## Reactions of 3-Acetyltropolone and Its Methyl Ethers with Hydrazine

Akio Yamane,<sup>†</sup> Masatoshi Nagayoshi,<sup>††</sup> Kimiaki Imafuku,\* and Hisashi Matsumura Department of Chemistry, Faculty of Science, Kumamoto University, Kurokami, Kumamoto 860 (Received December 1, 1978)

3-Acetyltropolone (1) was synthesized from 3-isopropenyltropolone by treatment with sodium azide in concentrated sulfuric acid. Methylation of 1 by diazomethane gave two isomers, 3-acetyl-2-methoxytropone (2a) and 2-acetyl-7-methoxytropone (2b). 1, 2a, and 2b reacted with hydrazine to give some 1,8-dihydrocycloheptapyrazol-8-one derivatives.

Doi reported that 4-isopropenyltropolone affords 4-acetyltropolone by treatment with one equivalent of hydrazoic acid in concentrated sulfuric acid.<sup>1)</sup> By the application of this reaction to 3-isopropenyltropolone, we successfully obtained 3-acetyltropolone (1). Schenck *et al.* synthesized chloro-substituted 3-acetyltropolone by another method.<sup>2)</sup> Since 3-acetyltropolone (1) has an active methyl group and  $\beta$ -diketone structure, several interesting reactions are expected. Its two isomeric methyl ethers, 2a and 2b, would behave as active troponoid. On the other hand, Matsumoto obtained 1,8-dihydrocycloheptapyrazol-8-one derivatives by the reactions of 3-formyltropolone derivatives with hydrazines.<sup>3)</sup> We have investigated the reactions of 1, 2a, and 2b with hydrazine.

## Results and Discussion

Synthesis of 3-Acetyltropolone (1) and Its Methyl Ethers (2a and 2b).

3-Isopropenyltropolone was obtained by the method of Asao et al.<sup>4)</sup> Treatment of 3-isopropenyltropolone with 1.5 equivalents of sodium azide in concentrated sulfuric acid gave 3-acetyltropolone (1) in a fairly good yield (70%). The structure was confirmed by means of spectral and analytical data. The IR spectrum showed a strong acetyl carbonyl band at 1715 cm<sup>-1</sup> and a characteristic band for tropolone at 1620 cm<sup>-1</sup>. The UV and NMR spectra also supported the structure of 3-acetyltropolone (1).

Being a  $\beta$ -diketone, 3-acetyltropolone (1) reacts with diazomethane to give two methyl ethers (2a and 2b). As to the two methyl ethers, one is 3-acetyl-2-methoxytropone and the other 2-acetyl-7-methoxytropone. Generally, in <sup>1</sup>H NMR spectrum the signal of the proton situated in a  $\beta$ -position to a carbonyl group appears in a lower field. The NMR spectrum of 2a

showed a multiplet for seven-membered ring protons at 6.9—7.3 ppm, while that of **2b** showed a doublet of doublets for H-3 at 7.58 ppm. Thus, **2a** was assigned to 3-acetyl-2-methoxytropone and **2b** to 2-acetyl-7-methoxytropone.

Reactions of 3-Acetyltropolone (1) and 3-Acetyl-2-methoxytropone (2a) with Hydrazine. Refluxing of a mixture of 1 and 2 equivalents of hydrazine hydrate in methanol afforded 3-methyl-1,8-dihydrocycloheptapyrazol-8-one (3) in 76% yield. The reaction at room temperature also gave 3 in 62% yield. Though 3 can be considered to be another tautomeric form, 3-methylcycloheptapyrazol-8-ol, its IR, NMR, and UV spectra support the form of 3-methyl-1,8-dihydrocycloheptapyrazol-8-one. 2a also reacted with hydrazine hydrate to give 3, the difference between the reactivities of 1 and 2a not being observed.

Reaction of 2-Acetyl-7-methoxytropone (2b) with Hydra-When a mixture of 2b and 2 equivalents of hydrazine hydrate in methanol was refluxed, 3methyl-1,8-dihydrocycloheptapyrazol-8-one azine (4) precipitated and 3-methyl-1,8-dihydrocycloheptapyrazol-8-one hydrazone (5) was obtained from the filtrate. The reaction with equimolar hydrazine hydrate gave only 4 (34%), 2b being recovered (36%). The azine (4) was purple plates and insoluble in organic solvents except acetic acid. Nozoe et al.5) obtained 3-phenyl-1,8-dihydrocycloheptapyrrol-8-one azine by the reaction of 8-chloro-3-phenylcycloheptapyrrole with hydrazine hydrate, the UV spectrum of which being very similar to that of 4. The IR spectrum of 4 has a few absorption bands because of high symmetry but no bands at near 1600 cm<sup>-1</sup> for tropone nor at near 1720 cm<sup>-1</sup> for acetyl group. However, it has an absorption band near 3200 and 3400 cm<sup>-1</sup> for NH. Mass spectrum of **4** shows a molecular ion peak at 326. The structure of 4 is reasonable from the above-mentioned evidence and analytical data. 5 has two tautomeric forms such as A and B. The UV spectrum of 5 in visible region shows absorption maxima at fairly shorter wavelength<sup>6)</sup> and the NMR spectrum the signal for NH of pyrazole ring at ca. 12.4 ppm, indicating that 5 exists mainly in the form B.

$$\begin{array}{c}
\stackrel{\mathsf{NH}_2}{\overset{\mathsf{NH}_2}{\overset{\mathsf{NH}_2}{\overset{\mathsf{NH}_2}{\overset{\mathsf{NH}_2}{\overset{\mathsf{NH}_2}{\overset{\mathsf{NH}_3}{\overset{\mathsf{NH}_2}{\overset{\mathsf{NH}_3}{\overset{\mathsf{N}}{\overset{\mathsf{N}}{\overset{\mathsf{NH}_3}{\overset{\mathsf{N}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{N}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset{\mathsf{N}}}{\overset{\mathsf{N}}}}{\overset$$

Treatment of 5 with acetone yielded a tricyclic compound (6). The NMR spectrum of 6 showed the

<sup>†</sup> Present address: Wakunaga Pharmaceutical Co., Ltd., Koda-machi, Takata-gun, Hiroshima 729-64.

<sup>††</sup> Present address: Iki Senior High School, Gonoura-machi, Iki-gun, Nagasaki 811-51.

presence of three methyl group  $[\delta \ 1.61 \ (6H), \ 2.36 \ (3H)]$ . Two methyl groups at 1.61 ppm suggest that they exist at sp³ carbon atom. The mass spectrum showed the parent peak at 214 and the elemental analysis also gave a satisfactory result. Accordingly, **6** was identified as 1,3,3-trimethyl-3,4-dihydro-2,2a,4,5-tetraazabenz[cd]azulene.

## Experimental

The melting points were determined with a Yanagimoto hot-stage apparatus and are uncorrected. All <sup>1</sup>H NMR spectra were recorded with a Hitachi R-24 spectrometer (60 MHz) with TMS as an internal standard. The IR and UV spectra were recorded with a JASCO IRA-1 and a Hitachi EPS-3T spectrophotometer, respectively. The mass spectra were taken on a JEOL JMS-OI-SG-2 spectrometer.

Preparation of 3-Acetyltropolone (1). Sodium azide (10 g) was added to a stirred mixture of 3-isopropenyltropolone (16.2 g), concentrated sulfuric acid (50 ml) and chloroform (50 ml) under cooling with water. The mixture was stirred at 60—70 °C for 2 h. After removal of the chloroform layer, the acid layer was diluted with water and then left to stand overnight. The crystals deposited were recrystallized from methanol to give 11.6 g (70%) of 3-acetyltropolone (1) as pale yellow needles: mp 131—132 °C; λmeon nm (log ε): 245 (4.25), 350 (3.80), 415 (3.75); IR (CHCl<sub>3</sub>): 1715 (C=O), 1620 cm<sup>-1</sup> (C=O); NMR (CDCl<sub>3</sub>): δ 9.0 (br, s, 1H, OH), 7.76 (d, 1H, J=9.0 Hz, H-4), 6.9—7.6 (m, 4H), 2.69 ppm (s, 3H, CH<sub>3</sub>). Found: C, 65.68; H, 4.94%. Calcd for C<sub>9</sub>H<sub>8</sub>O<sub>3</sub>: C, 65.85; H, 4.91%.

Methylation of 3-Acetyltropolone (1). An ethereal solution of diazomethane was slowly added to a solution of  ${\bf 1}$ (11.1 g) in chloroform until the resulting mixture gave no coloration with iron(III) chloride. After removal of the solvents in vacuo, the residue was chromatographed on a silica gel column (Wakogel C-100, 900 g) using ethyl acetate as eluant. The former fractions were combined and recrystallized from hexane-benzene to give 5.37 g (45%) of 3-acetyl-2-methoxytropone (2a) as colorless needles: mp 45—46 °C;  $\lambda_{\max}^{\text{MeOH}}$  nm (log  $\epsilon$ ): 235 (4.21), 330 (3.87); IR (CHCl<sub>3</sub>): 1730 (C=O), 1587 cm<sup>-1</sup> (C=O); NMR (CDCl<sub>3</sub>):  $\delta$  6.9—7.3 (m, 4H), 4.00 (s, 3H, OCH<sub>3</sub>), 2.53 ppm (s, 3H, COCH<sub>3</sub>). Found: C, 67.34; H, 5.69%. Calcd for  $C_{10}H_{10}O_3$ : C, 67.40; H, 5.66%. The latter fractions were also combined and recrystallized from hexane-benzene to give 4.92 g (41%) of 2-acetyl-7-methoxytropone (2b) as pale yellow needles: mp 105—106 °C;  $\lambda_{\text{max}}^{\text{MeOH}}$  nm(log  $\varepsilon$ ): 235 (4.24), 330 (3.84), 365 (3.83); IR (CHCl<sub>3</sub>): 1716 (C=O), 1601 cm<sup>-1</sup> (C=O); NMR (CDCl<sub>3</sub>):  $\delta$  7.58 (dd, 1H, J=11.2 and 2.0 Hz, H-3), 6.7— 7.4 (m, 3H), 3.96 (s, 3H, OCH<sub>3</sub>), 2.53 ppm (s, 3H, COCH<sub>3</sub>). Found: C, 67.22; H, 5.72%. Calcd for  $C_{10}H_{10}O_3$ : C, 67.40; H, 5.66%.

Reaction of 3-Acetyltropolone (1) with Hydrazine. a) A mixture of 1 (213 mg, 1.30 mmol) and 80% hydrazine hydrate (167 mg, 2.67 mmol) in methanol (10 ml) was refluxed for 1 h. After removal of the solvent the residue was recrystallized from benzene to give 158 mg (76%) of 3-methyl-1,8-dihydrocycloheptapyrazol-8-one (3) as orange plates: mp

183—184 °C;  $\lambda_{\rm max}^{\rm MooH}$  nm (log ε): 235 (4.35), 296 (3.82), 308 (3.81), 365 (3.85); IR (CHCl<sub>3</sub>): 3200 (NH), 1580 cm<sup>-1</sup> (C=O); NMR (CDCl<sub>3</sub>): δ: 12.0—13.5 (br, 1H, NH), 6.6—7.3 (m, 4H), 2.60 ppm (s, 3H, CH<sub>3</sub>). Found: C, 67.35; H, 5.07; N, 17.74%. Calcd for C<sub>9</sub>H<sub>8</sub>ON<sub>2</sub>: C, 67.48; H, 5.03; N, 17.49%. b) At room temperature, the reaction of 1 (205 mg, 1.25 mmol) with 80% hydrazine hydrate (80 mg, 1.28 mmol) in methanol (10 ml) gave the same product 3 (124 mg, 62%) after 48 h.

Reaction of 3-Acetyl-2-methoxytropone (2a) with Hydrazine.
a) A mixture of 2a (330 mg, 1.85 mmol) and 80% hydrazine hydrate (295 mg, 4.72 mmol) in methanol (10 ml) was refluxed for 1 h. After removal of the solvent the residue was recrystallized from benzene to give 206 mg (68%) of 3.
b) The same reaction of 2a (206 mg, 1.26 mmol) with 80% hydrazine hydrate (67 mg, 1.07 mmol) at room temperature for 48 h also gave 3 (90 mg, 51%).

Reaction of 2-Acetyl-7-methoxytropone (2b) with Hydrazine. a) When a mixture of **2b** (722 mg, 4.05 mmol) and 80% hydrazine hydrate (496 mg, 7.92 mmol) in methanol (40 ml) was refluxed, a purple precipitate was formed. After 1 h the precipitate was filtered off and recrystallized from dimethyl sulfoxide-water to give 82 mg (13%) of 3-methyl-1,8dihydrocycloheptapyrazol-8-one azine (4) as purple plates: mp >300 °C;  $\lambda_{\max}^{\text{MeOH}}$  nm (log  $\varepsilon$ ): 224 (4.32), 283 sh (3.69), 450 (4.13); IR (KBr): 3300 (NH), 1540 cm<sup>-1</sup>; NMR (CF<sub>3</sub> COOH):  $\delta$  7.4—8.1 (m, 8H), 2.98 ppm (s, 6H, CH<sub>3</sub>×2). Found: C, 67.98; H, 5.10; N, 26.28%. Calcd for C<sub>18</sub>H<sub>16</sub>N<sub>6</sub>: C, 68.33; H, 5.10; N, 26.57%. M+ 316. The filtrate was evaporated and the red residue was recrystallized from methanol to give 479 mg (68%) of 3-methyl-1,8-dihydrocycloheptapyrazol-8-one hydrazone (5) as red needles: mp 199-202 °C;  $\lambda_{\text{max}}^{\text{MeoH}}$  nm(log  $\varepsilon$ ): 215 (4.42), 335 (3.98); IR (KBr): 3340 (NH), 3200 (NH), 1640 cm<sup>-1</sup>; NMR (DMSO- $d_{\theta}$ ): δ 12.4 (br, 1H, NH), 5.6—6.7 (m, 6H), 2.28 ppm (s, 3H, CH<sub>3</sub>). Found: C, 61.98; H, 5.81; N, 31.90%. Calcd for  $C_9H_{10}N_4$ : C, 62.05; H, 5.79; N, 32.17%. b) Refluxing of **2b** (409 mg, 2.30 mmol) with 80% hydrazine hydrate (137 mg)mg, 2.19 mmol) in methanol (20 ml) for 1 h afforded 4 (130 mg, 34%), **2b** (148 mg, 36%) being recovered. c) At room temperature, the reaction of 2b (348 mg, 1.96 mmol) with 80% hydrazine hydrate (197 mg, 3.15 mmol) in methanol (20 ml) gave 4 (61 mg, 18%) and 5 (189 mg, 51%) after 24 h. d) At room temperature, 2b (405 mg, 2.28 mmol) reacted with 80% hydrazine hydrate (135 mg, 2.16 mmol) in methanol (20 ml) for 72 h to afford 4 (133 mg, 31%), **2b** (127 mg, 31%) being recovered.

Reaction of 3-Methyl-1,8-dihydrocycloheptapyrazol-8-one Hydrazone (5) with Acetone. Acetone (117 mg) was added to a solution of 5 (160 mg) dissolved in hot methanol (10 ml), and the mixture was refluxed for 2 h. The resulting solution was concentrated to dryness and the residue was recrystallized from benzene-petroleum ether to afford 127 mg (66%) of 1,3,3-trimethyl-3,4-dihydro-2,2a,4,5-tetraazabenz[cd]-azulene (6) as orange prisms: mp 152—155 °C;  $\lambda_{\max}^{\text{MoOH}}$  nm(log  $\varepsilon$ ): 222 (4.48), 293 (3.68), 350 (4.01); IR (CHCl<sub>3</sub>): 3270 (NH), 1630, 1600 cm<sup>-1</sup>; NMR (CDCl<sub>3</sub>): δ 5.8—6.6 (m, 4H), 5.7 (br, 1H, NH), 2.36 (s, 3H, CH<sub>3</sub>), 1.61 (s, 6H, CH<sub>3</sub>×2). Found: C, 67.33; H, 6.56; N, 25.96%. Calcd for C<sub>12</sub>H<sub>14</sub>N<sub>4</sub>: C, 67.26; H, 6.59; N, 26.15%. M+ 214.

We wish to express our thanks to Dr. Tetsuo Nozoe, Professor Emeritus of Tohoku University, for his advice and encouragement. We are also indebted to Dr. Kazu Kurosawa in this Laboratory for helpful suggestions and Mr. Shuichi Ueda of Taiho Pharmaceutical Co., Ltd. for the measurements of mass spectra.

## References

1974

- 1) K. Doi, Bull. Chem. Soc. Jpn., 34, 501 (1961).
- 2) G. O. Schenck, B. Brahler, and M. Cziesla, Angew. Chem., 68, 247 (1956).
  - 3) S. Matsumoto, Sci. Repts. Tohoku Univ., I, 42, 222

(1958).

- 4) T. Asao, T. Machiguchi, Y. Kitamura, and Y. Kitahara, Chem. Commun., 1970, 89.
- 5) T. Nozoe, Y. Kitahara, and T. Arai, Proc. Jpn. Acad., 30, 478 (1954).6) K. Yamane, K. Fujimori, J.-K. Sin, and T. Nozoe,
- Bull. Chem. Soc. Jpn., 50, 1184 (1977).