  Hz, C-4 Me), 3.2 (1H, *m*, H-7), 3.7 (1H, *m*, H-4), 4.72 (1H, *ddd*,  $J = 12, 8$  and 6 Hz, H-8), 5.47 (1H, *d*,  $J = 3$  Hz, H-13), 6.12 (1H, *d*,  $J = 3$  Hz, H-13). Carabrol (**2**) is unstable and deteriorates on storage, and the yield from the distillation was poor. The crystalline fraction was recrystallized from EtOH to give colourless prisms of ivaxilline (**1**, 0.03 g), mp 177–178°;  $[\alpha]_{\text{D}}^{18} - 130^\circ$ ,  $c$  0.28 (CHCl<sub>3</sub>). (Found: C, 67.34; H, 8.30. Calc. for C<sub>15</sub>H<sub>22</sub>O<sub>4</sub>: C, 67.64; H, 8.33 %.) Compound **1** was identified by comparison of its IR spectrum with that of an authentic sample.

**Isolation of 11(13)-dehydroivaxillin (3).** Fraction E (0.79 g) from the chromatography mentioned above was repeatedly chromatographed over silica gel in C<sub>6</sub>H<sub>6</sub>–EtOAc (5:1) followed by recrystallization from MeOH to give colourless needles of 11(13)-dehydroivaxillin (**3**, 0.088 g), mp 166–167°;  $[\alpha]_{\text{D}}^{18} - 57.0^\circ$ ,  $c$  0.33 (CHCl<sub>3</sub>). (Found: C, 67.96; H, 7.60. Calc. for C<sub>15</sub>H<sub>20</sub>O<sub>4</sub>: C, 68.16; H, 7.63 %.) IR  $\nu_{\max}^{\text{KBr}}$   $\text{cm}^{-1}$ : 1770, 1665; UV  $\lambda_{\max}^{\text{MeOH}}$  nm ( $\epsilon$ ): 214 (17 600);  $^1\text{H}$  NMR (100 MHz):  $\delta$  1.31 (3H, *s*, C-4 or C-10 Me), 1.47 (3H, *s*, C-4 or C-10 Me), 2.9 (3H, *m*, H-1, H-5 and H-7), 4.23 (1H, *ddd*,  $J = 10, 7$  and 1 Hz), 5.71 (1H, *d*,  $J = 3$  Hz), 6.36 (1H, *d*,  $J = 3$  Hz);  $^{13}\text{C}$  NMR (25 MHz):  $\delta$  15.95 (C-4 Me), 17.95 (C-10 Me), 23.60 (C-2), 32.09 (C-6), 35.91 (C-3), 45.86 (C-9), 46.35 (C-7), 57.75 (C-10), 60.73 (C-4), 63.40 and 64.37 (C-1 and C-5), 82.14 (C-8), 121.88 (C-13), 139.17 (C-11), 168.66 (C-12).

**Preparation of carabrol 3,5-dinitrobenzoate (6).** Treatment of **2** (200 mg) in pyridine (3 ml) with 3,5-dinitrobenzoyl chloride (400 mg) followed by recrystallization of the product from EtOH afforded the 3,5-dinitrobenzoate (**6**, 87 mg) as yellow prisms, mp 123–126°. (Found: C, 59.44; H, 5.49; N, 6.20. Calc. for C<sub>22</sub>H<sub>24</sub>O<sub>8</sub>N<sub>2</sub>: C, 59.45; H, 5.44; N, 6.30 %.) IR  $\nu_{\max}^{\text{KBr}}$   $\text{cm}^{-1}$ : 1750, 1730, 1625, 1550, 1350; UV  $\lambda_{\max}^{\text{MeOH}}$  nm ( $\epsilon$ ): 211 (32 000);  $^1\text{H}$  NMR (60 MHz):  $\delta$  0.35 (2H, *m*, H-1 and H-5), 1.10 (3H, *s*, C-10 Me), 1.40 (3H, *d*,  $J = 7$  Hz, C-4 Me), 3.2 (1H, *m*, H-7), 4.74 (1H, *m*, H-8), 5.3 (1H, *m*, H-4), 5.55 (1H, *d*,  $J = 3$  Hz, H-13), 6.23 (1H, *d*,  $J = 3$  Hz, H-13), 9.17 (3H, *br s*, aromatic).

**Oxidation of carabrol (2).** To a soln of **2** (200 mg) in DMF (16 ml) was added CrO<sub>3</sub> (250 mg) and H<sub>2</sub>SO<sub>4</sub> (3 drops). The reaction mixture was kept overnight at room temp., then poured on ice H<sub>2</sub>O. The product was extracted with Et<sub>2</sub>O and usual work-up of the extract afforded a crystalline mass which was

recrystallized from Et<sub>2</sub>O–petrol to give carabrone (**5**, 78 mg), which was characterized by mmp, co-TLC and IR comparison with an authentic sample.

**Hydrogenation of 3.** Compound **3** (106 mg) in EtOH (28 ml) was hydrogenated over Pt catalyst prepared from PtO<sub>2</sub> (50 mg) to absorb 9.55 ml (19°, 745 mm Hg) of H<sub>2</sub>. After removal of the catalyst, the soln was evapd *in vacuo* to dryness to give 105 mg of crystals which showed two spots on TLC (silica gel, Et<sub>2</sub>O). Fractional recrystallization of the product from EtOH gave ivaxillin (**1**, 20 mg), mp 165–168°, which was characterized by mmp, co-TLC and IR comparison with an authentic sample, and the epimer (**4**, 37 mg), mp 218–220°,  $[\alpha]_{\text{D}}^{18} - 46.5^\circ$ ,  $c$  0.2 (CHCl<sub>3</sub>). (Found: C, 67.67; H, 8.34. Calc. for C<sub>15</sub>H<sub>22</sub>O<sub>4</sub>: C, 67.64; H, 8.33 %.) IR  $\nu_{\max}^{\text{KBr}}$   $\text{cm}^{-1}$ : 1763;  $^1\text{H}$  NMR (60 MHz):  $\delta$  1.27 (3H, *s*, C-4 Me), 1.27 (3H, *d*,  $J = 7$  Hz, C-11 Me), 1.38 (3H, *s*, C-10 Me), 4.1 (1H, *br t*, H-8), which was identified as erioliol by comparison of its  $^1\text{H}$  NMR spectra.

**Acknowledgements**—We wish to thank Professor W. Herz of The Florida State University for the IR spectrum of ivaxillin, Mitsui Toatsu Chemicals Inc. for biological assays, Dr. J. Ishiyama of Tohoku University for  $^{13}\text{C}$  NMR measurements, the Micro-analysis Laboratory of Tohoku University for elemental analyses, and Mr. T. Sato of Akita Technical college for plant collection and extraction.

## REFERENCES

1. Maruyama, M. and Shibata, F. (1975) *Phytochemistry* **14**, 2247.
2. Maruyama, M. and Karube, A. (1976) *Phytochemistry* **15**, 2026.
3. Maruyama, M. and Omura, S. (1977) *Phytochemistry* **16**, 782.
4. Yoshioka, H., Mabry, T. J. and Timmerman, B. N. (1973) *Sesquiterpene Lactones*, p. 298. University of Tokyo Press.
5. Herz, W., Prasad, S. and Blount, J. F. (1982) *J. Org. Chem.* **47**, 3991.
6. Torrance, S. J., Geissman, T. A. and Chedekel, M. R. (1969) *Phytochemistry* **8**, 2381.
7. Yoshioka, H., Mabry, T. J. and Timmerman, B. N. (1973) *Sesquiterpene Lactones*, p. 187. University of Tokyo Press.