Methanolysis of Flambamycin. The Constitution of Methyl Eurekanate

By W. DAVID OLLIS* and CHRISTOPHER SMITH

(Department of Chemistry, The University, Sheffield S3 7HF)

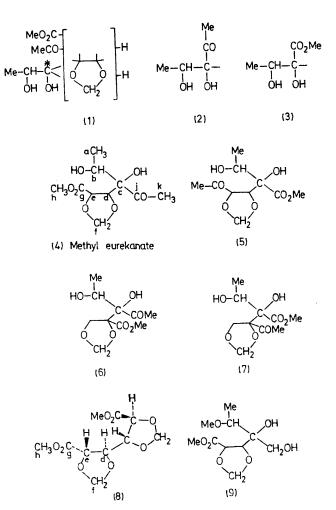
and DEREK E. WRIGHT

(Research Laboratories, May & Baker Ltd., Dagenham, Essex RM10 7XS)

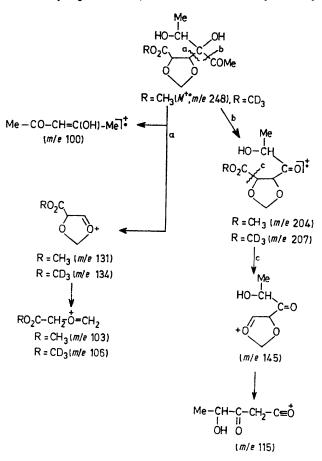
Summary Methyl eurekanate, one of the products from the mild acidic methanolysis of the antibiotic flambamycin, has been shown to have the constitution (4).

MILD acidic methanolysis¹ of the antibiotic flambamycin² yields a complex mixture of at least nine products of which eight have been fully identified.¹ We now report the constitution (4) for the ninth methanolysis product, methyl eurekanate, whose structure is crucial to the complete structure of the antibiotic and relevant to the presence of an orthoester grouping in the antibiotic.

The following functional assignments were initially considered for five of the seven oxygen atoms of methyl eurekanate, $C_{10}H_{16}O_7$ (M+, m/e 248) mainly on the basis of the following spectroscopic information [1H n.m.r.† and i.r. $(CHCl_3)$]: (a) CO₂Me (δ_{Me} 3.78, ν_{co} 1720 cm⁻¹); (b) MeCO



 $(\delta_{Me} 2.28, \nu_{CO} 1750 \text{ cm}^{-1});$ (c) OCH_AH_BO $(\delta_{A} 5.10, \delta_{B} 4.89,$ J_{AB} 0 Hz). The remaining two oxygen atoms were shown to be associated with two hydroxy-groups (δ 4.15 and 2.58, removed by D₂O addition) located in a secondary-tertiary



SCHEME. Part of the mass spectral fragmentation patterns of methyl eurekanate (4) and trideuteriomethyl eurekanate.

glycol system. Supplementary evidence reported below led eventually to the partial structure (1) for methyl eurekanate.

The presence of a methoxycarbonyl group in methyl eurekanate (1) was suspected by its transformation into ethyl eurekanate, $C_{11}H_{18}O_7$ (M+- CH₃CO, m/e 219) by ethanolic hydrogen chloride (1.5% w/v, room temp., 3 h). Its mild acid hydrolysis (5N HCl, room temp., 18 h) yielded eurekanic acid characterised (acetic anhydride, toluene-psulphonic acid, room temp., 18h) as eurekanic acid diacetate, ${\rm C_9H_{12}O_5(OAc)_2}~(M^+\!\cdot,~m/e~318).$

The methylenedioxy group in methyl eurekanate was confirmed by acid hydrolysis (5N HCl, 100 °C, 6 h) giving

† All compounds have been fully characterised spectroscopically. Unless otherwise indicated n.m.r. data refer to CDCl₃ solutions.

TABLE

Comparison of the ¹H and ¹³C chemical shifts (p.p.m. downfield from Me₄Si) for corresponding atoms in methyl eurekanate (4) and dimethyl 2,3:4,5-di-O-methylene-galactarate⁴ (8). The positions of the atoms are indicated by the letters in the formulae (4) and (8).

	Chemical shifts δ (CDCl ₃)									
	C_aH_a	C⊾H	Cc	CdH	C _e H	$C_{f}H_{2}$	Cg	$C_{h}H_{s}$	C ₁	C _k H ₃
(4) (¹ H)	1.03v	4·18 ^v		4.66₩	4 ·68₩	5.10, y $4.89y$		3.78		$2\cdot 28$
(8) (¹ H)				4·27×	4.61≖	5·24, ^z 5·06 ^z		3.78		
(4) (¹³ C)	17.4	68.4	$84 \cdot 2$	$74 \cdot 6$	81.5	$95 \cdot 9$	171.7	52.8	207.2	$26 \cdot 1$
(8) (¹³ C)				$75 \cdot 1$	78.9	96.8	170.6	52.6		

 $^{v}A_{3}X$ system, J_{AX} 6.5 Hz; ^{w}AB system, J_{AB} 6 Hz; $^{x}AA'BB'$ system, $J_{AA'}$ 5.5, $J_{BB'}$ 0, $J_{AB} = J_{A'B'} = 4.0$ Hz; 4 ^{y}AB system, J_{AB} 0 Hz; ^{z}AB system, $J_{AB} < 1.0$ Hz.

formaldehyde, isolated from the hydorlysate as its 2,4-dinitrophenylhydrazone (m.p. 164 °C, 55% yield). The ¹H n.m.r. spectral characteristics of the methylenedioxy-group $(\delta_{A} 5.10, \delta_{B} 4.89; J_{AB} 0 Hz)$ demonstrated its location in a 1,3-dioxolan ring.³

The presence of two hydroxy-groups in the function CH_3 -CH(OH)- $C^*(OH)$ was supported by the characterisation of methyl eurekanate as a mono-acetate (acetic anhydride-pyridine, room temp., 18 h) C₁₀H₁₅O₆(OAc), m.p. 87 °C, $(M^+ - CH_3CO, m/e 247)$ and a diacetate (acetic anhydride-toluene-p-sulphonic acid, room temp., 24 h), $C_{10}H_{14}O_5(OAc)_2$ (M⁺· - 1, m/e 331). Methyl eurekanate and trichloracetyl isocyanate gave a bis-trichloracetylcarbamate. The ¹H n.m.r. spectrum of methyl eurekanate clearly showed the presence of an A₃X system (δ_A 1.03, δ_X 4.18, J_{AX} 6.5 Hz) characteristic of the secondary-tertiary glycol grouping, CH₃-CH(OH)-C*(OH). This was confirmed by periodate cleavage (aqueous sodium metaperiodate, room temp., 35 min) yielding acetaldehyde, isolated as its 2,4-dinitrophenylhydrazone (m.p. 168 °C, 54% yield). The expected downfield shift of H_x was observed in the ¹H n.m.r. spectra of methyl eurekanate monoacetate (δ_x 5.39), diacetate (δ_x 5.57), and bis-trichloracetylcarbamate [δ_x 5.36, in (CD₃)₂CO]. These results proved that neither of the two unlocated hydrogen atoms in the partial structure (1) could be located on C*. Thus the presence of either the grouping (2) or the grouping (3) could now be considered in

relation to the partial structure (1). This leads to four possible constitutional formulae (4), (5), (6), or (7) for methyl eurekanate.

Comparison of the ¹H and ¹³C n.m.r. spectra (Table) of methyl eurekanate with that of the model, dimethyl 2,3:4,5di-O-methylene-galactarate⁴ (8) clearly demonstrated that the constitutions (6) and (7) could be excluded. However, it was still not possible on the basis of the foregoing evidence to determine the relative positions of the methoxycarbonyl and the acetyl groups associated with the remaining two constitutions (4) and (5). The elucidation of the constitution (4) for methyl eurekanate was finally achieved by a detailed examination of the high-resolution mass spectra of methyl eurekanate, trideuteriomethyl eurekanate, and methyl eurekanate diacetate. Only a small part of this extensive mass spectral study is recorded in the Scheme and the information given has been restricted to that which establishes the location of the methoxycarbonyl group.

It may be noted that a eurekanate residue does not occur as such in the antibiotic flambamycin. Flambamycin belongs to a family of structurally related antibiotics which includes the everninomycins,⁵ avilamycin,⁶ and curamycin⁷, and there is an interesting structural correspondence between methyl eurekanate (4) and the methyl ester⁵a,b (9) obtained by methanolysis of everninomycin-B and everninomycin-D.

(Received, 20th February 1976; Com. 179.)

¹ W. D. Ollis, C. Smith, and D. E. Wright, J.C.S. Chem. Comm., 1974, 882.

¹ W. D. Ollis, C. Smith, and D. E. Wright, J.C.S. Chem. Comm., 1974, 882.
² L. Ninet, F. Benazet, Y. Charpentie, M. Dubost, J. Florent, J. Lunel, D. Mancy, and J. Preud'homme, Experientia, 1974, 30, 1270.
³ R. C. Cookson and T. A. Crabb, Tetrahedron Letters, 1964, 679; R. C. Cookson, T. A. Crabb, J. J. Frankel, and J. Hudec, Tetrahedron Suppl. 7, 1966, 355; R. C. Cookson and T. A. Crabb, Tetrahedron 1968, 24, 2385; S. Sternhell Quart. Rev., 1969, 23, 236.
⁴ I. J. Burden and J. F. Stoddart, J.C.S. Perkin I, 1975, 675.
⁵ (a) A. K. Ganguly, O. Z. Sarre, D. Greeves, and J. Morton, J. Amer. Chem. Soc., 1975, 97, 1982; (b) A. K. Ganguly and A. K. Saksena, J. Antibiotics, 1975, 28, 707; (c) A. K. Ganguly and S. Szmulewicz, ibid., p. 710.
⁶ F. Buzzetti, F. Eisenberg, H. N. Grant, W. Keller-Schierlein, W. Voser, and H. Zähner, Experientia, 1968, 24, 320.
⁷ O. L. Galmarini and V. Deulofen. Tetrahedron, 1961, 15, 76; E. G. Gros, V. Deulofeu, O. L. Galmarini, and B. Frydman. Experi-

⁷ O. L. Galmarini and V. Deulofeu, Tetrahedron, 1961, 15, 76; E. G. Gros, V. Deulofeu, O. L. Galmarini, and B. Frydman, Experientia, 1968, 24, 323; V. Deulofeu and E. G. Gros, Anales de Quimica, 1972, 68, 789.

348