

Short Communication

The *Piperaceae* Amides II*:
 Synthesis of Pipericide, A New
 Insecticidal Amide from
Piper nigrum L.

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We previously reported the isolation and structure determination of a new insecticidal amide, pipericide $C_{22}H_{29}NO_3$, occurring in *Piper nigrum* L. and its remarkable toxicity against the adzuki bean weevil (*Callosobruchus chinensis* L.).¹⁾ The proposed structure for pipericide was deduced mainly on the basis of spectral data to be (2*E*,4*E*,10*E*)-*N*-isobutyl-11-(3,4-methylenedioxyphenyl)-2,4,10-undecatrienamide (1).

The major constituents of *Piperaceae* plants, piperine (2) and pellitorine (3) have been synthesized by Normant *et al.*²⁾ and Jacobson *et al.*,³⁾ respectively. As a more closely related compound to pipericide, Joshi *et al.*⁴⁾ isolated piperstachine from *Piper trichostachyon* C. DC. and deduced its structure to be (2*E*,8*E*,10*E*)-*N*-isobutyl-11-(3,4-methylenedioxyphenyl)-2,8,10-undecatrienamide (4). Later, Viswanathan *et al.*⁵⁾ synthesized the compound (4) and its 2*E*,4*E*,6*E*-isomer (5) and confirmed the proposed structure for piperstachine to be as reported (4) (see Fig. 1). In this communication we wish to report the synthesis of the amide of the proposed structure (1) and the identity with pipericide.

The present synthesis is based on a piperonyl + C_6 + C_4 principle employing piperonyl alcohol (13), 1,6-hexanediol (6) and methyl crotonate (8) which are readily available as starting

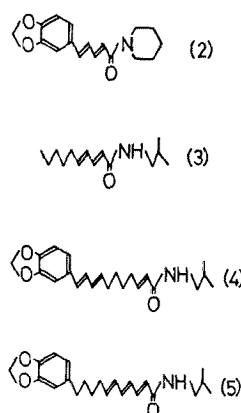


FIG. 1.

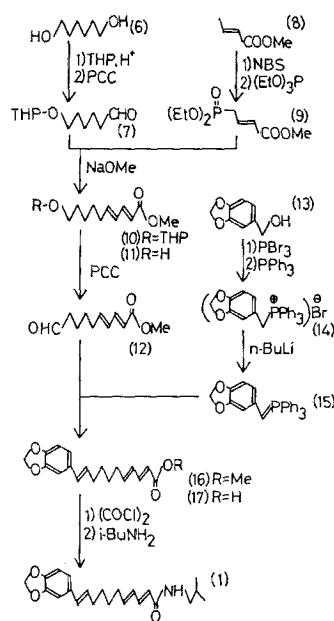


FIG. 2. Synthetic Schema of Pipericide (1).

materials (see Fig. 2). Firstly, the preparation of the C_6 construction unit: In the usual manner⁶⁾ 6-(tetrahydro-2'-pyraniloxy)-1-hexanol was prepared from 1,6-hexanediol (6), which was oxidized with pyridinium chlorochromate⁷⁾ affording the corresponding aldehyde (7) with a 95% yield. In order to obtain the C_4 construction unit, methyl crotonate (8) was brominated with NBS to give methyl 4-bromocrotonate⁸⁾ (92% yield), which was converted with triethyl phosphite to diethyl 3-methoxycarbonyl-2-propenylphos-

* The *Piperaceae* Amides I: M. Miyakado, I. Nakayama, H. Yoshioka and N. Nakatani, this Journal 43, 1609 (1979).

phonate (**9**) (77% yield) according to the method of Davis *et al.*⁹⁾

The 2,4-dienoate (**10**)*¹⁾ was prepared by condensation of the aldehyde (**7**) with the phosphonate (**9**) in the presence of sodium methoxide, whose configurations were predominantly *trans* at C-2 and exclusively *trans* at C-4.^{9,10,11)} The protecting group of the 2,4-dienoate (**10**) was removed by *p*-TsOH in MeOH to give (2*E*,4*E*)-methyl 10-hydroxy-2,4-decadienoate (**11**),*²⁾ which was oxidized with pyridinium chlorochromate⁷⁾ to (2*E*,4*E*)-methyl 10-oxo-2,4-decadienoate (**12**).

In order to obtain the piperonyl construction unit, piperonyl alcohol (**13**) was brominated with PBr₃ to give piperonyl bromide with a quantitative yield, which was converted with triphenylphosphine to 3,4-methylenedioxybenzyltriphenylphosphonium bromide (**14**) (93% yield) according to the method of Grewe *et al.*¹²⁾ The phosphonium bromide (**14**) in benzene was converted with an equimolar amount of *n*-BuLi (in *n*-hexane sol.) to the Wittig reagent (**15**),¹³⁾ which was coupled *in situ* with an equimolar amount of (2*E*,4*E*)-methyl 10-oxo-2,4-decadienoate (**12**) to give the condensation product (**16**) with a 70% yield. The ratio of the (*E*)-isomer to the (*Z*)-isomer at the C₁₀ double bond was about 8:2

*1 We tried Wittig reaction of the aldehyde (**7**) with (*E*)-3-methoxycarbonylallyltriphenylphosphonium bromide to obtain the 2,4-dienoate (**10**). A phosphonium salt was prepared from triphenylphosphine and methyl 4-bromocrotonate in benzene according to the method described by Buchta *et al.*¹⁴⁾ (mp, 178.5~179.5°C, lit. 179~180°C). However, the Wittig reaction whose procedure was in accordance with Buchta's description,¹⁴⁾ gave methyl 8-(tetrahydro-2'-pyraniloxy)-2-ethenyl-2-octenoate instead of the desired product, the 2,4-dienoate (**10**), as a result of an allylic phosphonium salt rearrangement. Similar γ -rearrangements have already been described by Font *et al.*¹⁵⁾ and E. J. Corey *et al.*¹⁶⁾

*2 $n_D^{21.5^\circ}$ 1.5113, NMR: $\delta_{TMS}^{CDCl_3}$ 1.45 (s, 6H), 2.15 (br.m, 2H), 3.08 (s, 1H, disappeared on D₂O addition), 3.49 (t, 2H, $J=6.0$ Hz), 3.68 (s, 3H), 5.68 (d, 1H, $J=15.5$ Hz), 6.00~6.23 (m, 2H), 6.95~7.43 (m, 1H), IR: ν_{max}^{film} 3400, 1710, 1635 cm⁻¹, MS: (*m/e*) 198 (rel. int., 7.0) M⁺, 113 (26), 111 (43), 107 (32), 91 (21), 81 (55), 79 (100, base peak), 67 (37), 59 (31), 55 (28), 53 (28).

as determined by GLC and NMR. The *E* and *Z* ester mixture (**16**) was hydrolyzed (KOH-MeOH) to give the acid (**17**). The crude acid was recrystallized twice from benzene to give (2*E*,4*E*,10*E*)-11-(3,4-methylenedioxyphenyl)-2,4,10-undecatrienoic acid (**17**)*³⁾ as fine needles (60% yield), which was chlorinated with oxalyl chloride. The resultant acid chloride was finally condensed with isobutylamine yielding (2*E*, 4*E*, 10*E*)-*N*-isobutyl-11-(3,4-methylenedioxyphenyl)-2,4,10-undecatrienamide (**1**),*⁴⁾ The synthetic product was identical in all respects (mixed mp., NMR, IR, MS and UV) with natural pipericide. Evaluation of detailed insecticidal properties of pipericide and co-occurring compounds is now in progress and will be reported elsewhere.

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REFERENCES

- 1) a) M. Miyakado, I. Nakayama, H. Yoshioka and N. Nakatani, Abstracts of Papers, Annual Meeting of the Agricultural Chemical Society of Japan, Tokyo, April 1979, p. 460.
b) M. Miyakado, I. Nakayama, H. Yoshioka and

*3 mp, 136~137°C, NMR: $\delta_{TMS}^{CDCl_3-DMSO-d_6}$ 1.47 (br.s, 4H), 2.18 (br.m, 4H), 5.59~6.23 (br.m, 5H), 5.92 (s, 2H), 6.73 (s, 2H), 6.87 (s, 1H), 6.98~7.50 (br.m, 1H), 11.53 (br.s, 1H, disappeared on D₂O addition), IR: ν_{max}^{nujol} 1680, 1660, 1625, 1600 cm⁻¹, MS: (*m/e*) 300 (rel. int., 60) M⁺, 192 (32), 187 (29), 173 (36), 148 (34), 135 (100, base peak), 131 (82), 103 (60), 77 (32), UV: λ_{max}^{EtOH} 212 nm (log ϵ 4.37), 260 (4.62), 270 (4.54), 305 (3.81), UV absorption at 270 and 305 nm reveals the presence of the 10-*E* double bond, see ref. (1b).

*4 mp, 114~115°C, NMR: $\delta_{TMS}^{CDCl_3}$ 0.92 (d, 6H, $J=6.5$ Hz), 1.44 (m, 4H), 1.60~2.00 (m, 1H), 2.12 (m, 4H), 3.12 (t, 2H, $J=6.5$ Hz), 5.60~6.20 (br.m, 6H, 1H disappeared on D₂O addition), 5.90 (s, 2H), 6.72 (s, 2H), 6.86 (s, 1H), 6.94~7.42 (br.m, 1H), IR: ν_{max}^{nujol} 3280, 1650, 1620, 1603 cm⁻¹, MS: (*m/e*) 355 (rel. int., 28) M⁺, 220 (38), 161 (30), 135 (100, base peak), 131 (68), 103 (47), UV: λ_{max}^{EtOH} 209 nm (log ϵ 4.34), 261 (4.68), 268 (4.63), 304 (3.80).

- N. Nakatani, *Agric. Biol. Chem.*, **43**, 1609 (1979).
- 2) H. Normant and C. Feugeas, *Compt. Rend.*, **258**, 2846 (1964).
- 3) M. Jacobson, *J. Am. Chem. Soc.*, **75**, 2584 (1953); M. Jacobson and D. G. Crosby (Ed.), "Naturally Occurring Insecticides," Marcel Dekker, N. Y., 1971, p. 137; J. Tsuji, H. Nagashima, T. Takahashi and K. Masaoka, *Tetrahedron Letters*, 1919 (1977) and literature cited within.
- 4) B. S. Joshi, N. Viswanathan, D. H. Gawad and W. von Philipsborn, *Helv. Chim. Acta*, **58**, 1551 (1975).
- 5) N. Viswanathan, V. Balakrishnan, B. S. Joshi and W. von Philipsborn, *Helv. Chim. Acta*, **58**, 2026 (1975).
- 6) O. P. Vig, A. K. Vig, J. S. Mass and K. C. Gupta, *J. Indian Chem. Soc.*, **52**, 538 (1975).
- 7) E. J. Corey and J. W. Suggs, *Tetrahedron Letters*, 2647 (1975).
- 8) C. Djerassi, *Chem. Revs.*, **43**, 271 (1948).
- 9) J. B. Davis, L. M. Jackman, P. T. Siddons and B. C. L. Weedon, *J. Chem. Soc., (C)*, 2154 (1966).
- 10) G. Pattenden and B. C. L. Weedon, *J. Chem. Soc., (C)*, 1984 (1968).
- 11) W. Stilz and H. Pommer, *U. S. Pat.*, 3163669 (Dec. 29, 1964).
- 12) R. Grewe, W. Freist, H. Neumann and S. Kersten, *Chem. Ber.*, **103**, 3752 (1970).
- 13) H. O. House, V. K. Jones and G. A. Frank, *J. Org. Chem.*, **29**, 3327 (1964).
- 14) E. Buchta and F. Andre, *Chem. Ber.*, **92**, 3111 (1959).
- 15) J. Font and P. March, *Tetrahedron Letters*, 3601 (1978).
- 16) E. J. Corey and B. W. Erickson, *J. Org. Chem.*, **39**, 821 (1974).
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