

β -Lactone formation during product release from a nonribosomal peptide synthetase

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Nonribosomal peptide synthetases (NRPSs) are multidomain modular biosynthetic assembly lines that polymerize amino acids into a myriad of biologically active nonribosomal peptides (NRPs). NRPS thioesterase (TE) domains employ diverse release strategies for off-loading thioester-tethered polymeric peptides from termination modules typically via hydrolysis, aminolysis, or cyclization to provide mature antibiotics as carboxylic acids/esters, amides, and lactams/lactones, respectively. Here we report the enzyme-catalyzed formation of a highly strained β -lactone ring during TE-mediated cyclization of a β -hydroxythioester to release the antibiotic obafluorin (Obi) from an NRPS assembly line. The Obi NRPS (ObiF) contains a type I TE domain with a rare catalytic cysteine residue that plays a direct role in β -lactone ring formation. We present a detailed genetic and biochemical characterization of the entire Obi biosynthetic gene cluster in plant-associated *Pseudomonas fluorescens* ATCC 39502 that establishes a general strategy for β -lactone biogenesis.

Strained β -lactone rings are found in diverse classes of natural products, including polyketides (PKs), nonribosomal peptides (NRPs), amino acids, terpenoids, and hybrid molecules¹. Little is known about the biosynthetic origins of β -lactones despite the wide therapeutic value of naturally occurring β -lactones as inhibitors of enzymes in the serine hydrolase superfamily². Tetrahydrolipstatin is a hybrid PK–NRP lipase inhibitor, known commercially as Orlistat, that is approved by the US Food and Drug Administration for the treatment of obesity³. Salinosporamide A, also known as marizomib and NPI-0052, is a hybrid PK–NRP proteasome inhibitor that is under clinical investigation for use as a treatment of multiple myeloma and other advanced malignancies⁴. Other natural β -lactones show promise as antimicrobial⁵, anticancer⁴, antiviral⁶, and anti-obesity³ agents (Supplementary Results, Supplementary Fig. 1). The genetic and biochemical basis for β -lactone ring formation is an unsolved problem in natural product biosynthesis⁷.

There is a precedent for enzymatic formation of strained four-membered rings in the closely related β -lactam family of antibiotics⁸. Three chemically distinct biosynthetic pathways leading to β -lactams have been reported. Penicillin and cephalosporin bicyclic β -lactam scaffolds form via oxidative cyclization of an NRP tripeptide precursor by isopenicillin N synthase⁹. The β -lactam rings in clavulanic acid and carbapenems arise via ATP-dependent cyclization of β -amino acid precursors catalyzed by the β -lactam synthetase enzyme family¹⁰. The nocardicin family of monocyclic β -lactam antibiotics are derived from an NRPS assembly line comprised of five catalytic modules that each covalently tether the evolving substrate in the form of a thioester on a peptidyl carrier protein, also known as the thiolation (T) domain. The condensation (C) domain of module 5 has a rare HHHxxDG motif that is important for dehydration of a T₄-serine thioester to a T₄-dehydroalanine thioester electrophile that undergoes a Michael addition–nucleophilic acyl substitution reaction cascade with the nucleophilic α -amino group of a downstream T₅-L-(*p*-hydroxyphenyl)glycine thioester to produce the nocardicin β -lactam warhead¹¹. We hypothesized that similar biosynthetic strategies could conceivably be employed to assemble β -lactone rings.

The β -lactone synthetase enzyme OleC produces β -lactone intermediates during olefinic hydrocarbon biosynthesis¹². OleC

is proposed to catalyze the ATP-dependent cyclization of *syn* and *anti* β -hydroxy acids to the corresponding *cis* and *trans* β -lactones, which undergo decarboxylation to give the *cis* and *trans* double bonds found in olefinic hydrocarbons. The biosynthetic gene clusters for the natural product β -lactones lipstatin¹³, ebelactone⁷, salinosporamide¹⁴, and oxazolomycin¹⁵ are known, and their biosynthetic pathways have been proposed. However, the mechanisms for β -lactone ring formation in these systems remain unclear, and no enzyme domains have been experimentally linked to β -lactone cyclization. Interestingly, the terminating PKS or NRPS module for each β -lactone antibiotic lacks an embedded TE domain¹⁶. TE domains can participate in acyl transfers¹⁷, epimerization of stereogenic centers¹⁸, proofreading¹⁹, and product release via hydrolysis or macrocyclization²⁰. The β -lactone rings of lactacystin and ebelactone can form nonenzymatically at pH 7 from precursor β -hydroxythioesters, which challenges the evolutionary need for enzyme catalysis of β -lactone ring formation^{7,21}. We hypothesized that enzyme catalysis could still be at play for these β -lactones, a factor that might have been overlooked as a result of the relative stability of the β -lactones. Thus, we turned our search for a ' β -lactone synthase' to scenarios in which the β -lactone is rapidly hydrolyzed in aqueous environments, where enzyme catalysis would be needed to increase the rate of β -lactone ring formation to maintain titer levels that are beneficial to the producing microbe.

We discovered that enzyme catalysis is required for strained ring formation during biosynthesis of the NRPS β -lactone antibiotic obafluorin (Obi, **1**), which rapidly hydrolyzes to the corresponding β -hydroxycarboxylic acid (Obi-COOH, **2**) at neutral pH^{22,23}. We used full genome sequencing and comparative sequence analysis to identify a putative Obi biosynthetic gene cluster in the known Obi producer *P. fluorescens* ATCC 39502. We biochemically characterized five recombinant enzymes, ObiD, ObiF, ObiG, ObiH, and ObiL, that convert chorismate-derived 2,3-dihydroxybenzoic acid (2,3-DHB, **3**) and *p*-NH₂-phenylpyruvic acid (PAPPA, **4**) to Obi by way of the biosynthetic intermediates *p*-NO₂-phenylpyruvic acid (PNPPA, **5**), *p*-NO₂-phenylacetaldehyde (PNPAA, **6**), and (R)- β -hydroxy-*p*-nitro-L-homophenylalanine (β -OH-*p*-NO₂-homoPhe, **7**). We found that the type I TE domain of ObiF catalyzed the cyclization of an active site-tethered Cys- β -hydroxythioester during product

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release from the NRPS assembly line. Our findings establish a genetic signature and biochemical mechanism for β -lactone biosynthesis from NRPS assembly lines that will guide future β -lactone natural product discovery efforts through genome mining, computational structure prediction, chemoenzymatic synthesis, and directed biosynthesis^{24,25}.

RESULTS

Identification of the obafluorin biosynthetic gene cluster

Obi is a *cis*-monocyclic β -lactone antibiotic produced by plant-associated strains of *P. fluorescens*^{22,23}. Under physiological conditions, Obi spontaneously hydrolyzes to the corresponding β -hydroxycarboxylic acid Obi-COOH (Fig. 1a)²⁶. The unique structure of Obi contains a 2,3-DHB unit coupled through an amide linkage to an α -amino- β -lactone ring of the unusual nonproteinogenic amino acid β -OH-*p*-NO₂-homoPhe. Stable-isotope feeding studies suggest that 2,3-DHB and β -OH-*p*-NO₂-homoPhe are advanced intermediates in the biosynthesis of Obi^{27,28}. We sequenced the genome of the known Obi producer *P. fluorescens* ATCC 39502 to search for a putative Obi biosynthetic gene cluster. We identified a ~20-kilobase-pair gene cluster with 14 candidate coding sequences (CDSs), including one encoding an AurF di-iron nonheme arylamine oxygenase homolog (*obiL*) predicted to install the aryl nitro functional group via oxidation of the precursor aniline (Fig. 1b and Supplementary Table 1). The Obi gene cluster encodes putative enzymes required for the biosynthesis of 2,3-DHB (*obiA*, *obiB*, *obiC*, and *obiE*) and the nonproteinogenic amino acid β -OH-*p*-NO₂-homoPhe (*obiG*, *obiH*, *obiI*, *obiJ*, *obiK*, and *obiL*). CDSs encoding an aryl acyl carrier protein (*obiD*) and an NRPS module (*obiF*) with condensation (C), acid adenylation (A), aryl acid adenylation (A_{Ar}), thiolation (T), and thioesterase (TE) domains make up the remaining biosynthetic genes. There are no obvious antibiotic resistance genes in the cluster. The genes *obiM* and *obiN* encode proteins that have strong homology to the acylhomoserine lactone (AHL) synthase LuxI and the AHL-binding transcriptional regulator LuxR, respectively. The LuxI and LuxR pair is a common quorum-sensing system found in many pseudomonads. A basic local alignment search tool (BLAST) search of the NCBI database revealed homologous Obi clusters in other pseudomonads including *Pseudomonas* sp. 37_R_15 and 34_E_7, the environmental chitin-degrading bacterium *Chitiniphilus shinanonensis* SAY3, and the plant-growth-promoting rhizobacterium *Burkholderia diffusa* RF8-non_BP2, which is also an opportunistic human pathogen (Supplementary Table 2; Supplementary Figs. 2 and 3).

2,3-DHB and PAPPa are precursors to Obi

Chorismate is the primary carbon source of the Obi biosynthetic pathway (Supplementary Fig. 4a)^{27,28}. The gene *obiB* encodes an extra copy of the first biosynthetic enzyme from the shikimate pathway, 3-deoxy-D-arabinoheptulosonate-7-phosphate (DAHP) synthase, which converts phosphoenolpyruvate (PEP) and erythrose-4-phosphate to chorismate²⁹. Chorismate is converted to 2,3-DHB by isochorismate synthase (*ObiA*), isochorismatase (*ObiE*), and a dehydrogenase (*ObiC*), which are analogous to EntC, EntB, and EntA, respectively, involved in biosynthesis of the siderophore enterobactin³⁰. Chorismate is also converted to PAPPa by the 4-amino-4-deoxychorismate (4-ADC) synthase (*ObiJ* and *ObiK*) and 4-ADC mutase/dehydratase (*ObiI*), a process analogous to that of the CmlB, CmlC, and CmlD and PapA, PapB, and PapC enzymes that lead to PAPPa in the chloramphenicol³¹ and pristinamycin³² biosynthetic pathways, respectively (Supplementary Fig. 4b). We undertook *in vitro* reconstitution of the remaining biosynthetic enzymes (*ObiD*, *ObiF*, *ObiG*, *ObiH*, and *ObiL*) to establish the timing and action of each step in the conversion of 2,3-DHB and PAPPa to Obi (Supplementary Fig. 5).

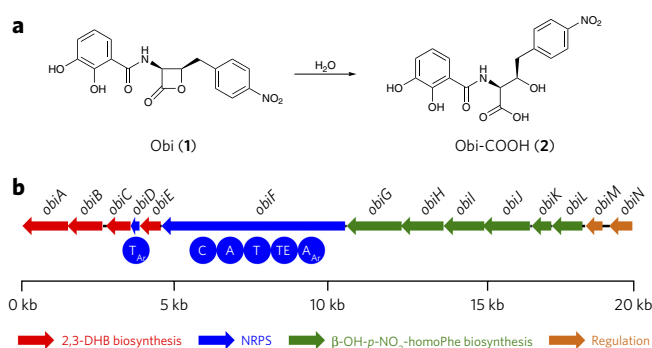


Figure 1 | Biosynthetic gene cluster for obafluorin β -lactone from

***P. fluorescens* ATCC 39502. (a)** Obafluorin β -lactone (Obi) undergoes nonenzymatic hydrolysis to form a β -hydroxycarboxylic acid (Obi-COOH). **(b)** Annotated biosynthetic gene cluster of Obi identified from whole-genome sequencing of *P. fluorescens* ATCC 39502. Arrows represent coding sequences. A, acid adenylation; A_{Ar}, aryl acid adenylation; C, condensation; T, acid thiolation; T_{Ar}, aryl acid thiolation; TE, thioesterase; NRPS, nonribosomal peptide synthetase.

Conversion of PAPPa to β -OH-*p*-NO₂-homoPhe

Three enzymes (*ObiL*, *ObiG*, and *ObiH*) are required to convert PAPPa to β -OH-*p*-NO₂-homoPhe (Fig. 2a). We found that *ObiL*, a putative nonheme di-iron oxygenase, catalyzed the six-electron oxidation of the arylamine PAPPa to the aryl nitro PNPPa. Related examples of arylamine oxidases are found in the biosynthetic pathways of aureothin (AurF; GenBank [EJZ60582.1](#)) and chloramphenicol (CmlI; PDB [5HYH](#)). *ObiL* showed 24% and 18% sequence homology to AurF and CmlI, respectively (Supplementary Table 3; Supplementary Fig. 6a)^{33,34}. Similarly to *ObiL*, AurF acts early in the biosynthetic pathway of aureothin to oxidize *p*-aminobenzoic acid (PABA) to *p*-nitrobenzoic acid (PNBA). CmlI catalyzes the oxidation of the penultimate aniline intermediate to the aryl nitro as the final step of chloramphenicol biosynthesis³⁴. Treatment of recombinant *ObiL* with PAPPa under reaction conditions similar to those used to functionally characterize AurF resulted in an immediate color change of the reaction mixture from clear and colorless to sky blue (Supplementary Fig. 7)³³. Subsequent treatment of the *ObiL*-PAPPa reaction mixture with recombinant *ObiG* and *ObiH* in the presence of thiamine diphosphate (ThDP), pyridoxal phosphate (PLP), and L-Thr produced β -OH-*p*-NO₂-homoPhe, which was detected by LC-MS (Fig. 2b). The results of the triple-enzyme *ObiL*-*ObiG*-*ObiH* reaction confirmed that all three recombinant enzymes were capable of converting PAPPa directly to β -OH-*p*-NO₂-homoPhe, but the timing of each reaction required further clarification.

We discovered that *ObiG*, a ThDP-dependent phenylpyruvate decarboxylase, catalyzed the conversion of PNPPa to PNPAa (Fig. 2a; Supplementary Figs. 6b and 8). A colorimetric Purpald assay suggested that PNPPa is decarboxylated to the aldehyde PNPAa in the presence of ThDP and *ObiG* (Supplementary Fig. 9). A double-enzyme reaction with *ObiG* and *ObiH* in the presence of ThDP, PLP, and L-Thr revealed that PNPPa is directly converted to β -OH-*p*-NO₂-homoPhe, which was detected by LC-MS (Fig. 2c) and ¹H-NMR (Fig. 2d). PAPPa was not accepted as a substrate by *ObiG* and *ObiH*, suggesting that *ObiL* oxidized PAPPa before catalysis by *ObiG* and *ObiH* (Fig. 2c,d and Supplementary Fig. 10).

BLAST searches suggested that *ObiH* was a PLP-dependent threonine aldolase³⁵. Similar aldolases are involved in the biosynthesis of natural products containing β -hydroxy- α -amino acids, including the peptidyl nucleoside antibiotics A090289 A, caprazamycin A, and muramycin A1 (ref. 36), the 4-methyl-oxazoline antibiotics JBIR-34

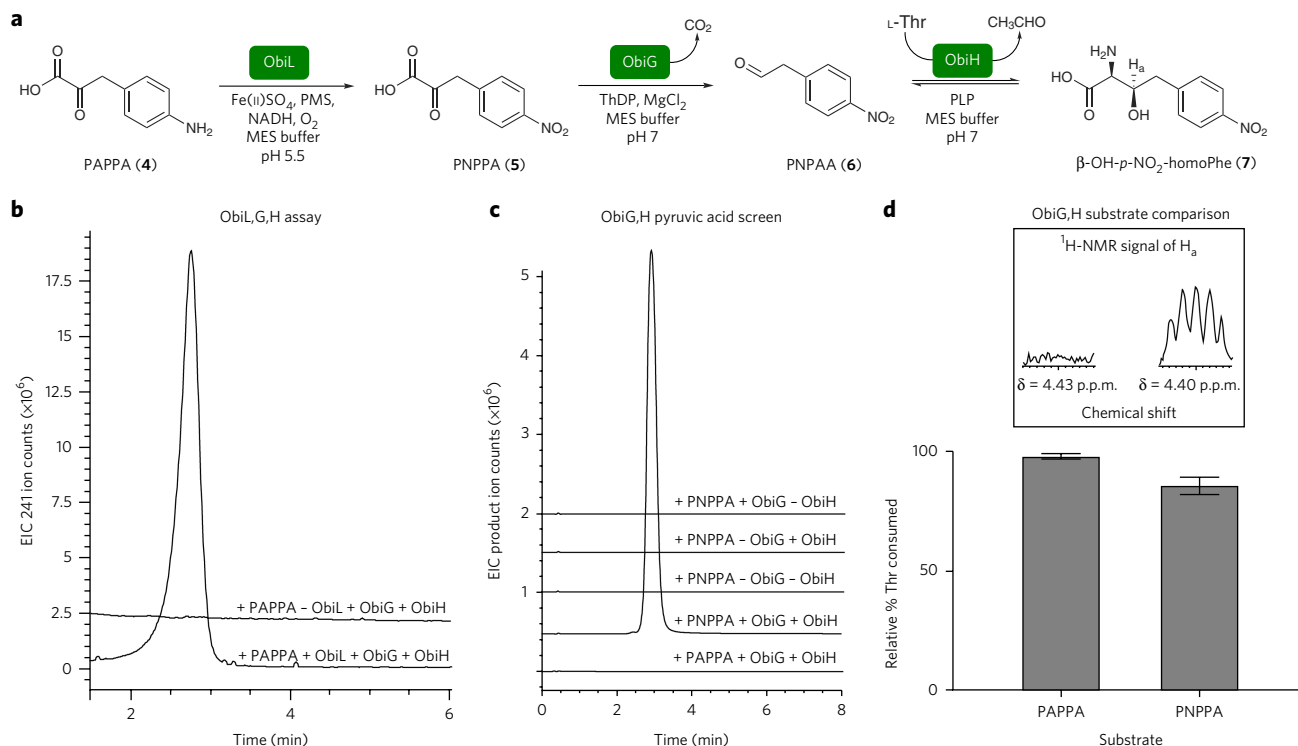


Figure 2 | Enzymatic conversion of PAPP to β -OH-*p*-NO₂-homoPhe using recombinant oxidase ObiL, decarboxylase ObiG, and aldolase ObiH.

(a) Scheme depicting the conversion of PAPP to β -OH-*p*-NO₂-homoPhe by tandem actions of ObiL, ObiG, and ObiH. (b) Extracted ion chromatograms (EICs) (β -OH-*p*-NO₂-homoPhe [M+H]⁺ ion m/z = 241) for the triple-enzyme reaction of PAPP with ObiL, ObiG, and ObiH. PMS, phenazine methosulfate. (c) EICs for the double-enzyme reaction of PAPP and PNPPA with ObiG and ObiH. EICs were generated by LC-MS in positive ion mode using single ion monitoring (SIM) for the predicted m/z values for the PAPP (m/z = 211) and PNPPA (m/z = 241) product [M+H]⁺ ions. Data represents a single measurement from two independent experiments. (d) The box shows a zoomed region of the ¹H-NMR spectra for the ObiG and ObiH double-enzyme reactions with L-Thr and PAPP or PNPPA after 3 h of equilibration in which H_a on C3 of β -OH-*p*-NO₂-homoPhe appears as a doublet of doublets. The bar graph depicts the relative percent of L-Thr remaining for each reaction, based on integration of the L-Thr methyl signal in the ¹H-NMR spectra. Error bars represent s.d. for three independent trials (P < 0.05).

and BE-32030 (ref. 37), and the α -methyl-L-Ser antibiotics amicetin and bamicitin³⁸. We showed that ObiH catalyzed a retro-aldol reaction on L-Thr, yielding acetaldehyde and a PLP-stabilized glycine enolate that underwent a crossed aldol reaction with PNPPA to yield (2S,3R)- β -OH-*p*-NO₂-homoPhe, setting the final stereochemistry found in the Obi β -lactone scaffold (Fig. 2a; Supplementary Figs. 6c and 11). Double-enzyme reactions with decarboxylase ObiG and aldolase ObiH monitored by LC-MS established that L-Thr is greatly preferred to D-Thr, L-allo-Thr, D-allo-Thr, L-Ser, D-Ser, and Gly as the amino acid substrate (Supplementary Fig. 10). The stereochemistry of β -OH-*p*-NO₂-homoPhe matches that of L-Thr, suggesting a preference of this substrate for giving the required enolate stereochemistry for the asymmetric aldol reaction (Supplementary Fig. 11). The ObiH PLP cofactor formed a glycine enolate quinonoid intermediate upon exposure to L-Thr, as observed by optical absorption spectroscopy (λ = 495 nm), which was stable throughout the entire protein expression and purification process, including dialysis (Supplementary Figs. 12 and 13). Kinetic analysis of the ObiH reaction by ¹H-NMR suggested strong product feedback inhibition of ObiH (Fig. 2d and Supplementary Fig. 10). We hypothesized that downstream utilization of the nonproteinogenic amino acid β -OH-*p*-NO₂-homoPhe by the ATP-consuming NRPS ObiF would drive the ObiH equilibrium toward product.

ObiF is a β -lactone-forming NRPS

The NRPS ObiF is a single-module, 209-kDa protein with C-A-T-TE-A_{Ar} catalytic domains (Supplementary Figs. 6e,f and 14), and ObiD is a separate aryl acyl carrier protein (T_{Ar}) (Supplementary Fig. 6d).

We examined whether recombinant ObiF and ObiD, together, could convert β -OH-*p*-NO₂-homoPhe and 2,3-DHB to Obi (Fig. 3a). Prior to running *in vitro* enzyme reactions, we primed recombinant ObiD and ObiF with 4-phosphopantetheine transferase (Sfp) and coenzyme A to install the phosphopantetheine post-translational modification on the T domains. Treatment of β -OH-*p*-NO₂-homoPhe and 2,3-DHB with holo-ObiF and holo-ObiD, together, in the presence of ATP and Mg²⁺ gave clean conversion to the ring-closed Obi β -lactone and the corresponding ring-opened hydrolysis product Obi-COOH, presumably originating from the nonenzymatic hydrolysis of Obi (Fig. 3b; Supplementary Fig. 15; Supplementary Table 4).

Unlike the lipstatin¹³, ebelactone⁷, salinosporamide¹⁴, and oxazolomycin¹⁵ PKS and NRPS assembly lines, ObiF has a TE domain in the termination module. Primary sequence analysis of the ObiF TE domain revealed that the expected catalytic serine of the conserved GxSxG motif is replaced by a cysteine at residue 1141; this mutation has been found previously in type II TEs but is rarely reported for type I TEs^{16,39,40}. We prepared phosphopantetheinylated C1141S and C1141A ObiF variants to test the importance of C1141 in β -lactone formation. The C1141S mutation converted ObiF from a ' β -lactone synthase' to a classic hydrolase that produced only Obi-COOH from the *in vitro* reaction of the holo-ObiF C1141S mutant and holo-ObiD with β -OH-*p*-NO₂-homoPhe and 2,3-DHB (Fig. 3b,c; Supplementary Fig. 15; Supplementary Table 5). The C1141A mutation brought the catalytic activity of ObiF to a near halt, with only trace amounts of Obi-COOH detected by LC-MS (Fig. 3b,c; Supplementary Fig. 15; Supplementary Table 6). Presumably, the

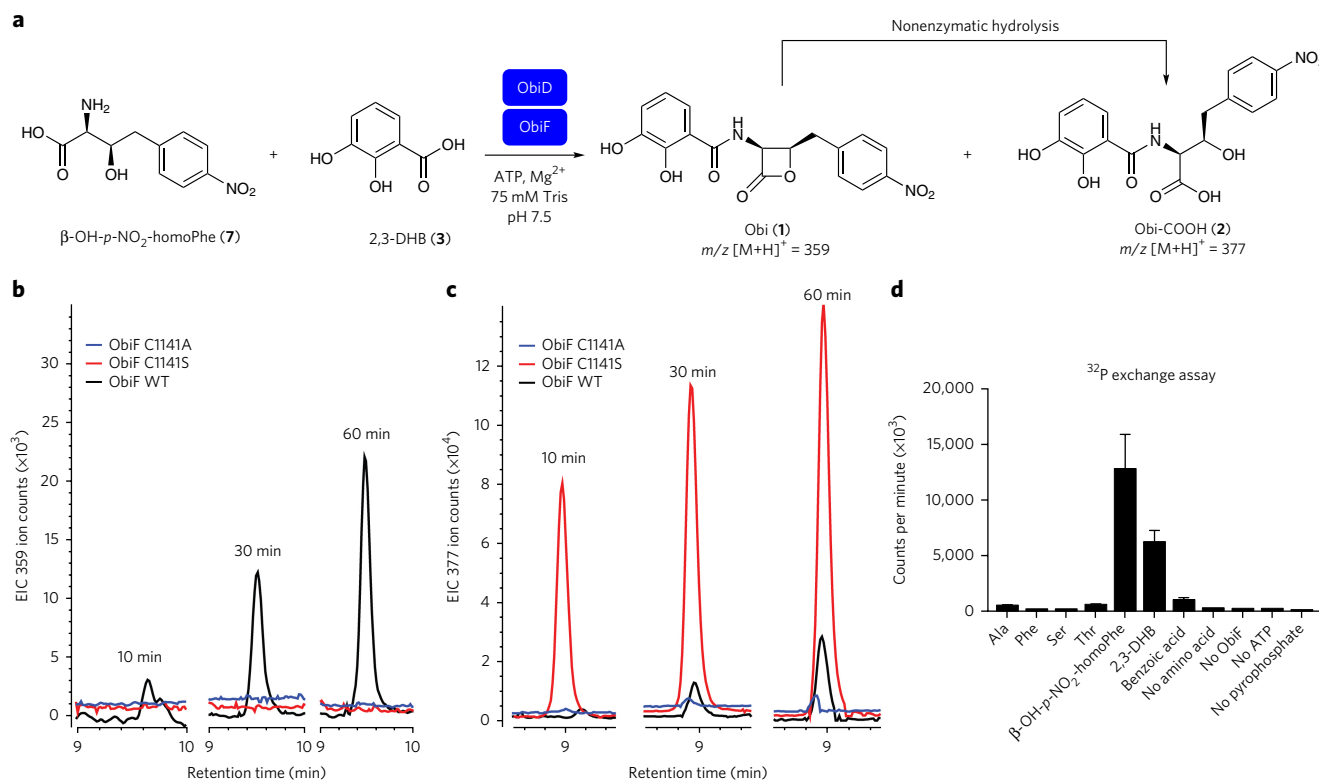


Figure 3 | Enzymatic conversion of β -OH-*p*-NO₂-homoPhe and 2,3-DHB to Obi β -lactone using recombinant NRPS assembly line ObiF and ObiD.

(a) Scheme depicting the enzymatic conversion of β -OH-*p*-NO₂-homoPhe and 2,3-DHB to Obi (m/z = 359 for [M+H]⁺) and Obi-COOH (m/z = 377 for [M+H]⁺). (b) EICs (m/z = 359) for reactions of 2,3-DHB, β -OH-*p*-NO₂-homoPhe, and holo-ObiD with holo-ObiF wild type (WT; black), holo-ObiF C1141S mutant (red), and holo-ObiF C1141A mutant (blue). (c) EICs (m/z = 377) for reactions of 2,3-DHB, β -OH-*p*-NO₂-homoPhe, and holo-ObiD with holo-ObiF WT (black), holo-ObiF C1141S mutant (red), and holo-ObiF C1141A mutant (blue). (d) ATP-[³²P]PP_i exchange assay to detect carboxylic acid adenylation by ObiF. The graph shows counts per minute (c.p.m.) for each carboxylate substrate. Additional amino acid substrates (Arg, Asn, Asp, Cys, Glu, Gln, Gly, His, Ile, Leu, Lys, Met, Orn, Pro, Trp, Val) are not included in the graph because of the negligible c.p.m. observed for them. Error bars represent s.d. for three independent trials ($P < 0.05$).

penultimate 2,3-DHB- β -OH-*p*-NO₂-homoPhe thioester stalled on the T domain, as acyl transfer to the TE domain of the holo-ObiF C1141A mutant was blocked. The trace Obi-COOH that was detected by LC-MS is predicted to form via slow hydrolysis of the T-domain thioester. Prolonged exposure (24 h) of the holo-ObiF C1141A mutant and holo-ObiD to 2,3-DHB, β -OH-*p*-NO₂-homoPhe, and ATP resulted in steady production of Obi-COOH, supporting the theory that the NRPS module was functional but required slow, nonenzymatic hydrolysis of the 2,3-DHB- β -OH-*p*-NO₂-homoPhe T-domain thioester for turnover. We also expressed, purified, and functionally characterized recombinant ObiF and ObiD homologs from *C. shinanonensis* SAY3 (Supplementary Fig. 16; Supplementary Table 2). Reaction of holo-ObiF and holo-ObiD from *C. shinanonensis* with β -OH-*p*-NO₂-homoPhe and 2,3-DHB gave enzyme-, time-, and ATP-dependent conversion to Obi and Obi-COOH (Supplementary Fig. 17; Supplementary Table 7). The ability of recombinant holo-ObiF and holo-ObiD from *C. shinanonensis* to catalyze the *in vitro* conversion of β -OH-*p*-NO₂-homoPhe and 2,3-DHB to Obi with similar efficiency to the *P. fluorescens* enzymes suggests that the NRPS module would be functional when expressed and phosphopantetheinylated in *C. shinanonensis*.

We employed an ATP-[³²P]pyrophosphate (PP_i) exchange assay to test for reversible acyl-adenylate intermediate formation and showed that the ObiF A domains are highly selective for the activation of β -OH-*p*-NO₂-homoPhe and 2,3-DHB (Fig. 3d). Proteinogenic amino acids, including the β -hydroxy (L-Thr and L-Ser) and aromatic (L-Phe) amino acids, produced signals comparable to those of control reactions lacking an amino substrate.

Analysis of the ObiF primary sequence with NRPSpredictor2 software predicted that the embedded A domain would adenylate L-Thr and that the terminal A_{Ar} domain would adenylate 2,3-DHB (Supplementary Table 8)⁴¹. The specificity-conferring sequence⁴² of the embedded ObiF A domain (D/A/W/G/C/G/L/I) showed similarity to the consensus signature sequences for A domains that activate L-Thr (D/F/W/N/I/G/M/V) and L-Phe (D/A/W/T/I/A/V), with some key differences that help account for accommodation of the sterically larger β -OH-*p*-NO₂-homoPhe substrate. In comparison to L-Phe A domains, Thr at position 4 is changed to a Gly, extending the active site pocket to accommodate the chain extended homoPhe. In comparison to L-Thr A domains, Phe at position 2 is changed to Ala, and Asn at position 4 is changed to Gly, which leaves room to accommodate the benzyl group. Both L-Phe and L-Thr A domains have an Ile at position 5, which is changed to Cys in ObiF. The Cys might interact favorably with the electron-deficient nitrophenyl group (Supplementary Fig. 18).

β -Lactone formation requires Cys1141

We propose a direct role for the ObiF TE domain in the formation of the Obi β -lactone ring as the final biosynthetic step, releasing the mature cyclized product from the NRPS assembly line. The ObiF 2,3-DHB- β -OH-*p*-NO₂-homoPhe T thioester must be shuttled from the T domain and loaded to the TE domain via energy-neutral transthioesterification, where it resides as the C1141 thioester and undergoes C3-OH-to-C1 cyclization to release Obi β -lactone. The weaker C-S bond of the thioester, compared to the C-O bond of an oxo-ester, is likely required to make strained β -lactone ring

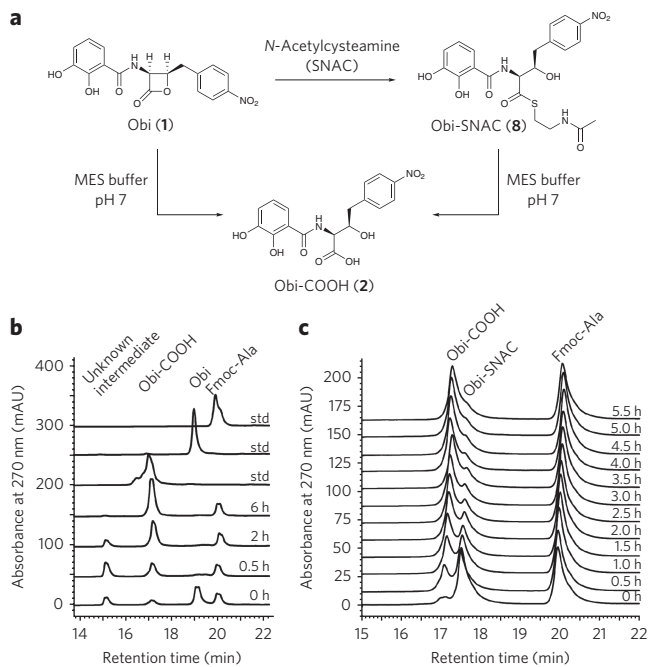


Figure 4 | ObiF catalysis is required for β -lactone ring formation.

(a) Scheme depicting the synthesis of Obi-*N*-acetylcysteamine (SNAC) and the hydrolysis of Obi and Obi-SNAC to Obi-COOH. Treatment of Obi with neat SNAC results in quantitative conversion to Obi-SNAC. Hydrolysis of Obi (b) and Obi-SNAC (c) at pH 7 in MES buffer was monitored by HPLC with detection by optical absorbance spectroscopy at 270 nm using an Fmoc-L-Ala internal standard. Peak identities were confirmed using LC-MS and comparison to authentic standards. Std, pure analytical standard. The traces represent single measurements from two independent experiments.

formation thermodynamically favorable⁴³. The higher ground state energy of the Cys thioester, compared to that of the Ser oxo-ester, is also predicted to increase the rate of alcohol addition to the carbonyl group, which is likely the rate-determining step. The thermodynamic requirements for β -lactone formation are not met by the oxo-ester intermediate created by the ObiF C1141S mutant, resulting in the accumulation of only the hydrolysis product Obi-COOH (Fig. 3c). The ObiF C1141A mutant also failed to generate detectable concentrations of Obi β -lactone, which supports the theory that TE catalysis is required in β -lactone ring formation.

To further probe the role of the TE domain in β -lactone formation, we prepared an *N*-acetylcysteamine (SNAC) thioester (Obi-SNAC, 8) and compared its rate of hydrolysis with that of Obi β -lactone. Dissolving purified Obi from *P. fluorescens* culture supernatant in neat SNAC thiol resulted in quantitative conversion to Obi-SNAC thioester (Fig. 4a; Supplementary Fig. 19). We monitored the hydrolysis of Obi (Fig. 4b) and Obi-SNAC (Fig. 4c) in MES buffer at pH 7 by HPLC with detection by optical absorbance spectroscopy at 270 nm. We observed rapid hydrolysis of Obi in MES buffer at pH 7 with a half-life <30 min and immediate formation of two products (retention times of 15.2 min and 17.2 min) with optical absorbance spectra that are consistent with the presence of an aryl nitro group (Fig. 4b). Comparison to an analytical standard of Obi-COOH purified from *P. fluorescens* culture supernatant confirmed the peak at retention time 17.2 min to be Obi-COOH. The peak at retention time 15.2 min is an uncharacterized intermediate that converted directly to Obi-COOH. Previous studies of Obi hydrolysis using ¹H-NMR in D₂O/CD₃CN (1:4) showed that an oxazoline intermediate forms and hydrolyzes to give *O*-(2,3-dihydroxybenzoyl)- β -OH-*p*-NO₂-homoPhe²⁶. We observed that Obi-SNAC hydrolyzed directly to Obi-COOH in MES buffer at pH 7 with a half-life of

3.2 h (Fig. 4c). The rate of Obi-SNAC hydrolysis was comparable to that of the ObiF C1141A mutant (Fig. 3c; Supplementary Fig. 15), which further supports the role of the ObiF TE domain as a catalyst for β -lactone ring formation.

A model for Obi biosynthesis

Our *in vitro* characterization of ObiD, ObiF, ObiG, ObiH, and ObiI; computational prediction of ObiA, ObiB, ObiC, ObiE, ObiJ, ObiK, and ObiL function; and previously reported isotope labeled precursor studies^{27,28} enabled us to propose a complete model for Obi biosynthesis in *P. fluorescens* (Fig. 5). Carbon flux to Obi biosynthesis starts from the primary metabolites erythrose-4-phosphate and PEP, which are converted to chorismate by the upregulated DAHP synthase ObiB and the endogenous shikimate pathway. ObiA, ObiC, and ObiE convert chorismate to 2,3-DHB, and ObiI, ObiJ, and ObiK convert it to PAPP. Aryl amine oxidase ObiL converts PAPP to PNPPA, and ThDP-dependent decarboxylase ObiG converts PNPPA to PNPA. Threonine aldolase ObiH establishes equilibrium between PNPA, L-Thr, acetaldehyde, and β -OH-*p*-NO₂-homoPhe. The ATP-consuming NRPS ObiF activates β -OH-*p*-NO₂-homoPhe as the acyl adenylate with the embedded A domain and covalently loads the amino acid as a phosphopantetheinyl thioester on the embedded T domain. 2,3-DHB is similarly activated as the acyl adenylate by the C-terminal A_{Ar} domain and loaded on the stand-alone T_{Ar} domain ObiD as a phosphopantetheinyl thioester. The C domain is predicted to catalyze amide bond formation between the α -amino group of the ObiF β -OH-*p*-NO₂-homoPhe-T-thioester and the carbonyl of the ObiD 2,3-DHB-T_{Ar}-thioester, releasing free T_{Ar} (ObiD) and forming the 2,3-DHB- β -OH-*p*-NO₂-homoPhe-T-thioester. Transthioesterification to the active site C1141 of the TE domain leads to cyclization of the β -hydroxythioester to the corresponding β -lactone, releasing Obi and turning over the NRPS assembly line for another round of catalysis (Supplementary Fig. 20).

DISCUSSION

There are some general features from the ObiF TE domain that might be genetic and biochemical signatures for β -lactone biosynthesis. A GxCxG motif is likely required for β -lactone ring formation. The active site Cys provides a favorable kinetic and thermodynamic scenario for strained-ring formation, proceeding through a more reactive thioester compared to the traditional oxo-ester intermediate. The advantages of activated TE domain thioesters as substrates for NRPS product release might extend to other strained-ring systems including β -lactams. Recently, the biosynthetic gene cluster for the monobactam sulfazecin was reported to encode an NRPS with a terminal type I TE domain that contains an active site Cys residue⁴⁴. GxCxG motifs are rare among type I TE domains¹⁶. Analysis of the TE protein family (PF00975) revealed that 264 out of 3,863 members (6.8%) contained a GxCxG motif, whereas the remaining 93.2% contained a GxSxG motif. Analysis of the top 20,000 hits from standard protein BLAST search of the full ObiF sequence against nonredundant protein sequences in the NCBI database revealed that 13,944 sequences possessed a GxSxG motif, 1,240 sequences contained a GxCxG motif, and 194 sequences contained a GHCxG motif. Homologous ObiF TE domains from *P. fluorescens*, *C. shinanonensis*, and *B. diffusa* all contain a GHCAAG motif and show >50% total sequence homology (Supplementary Fig. 16; Supplementary Table 9). Comparison to a diverse sampling of PKS and NRPS TE domains, including pyochelin and AB3403 TE domains with GxCxG motifs, showed that the ObiF TE domain is sequence unique (Supplementary Fig. 21; Supplementary Table 10)^{39,40}. Homology modeling predicted overall structural similarities to canonical type I α/β hydrolases, with the conserved Asp residue of the catalytic triad moved from β -strand 6 to β -strand 7 (Supplementary Fig. 22)¹⁶. Future mechanistic and structural studies will help elucidate the catalytic roles of the His-Cys-Asp triad

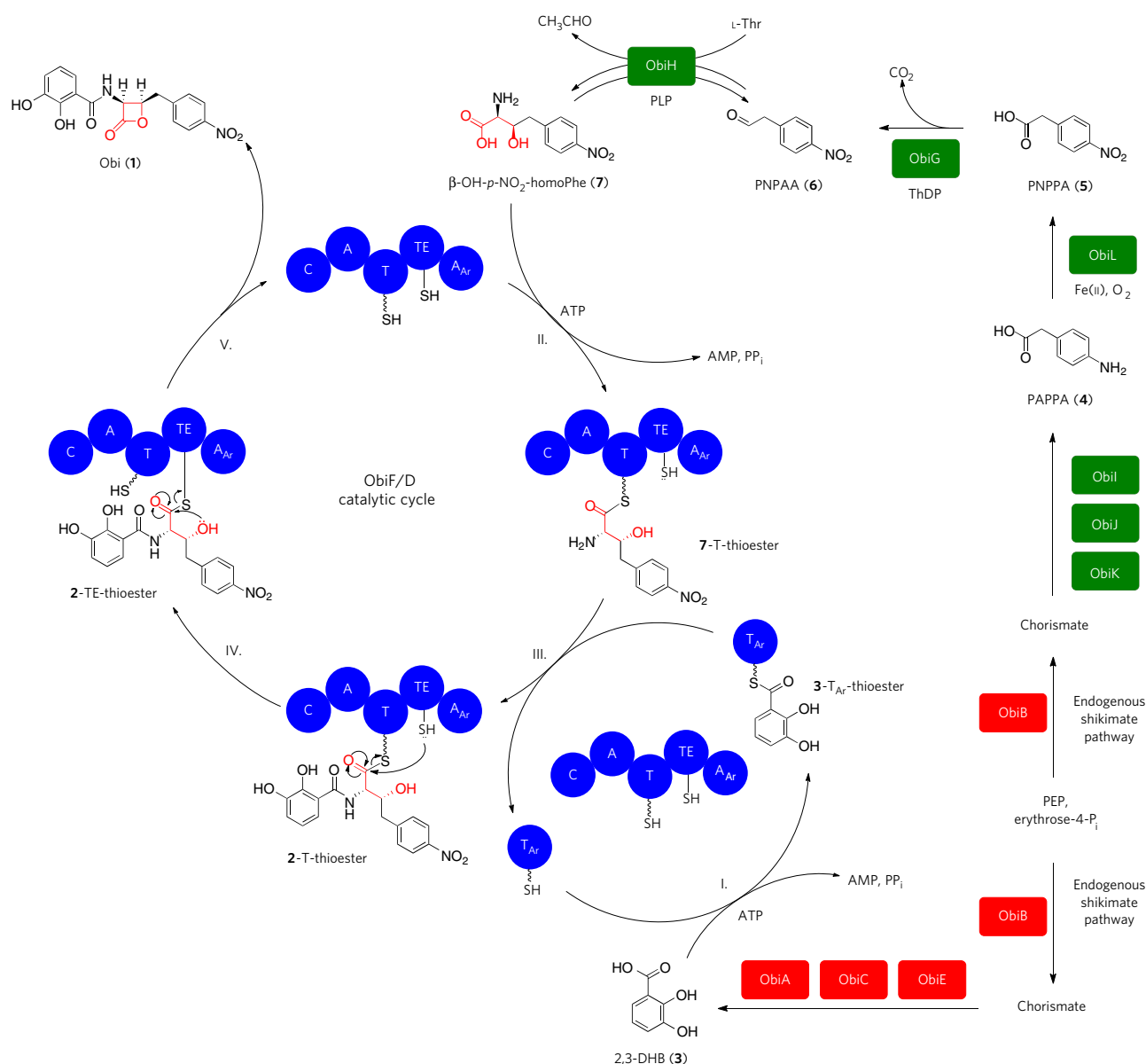


Figure 5 | Model for Obi biosynthesis and the catalytic cycle of NRPS assembly line ObiF. Obi directs carbon flux from the primary metabolic pool to chorismate via the endogenous shikimate pathway. ObiA, ObiC, and ObiE convert chorismate to 2,3-DHB. ObiL, ObiJ, and ObiK convert chorismate to PAPPa. ObiL converts PAPPa to PNPPa. ObiG converts PNPPa to PNPPaA. ObiH establishes equilibrium between L-Thr, PNPPaA, acetaldehyde, and β -OH-*p*-NO₂-homoPhe. ObiF and ObiD are the NRPS assembly line components that convert β -OH-*p*-NO₂-homoPhe and 2,3-DHB to Obi at the cost of two ATP molecules. The NRPS catalytic cycle starts (I) with A_{Ar} catalyzed activation and loading of 2,3-DHB to the T_{Ar} domain as the phosphopantetheinyl thioester. Similarly, the ObiF embedded A domain activates and loads β -OH-*p*-NO₂-homoPhe to the ObiF T domain (II). The C domain catalyzes amide bond formation giving the 2-T-thioester (III). Transthioesterification to active site Cys1141 of the TE domain (IV) leads to cyclization releasing Obi β -lactone (V).

and will clarify whether substrate preorganization plays a role in formation of the strained four-membered β -lactone ring^{10,20}.

Inclusion of the 2,3-DHB-activating A_{Ar} domain as part of the ObiF NRPS module might be optional. In the *P. fluorescens* and *C. shinanonensis* ObiF NRPSs the A_{Ar} domain is included in the CDS as the C-terminal catalytic domain. In the *B. diffusa* NRPS, the A_{Ar} domain is predicted to be a stand-alone enzyme that activates 2,3-DHB as the acyl adenylate for *in trans* transfer to the T_{Ar} carrier protein in analogy to the enterobactin NRPS machinery (EntF, EntE, and EntB) (Supplementary Fig. 23). An MbtH domain is encoded in the ObiF primary sequence between the TE and A_{Ar} domains. The MbtH-like protein YbdZ is encoded in the enterobactin biosynthetic gene cluster and has been shown to enhance the adenylation activity

of EntF⁴⁵. The ObiF MbtH-like domain might play a similar role to YbdZ. Recent structural and biophysical studies of NRPS modules with C-A-T-TE domains have provided insight into the motions and dynamics of individual domains during coordinated catalysis by the enzyme assembly lines^{40,46}. The TE domain is highly dynamic and is distal from the T domain until invoked for final product release. In the case of ObiF, this would leave A_{Ar} free to catalyze the formation of the 2,3-DHB acyl adenylate and keep the T_{Ar} carrier protein loaded as a 2,3-DHB phosphopantetheinyl thioester. In the case of enterobactin biosynthesis, the A_{Ar} and T_{Ar} domains, EntE and EntB, respectively, are separate from the main NRPS module, EntF, which activates and loads Ser as a T-domain phosphopantetheinyl thioester⁴⁵. The full-length EntF is required for trimerization of Ser,

amide bond formation with 2,3-DHB, and cyclization to the macrolactone⁴⁷. Thus, only A_{Ar} and T_{Ar} domains have been shown to be active *in trans*, supporting the existence of ObiF NRPS modules with and without the A_{Ar} domain as part of the main NRPS module.

It is conceivable that other catalytic domains with active site Cys residues might catalyze cyclization of β -hydroxythioesters to β -lactones. For example, an uncharacterized ketosynthase (KS) domain resides at the C terminus of the ebelactone PKS termination module⁷. KS domains undergo transthioesterification reactions with T domain thioesters to form intermediate KS-Cys thioesters on a conserved Cys–His–His triad⁴⁸. KS domains typically catalyze thio-Claisen condensations with malonyl T-domain thioesters, but might also be capable of β -lactone ring forming condensations. The termination module of the hybrid PKS–NRPS assembly line for oxazolomycin features a C-terminal C domain of unknown function¹⁵. Condensation domains typically catalyze amide bond formation between two T domain thioesters, but there are documented cases of C domains catalyzing cyclization during product release such as in cyclosporin A and enniatin biosynthesis²⁰. The oxazolomycin terminal C domain might catalyze the intramolecular condensation of a T domain thioester to the spirocyclic β -lactone. The salinosporamide biosynthetic gene cluster encodes for several uncharacterized enzymes, including a stand-alone KS domain, SalC, that might participate in bicyclic β -lactone formation^{4,14}. We predict that more β -lactone-forming “logic gates” will be discovered as part of NRPS–PKS modular assembly lines and possibly as autonomous catalysts^{10,16,49}.

Obi represents a novel structural class of β -lactone antibiotics from nature's chemical inventory. The Obi biosynthetic machinery characterized in this study offers a versatile platform for the chemoenzymatic synthesis of β -lactones from β -hydroxy- α -amino acid precursors. Here, the new amino acid β -OH-*p*-NO₂-homoPhe was produced by a threonine aldolase, possibly representing a general strategy for producing β -hydroxy- α -amino acid precursors to peptide β -lactones. Fe(II)/ α -KG-dependent oxygenases and non-heme di-iron oxygenases are also known to produce β -hydroxy- α -amino acids via direct C–H bond activation chemistry on free amino acids^{50,51}. Hydroxylation of amino acid side chains can also take place while loaded as thioesters on NRPS T domains, making A domain substrate prediction insufficient for the potential incorporation of β -hydroxy- α -amino acids into NRP scaffolds⁵². Our work shows that β -hydroxy- α -amino acids can be directly incorporated into peptide scaffolds on an NRPS assembly line, leading to peptides capped at the C terminus with a β -lactone warhead. Our genetic and biochemical characterization of the Obi biosynthetic pathway in *P. fluorescens* will guide future genome mining efforts and expands the known chemistry of NRPS and PKS product release mechanisms to include strained 4-membered ring formation^{10,20,24}. Our chemoenzymatic approach to C-terminal β -lactone peptides has the potential for applications in precursor-directed feeding studies and NRPS engineering to produce targeted peptide β -lactone inhibitors of proteases for the treatment of diseases associated with microbial and viral infections^{5,6,25}, cancer⁴, and obesity³.

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METHODS

Methods, including statements of data availability and any associated accession codes and references, are available in the [online version of the paper](#).

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Author Contributions

T.A.W., J.E.S., and M.R.R. wrote the paper and prepared the supplementary information. T.A.W. oversaw all of the experiments. J.E.S. and M.R.R. cloned and purified ObiG, ObiH, and ObiL. J.E.S. functionally characterized ObiG, ObiH, and ObiL. M.R.R. cloned, purified, and functionally characterized ObiD and ObiF. J.E.S. and M.R.R. purified and characterized all compounds. T.A.W. and N.K.P. isolated *Pseudomonas fluorescens* gDNA, analyzed sequencing data, and annotated the Obi biosynthetic gene cluster. N.K.P. performed protein homology modeling and helped with preparation of the supplementary information.

Competing financial interests

The authors declare no competing financial interests.

Additional information

Any supplementary information, chemical compound information and source data are available in the [online version of the paper](http://www.nature.com/reprints/index.html). Reprints and permissions information is available online at <http://www.nature.com/reprints/index.html>. Publisher's note: Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations. Correspondence and requests for materials should be addressed to T.A.W.

ONLINE METHODS

Strains, materials, and instrumentation. *P. fluorescens* ATCC 39502 was purchased from ATCC, and glycerol stocks were made from cells grown in Bennett's media and stored at -80°C . DNA primers were purchased from Integrated DNA Technologies. Herculase II DNA polymerase kit was purchased from Agilent. T4 DNA ligase and NheI were purchased from New England BioLabs. T4 Polynucleotide Kinase and FastDigest (FD) restriction enzymes NdeI, HindIII, and DpnI were purchased from Thermo Scientific. TOP10 *Escherichia coli* cells were purchased from Invitrogen and BL21-Gold(DE3) *E. coli* cells were purchased from Agilent. PCR was performed on a Bio-Rad MyCycler thermal cycler. Nickel-nitrilotriacetic acid (Ni-NTA) agarose was purchased from Invitrogen. Any kD SDS-PAGE gels were purchased from Bio-Rad. Plasmid DNA sequencing was performed by Genewiz. Buffers, salts, and chemical reagents were purchased from Sigma-Aldrich, unless otherwise stated. Fully deuterated MES buffer and D_2O were purchased from Cambridge Isotope Laboratories. PAPPa was purchased from Tractus Chemistry. PNPPa was purchased from Frinton Laboratories, Inc. PPA and phenazine methosulfate were purchased from Acros Organics. ^{32}P -radiolabeled sodium pyrophosphate was purchased from PerkinElmer. Samples for HPLC and LC-MS were prepared in $0.45\text{ }\mu\text{m}$ PTFE MiniUniPrep vials from Agilent. All pH measurements were recorded using an Orion Star A111 pH meter and a PerpHecT ROSS micro combination pH electrode from Thermo Scientific. Analytical HPLC was performed using Beckman Coulter SYSTEM GOLD 126 solvent module, 508 autosampler, and 168 detector with a Phenomenex Luna $5\text{ }\mu\text{m}$ C18(2) 100A column, $250 \times 4.6\text{ mm}$ with guard column. All preparatory HPLC was performed using a Beckman Coulter SYSTEM GOLD 127P solvent module and 168 detector with a Phenomenex Luna 10u C18(2) 100A column, $250 \times 21.20\text{ mm}$, $10\text{ }\mu\text{m}$ with guard column. Analytical and prep HPLC were performed with mobile phases of 0.1% TFA in (A) water and (B) acetonitrile, and data were processed using 32 Karat software, version 7.0. Low-resolution LC-MS was performed on an Agilent 6130 single quadrupole LC-MS with G1313 autosampler, G1315 diode array detector, and 1200 series solvent module. A Phenomenex Gemini C18 column, $50 \times 2\text{ mm}$, $5\text{ }\mu\text{m}$ with guard column was used for LC-MS separations. LC-MS mobile phases were 0.1% formic acid in (A) water and (B) acetonitrile, and data were processed using G2710 ChemStation software. NMR was performed on Varian Unity Plus-300 MHz, Varian Unity Inova-500 MHz, and Varian Unity-600 MHz (with cold probe) instruments. Optical absorption spectroscopy was performed on a Cary 50 fit with an autosampler and water Peltier thermostat system using 1 cm quartz cuvettes. High-resolution MS spectra were collected using an LTW-Velos Pro Orbitrap at the Donald Danforth Plant Science Center, St. Louis, MO.

Sequencing of *P. fluorescens* ATCC 39502 Genomic DNA. The *P. fluorescens* ATCC 39502 gDNA was isolated using a Qiagen DNeasy Blood & Tissue kit following the provided instructions. The gDNA was sequenced by Ambry Genetics (Aliso Viejo, CA) using Illumina MiSeq. Contigs containing ORFs were assembled, annotated, and deposited in GenBank (Accession #s [KX134682–KX134695](#)) with assistance from Cofactor Genomics (St. Louis, MO).

Cloning, expression, and purification of ObiG, ObiH, ObiL, ObiD, ObiF, ObiF C1141S, ObiF C1141A, CS-ObiD and CS-ObiF. Gene sequences were amplified from either *P. fluorescens* ATCC 39502 gDNA or *C. shinanonensis* DSM 23277 (CS) gDNA (isolated using Qiagen DNeasy Blood & tissue kit following provided instructions) using forward and reverse primers designed for each gene (Supplementary Tables 11–13) and a Herculase II Fusion DNA Polymerase kit. The resulting PCR fragments were purified by gel electrophoresis. PCR products and pET28a vector were separately digested with either FD-NdeI and FD-HindIII or NheI and FD-HindIII. Cut DNA was purified using a QIAquick Gel Extraction Kit. Ligation reactions were carried out overnight at 16°C using T4 DNA Ligase. Ligation mixtures were purified and transformed into electrocompetent *E. coli* TOP10 cells. Clones were selected for on LB agar containing $50\text{ }\mu\text{g/ml}$ kanamycin. Sequencing of selected colonies revealed clones containing the desired constructs. Plasmids were purified and used to transform electrocompetent *E. coli* BL21 (DE3) cells.

For protein expression, a 5 ml culture of *E. coli* BL21 harboring the appropriate plasmid was grown overnight in LB containing $50\text{ }\mu\text{g/ml}$ kanamycin with

agitation at 37°C . A $200\text{ }\mu\text{l}$ aliquot of this culture was used to inoculate 500 ml of terrific broth (12 g/L tryptone, 24 g/L yeast extract, 5 g/L glycerol, 17 mM KH_2PO_4 , and 72 mM K_2HPO_4) containing $50\text{ }\mu\text{g/ml}$ kanamycin. The culture was grown at 37°C with agitation until OD_{600} reached approximately 0.4 . The culture was cooled in an ice bath for 20 min , then $500\text{ }\mu\text{l}$ of a sterile 0.5 M IPTG solution was added. Culture was then incubated with agitation for 18 h (at 15°C for ObiG/ObiF/CS-ObiF, and at 20°C for ObiD/CS-ObiD/ObiH/ObiL). From this point on, all protein purification steps were performed at 4°C . Cells were harvested by centrifugation at $5,000\text{ r.p.m.}$ for 20 min . Supernatant was discarded, and cell pellets were each suspended in 40 ml cold lysis buffer (50 mM K_2HPO_4 pH 8.0 , 500 mM NaCl, 5 mM β -mercaptoethanol, 20 mM imidazole, and 10% glycerol). Cell suspensions were transferred to 50 ml Falcon tubes and flash frozen in liquid nitrogen. Frozen cells were thawed and gently rocked for 30 min before being mechanically lysed using an Avestin EmulsiFlex-C5 cell disruptor. Cell lysate was centrifuged at $45,000\text{ r.p.m.}$ for 35 min and supernatant was incubated with pre-washed Ni-NTA resin for 30 min . Resin was washed twice with 40 ml lysis buffer and then eluted five times with 10 ml elution buffer (50 mM K_2HPO_4 pH 8.0 , 500 mM NaCl, 5 mM β -mercaptoethanol, 300 mM imidazole, and 10% glycerol). Fractions containing the majority of protein, as judged by SDS-PAGE with Coomassie blue visualization (Supplementary Fig. 5), were combined in $10,000\text{ MWCO}$ SnakeSkin dialysis tubing from Thermo Scientific and soaked overnight in 1.8 L phosphate buffer (50 mM K_2HPO_4 pH 8.0 , 150 mM NaCl, 1 mM DTT). Dialyzed protein solution was concentrated using an appropriately sized spin filter (EMD Millipore Amicon Ultra 15 ml Centrifugal Filters). Concentrated protein solutions were flash frozen in liquid nitrogen and stored at -80°C .

Mutagenesis of WT-ObiF to ObiF C1141S and ObiF C1141A. Mutagenesis primers (Supplementary Tables 11 and 12) were phosphorylated using T4 Polynucleotide Kinase according to product instructions. Primers were taken directly from the phosphorylation mixture for use in PCR amplification. PCR was carried out using a Herculase II Fusion DNA Polymerase kit. After amplification, $1\text{ }\mu\text{l}$ FastDigest DpnI was added to PCR reaction mixture to digest template plasmid with wild-type gene. The reaction was incubated at 37°C for 60 min , and plasmids were then purified using gel electrophoresis. Purified plasmids were ligated overnight at 16°C using T4 DNA ligase, then transformed into electrocompetent *E. coli* TOP10 cells. Plasmids were sequenced to confirm the desired mutations.

ObiG Purpald reactions. A pre-incubation solution was made with $10\text{ }\mu\text{M}$ ThDP, $10\text{ }\mu\text{M}$ MgCl_2 , 25 mM MES buffer pH 7.5 , and $10\text{ }\mu\text{M}$ ObiG from a freshly thawed frozen stock. The enzyme pre-incubation solution incubated at room temperature for 5 min . In a separate 1.5 ml Eppendorf tube, the pyruvic acid substrate (PAPPa, PNPPa, PHPPa, or PPA) was diluted to 1 mM in 25 mM MES pH 7.5 buffer. The substrate mix was transferred in one aliquot into the pre-incubation solution. The final concentrations of reagents were $10\text{ }\mu\text{M}$ ObiG, $5\text{ }\mu\text{M}$ ThDP, $5\text{ }\mu\text{M}$ MgCl_2 , 1 mM pyruvic acid substrate, and 25 mM MES buffer pH 7.5 . Aliquots of the enzyme solution were added to an equal volume of a 20 mg/ml solution of Purpald in 1 M aqueous NaOH at various time points. The resulting solution was mixed and then left open to air to develop the purple color (Supplementary Fig. 9). All reactions and time points were single trials.

Detection of ObiH quinonoid by optical absorption spectroscopy. In a 1.5 ml Eppendorf tube, a master mix of $50\text{ }\mu\text{M}$ ObiH from a freshly thawed frozen stock, $50\text{ }\mu\text{M}$ PLP, and 25 mM MES buffer was prepared and analyzed by optical absorption spectroscopy scanning from 200 to 600 nm . Three solutions at a final volume of $500\text{ }\mu\text{l}$ were analyzed as single trials: (1) L-Thr was added to the master mix at 1 mM final concentration; (2) PAA was added to the master mix at 1 mM final concentration; (3) L-Thr and PAA were added to the master mix at 1 mM final concentration each. Samples were then analyzed by optical absorption spectroscopy scanning from 200 – 600 nm in order to observe the quinonoid peak, which absorbs at $\sim 495\text{ nm}$ (Supplementary Fig. 12).

Combined ObiG and ObiH double enzyme reactions. Three solutions were prepared using individual Eppendorf tubes. First, an ObiG solution containing

10 μM ThDP, 10 μM MgCl_2 , and 10 μM ObiG was adjusted to 333 μl final volume using 25 mM MES buffer pH 7.5 and allowed to pre-incubate for 5 min. Second, a pyruvic acid solution, 1 mM final concentration, was adjusted to 333 μl with 25 mM MES buffer pH 7.5. Third, an ObiH solution containing 10 μM PLP, 1 mM amino acid (L-Thr, D-Thr, L-Ser, D-Ser, Gly, L-alloThr, or D-alloThr), and 10 μM ObiH was adjusted to 333 μl final volume. After the pre-incubation time was complete, the three solutions were combined in a single Eppendorf tube and allowed to rest for 3 h at room temperature before quenching with MeCN to crash the enzyme. The mixtures were centrifuged and the supernatants were analyzed by LC-MS using a gradient of 0% B held for 5 min then 0% B to 95% B over 10 min using single ion monitoring for the expected ions in positive ion mode (retention times and observed product ions: $\beta\text{-OH-homoPhe}$, 1.7 min, $[\text{M}+\text{H}]^+ = 196$; $\beta\text{-OH-}p\text{-OH-homoPhe}$, 0.8 min, $[\text{M}+\text{H}]^+ = 212$; $\beta\text{-OH-}p\text{-NO}_2\text{-homoPhe}$, 2.9 min, $[\text{M}+\text{H}]^+ = 241$; $\beta\text{-OH-}p\text{-NH}_2\text{-homoPhe}$, not observed; **Fig. 2c** and **Supplementary Fig. 10**). Product masses were confirmed by high-resolution LC-MS analysis. For NMR analysis the same samples were prepared using fully deuterated MES buffer. After equilibration for 3 h, the reactions were quenched with TFA to a pH ~ 2 . The samples were then flash frozen in liquid nitrogen, lyophilized to dryness, resuspended in 750 μl of D_2O , and centrifuged to pellet any insoluble particulate. The soluble supernatants were then transferred to NMR tubes and analyzed (**Fig. 2d**; **Supplementary Fig. 10**). All reactions were performed in triplicate.

Purification of $\beta\text{-OH-}p\text{-NO}_2\text{-homoPhe}$ from the ObiG/ObiH reaction. The ObiG/ObiH double-enzyme reaction described above was scaled up with respect to the substrates to produce enough material for NMR analysis. The concentrations of reaction components were 10 μM ObiG, 10 μM ThDP, 10 μM MgCl_2 , 10 μM ObiH, 10 μM PLP, 5 mM L-Thr, 5 mM PNPPA, and 25 mM pH 7.5 MES buffer, with a final reaction volume of 1 ml. After 3 h the reaction was quenched with TFA to a pH ~ 2 , centrifuged at 13,000 r.p.m. for 2 min, flash frozen in liquid nitrogen, and lyophilized to dryness. The resulting solid was dissolved in 1:10 MeCN:H₂O, filtered, and purified by preparatory HPLC using a solvent gradient of 0% B held for 5 min then 0% B to 100% B over 10 min ($\beta\text{-OH-}p\text{-NO}_2\text{-homoPhe}$ retention time = 14.3 min). The desired TFA salt of $\beta\text{-OH-}p\text{-NO}_2\text{-homoPhe}$ product was isolated as a white powder. ¹H NMR (500 MHz, D_2O) δ (p.p.m.) = 8.24 (d, $J = 8.4$ Hz, 2 H), 7.55 (d, $J = 9.0$ Hz, 2 H), 4.49 – 4.43 (ddd, $J = 3.6, 4.8, 10.2$ Hz, 1 H), 3.98 (d, $J = 4.8$ Hz, 1 H), 3.20 (dd, $J = 3.6, 14.4$ Hz, 1 H), 2.99 (dd, $J = 10.2, 13.8$ Hz, 1 H); ¹³C NMR (126 MHz, D_2O) δ (p.p.m.) = 160.6, 136.0, 134.9, 119.7, 113.2, 59.1, 47.5, 28.9; HRMS (ESI) calculated for $\text{C}_{10}\text{H}_{11}\text{N}_2\text{O}_5$: 241.0819 $[(\text{M}+\text{H})^+]$, found peak when extracting ions for 241.0795–241.0843. See **Supplementary Note** for tabulated NMR data and copies of spectra.

Combined ObiG, ObiH, and ObiL triple enzyme reaction. A solution of 100 μM ObiL from a freshly thawed frozen stock, 3 mM phenazine methosulfate (PMS), 1 mM PAPPa, 100 μM iron(II) sulfate, and 25 mM NADH was prepared in 25 mM MES buffer at pH 5.5 at a final volume of 500 μl (refs. 33,53). NADH was added last to initiate the reaction and the mixture was incubated at room temperature for 1 h. Upon addition of NADH, the color of the reaction changed from light yellow to sky blue (**Supplementary Fig. 7**). In a separate Eppendorf tube, a 500 μl solution was prepared containing 10 μM ObiG, 10 μM ThDP, 10 μM MgCl_2 , 10 μM ObiH, 10 μM PLP, 1 mM L-Thr, and 25 mM MES buffer at pH 7. The pH of the ObiL solution changed from 5.5 to 7.0 over the course of the reaction. The control assay reached an end pH of 6.0 and was adjusted to a pH of 7 using 1 M NaOH before addition of the ObiG/ObiH solution. The ObiG/ObiH solution was added directly to the ObiL solution, allowed to react for 2 h at room temperature, and quenched with MeCN to crash the enzymes. The mixtures were centrifuged and the supernatants were analyzed by LC-MS using a gradient of 0% B held for 5 min then 0% B to 95% B over 10 min. Ion counts for expected products were extracted from total ion chromatograms in positive ion mode (**Fig. 2b**; retention times and observed product ions: $\beta\text{-OH-}p\text{-NO}_2\text{-homoPhe}$, 1.7 min, $[\text{M}+\text{H}]^+ = 241$). The $\beta\text{-OH-}p\text{-NO}_2\text{-homoPhe}$ ion was confirmed by HRMS. Reaction was performed as two independent trials.

Combined ObiF and ObiD double enzyme reaction. Phosphopantetheinylation of apo-ObiD and apo-ObiF (wild type, C1141S and C1141A mutants, and

C. shinanonensis homologs) was carried out in separate 500 μl solutions containing 180 μM CoASH, 5 mM DTT, 10 mM MgCl_2 , 400 nM Sfp, 25 μM apo enzyme, and 75 mM Tris-HCl at pH 7.5 (ref. 54). Reactions were left at room temperature for 2 h, and then used directly as the source of holo-ObiD and holo-ObiF. Reactions were performed in duplicate at 500 μl total volume. Full reactions contained 5 mM ATP, 5 mM DTT, 1 mM 2,3-DHB, 1 mM $\beta\text{-OH-}p\text{-NO}_2\text{-homoPhe}$, 1 μM holo-ObiF (ObiF, CS-ObiF, ObiF-C1141A, or ObiF-C1141S), 1 μM holo-ObiD, and 75 mM Tris-HCl at pH 7.5. For reactions with CS-ObiF, the holo-ObiD homolog from *C. shinanonensis* (CS-ObiD) was used as the aryl acyl adenylating enzyme. Control reactions were also prepared by replacing various components with an equivalent volume of 75 mM Tris-HCl buffer. Three control experiments were performed for the ObiF, ObiF-C1141S, ObiF-C1141A, CS-ObiF reactions: (–) ObiD; (–) 2,3-DHB/(–) $\beta\text{-OH-}p\text{-NO}_2\text{-homoPhe}$; and (–) ATP. Two additional controls were prepared as either (–) ObiF or (–) ObiF/(–) ObiD (**Supplementary Tables 4–7**). Reactions were performed at room temperature. At 10, 30, and 60 min time points, a 50 μl aliquot from each reaction was quenched with 50 μl acidic MeCN (acidified with HCl) to give a final pH of approximately 3.5. Mixtures were centrifuged for 2 min at 13,000 r.p.m. to remove precipitated enzyme, and the supernatants were analyzed by LC-MS using a gradient of 5% solvent B to 95% solvent B over 20 min. Ion counts for Obi and Obi-COOH were extracted from total ion chromatograms in positive ion mode (retention times and observed ions: Obi, 9.5 min, $[\text{M}+\text{H}]^+ = 359$; Obi-COOH, 9.0 min, $[\text{M}+\text{H}]^+ = 377$) (**Fig. 3b,c**). Reaction mixtures were also analyzed by HPLC using a gradient of 5% B to 95% B over 20 min, 95% B to 100% B over 3 min, and 100% B to 5% B over 2 min at a flow rate of 1 ml/min with optical absorbance detection at 270 nm (**Supplementary Fig. 15**). Product ions and retention times were confirmed using purified standards of Obi and Obi-COOH isolated from *P. fluorescens* ATCC 39502 fermentations (**Supplementary Fig. 19**).

[³²P]PPI exchange assay. Reaction mixtures (650 μl) contained 2 μM ObiF, 5 mM acid substrate, 1 mM ATP, 1 mM MgCl_2 , 40 mM KCl, 1 mM DTT, 5 mM $\text{Na}^{32}\text{P}[\text{PPI}]$ (3.3×10^5 c.p.m./ml), and 50 mM Tris-HCl (pH 8)⁵⁵. Mixtures were incubated at room temperature for 30 min, then three 200 μl aliquots were removed and each were quenched with 500 μl of a charcoal suspension (100 mM NaPP₆, 350 mM HClO_4 , and 16 g/L powdered charcoal). The mixtures were shaken, then centrifuged at 13,000 r.p.m. for 3 min. Charcoal pellets were washed with 750 μl of wash solution (100 mM NaPP₆, 350 mM HClO_4) and centrifuged again for 3 min. Washing step was repeated once more, followed by suspension of charcoal pellets in 1.5 ml EcoLite(+) scintillation fluid from MP Biomedicals. Charcoal-bound radioactivity was measured on a Beckman Coulter LS 6500 scintillation counter. Graph of results indicates measured c.p.m. (**Fig. 3d**). All proteinogenic amino acids were L-stereoisomers.

Isolation of Obi and Obi-COOH. Obi and the hydrolysis product Obi-COOH were isolated from *Pseudomonas fluorescens* ATCC 39502 fermentations as previously described^{22,23}. Bennet's Agar slants (1 g/L yeast extract, 1 g/L beef extract, 2 g/L NZ amine, 10 g/L glucose, 15 g/L agar) were inoculated with streaks of *P. fluorescens* ATCC 39502 from a frozen glycerol stock. The slants were incubated at 25 °C for 48 h. 5 ml of sterile saline was added to the top of the slant and shaken gently. Inoculated saline (1 mL) was transferred to 100 ml of sterile media (5 g/L yeast extract, 5 g/L glucose, 0.1 g/L $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$, 0.1 g/L $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$, 200 ml soil filtrate extract from campus of Washington University in St. Louis, 800 ml tap water, autoclaved). This starter culture was incubated with shaking at 225 r.p.m. at 25 °C for 24 h. 5 ml of starter culture was transferred to each 3 L baffled flask containing 500 ml of the same media, and the cultures were incubated with shaking at 225 r.p.m. at 25 °C for 17 h. The cultures were pooled and centrifuged at 5,000 r.p.m. at 4 °C for 25 min to pellet cells. The supernatant was pooled and acidified to pH 3 with 1M aqueous HCl. The pooled supernatant was saturated with EtOAc then extracted with three 100 ml volumes of EtOAc. Extractions were pooled and dried using rotary evaporation to yield a gray-brown solid. The solid was dissolved in 10 ml acetonitrile, filtered, and concentrated by rotary evaporation under reduced pressure to yield 221 mg of clear, brown oil. The oil was dissolved in 10 ml of acetonitrile, filtered, and purified with Prep HPLC with a gradient of 5% solvent B to 95% solvent B over 20 min. The two largest peaks were isolated

which upon NMR analysis proved to be Obi (retention time = 23.0 min) and Obi-COOH (retention time = 20.5 min), both isolated as brown oils. Obi:¹H NMR (500 MHz, methanol-*d*₄) δ (p.p.m.) = 8.12 (d, *J* = 9.0 Hz, 2 H), 7.49 (d, *J* = 8.4 Hz, 2 H), 7.28 (d, *J* = 9.6 Hz, 1 H), 6.99 (d, *J* = 6.6 Hz, 1 H), 6.78 (t, *J* = 1.0 Hz, 1 H), 5.83 (d, *J* = 6.0 Hz, 1 H), 5.08 (td, *J* = 5.7, 9.0 Hz, 1 H), 3.40 (dd, *J* = 9.0, 15.0 Hz, 1 H), 3.21 (dd, *J* = 5.1, 14.7 Hz, 1 H); ¹³C NMR (126 MHz, methanol-*d*₄) δ (p.p.m.) = 171.8, 170.2, 150.5, 148.7, 147.8, 145.8, 131.6, 130.5, 129.9, 128.1, 124.9, 120.8, 120.4, 119.5, 116.4, 79.2, 60.5, 36.8; HRMS (ESI) calculated for C₁₇H₁₅N₂O₇: 359.0874 [(M+H)⁺], found peak when extracting ions for 359.0858–359.0930. See **Supplementary Note** for tabulated NMR data and copies of spectra. Obi-COOH:¹H NMR (500 MHz, methanol-*d*₄) δ (p.p.m.) = 8.15 (d, *J* = 8.4 Hz, 2 H), 7.50 (d, *J* = 8.4 Hz, 2 H), 7.37 (d, *J* = 9.6 Hz, 1 H), 6.98 (d, *J* = 9.0 Hz, 1 H), 6.78 (t, *J* = 8.1 Hz, 1 H), 4.75 (d, *J* = 2.4 Hz, 1 H), 4.54 (ddd, *J* = 2.1, 5.6, 8.0 Hz, 1 H), 3.03 (t, *J* = 1.0 Hz, 1 H), 2.96 (t, *J* = 1.0 Hz, 1 H); ¹³C NMR (126 MHz, methanol-*d*₄) δ (p.p.m.) = 173.8, 170.7, 149.3, 148.3, 147.8, 147.4, 131.8, 124.7, 124.6, 124.5, 120.2, 120.1, 120.0, 117.8, 73.3, 63.4, 57.7, 41.8; HRMS (ESI) calculated for C₁₇H₁₅N₂O₈: 375.0834 [(M+H)⁺], found peak when extracting ions for 375.0796–375.0872. See **Supplementary Note** for tabulated NMR data and copies of spectra. See **Supplementary Figure 19** for LC–MS analysis of *P. fluorescens* culture supernatant, purified Obi, and purified Obi-COOH.

Synthesis of Obi-SNAC. 1 mg purified Obi β-lactone was dissolved in 5 mg SNAC. The solution was left at room temperature overnight, and then diluted with 0.5 ml MeCN and purified by reverse phase preparatory HPLC (gradient of 5% B to 95% B over 20 min at flow rate of 10 ml/min). Product-containing fraction (retention time 21.2 min) was concentrated by rotary evaporation, dissolved in 0.5 ml DMSO, and transferred to a 1.5 ml Eppendorf tube. Solution was lyophilized overnight and dissolved in methanol-*d*₄ for NMR analysis. Obi-SNAC forms a 2:1 mixture of rotamers in methanol-*d*₄. The NMR data for the major rotamer is given below. ¹H NMR (600 MHz, methanol-*d*₄) δ (p.p.m.) = 9.08 (d, *J* = 8.7 Hz, 1 H), 8.13 (d, *J* = 8.7 Hz, 1 H), 7.48 (d, *J* = 7.9 Hz, 1 H), 7.41 (d, *J* = 7.9 Hz, 1 H), 6.99 (d, *J* = 7.9 Hz, 1 H), 6.80 (t, *J* = 7.9 Hz, 1 H), 4.75

(d, *J* = 8.7 Hz, 1 H), 4.59 (t, *J* = 6.7 Hz, 1 H), 3.05 – 2.94 (m, 4 H), 1.89 (s, 3 H); ¹³C NMR (151 MHz, methanol-*d*₄) δ (p.p.m.) = 201.7, 170.3, 148.6, 148.2, 147.3, 147.2, 131.7, 124.4, 120.8, 120.2, 119.9, 72.7, 64.2, 41.6, 39.8, 29.4, 22.5; LC-MS (ESI) calculated for C₂₁H₂₄N₃O₈S: 478.1 [(M+H)⁺], found: 478.1 (retention time 9.1 min). See **Supplementary Note** for tabulated NMR data and copies of spectra. See **Supplementary Figure 19** for LC–MS analysis of purified Obi-SNAC.

Hydrolysis of Obi and Obi-SNAC to Obi-COOH. Solutions of purified Obi (1 mM) and Obi-SNAC (1 mM) were prepared in MES buffer at pH ~7. Aliquots (100 μl) were taken at various time points analyzed by HPLC (gradient of 5% B to 95% B over 20 min, 95% B to 100% B over 3 min, and 100% B to 5% B over 2 min at a flow rate of 1 ml/min) with detection by optical absorbance spectroscopy at 270 nm (**Fig. 4b,c**). HPLC peak identities were confirmed by LC–MS and retention times were normalized using an Fmoc-Ala internal standard. Experiments were performed in duplicate.

Data availability. *P. fluorescens* ATCC 39502 sequencing data for the Obi biosynthetic gene cluster is deposited in GenBank (Accession numbers [KX134682](#), [KX134683](#), [KX134684](#), [KX134685](#), [KX134686](#), [KX134687](#), [KX134688](#), [KX134689](#), [KX134690](#), [KX134691](#), [KX134692](#), [KX134693](#), [KX134694](#), [KX134695](#)). All other data supporting the findings of this study are contained in the published article (and its supplementary information files) or are available from the corresponding author upon request.

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