A NEW SYNTHETIC METHOD OF 1-AZABICYCLO[4.n.0]SYSTEMS¹⁾

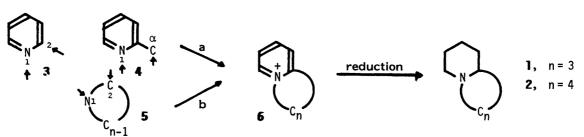
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A new method for the synthesis of bicyclic pyridinium salts from alicyclic amines and reduction of the salts to 1-azabicyclo[4.n.0]systems has been exploited.

Since compounds possessing the skeletones of indolizidine 1 and quinolizidine 2 are widely found in a variety of physiologically important alkaloids, much effort has been devoted to the synthesis of such compounds.²⁾ Among them, 1,2-annulation on the pyridine ring 3 and 1, α -annulation on the pyridine derivative 4 (route a in Scheme I) followed by reduction of the resulting pyridinium salts 6 seem to be convenient routes to prepare 1 and 2,^{3,4)} whereas multisteps often required in these routes, and the limited availability of the starting pyridine derivatives do not always allow the easy synthesis of 6 bearing functional groups on the desired positions.

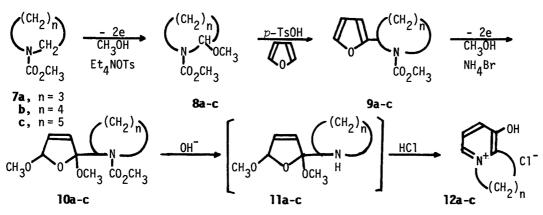
Scheme I



We describe herein a new synthetic method of **6**, which is characterized by formation of the pyridinium ring at the 1,2-positions of the starting cyclic secondary amines **5** (route b in Scheme I). The key reaction involved in the method is the conversion of dihydrofuran derivatives **10** to pyridinium salts **12**.⁵⁾

A typical procedure is exemplified by the preparation of 2, 3-dihydro-8-hydroxy-1Hindolizinium chloride (12a). Anodic oxidation of 1-carbomethoxypyrrolidine 7a followed by the acidcatalyzed coupling reaction of the resulting α -methoxylated carbamate 8a with furan was carried out according to the reported procedure,⁶ the overall yield being 71%. Subsequent anodic oxidation of 9a (30 mmol) in methanol (30 ml) containing ammonium bromide (20.4 mmol) gave 1-carbomethoxy-2-(2,5-dimethoxy-2,5-dihydrofuryl)pyrrolidine (10a) in 95% yield. After 10a (6 mmol) was refluxed for 2 h in ethylene glycol (25 ml) containing potassium hydroxide (89mmol) and hydrazine hydrate (20.6 mmol), the mixture was cooled and extracted with methylene chloride to afford crude 11a, which was then heated without purification with 1 N HCl (20 ml) followed by the removal of the solvent in vacuo to give **12a** (70% yield from **10a**). In a similar way, piperidine, hexamethyleneimine, and morphorine gave the corresponding pyridinium salts (**12b-d**) (Table I).

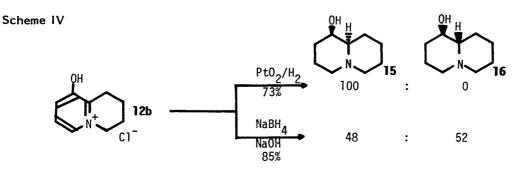




This method is advantageous for regioselective synthesis of substituted bicyclic pyridinium salts 12. Thus, 12 possessing an alkyl substituent on the desired position of the rings can be regioselectively prepared as exemplified by the synthesis of 1,2,3,4-tetrahydro-9-hydroxy-6-methylquinolizinium chloride (12e) and 2,3-dihydro-8-hydroxy-5-methyl-1 \mathcal{H} -indolizinium chloride (12f). The former compound was obtained starting from α -pipecoline, and the latter was synthesized by the reaction of **8a** with methylfuran instead of furan. Also, the synthesis of hydroxyindolizidine and -quinolizidine from 12 was reasonably stereoselective. The catalytic hydrogenation of 12a gave 8-hydroxyindolizidine (73% yield), in which the main isomer 13⁷ (distribution, 81%) has a trans configuration between the hydroxyl group and the bridgehead hydrogen. On the other hand, the predominant formation of epimer 14 was achieved by the reduction of 12a with sodium borohydride in alkaline solution (Scheme III).

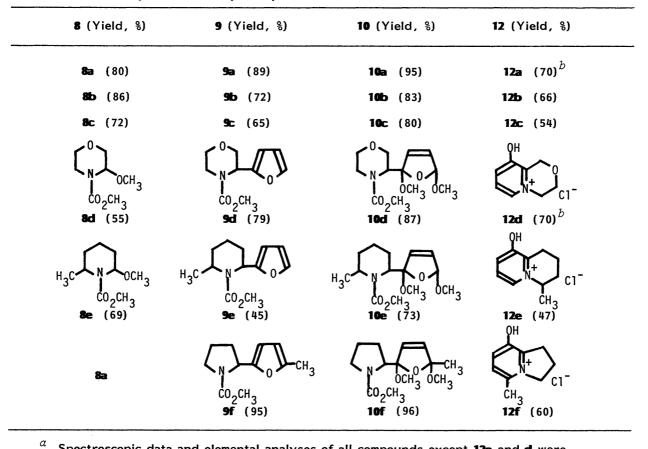
Scheme III

Furthermore, in the catalytic hydrogenation of **12b**, the compound **15**⁸ was obtained as a sole product with indicating the perfect stereoselectivity of hydrogenation, while the reduction with sodium borohydride resulted in the formation of a mixture of **15** and its epimer **16** in a ratio of 48 : 52 (Scheme IV). The assignment of the stereochemistry of **13-16** was carried out by comparison of their physical and spectroscopic data with those of authentic samples which were prepared by the reduction of 8-ketoindolizidine⁷⁾ and 1-ketoquinolizidine.⁸⁾



Accordingly, our method described above provides a general route to synthesize 1-azabicyclo[4.n.0]system with high regio- and stereoselectivity. Application of this method to the synthesis of natural alkaloids is now in progress.

Table I. Preparation of Bicyclic Pyridinium Chlorides	Table I.	Preparation of	Bicyclic P	yridinium	Chlorides
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² Spectroscopic data and elemental analyses of all compounds except **12a** and **d** were satisfactory for assigned structures.

^b Because of their highly hygroscopic nature, satisfactory results were not obtained in the elemental analyses of **12a** and **d**, whereas their ir and nmr spectra were reasonable with the assigned structures. Also, the successful conversion of **12a** to the known **13** and **14** substantiates the structure of **12a**. Acknowledgment. One of the authors (Y.M.) wishes to thank the Kurata Foundation for the Kurata Research Grant.

References

- 1) Electroorganic Chemistry. 65.
- 2) T. L. Macdonald, J. Org. Chem., 45, 193 (1980), and references cited therein.
- (a) W. L. Mosby, "The Chemistry of Heterocyclic Compounds," ed. by A. Weissberger, Interscience Publishers, New York (1961), Vol. 15.
 - (b) K. B. Prasad and S. C. Schaw, Indian J. Chem., 11, 621 (1973).
 - (c) P. E. Sonnet and J. E. Oliver, J. Org. Chem., 39, 2662 (1974).
- Some synthetic routes without utilizing pyridinium intermediates have been reported: for example,
 - (a) From imides as starting compounds: H. E. Schoemaker, J. Dijkink, and W. N. Speckamp, Tetrahedron, 34, 163 (1978). B. D. Wijnberg and W. N. Speckamp, Tetrahedron Lett., 1980. 1987.
 - (b) Intramolecular cyclization of azaolefins: S. R. Wilson and R. A. Sawicki, J. Org. Chem.,
 44, 330 (1979).
 - (c) The Hoffmann-Loffler reaction: M. E. Wolff, Chem. Rev., 63, 55 (1963).
- 5) The synthesis of some simple quarternary 2-substituted 3-hydroxypyridinium chlorides has been reported with using chlorine as the oxidizing agent in oxidation of N-monoalkyl-2-(αaminoalkyl)furans: J. B. Petersen, K. Norris, N. C. Kaas, and K. Svanholt, Acta Chem. Scand., 23, 1785 (1969), and references cited therein.
- 6) T. Shono, Y. Matsumura, K. Tsubata, and J. Takata, Chem. Lett., 1981, 1121.
- 7) C. P. Rader, A. L. Young, Jr., and H. S. Aaron, J. Org. Chem., 30, 1536 (1965).
- 8) H. S. Aaron, G. E. Wicks, Jr., and C. P. Rader, J. Org. Chem., 29, 2248 (1964).

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