HIGHLY STEREOSELECTIVE SYN-HYDROXYLATION OF SPIROKETALS

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Summary: The highly stereoselective syn-hydroxylation of unsaturated spiroketals (3,5) is reported.

Griseusin A $(1)^1$, a pyranonaphthoquinone antibiotic, is structurally unique compared to simpler members of the family due to the presence of the highly oxygenated 1,7-dioxaspiro[5.5]undecane ring system. We envisaged a key step in our synthetic route² to griseusin A to involve functionalisation of an unsaturated spiroketal (2) (Scheme). Towards this end, we examined the stereochemical outcome of the *syn*-hydroxylation of a series of model unsaturated spiroketals (3,5) and herein report our results (Table).



Using a catalytic amount of osmium tetraoxide and *N*-methylmorpholine-*N*-oxide (NMO)³ in aqueous acetone at room temperature for 16 h., the spiroketals (3)⁴ underwent smooth hydroxylation to the *syn*-diols (4)⁵. In all cases, hydroxylation occurred from the β -face giving the diols (4) in which the hydroxyl group at C-5 was axial and *anti* to the C-O bond of the neighburing tetrahydropyran ring.⁶ Despite the fact that this was the opposite stereochemistry to that required for the synthesis of griseusin A, no evidence for formation of other diastereomeric diols was observed.⁷ Whilst the addition of other electrophiles such as peracids and *t*-butyl hypochlorite to related 1,7-dioxaspiro[5 5]undec-4-enes⁸ is much less stereoselective, the preferred site of attack was the same as that observed in the present case for osmylation.

Conversion of the spiroketal diols (4) into open chain derivatives may provide access to compounds of known relative configuration 9 Thus, osmylation of the more functionalised equatorial allylic acetates (5)⁴ was effected providing the diols (6) in moderate yield. These latter examples yield highly oxygenated spiroketals with the potential to provide acyclic derivatives of predetermined relative stereochemistry for use in complex natural products synthesis

Diols (4) underwent selective monoacetylation on the least hindered hydroxyl group at C-4 affording acetates (7) in high yield. This latter reaction provides the opportunity for further manipulation of the diol functionality either before or after the ring opening reaction.

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Alkene	Diol	Yield (%)	М.р. (⁰ С)
(3a) $R^1 = H$ (3b) $R^1 = Me$	$(4a) R^{1} = H, R^{2} = H$ $(4b) R^{1} = Me, R^{2} = H$	(4a) 78% (4b) 80% (7a) 89% (7b) 91%	(4a) 124-125 ⁰ C (7a) 111-113 ⁰ C (R ¹ -H, R ² - Ac) (4b) oil (7b) 129-130 ⁰ C (R ¹ -Me, R ² - Ac)
(3c)	Heed S OR Heed S OR Me (4c) R = H	(4c) 73% (7c) 88%	(4c) 94-96 ⁰ C (7c) oil, (R = Ac)
(5a) R = H (5b) R = Me	$(6a) \begin{array}{c} OH \\ H_{eq} & OH \\ O \\ O \\ (6a) \end{array} \begin{array}{c} R = H \\ (6b) \end{array} \begin{array}{c} R = H \\ R = Me \end{array}$	(6a) 59% (6b) 61%	(6a) 145-146 ⁰ C (6b) , oil

References and Notes

- N. Tsuji, M. Kobayashi, Y. Wakisaka, Y. Kawamura, M. Mayama and K. Matsumoto, J. Antibiot., 1976, 29, 7
- 2. M.A. Brimble and M.R. Nairn, J Chem Soc. Perkin Trans. I, 1990, 169.
- 3. V. Van Rheenen, R C. Kelly and D Y. Cha, Tetrahedron Lett., 1976, 1973.
- 4 M A. Brimble, M K. Edmonds and G.M. Williams, Tetrahedron Lett, 1990, 31, 7509.
- 5. All new compounds gave satisfactory spectroscopic and analytical data.
- 6. This is consistent with the model proposed by Kishi for allylic alcohols see: J.K Cha, W.J Christ and Y Kishi, *Tetrahedron Lett.*, 1983, 24, 3943.
- 7. Whilst no examples of *syn*-hydroxylation of spiroketals have been reported, a highly stereoselective osmylation of an unsaturated acetal was published during the course of this work see: E.A. Mash, J B Arterburn, J.A. Fryling and S.H Mitchell, *J Org Chem*, 1991, **56**, 1088.
- 8. R Baker, J.C. Head and C J. Swain, J. Chem. Soc, Perkin Trans 1, 1988, 85.
- 9. Spiroketals have been converted to their open-chain derivatives see: F. Perron and K.F Albizati, *Chem. Rev.*, 1989, **89**, 1617.

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