Chemistry Letters 1997 811

## A Sequence of Palladium-Catalyzed Borylation of Allyl Acetates with Bis(pinacolato)diboron and Intramolecular Allylboration for the Cyclization of Oxo-2-alkenyl Acetates

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(Received May 12, 1997; CL-970353)

The cross-coupling reaction of bis(pinacolato)diboron with oxo-2-alkenyl acetates in toluene at 50  $^{\circ}$ C in the presence of Pd(dba)<sub>2</sub>/2 AsPh<sub>3</sub> (3 mol%) to give allylboronates was followed by the intramolecular allylboration at 100  $^{\circ}$ C. The sequence provided cyclic homoallyl alcohols in high yields.

Intramolecular addition of allylmetal reagents to carbonyl substrates is a powerful tool for the synthesis of cyclic homoallyl alcohols with high regio- and stereoselectivity. Lewis acid- or fluoride anion-promoted cyclization of oxo-2-alkenylsilanes<sup>1</sup> or stannanes,<sup>2</sup> and Barbier type reductive cyclization of oxo-2-alkenyl halides by the use of Mg,<sup>3</sup> In,<sup>4</sup> and SmI<sub>2</sub><sup>5</sup> have been extensively studied. However, the corresponding reaction of allylboranes has not been well developed mainly due to the lack of a general method for introduction of a boryl group into carbonyl substrates. 6 Recently, we reported the regio- and stereoselective synthesis of allylboron reagents by the palladium(0)-catalyzed cross-coupling reaction of bis(pinacolato)diboron (1) with allyl acetates. The method provides an efficient and convenient access to variously functionalized allyboranes because a variety of allyl acetates are easily available and the reaction conditions for the coupling are sufficiently mild (Pd(dba)2/DMSO/50°C). In the present paper, we wish to report the direct synthesis of oxo-2alkenylboranes (3) by the cross-coupling of 1 with oxo-2-alkenyl acetates (2) and their cyclization to cyclic homoallyl alcohols (4)

$$\begin{bmatrix} R^1 & COR^3 \\ R^2 & B \end{bmatrix} \xrightarrow{100 \circ C} \begin{bmatrix} R^1 & OH \\ R^2 & COR^3 \end{bmatrix}$$
 (1)

The sequential reaction, involving the cross-coupling between 1 (1.1 mmol) and 2a ( $R^1$ ,  $R^2 = -(CH_2)_4$ -,  $R^3 = OEt$ , Entry 1 in Table 1) (1.0 mmol) and the intramolecular allylboration, was investigated under various conditions. The best yield was achieved (4a, 88%) in toluene when the coupling with  $Pd(dba)_2/2$  AsPh<sub>3</sub> (dba is dibenzylideneacetone) (3 mol%) at 50 °C for 16 h was followed by the cyclization at 100 °C for 24 h. Although we have previously reported that  $Pd(dba)_2$  in DMSO exhibits excellent catalytic activity in the cross-coupling of 1 with allyl acetates, the intramolecular allylboration of the carbonyl was extremely slow in this solvent, presumably due to its coordination

**Table 1.** Synthesis of **4** (Eq. 1) $^a$ 

	e 1. Synthesis of 4 (Eq. Acetate (2)	Product (4)	Yield / % <sup>b</sup>
1	OAc OAc	OH CO <sub>2</sub> Et <b>4a</b>	72 (88)
2	CO <sub>2</sub> Et OAc	OH CO <sub>2</sub> Et <b>4b</b>	82
3	CO <sub>2</sub> Me OAc	CO <sub>2</sub> Me	71
4	Me OAc	OH Me 4d	77
5	Me CO <sub>2</sub> Et OAc	Me OH CO <sub>2</sub> Et <b>4e</b>	62
6	Me COMe OAc	Me COMe	71
7	COMe OAc	oMe \$ H	52

<sup>&</sup>lt;sup>a</sup>The experimental procedures, see the text.

to the boron atom (63%, 120 °C/24 h). Thus, less polar solvents

<sup>&</sup>lt;sup>b</sup>Isolated yields based on 2 and GLC yield in parenthesis.

812 Chemistry Letters 1997

were desirable for the cyclization; however, a similar cross-coupling in toluene was unsuccessful because of the catalyst decomposition precipitating palladium black. Fortunately, the addition of AsPh<sub>3</sub> or PPh<sub>3</sub> (2 equivs) to Pd(dba)<sub>2</sub> was found to be effective to stabilize the active palladium(0) species during the cross-coupling at 50 °C. 1,1'-Bis(diphenylphosphino)ferrocene (dppf) did not give any good results whereas this ligand has been successfully utilized in the coupling of 1 with aryl halides and triflates.<sup>8</sup>

The  $^1$ H NMR spectra of  $\mathbf{4a}$  in CDCl<sub>3</sub> revealed a down-field resonance at 3.74 ppm for the hydroxyl proton due to the intramolecular hydrogen bonding between OH and C=O, suggesting a *cis*-cyclization. The stereochemistry is further supported by its conversion into  $\mathbf{5}$  and the presence of NOE (2.3 %) in DMSO-d<sub>6</sub> between OH (4.14 ppm) and CH<sub>3</sub> (0.91 ppm) (Eq. 2).

The representative results are summarized in Table 1. A variety of oxo-2-alkenyl acetates 2 smoothly underwent the cyclization to the corresponding cyclic homoallyl alcohols 4 under the conditions optimized above. The 5-5, 6-5, and 7-5 *cis*-fused alcohols 4a-4d were readily obtained from cyclic β-ketoesters and diketone (Entries 1-4). Interestingly, acyclic oxo-2-alkenyl acetates, such as 2e and 2f, stereoselectively provided a single stereoisomer (4e and 4f) (Entries 5 and 6). <sup>1</sup>H NMR analyses of 4e,f exhibited a down-field resonance for the hydroxyl proton and NOE (1.4%) between two methyl groups. <sup>9</sup> Spirocyclic alcohols 4g was similarly obtained as a mixture of two stereoisomers (Entry 7).

The representative procedure: A dry 25-ml flask equipped with a magnetic stirring bar, a septum inlet, an oil bubbler, and a reflux condenser was charged with  $Pd(dba)_2$  (dba is dibenzylideneacetone) (0.03 mmol),  $AsPh_3$  (0.06 mmol), and toluene (6 ml) under nitrogen. After being stirred at room temperature for 30 min, 1 (1.1 mmol) and 2a (1.0 mmol) were successively added. The mixture was heated at 50 °C for 16 h and then at 100 °C for 24 h. The resulting mixture was treated with

saturated ammonium chloride solution (10 ml) at room temperature for 1 h, extracted with ether (10 ml, three times), and dried over anhydrous magnesium sulfate. An analytically pure product was isolated by column chromatography over silica gel. **4a**: IR (film) 3500, 1710 cm<sup>-1</sup>; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  1.15-1.45 (m, 3 H), 1.28 (t, 3 H, J = 7.1 Hz), 1.54 (dt, 1 H, J = 3.6 and 13.2 Hz), 1.70-1.75 (m, 1 H), 1.82 (dd, 2 H, J = 4.0 and 9.6 Hz), 2.02 (d, 1 H, J = 12.7 Hz), 2.32 (t, 2 H, J = 15.7 Hz), 2.69 (dt, 1 H, J = 2.5 and 17.6 Hz), 2.99 (dd, 1 H, J = 2.6 and 16.5 Hz), 3.74 (d, 1 H, J = 2.0 Hz), 4.19 (dq, 2 H, J = 1.8 Hz and 7.1 Hz), 4.95 (s, 1 H), 4.99 (s, 1 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  14.10, 22.28, 23.65, 33.22, 34.05, 41.79, 42.18, 55.17, 60.58, 80.31, 108.80, 146.45, 176.74.

## References

- For recent reviews, see: a) C. E. Masse and J. S. Panek, *Chem. Rev.*, **95**, 1293 (1995). b) Y. Yamamoto and N. Asao, *Chem. Rev.*, **93**, 2207 (1993). c) I. Fleming, J. Dunogés, and R. Smithers, *Org. React.*, **37**, 57 (1989).
- 2 a) G. E. Keck, S. M. Dougherty, and K. A. Savin, J. Am. Chem. Soc., 117, 6120 (1995). b) V. Gevorgyan, I. Kadota, and Y. Yamamoto, Tetrahedron Lett., 34, 1313 (1993). c) S. E. Denmark, E. J. Weber, T. M. Wilson, and T. M. Wilson, Tetrahedron, 45, 1053 (1989). d) J. A. Marshall, S. L. Crooks, and B. S. DeHoff, J. Org. Chem., 53, 1616 (1988).
- 3 H. Felkin, Y. Gault, and G. Roussi, *Tetrahedron*, 26, 3761 (1970).
- 4 a) C.-J. Li, D.-L. Chen, Y.-Q. Lu, J. X. Haberman, and J. T. Mague, J. Am. Chem. Soc., 118, 4216 (1996). b) Y.-Q. Lu and C.-J. Li, Tetrahedron Lett., 37, 471 (1996).
- 5 M. Kito, T. Sakai, H. Shirahama, M. Miyashita, and F. Matsuda, Synlett, 1997, 219
- a) R. W. Hoffmann and A. Hense, *Liebigs Ann. Chem.*,
   1996, 1283. b) T. Sander and R. W. Hoffmann, *Liebigs Ann. Chem.*, 1993, 1193. c) R. W. Hoffmann and G. Niel,
   *Liebigs Ann. Chem.*, 1991, 1195
- 7 T. Ishiyama, T.-a. Ahiko, and N. Miyaura, *Tetrahedron Lett.*, **37**, 6889 (1996).
- a) T. Ishiyama, M. Murata, and N. Miyaura, J. Org. Chem.,
  60, 7508 (1995). b) T. Ishiyama, Y. Itoh, T. Kitano, and N. Miyaura, Tetrahedron Lett., 38, 3447 (1997).
- G. A. Molander and S. W. Andrews, *Tetrahedron*, 40, 3869 (1988).