Stereospecific *cis*-Reduction of Hexahydro-3-oxophenanthrene Derivatives. Total Synthesis of a New Class of Ring-C Aromatic Steroid

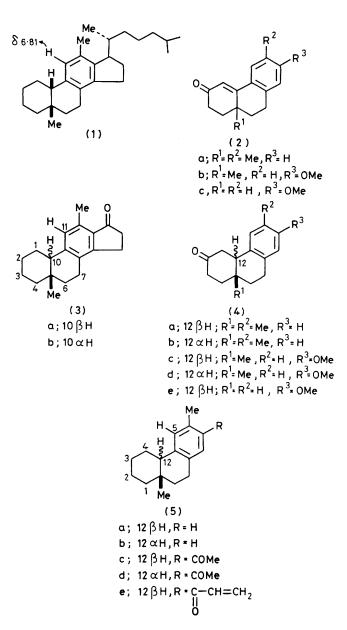
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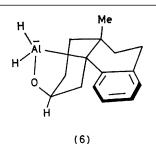
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Catalytic hydrogenation (Pd/C, 10%) of the hexahydro-3-oxophenanthrene (2a) in acidic medium gave the *trans*-fused octahydrophenanthrene ring system (5b) predominantly, whereas reduction of (2a—c) with LiAlH₄ followed by oxidation gave exclusively the *cis*-fused systems (4a), (4c), and (4e), respectively; the products (4a) and (5b) have been elaborated to the 10β H and 10α H isomers, respectively, of 5,12-dimethyl-5 β -gonane-8,11,13-triene-17-one.

The presence of partially aromatised hydrocarbons related to steroids in geological samples has been noticed by several authors,¹ and the first identification of one type, *e.g.* (1) has recently been reported by Ourisson and his associates.¹

In connection with the total synthesis of (1) and related compounds,¹ we searched for a method for the exclusive *cis*-reduction of the hexahydrophenanthrene derivative (2a). We report here a general and efficient stereospecific *cis*-reduction procedure for (2a) and analogous compounds, leading to a total synthesis of the title compounds.





Catalytic reduction (Pd-C, 10%) of (2a)[†] under neutral conditions afforded 50% (isolated yield)[‡] of a 7:93 mixture (g.l.c.) of the saturated hydrocarbons (5a) and (5b); and 40%of 42:58 mixture of the saturated ketones (4a) and (4b), respectively. Similar reduction of (2a) in pyridine solution was encouraging as it provided 90% of a 85:15 mixture of the cis- (4a) and the trans-ketone (4b). Hydrogenation of (2a) in acetic acid containing perchloric acid afforded directly 78% of a 10:90 mixture of cis- (5a) and the trans-hydrocarbon (5b), respectively. Lithium-liquid ammonia reduction of (2a), using ammonium chloride as the proton donor, furnished 73% of a 65:35 mixture of (4a) and (4b). The methoxy-ketone (2b)² on similar reduction provided the same mixture of the cis- (4c) and the trans-ketone (4d), indicating thereby that the electronic effect of the p-methoxy-group has nothing to do with stereochemical outcome of this reduction.

The reduction of the double bond of α,β -unsaturated carbonyl compounds with LiAlH₄ is well documented in the literature,³ and the reduction proceeds through the alkoxyaluminium hydride. Reductions of the unsaturated ketones (2a, b) with LiAlH₄ in refluxing tetrahydrofuran, and subsequent oxidation of the resulting alcohols with pyridinium dichromate afforded in 76 and 52% yield, respectively, the saturated *cis*-ketones (4a) and (4c), homogeneous by g.l.c. and ¹H n.m.r. spectroscopy. The methoxy-ketone (2c),⁴ lacking the angular methyl group, after a similar reduction-oxidation sequence also provided, in comparable yield,§ (50%) the *cis*-ketone (4e), identical to an authentic sample.⁴ Not a trace of the *trans*-ketones could be detected in these hydride reductions.

Stereochemical evidence for the above hydride reduction suggests that the reaction is intramolecular, and the formation of a carbon-aluminium bond^{3b,5} as shown in (6) (Al not tetrahedral) is probably necessary. The results so far obtained indicate that the generation of the intermediate (6) may not be totally dependent on the presence or absence of the angular methyl group.

[†] All new compounds described herein are racemic and provided the expected spectral data and elemental analyses.

[‡] The percentage yields mentioned in other cases are all isolated ones.

[§] The lower yield of the saturated *cis*-ketones (4c) and (4e) may possibly be due to the resonance effect of the *p*-OMe which impedes conventional 5-centre hydride transfer to the styrene double bond, and destabilises to a certain extent the intermediate benzylic carbanion.

Huang-Minlon reduction of the *cis*-ketone (4a) afforded in 80% yield the *cis*-hydrocarbon (5a). Friedel-Crafts acetylation¶ of this in the presence of tin(rv) chloride in methylene chloride provided regiospecifically the acetyl derivative (5c) (70%). The crude vinyl ketone (5e), prepared from (5c) by a recently developed procedure,⁶ was directly cyclised with conc. H₂SO₄ to afford the key tetracyclic *cis*-ketone (3a) [42% based on (5c)], a potential intermediate for the synthesis of (1) and related compounds.¹

The mixture of (5a) and (5b), obtained directly from (2a) through hydrogenation in acid medium, afforded the tetracyclic *trans*-ketone (3b) in comparable yield (42%) following the sequence used for $(5a) \rightarrow (3a)$.

The yields of (3a, b) were slightly improved (52%) when vinyl ketones were prepared from the acetyl derivatives through a Mannich base procedure.

Careful ¹H n.m.r. study of the octahydrophenanthrene system related to (5a, b) revealed⁷ that the aromatic C-5 proton in the *trans*-isomer shows a signal at lower field than does the corresponding hydrogen in the *cis*-isomer; we observed this also in the present study. The aromatic C-5 protons in our *cis*- (5c) and *trans*-acetyl derivative (5d) show

 \P This reaction in the presence of anhydrous AlCl₃ gave a mixture which showed no characteristic singlet for the angular methyl group.

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References

- 1 J. Schaefle, B. Ludwig, P. Albrect, and G. Ourisson, *Tetra*hedron Lett., 1978, 4163, and references therein.
- 2 G. Stork, H. J. E. Loewenthal, and P. C. Mukharji, J. Am. Chem. Soc., 1956, 78, 501.
- 3 (a) R. F. Nystrom and W. G. Brown, J. Am. Chem. Soc., 1947, 69, 2548; (b) W. T. Borden, *ibid.*, 1970, 92, 4898; (c) M. Fetizon, Hoa Tran Huy, and P. Mourgues, *Tetrahedron*, 1978, 34, 209; (d) A. Chatterjee, L. N. Dutta, and S. K. Chatterjee, *Indian J. Chem.*, Sect. B, 1980, 19, 955.
- 4 K. E. Fahrenholtz, A. Capomaggi, M. Lurie, M. W. Goldberg, and R. W. Kierstead, J. Med. Chem., 1966, 9, 304.
- 5 Cf. B. Franzus and E. I. Snyder, J. Am. Chem. Soc., 1965, 87, 3423; H. W. Thompson and E. McPherson, J. Org. Chem., 1977, 42, 3350; E. I. Snyder, *ibid.*, 1967, 32, 3531.
- 6 J.-L. Gras, Tetrahedron Lett., 1978, 2111 and 2955.
- 7 W. Nagata, T. Terasawa, and K. Tori, J. Am. Chem. Soc., 1964, 86, 3746.