Synthesis of 1,6-anhydro-3-bromo-3,4-dideoxy-β-D-*threo*-hex-3-enopyranose

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A preparative method for chemo- and stereoselective reduction of 3-bromolevoglucosenone into 1,6-anhydro-3-bromo-3,4-dideoxy- β -D-*threo*-hex-3-enopyranose using $Zn(BH_4)_2$ was developed.

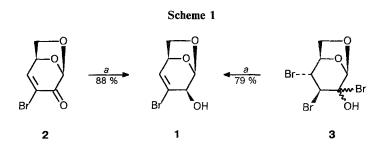
Key words: 3-bromolevoglucosenone, zinc borohydride, reduction.

In this work, we report a synthesis of unsaturated vic-bromohydrine (1), a novel versatile chiral block functionalized by groups of different types. This allows one to perform, in addition to reactions involving O-containing centers, cross-coupling with participation of a Br atom, generation of the corresponding anionic and radical vinyl-type intermediates, etc. This opens new prospects in implementing "chiron approaches" to complex natural compounds.

To prepare the target compound, we intended to perform a one-step $2 \rightarrow 1$ transformation by regio- and stereoselective reduction of ketone 2, a readily accessible product of levoglucosenone bromination¹ (Scheme 1). However, taking into account that structure 2 contains a strongly activated double bond with a highly reactive Br atom, we anticipated that the reduction would occur somewhat ambiguously. Therefore, we assumed that we would also have to use, as an alternative to 2, its latent equivalent 3 (see Ref. 1). Considering the predicted stereoselectivity of the reduction of compound 2, it should be noted that we expected preferential reduction into *threo*-alcohols to occur, in analogy with the known data for levoglucosenone,^{2,3} with which the stereochemical result of the reaction is exclusively determined by steric control of the direction of attack of the reagent by the 1,6-anhydro bridge.

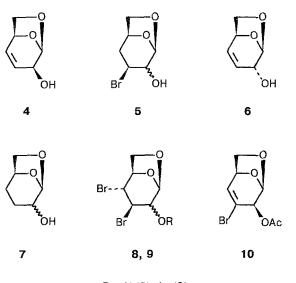
To perform the intended transformations of compounds 2 and 3 into alcohol 1, we examined a series of reducing agents and found that $Zn(BH_4)_2$ (see Ref. 4) is the most appropriate, soft reagent, which allows exceedingly smooth and chemoselective transformation of the starting substrates into synthon 1 (see Scheme 1).

It should be noted that the above optimal transformations were preceded by experiments using readily available LiAlH₄ and NaBH₄; these preliminary experiments did not give results acceptable from the preparative viewpoint. For example, reduction of compound 2 with LiAlH₄ in ether at 0 °C afforded two *threo*-alcohols 1 and 4 easily separable on SiO₂. The reaction of the less nucleophilic NaBH₄ with 2 gave the target alcohol 1 (39 %) along with isomeric bromohydrins 5 (yield 50 %). Under similar conditions, the reaction involving tribromo-derivative 3 also gave an isomeric pair 4 and 6 as well as epimeric saturated alcohols 7 and 8, also characterized as acetates 9. The ease of blocking the hydroxyl group in compound 1 was demonstrated by its transformation into acetate 10.



Reagents and conditons: a. Zn(BH₄)₂, Et₂O, 0 °C.

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R = H (8), Ac (9)

Experimental

IR spectra were obtained on a UR-20 spectrophotometer in thin layers or in Nujol. ¹H and ¹³C NMR spectra were recorded on a Bruker AM-300 spectrometer with working frequencies of 300 and 75.47 MHz, respectively, using SiMe₄ as the internal standard and CDCl₃ as the solvent. TLC analyses were carried out on Silufol chromatographic plates. Chromatographic analyses were performed on a Chrom-5 chromatograph [column length 1.2 m, SE-30 stationary phase, helium as the carrier gas, working temperature 50–300 °C (12 °C min⁻¹)]. Optical rotations were recorded on a Perkin-Elmer 141 polarimeter. Mass spectra were measured on an MKh-1306 instrument, ionizing voltage 70 eV, temperature in the ionization chamber 50–100 °C.

1,6-Anhydro-3-bromo-3,4-dideoxy-β-D-threo-hex-3-enopyranose (1). A. A solution of $Zn(BH_4)_2$ in ether [8 mL; obtained from $ZnCl_2$ (4.0 g) and $NaBH_4$ (2.7 g) in Et_2O (200 mL)] was added dropwise at 0 °C to a solution of bromide 2 (0.45 g, 2.2 mmol) in dry Et₂O (4 mL). After the reaction was complete (TLC), the reaction mixture was cautiously diluted with 3 % aqueous HCl (5 mL), extracted with ether, dried with Na₂SO₄, and concentrated. The residue was recrystallized from an ether-hexane mixture (3 : 1) to give bromohydrin 1 (0.40 g, 88 %). $R_{\rm f}$ 0.37 (ethyl acetatepentane, 3 : 7), m.p. 72-74 °C (in a sealed capillary), $[\alpha]_D^{20}$ -69.3° (c 1.0, CHCl₃). IR, v/cm⁻¹: 3430 (COH); 1645 (C=C). ¹H NMR (CDCl₃), δ : 2.46 (d, OH, J = 11.5 Hz); 3.77 (dd, 1 H, 6-H_{exo}, J = -6.8 and 4.5 Hz); 3.89 (d, 1 H, 6-H_{endo}, J = -6.8 Hz); 4.30 (dd, 1 H, 2-H, J = 2.8and 11.5 Hz); 4.69 (dd, 1 H, 5-H, J = 4.5 and 4.9 Hz); 5.59 (d, 1 H, 1-H, J = 2.8 Hz); 6.50 (d, 1 H, 4-H, J = 4.9 Hz).¹³C NMR (CDCl₃), δ: 70.69 (C-6); 72.74 (C-2); 72.92 (C-5); 100.78 (C-1); 126.43 (C-3); 132.35 (C-4). MS (EI), m/z $(I_{rel}(\%))$: 206 [M]⁺ (2), 188 [M-H₂O]⁺ (77), 175 [M-CH₂OH]⁺ (79), 160 [M-CHO-OH]⁺ (65), 132 [M-CHO-OH-CO]⁺ (18), 81 [M-CHO-OH-Br]⁺ (13), 53 [M-CHO-OH-Br-CO]⁺ (100).

B. Treatment of tribromide 3 with a $Zn(BH_4)_2$ solution (2 mL) by the above procedure gave 0.06 g (79 %) of compound 1.

1,6-Anhydro-3-bromo-3,4-dideoxy- β -D-*glycero*-hex-3-enopyranose (2) was obtained by the known procedure¹: m.p. 49 °C (from a Et₂O-hexane mixture, 3 : 7), $[\alpha]_D^{18}$ -496.1° (*c* 1.0, CHCl₃). ¹H NMR (CDCl₃), δ : 3.82 (d, 1 H, 6-H_{endo}, J = -7.1 Hz); 3.90 (dd, 1 H, 6-H_{exo}, J = -7.1 and 4.6 Hz); 5.09 (dd, 1 H, 5-H, J = 4.6 and 5.0 Hz); 5.55 (s, 1 H, 1-H); 7.69 (d, 1 H, 4-H, J = 5.0 Hz). ¹³C NMR (CDCl₃), δ : 66.52 (C-6); 73.49 (C-5); 100.81 (C-1); 122.25 (C-3); 147.94 (C-4); 182.36 (C-2).

1,6-Anhydro-2,3,4-tribromo-3,4-dideoxy-β-**D**manno(gluco)-pyranose (3): m.p. 103–105 °C (from Me₂CO). ¹³C NMR (DMSO-d₆), δ of the main manno-isomer 3: 53.14 (C-4); 61.36 (C-3); 64.29 (C-6); 76.11 (C-5); 93.15 (C-2); 102.87 (C-1).

Reduction of 3-bromolevoglucosenone. *A.* LiAlH₄ (0.09 g, 2.4 mmol) was added in portions to a cooled solution (0 °C) of ketone 2 (0.50 g, 2.4 mmol) in Et₂O (5 mL). The mixture was stirred for 15 min with TLC monitoring. When compound 2 was consumed, the reaction was terminated by adding 3 % aqueous HCl (5 mL), the Et₂O was distilled off, and the mixture was extracted with CH₂Cl₂ (3 × 50 mL). The combined extracts were dried with Na₂SO₄ and concentrated. The residue was chromatographed on SiO₂ using ethyl acetate—pentane (3 : 7) as the eluent to give 0.3 g of a mixture of alcohols 1 and 4 in a 5 : 1 ratio. Alcohols 1 and 4 were isolated in pure state by repeated chromatography of the mixture on a column with SiO₂.

B. A mixture of NaBH₄ (0.08 g, 2.2 mmol) and compound 2 (0.30 g, 1.4 mmol) in MeOH (4 mL) was added to MeOH (6 mL) precooled to 0 °C. The mixture was stirred for 20 min and diluted with H₂O (10 mL). The MeOH was distilled off, and the residue was extracted with CH₂Cl₂ (3 × 25 mL). The combined extracts were dried with Na₂SO₄ and concentrated. Chromatography gave 0.21 g of alcohols 1 and 5 in a 4 : 5 ratio.

1,6-Anhydro-3,4-dideoxy-\beta-D-threo-hex-3-enopyranose (4): $R_{\rm f}$ 0.19 (ethyl acetate—hexane, 3 : 7), m.p. 68—69 °C, $[\alpha]_{\rm D}^{13}$ -30.8° (c 1.0, CHCl₃) [cf. Ref. 2: m.p. 67—69 °C, $[\alpha]_{\rm D}^{20}$ -34° (c 1.0, CHCl₃); cf. Ref. 3: m.p. 70—70.5 °C, $[\alpha]_{\rm D}^{20}$ -30° (c 1.0, CHCl₃)]. ¹H NMR (CDCl₃), δ : 2.60 (br.s, OH); 3.72 (dd, 1 H, 6-H_{exo}, J = -6.6 and 4.2 Hz); 3.80 (d, 1 H, 6-H_{endo}, J = -6.6 Hz); 4.30 (br.s, 1 H, 2-H); 4.62 (t, 1 H, 5-H, J = 4.2 Hz); 5.48 (dd, 1 H, 1-H, J = 2.0 and 2.5 Hz); 5.67 (ddd, 1 H, 3-H, J = 2.0, 2.3 and 9.8 Hz); 6.08 (dd, 1 H, 4-H, J = 9.8 and 4.2 Hz). The ¹H NMR spectrum obtained agrees with that reported in the literature.⁵ ¹³C NMR (CDCl₃), δ : 68.69 (C-2); 70.65 (C-6); 71.12 (C-5); 101.21 (C-1); 129.04 (C-3); 130.64 (C-4).

1,6-Anhydro-3-bromo-3,4-dideoxy-β-D-*lyxo(xylo)*-hexopyranose (5): R_f 0.35 (ethyl acetate—hexane, 3 : 7). xylo-Isomer of 5: ¹H NMR (CDCl₃), δ: 2.10 (m, 1 H, 4-H); 2.45 (m, 1 H, 4-H); 2.40 (br.s, OH); 3.70–3.90 (m, 3 H, 6-H_{endo}, 6-H_{exo}, 2-H); 4.05 (ddd, 1 H, 3-H, J = 11.5, 8.9 and 7.1 Hz); 4.3 (dd, 1 H, 5-H, J = 11.5 and 3.0 Hz); 5.52 (d, 1 H, 1-H, J = 2.7 Hz). ¹³C NMR (CDCl₃), δ: 36.20 (C-4); 46.97 (C-3); 66.61 (C-6); 69.56 (C-2); 73.42 (C-5); 101.32 (C-1). *lyxo*-Isomer of 5: ¹H NMR (CDCl₃), δ: 2.30–2.40 (m, 2 H, 4-H); 2.40 (br.s, OH); 3.70–3.90 (m, 3 H, 6-H_{endo}, 6-H_{exo}, 2-H); 4.46 (ddd, 1 H, 3-H, J = 12.5, 6.1, and 3.7 Hz); 4.55 (m, 1 H, 5-H); 5.40 (br.s, 1 H, 1-H, J = 3.7 Hz). ¹³C NMR (CDCl₃), δ: 39.80 (C-4); 48.82 (C-3); 67.82 (C-6); 73.62 (C-5); 76.09 (C-2); 102.12 (C-1).

C. Reduction of compound **3** (0.53 g) with LiAlH₄ (0.09 g) in 5 mL of Et₂O by a procedure similar to that for **2** gave 0.31 g of a mixture containing epimers of compounds **4** and **6** in a 1 : 1 ratio, saturated alcohols 7 (*threo* : *erythro* = 2 : 1),

and a mixture of *manno*- and *gluco*-isomers of dibromides $\mathbf{8}$ in a 10 : 3 ratio.

1,6-Anhydro-3,4-dideoxy-\beta-D-*erythro*-hex-3-enopyranose (6): $R_{\rm f}$ 0.11 (ethyl acetate—pentane, 3 : 7), m.p. 52.5–53 °C, $[\alpha]_{\rm D}^{23}$ -229° [*cf.* Ref. 3: m.p. 53–54 °C, $[\alpha]_{\rm D}^{20}$ -236° (*c* 1.0, CHCl₃)]. ¹H NMR (CDCl₃), δ : 2.59 (br.s, OH); 3.58–3.67 (m, 2 H, 6-H_{exo}, 6-H_{endo}); 4.66 (dd, 1 H, 5-H, *J* = 1.0 and 3.8 Hz); 5.48 (br.s, 1 H, 1-H); 5.78 (ddd, 1 H, 3-H, *J* = 2.1, 4.1, and 9.8 Hz); 6.14 (dd, 1 H, 4-H, *J* = 4.3 and 9.8 Hz). The ¹H NMR spectrum agrees with that reported previously.⁵ ¹³C NMR (CDCl₃), δ : 55.77 (C-2); 68.89 (C-6); 70.51 (C-5); 102.60 (C-1); 126.43 (C-3); 130.64 (C-4).

1,6-Anhydro-3,4-dideoxy-β-D-*threo(erythro)*-hexopyranose (7): $R_f 0.19$ (ethyl acetate—hexane, 3 : 7). ¹H NMR (CDCl₃), δ: 1.40—1.60 (m, 2 H, 3-H₂C); 1.78—2.04 (m, 2 H, 4-H₂C); 2.40 (br.s, OH); 3.55—3.60 (m, 1 H, 5-H); 3.75—3.90 (m, 2 H, 6-H₂C); 4.43 (br.s, 1 H, 2-H); 5.27 (br.s, 1 H, 1-H). *threo*-Isomer of 7: ¹³C NMR (CDCl₃), δ: 25.92 (C-3); 27.79 (C-4); 68.13 (C-6); 68.95 (C-2); 72.77 (C-5); 102.92 (C-1). *erythro*-Isomer of 7: ¹³C NMR (CDCl₃), δ: 23.23 (C-3); 24.76 (C-4); 66.72 (C-2); 66.79 (C-6); 73.26 (C-5); 102.06 (C-1).

1,6-Anhydro-3,4-dibromo-3,4-dideoxy- β -**D**-*manno(gluco)*hexopyranose (8): R_f 0.25 (ethyl acetate—hexane, 3 : 7). gluco-Isomer of 8: ¹H NMR (CDCl₃), δ : 3.20 (br.s, OH); 3.65 (dd, 1 H, 6-H, J = -8.2 and 3.5 Hz); 3.80 (dd, 1 H, 2-H, J =3.9 and 2.1 Hz); 4.17 (d, 1 H, 6-H, J = -8.2 Hz); 4.31 (dd, 1 H, 4-H, J = 11.1 and 3.3 Hz); 4.36 (dd, 1 H, 3-H, J =11.1 and 3.9 Hz); 4.65 (dd, 1 H, 5-H, J = 3.5 and 3.3 Hz); 5.45 (d, 1 H, 1-H, J = 2.1 Hz). ¹³C NMR (CDCl₃), δ : 51.13 (C-3); 55.19 (C-4); 64.39 (C-6); 71.98 (C-2); 76.63 (C-5); 101.18 (C-1). *manno*-Isomer of 8: ¹³C NMR (CDCl₃), δ : 51.36 (C-3); 56.27 (C-4); 65.64 (C-6); 76.63 (C-5); 77.00 (C-2); 101.79 (C-1).

1,6-Anhydro-2-*O***-acetyl-3,4-dibromo-3,4-dideoxy**β-D-*manno(gluco)*-hexopyranose (9) was obtained in 81 % yield by acylation of compound 8 in an Ac₂O--Py mixture (1 : 2) at 0 °C. $R_{\rm f}$ 0.60 (ethyl acetate-hexane, 3 : 7). *manno*-Isomer of 9: ¹H NMR (CDCl₃), δ : 2.14 (s, 3 H, CH₃); 3.80 (dd, 1 H, 6-H_{exo}, J = -8.3 and 4.9 Hz); 4.11 (dd, 1 H, 3-H, *J* = 9.2 and 10.2 Hz); 4.30 (d, 1 H, 6-H_{endo}, *J* = -8.3 Hz); 4.38 (dd, 1 H, 4-H, *J* = 2.5 and 10.2 Hz); 4.75 (dd, 1 H, 5-H, *J* = 2.5 and 4.9 Hz); 5.00 (dd, 1 H, 2-H, *J* = 1.3 and 9.2 Hz); 5.49 (d, 1 H, 1-H, *J* = 1.3 Hz). ¹³C NMR (CDCl₃), 8: 20.71 (CH₃); 49.31 (C-3); 51.22 (C-4); 65.69 (C-6); 76.09 (C-2); 76.78 (C-5); 99.74 (C-1); 169.66 (CO). gluco-Isomer of **9**: ¹H NMR (CDCl₃), 8: 2.19 (s, 3 H, CH₃); 3.85 (dd, 1 H, 6-H_{exo}, *J* = -8.3 and 4.4 Hz); 4.25 (m, 1 H, 3-H); 4.30 (d, 1 H, 6-H_{endo}, *J* = -8.3 Hz); 4.38 (m, 1 H, 4-H); 4.68 (dd, 1 H, 5-H, *J* = 2.5 and 4.4 Hz); 5.08 (d, 1 H, 2-H, *J* = 2.2 Hz); 5.51 (d, 1 H, 1-H, *J* = 2.2 Hz). ¹³C NMR (CDCl₃), 8: 20.71 (CH₃); 50.21 (C-3); 51.22 (C-4); 64.52 (C-6); 71.56 (C-2); 76.94 (C-5); 99.45 (C-1); 169.66 (CO).

1,6-Anhydro-2-*O*-acetyl-3-bromo-3,4-dideoxy-β-D-*threo*hex-3-enopyranose (10) was obtained in 75 % yield by acylation of compound 1 in an Ac₂O-Py mixture (1 : 2) at 0 °C. $[\alpha]_D^{20}$ -67.2 (*c* 1.0, CHCl₃). ¹H NMR (CDCl₃), δ: 2.15 (s, 3 H, CH₃); 3.52 (ddd, 1 H, 6-H, J = -6.8, 4.2, and 0.9 Hz); 3.95 (d, 1 H, 6-H, J = -6.8 Hz); 4.63 (dd, 1 H, 5-H, J =4.8 and 4.2 Hz); 5.55 (dd, 1 H, 2-H, J = 2.6 and 1.1 Hz); 5.60 (d, 1 H, 1-H, J = 2.6 Hz); 6.53 (ddd, 1 H, 4-H, J =4.8, 1.1, and 0.9 Hz). ¹³C NMR (CDCl₃), δ: 20.62 (CH₃); 70.93 (C-6); 72.51 (C-2); 73.40 (C-5); 98.56 (C-1); 120.35 (C-3); 134.17 (C-4), 170.09 (CO).

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