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EPOXY-ISOSIDOL: A NEW DITERPENOID FROM SIDERITIS BIFLORA*

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Key Word Index-Sideritis biflora; Labiatae; tetracyclic diterpenes; new ent-15,6,16-epoxykaurane derivative.

Abstract—The new diterpenoid ent- 3β -acetoxy- 7α ,18-dihydroxy- 15β ,16-epoxykaurane has been isolated from Sideritis biflora. Twelve previously known ent-kaurane derivatives have also been obtained from the same source.

Plant. Sideritis biflora Port. et Rig. = S. bourgaeana (Boiss. et Reut.) Font Quer = S. leucantha Cav. var. paucidentata Wk. Source. Low areas in the S.E. of the Iberian Peninsula (near Tobarra, Albacete). Previous work. None.

Present work. From the aerial parts of this plant have been isolated 12 diterpenoids previously described: ent-18-acetoxy-3β,7α-dihydroxykaur-16-ene (linearol), ent-18acetoxy-3 β ,7 α -dihydroxykaur-15-ene (isolinearol), ent-3 β acetoxy-7 α ,18-dihydroxykaur-16-ene (sidol), ent-3 β -acetoxy-7 α ,18-hydroxykaur-15-ene (isosidol), ent-3 β ,7 α ,18trihydroxykaur-16-ene (foliol), ent-3β,7α,18-trihydroxykaur-15-ene (isofoliol) [1], ent-3β,7α,15β,18-tetrahydroxykaur-16-ene (leucanthol), ent-3 β , 7α , 17, 18-tetrahydroxykaur-15-ene (isoleucanthol) [2], ent-18-acetoxy- 3β , 7α , 15 β -trihydroxykaur-16-ene, ent-3 β -acetoxy-7 α , 15 β , 18-trihydroxykaur-16-ene [3], ent-18-acetoxy- 3β , 7α -dihydroxy- 15β , 16-epoxykaurane (epoxy-isolinearol) [4] and ent-3 β ,7 α ,18-trihydroxy-15 β ,16-epoxykaurane (epoxy-isofoliol) [5], plus a new compound (1, $C_{22}H_{34}O_5$). The NMR spectrum of 1 showed signals for two tertiary methyls, one acetyl, one > CHOH and one methyl and one proton both on an epoxide ring. The chemical shifts and shapes of the above signals were identical with those previously observed for epoxy-isolinearol [4]; on the other hand, the signals for the C-3 and C-18 protons appeared at δ 5.00 and 3.18 respectively, indicating attachment of the acetoxyl group on C-3; this was also confirmed by the small diamagnetic shift ($\Delta\delta - 0.07$) undergone by the C-4 axial methyl group in compound 1 with respect to epoxy-isolinearol.



The new diterpenoid (1) is thus $ent-3\beta$ -acetoxy- 7α , 18dihydroxy- 15β , 16-epoxykaurane (epoxy-*iso*sidol). The above structure was confirmed by alkaline hydrolysis of 1 to give compound 2, identical (physical and spectroscopic data) with epoxy-*iso*foliol [5].

EXPERIMENTAL

For general details on extraction and separation of diterpenes from Sideritis, see Ref. [4]; for most experimental details, see part 13 [6] of this series. The previously described compounds have been identified by their physical (mp $[\alpha]_D$) and spectroscopic (IR. NMR) data and by comparison with authentic samples. Epoxy-isosidol (1). Mp 177-181° (Me₂CO-n-hexane); $[\alpha]_0^{20} - 3.6^\circ$ (EtOH: c 0.74). IR v_{max}^{KBr} cm⁻¹: 3450, 3300 (-OH); 1730, 1250 (-OAc). NMR (60 MHz, CDCl₃): δ 5.00 (1 H, m, $W_4 = 18$ Hz, C-3), 3.80 (1 H, m, $W_4 = 6$ Hz, C-7), 3.18 (2 H, AB q, J 12 Hz, C-18 methylene), 3.10 (1 H, s, C-15, epoxide ring), 2.06 (3 H, s, -OAc), 1.44 (3 H, s, Me on epoxide ring), C-Me singlets at 1.06 and 0.68. MS (70 eV) m/e rel. int.): 378 M⁺ (13), 363 (3), 360 (5), 318 (5), 300 (16), 270 (100. base peak), 255 (56), 242 (17), 105 (60), 95 (64). (Found; C, 69.93; H, 9.03. C₂₂H₃₄O₅ requires: C, 69.81; H, 9.05%

Alkaline hydrolysis of 1. Treatment of compound 1 (30 mg) with 5% KOH-EtOH (15 ml) for 24 hr at room temp. gave 2 (23 mg): mp 248-250° (Me₂CO); $[\alpha]_D^{20°} + 27.8°$ (EtOH; c 0.72). IR ν_{max}^{KBt} cm⁻¹: 3460, 3380 (-OH). MS: 336 m/e (M⁺). (Found: C, 71.31; H, 9.37. Calc. for C₂₀H₃₂O₄: C, 71.39; H, 9.59%). Compound 2 was identical in all respects with epoxy-isofoliol [5].

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ISOPRENOID TRIETHER LIPIDS FROM CALDARIELLA

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In the Caldariella group of extreme thermoacidophile bacteria [1], the major lipids [2,3] are macrocyclic sn-1,2-diethers of glycerol derived from a range of C_{40} diols with unique ω, ω' -biphytanyl structures. In the hope of shedding some light on their biosynthesis, we investigated minor lipid components, here shown to be triethers of glycerol with partly-or fully-saturated C_{20} (phytanyl) alcohols.

Separation of minor lipids

From ca 350 g of lyophilized cells of the MT-3 and MT-4 strains of Caldariella acidophila, some 25 g of total lipid was obtained, of which ca 0.5 g (accumulated over a series of experiments) was directly soluble in hexane and separated accordingly. Chromatography of this fraction afforded the thianaphthenequinone described elsewhere [4] as major component, a complex mixture of hydrocarbons, and some 30 mg of glycerol triethers. These on TLC (Si gel; hexane-Et₂O, 7:3) afforded two components, R_f 0.5 and 0.4; the former predominated in the triether fraction from MT-4.

Tri-O-phytanylglycerol (1)

The IR spectrum of the less polar triether showed only saturated alkyl and ether bands (v_m 1110, 1375-1385, 1465, 2850-2950 cm⁻¹) and the ¹H NMR spectrum showed, per 11 CH₂-O and CH-O protons (δ 3.40, br) some 72 saturated CH₂ and CH protons (δ 1.22–1.60, br) and 45 CH₃ protons (δ 0.86, d, J = 9Hz). Satisfactory MS were not obtained. Treatment with BCl3 gave glycerol (TLC) and an alkyl chloride C₂₀H₄₁Cl (by MS); similarly HI cleavage gave C₂₀H₄₁l. The ¹H NMR spectra of these halides showed, per 2 CH₂X protons (δ 3.15, t) 24 CH₂ and CH protons (δ 1.22–1.60) and 15 CH₃ protons (δ 0.86, d, J = 9 Hz) and their identification as phytanyl halides was confirmed by conversion into the acetate which was identical (MS, NMR, and GLC) with authentic phytanyl acetate. Thus structure (1) for the less polar triether is established.

The more polar component

The more polar triether component was very similar to (1) but contained some unsaturation (addition v_m 1665 cm⁻¹; NMR data below). Hydrogenation gave a product identical with (1) which was similarly converted into



(2) (unsaturation as broken lines, see text)

authenticated phytanyl acetate. The extent and location of unsaturation in this component (which is probably a mixture) were in part extablished by ¹H NMR methods. If we regard (1) biogenetically as a tri-O-geranylgeranylglycerol which has been completely reduced, then in a partly-reduced analogue there are 12 possible double bond sites and these are of three distinguishable kinds; (i) in the α -residue, bearing allylic CH₂-O; (ii) in the β - and γ -residues; (iii) in the δ -residue, with terminal =CMe₂.

The resolved signals, assignments, and approximate integrals (calibrated from the total of the CH₂-O and CHO signals, δ 3.40 and 3.95; 11 protons) in the ¹H NMR spectrum of the more polar triether fraction are: δ 0.86 (21H, d, J = 9 Hz, CH₃-CH sat); 1.10-1.40 (36H, br, CH₂ and CH, sat); 1.57 (6H, s, one CH₃-CMe= in δ -residue); 1.66 (18H, s, CH₃-C= in α , β , γ and one CH₃-CMe= in δ -residue); 2.00 (18H, br, CH₂-CH=); 3.40 (9H, br, sat CH₂-O and CH-O); 3.95 (2H,d, J = 8 Hz, =CH-CH₂-O); 5.05 (5H, br, =CH-CH₂); 5.35 (1H, t, J = 8 Hz, =CH-CH₂O). The data are satisfied if in the more polar triethers there are (approximately) 6 double

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