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## Diazo-compounds in Heterocyclic Synthesis. Aziridine Ring Expansion in a Retro-Diels-Alder Reaction

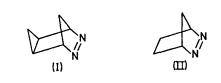
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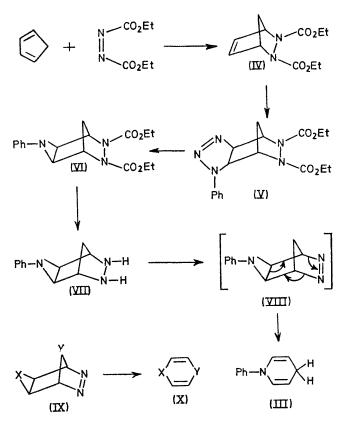
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Summary Hydrolysis and decarboxylation of exo-diethyl 3-phenyl-3,6,7-triazatricyclo $[3,2,1,0^{2,4}]$ octane-6,7-dicarboxylate followed by mild oxidation gives N-phenyl-1,4-dihydropyridine (50-70%), thus affording a simple route for the conversion of cyclopentadiene into 1,4-dihydropyridines.

THE remarkable rate enhancement of  $10^{11}$  for loss of molecular nitrogen from diazo-compound (I) compared with (II) has been attributed largely to relief of strain and steric interactions on opening the *exo*-cyclopropane ring and the near ideal geometry of the transition state for this ready retro-Diels-Alder reaction.<sup>1</sup> Synchronous loss of nitrogen and double-bond formation from (I) gave 1,4cyclohexadiene quantitatively. The synthetic potential of this reaction has been developed in a synthesis of semibullvalene.<sup>2</sup>

We now report the conversion of cyclopentadiene into the dihydropyridine (III) (44-61% overall) via the analogous aziridine ring expansion. Bicyclic precursor (IV) was prepared from cyclopentadiene and diethyl azodicarboxylate (95%).<sup>3</sup> Subsequent 1,3-dipolar addition of phenyl azide to (IV) gave exclusively the exo-adduct (V), m.p. 124-125°4 (92.5%). Photolysis of this triazoline in acetone (sun lamp; Pyrex filter)<sup>5</sup> gave exclusively the exodicarboxylate (VI) (99%), m.p.  $124-124\cdot5^{\circ}$ ;  $M^+$  331;  $v_{max}$  (Nujol) 1740 and 1720 cm<sup>-1</sup>;  $\delta$  (CDCl<sub>3</sub>) 1.12-1.45 (7H, t, J 7 Hz), 2.00-2.28 (1H, d, J 10 Hz), 2.81 (2H, s), 4.01-4.41 (4H, q, J 7 Hz), 4.85 (2H, s), and 6.72-7.36(5H, m) p.p.m. Hydrolysis and decarboxylation of (VI) (hot KOH) gave the hydrazine (VII) as an unstable brown oil  $v_{max}$  (neat) 3240 and 1600 cm<sup>-1</sup>;  $\delta$  (CCl<sub>4</sub>) 1.05–1.32 (1H, m), 1.91-2.19 (1H, d, J 10 Hz, further split), 2.33-2.42 (2H, s), 3.26 br (2H, s, disappears on addition of  $D_2O$ ), 3.69 (2H, s), and 6.58-7.33 (5H, m) p.p.m., which was readily oxidized by oxygen in CH2Cl2 or better by yellow mercuric oxide to the dihydropyridine (III) [50-70% from (VI)], presumably by spontaneous decomposition of the diazo-compound (VIII). The product was identical in all respects with that published.6





The successful synthesis of (III) suggests that the conversion of (IX) into (X) may have general application to the synthesis of novel heterocyclic systems.

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<sup>1</sup> E. L. Allred, J. C. Hinshaw, and A. L. Johnson, J. Amer. Chem. Soc., 1969, 91, 3382; E. L. Allred and J. C. Hinshaw, Chem. Comm., 1969, 1021.