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Enzyme-catalyzed Asymmetric Hydrolysis of meso-Substrate. The Facile Synthesis of Both Enantiomers of cis-2,5-Disubstituted Tetrahydrofuran Derivatives

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The enzyme-catalyzed asymmetric hydrolysis of the meso-diesters derived from cis-2,5-bis(hydroxymethyl)tetrahydrofuran in presence of PLE, PPL, and CCL afforded the optically active half-esters. The both enantiomers of cis-5-(hydroxymethyl)tetrahydrofuran-2-carboxylic acid with high optical purity were prepared from the half-esters.

The use of hydrolytic enzymes as practical chiral catalysts have well been documented<sup>1)</sup> and of particular value in this regard are the abilities of hydrolytic enzymes to induce stereospecific transformations of meso-substrates to chiral products. Many naturally occuring antibiotics have cis-2,5-disubstituted tetra-hydrofuran unit as a common structural feature.<sup>2)</sup> In this context we recently described the preparation of optically active trans-2,5-disubstituted tetrahydro-furan derivative and of chiral ionophores having this derivative as a chiral unit.<sup>3)</sup> We report here the synthesis of the optically active cis-2,5-disubstituted tetrahydrofuran derivatives by enzyme-catalyzed asymmetric hydrolysis of meso-diesters.

cis-2,5-Bis(hydroxymethyl)tetrahydrofuran (<u>1</u>), bp 148-151 °C (10 mmHg), was prepared from cis-2,5-bis(hydroxymethyl)furan, prepared according to the known procedure, <sup>4</sup>) by catalytic hydrogenation with Raney-Ni. Treatment <u>1</u> with acetyl, propionyl, isobutyl, and vareryl chloride in ether-pyridine gave the diesters <u>2a</u> (bp 105.5-106.5 °C (0.5 mmHg)), <u>2b</u> (bp 115-116.5 °C (0.4 mmHg)), <u>2c</u> (bp 122-124 °C (0.5 mmHg)), and 2d (bp 120.5-121.5 °C (0.4 mmHg)), respectivery.

Asymmetric hydrolysis of 2a-2d catalyzed by pig liver esterase (PLE)<sup>5)</sup> were performed at pH 8.0 in 0.1 M (1 M=1 mol dm<sup>-3</sup>) phosphate buffer solution (2.4 L) at 0 °C on 19 mmol scale. In every cases, the product was extracted with chloroform and purified by alumina column chromatography followed by distillation to give the corresponding half-esters, and the resulting meso-diol <u>1</u> was not isolated, because it is readily soluble in water. Asymmetric hydrolysis catalyzed by lipase from porcine pancreas (PPL)<sup>5)</sup> and lipase from Candida cylindracea (CCL)<sup>5)</sup> were performed at pH 8.0 and 7.0 in 0.1 M phosphate buffer solution at room temperature, respectively. The results are summarized in Table 1.

Oxidation of <u>3a</u>,  $[\alpha]_D$  -4.45°, with CrO<sub>3</sub> in acetone followed by esterification with diazomethane gave <u>4a</u>, bp 125-127 °C (0.5 mmHg);  $[\alpha]_D$  -15.3° (CHCl<sub>3</sub>). The ester <u>4a</u> was hydrolyzed with aqueous solution of sodium carbonate at room temperature to provide (-)-cis-5-(hydroxymethyl)tetrahydrofuran-2-carboxylic acid (<u>5</u>),  $[\alpha]_D$  -2.82° (CHCl<sub>3</sub>). Treatment of <u>5</u> with diazomethane in ether provided (+)-  $(2S,5R)-\underline{6}$ ,  $[\alpha]_{D} +21.4^{\circ}$  (CHCl<sub>3</sub>) (96% enantiomeric excess (e.e.)), whose absolute configuration and e.e. value have been reported by Jones.<sup>6</sup>) The information on the absolute configuration and e.e. value of  $(+)-\underline{6}$  permitted us to assign the 2R,5R configuration to  $(-)-\underline{3a}$  and estimate 96% e.e. for  $(-)-\underline{3a}$  obtained by PLE-catalyzed hydrolysis. Analogously, the absolute configuration and the e.e. value of the half esters  $\underline{3b},\underline{3c}$ , and  $\underline{3d}$  were determined by conversion into  $\underline{6}$  via  $(-)-\underline{4b}$ ,  $[\alpha]_{D}$  -6.81° (CHCl<sub>3</sub>),  $(-)-\underline{4c}$ ,  $[\alpha]_{D}$  -8.81° (CHCl<sub>3</sub>), and  $(-)-\underline{4d}$ ,  $[\alpha]_{D}$  -9.05° (CHCl<sub>3</sub>), respectively. By the same procedure described above,  $(+)-\underline{5}$  with 80% e.e. was also prepared from  $(+)-\underline{3b}$  obtained by CCL-catalyzed hydrolysis of  $\underline{2a}$ .

Entry	Substrate	Enzyme	Product <sup>7</sup> ) (% isolated) <sup>a)</sup>	e.e. (%)	Time/h	[a] <sub>D</sub> /°(CHC1 <sub>3</sub> )
1	<u>2a</u>	PLE	(2S,5R)- <u>3a</u> (86)	96	18	-4.45
2	<u>2b</u>	PLE	(2S, 5R) - 3b (54)	42	2.3	-2.84
3	<u>2c</u>	PLE	(2S, 5R) - 3c (62)	94	2.8	-3.17
4	2d	PLE	(2S, 5R) - 3d (72)	55	7.5	-1.20
5	2a	PPL	(2S,5R)- <u>3a</u> (53)	41	305	-1.91
6	<u>2b</u>	PPL	(2S,5R)- <u>3b</u> (57)	33	22	-2.25
7	<u>2c</u>	PPL	(2S,5R)- <u>3c</u> (68)	32	20	-1.08
8	<u>2d</u>	PPL	(2S,5R)- <u>3d</u> (71)	16	56 (d)	-0.34
9	<u>2a</u>	CCL	(2R,5S)- <u>3a</u> (50)	70	96	+3.26
10	<u>2a</u>	CCL	(2R,5S)- <u>3a</u> (20)	80	214	+3.72
11	<u>2b</u>	CCL	(2R,5S)- <u>3b</u> (69)	26	22	+1.77
12	<u>2c</u>	CCL	(2R,5S)- <u>3c</u> (75)	28	3.5	+0.94
13	2d	CCL	(2R,5S)-3d (71)	4	41 (d)	+0.09

Table 1. Asymmetric hydrolysis of the meso-diesters

a) After alumina column chromatography.

As can be seen from Table 1, among the asymmetric hydrolysis of the diesters, hydrolysis of 2a with PLE provided the half-ester 3a with the highest e.e. value. In the cases of PLE-catalyzed and PPL-catalyzed hydrolysis, the CH<sub>2</sub>OCOR-group located at the (S)-chirality center was dominantly hydrolyzed. In CCl-catalyzed hydrolysis, a clear reversal of the stereospecificity was observed and elongation of the reaction time raised the e.e. value of the product 3a from 70% to 80% (Entries 9, 10). The rate of hydrolysis of these diesters was considerably influenced by the nature of the acyl group. The diesters having relatively large acyl graoups; 2b and 2c were hydrolyzed faster than 2a with all enzymes examined here, but the 2,2-dimethylpropionyloxy group reduced remarkably the rate of hydrolysis of the diester.

Next we turned attention to resolve the racemic cis-monoacetate  $\underline{3a}$  and transdiacetate  $\underline{8}$  by enantioselective hydrolysis with PLE and CCL. The hydrolysis of  $(\pm)-\underline{3a}$  and  $(\pm)-\underline{8}$  was carried out under the similar conditions described for the hydrolysis of meso-substrates and terminated at, or close to, the 50%-of-hydrolysis point. The results are summarized in Table 2.

Entry	Substrate	Enzyme	Product <sup>7)</sup> (% yield)	e.e. (%)	Time/h	[α] <sub>D</sub> /°
14	(±)- <u>3a</u>	PLE	(-)-(2S,5R)- <u>3a</u> (47)	53	17	-2.46
15	(±)- <u>3a</u>	CCL	(+)-(2R,5S)- <u>3a</u> (54)	7	115	+0.31
16	(±)– <u>8</u>	PLE	(-)-(2R,5R)- <u>8</u> (47)	58	3.5	-24.8
			(+)-(2 <b>s,</b> 5 <b>s</b> )- <u>9</u> (48)	58		+22.8
17	(±)– <u>8</u>	CCL	(+)-(2 <b>S,</b> 5S)- <u>8</u> (38)	<1	52.5	+0.18
			(-)-(2R,5R)- <u>9</u> (49)	<1		-0.10

Table 2. Enantioselective hydrolysis of racemic substrates

In the case of PLE-catalyzed hydrolysis, <u>3a</u> having the acetoxy group located at the (S)-chirality center was dominantly hydrolyzed to give <u>1</u>, and (-)-(2S,5R)-<u>3a</u> with 53% e.e. was recovered after alumina chromatography. A reversal of the stereospecificity was also observed in CCL-catalyzed hydrolysis of (±)-<u>3a</u>, and (+)-<u>3a</u> with 7% e.e. was recovered. The hydrolyzed product <u>1</u> was not isolated in both cases. Both acetoxy groups of (+)-<u>8</u> locate at the (S)-chilality center, however, the PLE-catalyzed hydrolysis of <u>8</u> terminated exclusively at the half-ester stage to give (+)-(2S,5S)-<u>9</u> in 48% yield, and (-)-(2R,5R)-<u>8</u> was recovered. The absolute configuration and e.e. value of <u>8</u> and <u>9</u> were determined by conversion to the known diol <u>7</u>.<sup>3</sup> In CCL-catalyzed hydrolysis of (±)-<u>8</u>, the half-ester <u>9</u> undergo further cleavage to yield the diol <u>7</u> which was not isolated, and the enantiomer selectivity of this reaction was extremely low. Oxidation of (+)-<u>9</u> with CrO<sub>3</sub> in acetone followed by hydrolysis with sodium carbonate in water gave (+)-trans-5-(hydroxymethyl)tetrahydrofuran-2-carboxylic acid (<u>10</u>), [ $\alpha$ ]<sub>D</sub> +3.1° (CHCl<sub>3</sub>) (58% e.e.).



As mentioned above, the enzyme-catalyzed asymmetric hydrolysis of the mesodiesters provided the method of the facile preparation of both enantiomers of the chiral cis-2,5-disubstituted tetrahydrofuran derivatives, and PLE-catalyzed hydrolysis of the racemic diacetate was useful for the convenient synthesis of the half-ester in an optically active form.

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- 5) P.L.E.: BMY 104698 100 Units/mg; P.P.L.: Sigma, Type II, 16 Units/mg; C.C.L.: Sigma, Type VII, 400-900 Units/mg.
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- 7) All new compounds gave satisfactory spectral data and elemental analyses. Boiling points of the esters, <u>3a</u>; bp 92.0-92.5 °C (0.3 mmHg), <u>3b</u>; bp 99.0 °C (0.3 mmHg), <u>3c</u>; bp 96.0-97.0 °C (0.2 mmHg), <u>3d</u>; bp 93.0 °C (0.2 mmHg), <u>8</u>; bp 87-90 °C (0.2 mmHg), <u>9</u>; bp 96.5-97.5 °C (0.2 mmHg).

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