

4,17a-Diaza-D-homoandrostane Lactams,¹ a New Hetero-steroid System

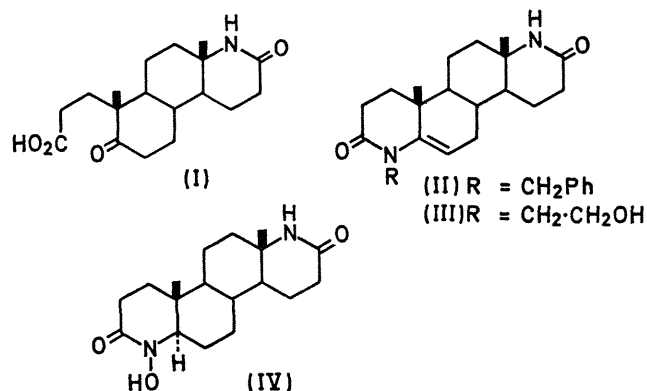
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Summary Compounds related to a new 4,17a-diaza-D-homosteroid system have been prepared by reactions with 17a-aza-5,17-dioxo-D-homo-3,5-seco-4-norandrostane-3-oic acid.

IN continuation of our work on the synthesis of new hetero-steroid systems the analogues of which may be of medicinal significance, we report a new diaza-D-homo-skeleton achieved through the preparation of 4,17a-diaza-D-homoandrostane lactams.

17a-Aza-D-homoandrost-4-en-3-one,² on oxidation with



the periodate-permanganate reagent of Lemieux and von Rudloff^{3,4} gave the seco-keto-acid (I), m.p. 266—267° (77% yield), C₁₈H₂₇NO₄, ν_{\max} (KBr) 3320, 2970, 2930, 2860, 1700, and 1610 cm⁻¹. The compound (I) was also characterized by preparation of its oxime by heating with hydroxylamine hydrochloride and pyridine; m.p. 247—248° (84% yield), ν_{\max} (KBr) 3270 (broad), 1680, and 1610 cm⁻¹.

The seco-keto-acid (I), on heating under reflux with benzylamine, afforded a compound, m.p. 278—280° (62% yield), $[\alpha]_D^{20}$ -127.2° (*c* 1.05, CHCl₃), analysing for C₂₅H₃₂N₂O₂, which was assigned the structure 4-benzyl-4,17a-diaza-D-homoandrost-5-ene-3,17-dione (II) on the basis of the spectral data, λ_{\max} (MeOH) 234 nm (ϵ 10,110) indicative of an enamine system, ν_{\max} (KBr) 1655 and 1625 cm⁻¹. Similarly, the dilactam (III) was prepared by treatment of (I) with ethanolamine, m.p. 238—240° (21% yield), $[\alpha]_D^{20}$ -109.3° (*c* 1.16, CHCl₃), λ_{\max} (MeOH) 235 nm (ϵ 10,200), ν_{\max} (KBr) 3335 and 1645 cm⁻¹, and showing satisfactory elemental composition.

Reduction of the oxime of (I) with zinc and acetic acid gave the hydroxamic acid (IV), m.p. >300° (70% yield), ν_{\max} (KBr) 3310, 3230, 3175, and 1640 (broad) cm⁻¹, giving with methanolic ferric chloride an immediate pink-violet colour, turning to green. The structure (IV) was assigned by analogy with products of similar reductions.⁵

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¹ For previous paper, see H. Singh and S. Padmanabhan, *Indian J. Chem.*, 1969, **7**, 1084.

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³ R. U. Lemieux and E. von Rudloff, *Canad. J. Chem.*, 1955, **33**, 1701 and 1710.

⁴ E. von Rudloff, *Canad. J. Chem.*, 1955, **33**, 1714; 1956, **34**, 1413.

⁵ J. T. Edward and P. F. Morand, *Canad. J. Chem.*, 1960, **38**, 1316.