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1) LDA, 2 equiv., -30°C
 2) 1.5 equiv., 0°C, 4h
 2, 65-88% yield
 a, R₁=H, R₂=Me
 b, R₁=Me, R₂=Me
 c, R₁=H, R₂=Ph

1) LDA, 1.1 equiv., THF, -78°C
 2) CF₃SO₂TBDMS, 1.2 equiv., -78°C
 (1Z,3E)3, 70-85% yield

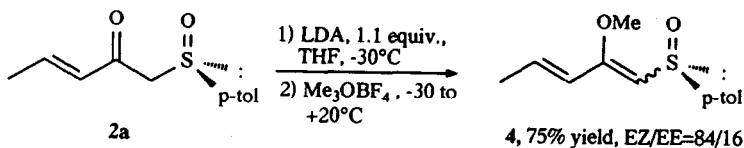
The pure (1Z, 3E) diene **3**¹¹ was easily obtained by flash chromatography. The stereochemistry was attributed by NMR from NOE experiments : irradiation of the vinylic proton H-1 lead to a clear NOE on H-3.

Table I: β -ketosulfoxides 2 and sulfinyldienes 3

	R ₁	R ₂	R ₃	Yield % 2	[α] _D ^a 2	Yield % ^b (1Z,3E)3	[α] _D ^c (1Z,3E)3
a	H	Me	Me	65	+241	85	-201 ^d
b	Me	Me	Et	88	+278	70	-119 ^e
c	H	Ph	Me	85	+174	70	-118 ^f

a) isolated yields, b) CHCl₃, c=1, c) CHCl₃, d) c=1.8, e) c=0.5, f) c=1.3

It was also attempted to quench the enolate with an alkyl instead a silyl. This was carried out from the lithium enolate of the β -ketosulfoxide 2a and trimethyloxonium tetrafluoroborate as a quenching agent. The enol ether was obtained in 75% yield as a 84/16 mixture of (1Z, 3E/1E,3E) isomeric dienes 4, determined by NMR from the vinylic proton H-1 (5.5 ppm in the ZE isomer and 5.4 in the EE) and from the methoxy group (3.9 and 3.6 ppm respectively).



However the separation of these two isomers was impossible by chromatography.

Acknowledgments : We thank dirección general de Investigación Científica y Técnica and CNRS for financial support and the Ministerio de Educación y Ciencia for a scholarship to A.A.

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- 11) Typical example of ¹H NMR of sulfinyl diene: 1Z,3E 3a (200 MHz,CDCl₃): δ : 0.3 and 0.4 (2s, 6H, Me₂Si), 1.1 (s, 9H, t-BuSi), 1.8 (dd, J=7 Hz, 3H, CH₃), 2.4 (s, 3H, CH₃-Ar), 5.6 (s, 1H, H-2), 5.9 (qd, 1H, H-4, J_{4,5}=15 Hz, J_{4,6}=1Hz), 6.1 (qd, 1H, H-5, J_{5,6}=7 Hz, J_{5,4}=15 Hz), 7.5-7.3 (AA'BB', 4H, J=8 Hz, arom.).