

# Activated Nitriles in Heterocyclic Synthesis: Novel Syntheses of Pyrano[2,3-*b*]pyridines and Pyrano[2,3-*d*]pyrimidines

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Reactions of  $\beta$ -(2-furanyl)- and  $\beta$ -(2-thienyl)-acrylonitrile with ethyl acetoacetate are reported. They lead to new polyfunctional derivatives of pyrano[2,3-*b*]pyridine and pyrano[2,3-*d*]pyrimidine. The structures of these products and the mechanisms of their formation are reported.

## Aktiviert Nitrile in der Heterocyklen-Synthese: Neue Synthese von Pyrano[2,3-*b*]pyridinen und Pyrano[2,3-*d*]pyrimidinen

Es wird über die Reaktionen von  $\beta$ -(2-Furanyl)- und  $\beta$ -(2-Thienyl)-acrylonitril mit Ethylacetoacetat berichtet. Einige neue Pyrano[2,3-*b*]pyridine und Pyrano[2,3-*d*]pyrimidine wurden synthetisiert. Über die Strukturen und die Bildungsmechanismen wird berichtet.

In the last decade we have been involved in a program aiming to explore the synthetic potential, scope and limitations of  $\alpha$ ,  $\beta$ -unsaturated nitriles in heterocyclic synthesis<sup>1</sup>. Several new approaches to five- and six-membered and fused heterocyclic derivatives could be achieved during this work<sup>2</sup>.

As a part of a medicinal chemistry program<sup>3,4</sup>, in our laboratories the synthesis of several substituted pyrans was required. Several new, otherwise difficultly accessible, pyrans have been prepared. Moreover, the results offer a new and efficient route for the synthesis of pyranopyridines and pyranopyrimidines.

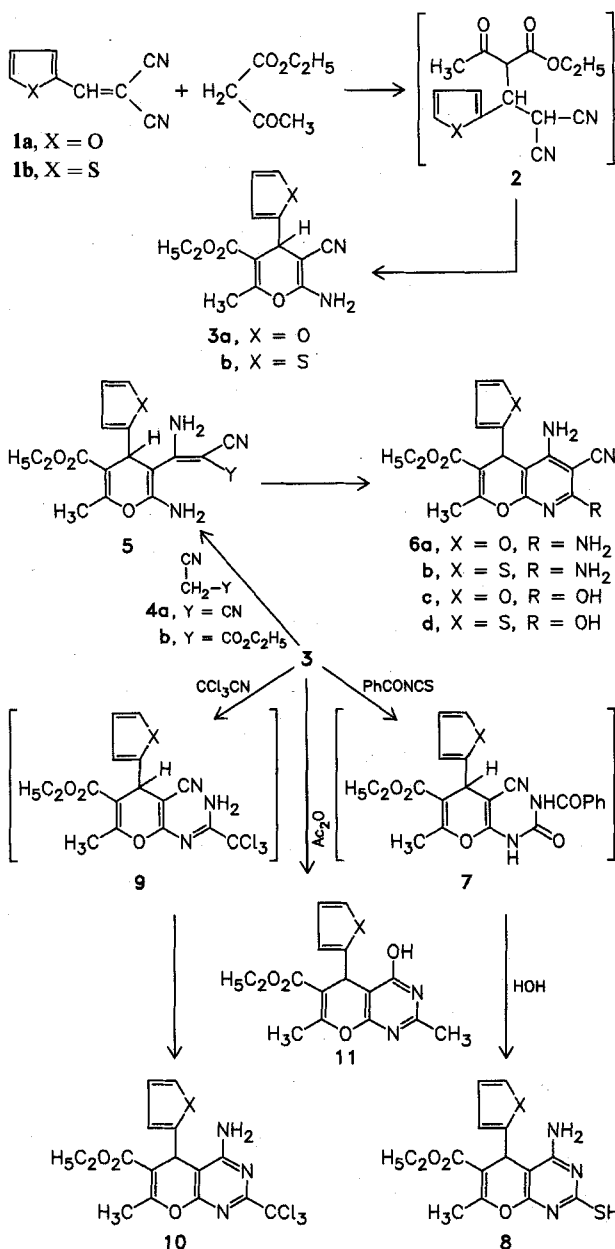
$\alpha$ -Cyanofuran-2-yl-acrylonitrile (**1a**) and  $\alpha$ -cyanothiophen-2-yl-acrylonitrile (**1b**) reacts with ethyl acetoacetate to yield 1:1 adducts. <sup>1</sup>H-NMR- and IR-spectra establish the pyran structure **3** for these products. Thus, it revealed for **3a** the ethyl ester protons, the methyl singlet at  $\delta$  = 3.3 ppm, the pyran 4-H at  $\delta$  = 4.4 ppm, two amino protons at  $\delta$  = 6.8 ppm and furan protons at  $\delta$  = 6.9–7.5 ppm (m).

The reactivity of compound **3** towards some cyano methylene reagents was investigated. Thus, **3** reacted with malononitrile (**4a**) and ethyl cyanoacetate (**4b**) in ethanol in the presence of triethylamine to yield the pyrano[2,3-*b*]pyridines **6a–d**. The formation of **6a, b** from **3** with **4a** is assumed to proceed via addition of the methylene group in **4a** to the cyano group in **3** to yield **5** followed by cyclization. The formation of **6c, d** from **3** with **4b** is assumed to proceed via addition followed by ethanol elimination.

Compound **3** also reacted with the highly activated cyano group of trichloroacetonitrile to give the pyranopyrimidine derivatives **10**. Compounds **10** are assumed to be formed via the intermediacy of **9**.

Compound **3a** reacted with benzoylisothiocyanate to yield a product C<sub>15</sub>H<sub>15</sub>N<sub>3</sub>O<sub>4</sub>S. Structure **8a** was established for this compound based on its <sup>1</sup>H-NMR spectrum which revealed a NH<sub>2</sub> singlet at  $\delta$  = 8.0 ppm. The formation of **8a** might be assumed to proceed via addition of benzoylisothiocyanate to yield the intermediate **7a**, which cyclises under the reaction condition and then hydrolyses into the finally isolated product **8a**.

Compound **3** reacted with acetic anhydride to afford the pyranopyrimidines **11**. These results indicate that the reac-



tion of **3** with suitable active functionally substituted nitriles and isothiocyanates can be utilised as an effective route for the synthesis of several, otherwise difficultly accessible heterocyclic derivatives.

## Experimental

Melting points uncorrected. – IR spectra: (KBr) Pye Unicam sp-1000 spectrophotometer. – <sup>1</sup>H-NMR spectra: Varian EM-390 90 MHz spectrometer, DMSO as solvent, TMS as int. reference. Chemical shifts in δ units (ppm). – Mass spectra: MS 30 (AEI), 70 ev. – Analytical data: Microanalytical Centre, Cairo University.

### Reaction of **1** with ethyl acetoacetate

To a suspension of each of **1a**, **b** (0.01 mole) in ethanol (30 ml) and ethyl acetoacetate (0.01 mole) 1 ml of piperidine was added. The mixture was refluxed for 6 h then cooled and poured into water. The solid product was collected and crystallised from the proper solvent.

#### Ethyl 6-amino-5-cyano-4-furanyl(2)-2-methyl-4H-pyran 3-carboxylate (**3a**)

Violet crystals from ethanol; 85 %; m.p. 208 °C. – IR: 3430, 3360 (NH<sub>2</sub>), 2220 (CN), 1710 (CO) cm<sup>-1</sup>. – <sup>1</sup>H-NMR: 1.1 (t, 3H, CH<sub>3</sub>), 3.3 (s, 3H, CH<sub>3</sub>), 4.1 (q, 2H, CH<sub>2</sub>), 4.4 (s, 1H, pyran H-4), 6.8 (s, br, 2H, NH<sub>2</sub>), 6.9 (dd, 1H, furan H-4), 7.3 (dd, 1H, furan H-3), 7.5 (m, 1H, furan H-5). – <sup>13</sup>C-NMR: 166.2 (CO), 155.6 (pyran C-2), 154.8 (C-6), 146.2 (C-5), 142–138.2 (furan carbons), 118.1 (CN), 104.2 (C-3), 58.2 (CH<sub>2</sub> ester), 16.8 (CH<sub>3</sub>), 12.8 (CH<sub>3</sub> ester). – C<sub>14</sub>H<sub>14</sub>N<sub>2</sub>O<sub>4</sub> (274) Calcd. C 61.3 H 5.1 N 10.2 Found C 61.0 H 4.8 N 9.9.

#### Ethyl 6-amino-5-cyano-2-methyl-4-thienyl(2)-4H-pyran 3-carboxylate (**3b**)

Yellow crystals from ethanol; 80 %; m.p. 196 °C. – IR: 3450, 3380 (NH<sub>2</sub>), 2210 (CN), 1720 (CO) cm<sup>-1</sup>. – <sup>1</sup>H-NMR: 1.2 (t, 3H, CH<sub>3</sub>), 3.1 (s, 3H, CH<sub>3</sub>), 4.2 (q, 2H, CH<sub>2</sub>), 4.6 (s, 1H, pyran H-4), 6.4 (s, br, 2H, NH<sub>2</sub>), 6.8 (dd, 1H, thiophene H-4), 7.5 (m, 1H, thiophene H-3), 7.8 (dd, 1H, thiophene H-5). – MS: m/z = 290 (M<sup>+</sup>). – C<sub>14</sub>H<sub>14</sub>N<sub>2</sub>O<sub>3</sub>S (290) Calcd. C 57.9 H 4.8 N 9.3 Found C 58.0 H 4.5 N 9.1.

### Reaction of **3** with **4a**, **b**

A solution of **3** (0.01 mole) in ethanol (30 ml), malononitrile or ethyl cyanoacetate (0.01 mole) and 1 ml of triethylamine was refluxed for 5 h, then cooled and poured into water. The solid product was collected and crystallised from the proper solvent.

#### Ethyl 6-cyano-5,7-diamino-4-furanyl(2)-2-methyl-4H-pyranol 2,3-bispyridine-3-carboxylate (**6a**)

Yellow crystals from dioxane; 65 %; m.p. 242–4 °C. – IR: 3450, 3360 (NH<sub>2</sub>); 2220 (CN); 1710 (CO) cm<sup>-1</sup>. – <sup>1</sup>H-NMR: 1.1 (t, 3H, CH<sub>3</sub>); 3.2 (s, 3H, CH<sub>3</sub>); 3.6 (s, br, 2H, NH<sub>2</sub>); 4.2 (q, 2H, CH<sub>2</sub>); 4.6 (s, 1H, pyran H-4); 5.6 (s, br, 2H, NH<sub>2</sub>); 6.8–7.9 (m, 3H, furan H-3,4,5). – C<sub>17</sub>H<sub>14</sub>N<sub>4</sub>O<sub>4</sub> (338) Calcd. C 60.0 H 4.7 N 16.5 Found C 59.8 H 4.6 N 16.2.

#### Ethyl 6-cyano-5,7-diamino-2-methyl(4)-thienyl-2-4H-pyranol 2,3-bispyridine-3-carboxylate (**6b**)

Buff crystals from dioxane; 60 %; m.p. 236 °C. – IR: 3400, 3360 (NH<sub>2</sub>), 2210 (CN), 1700 (CO) cm<sup>-1</sup>. – <sup>1</sup>H-NMR: 1.3 (t, 3H, CH<sub>3</sub>); 2.6 (s, 3H, CH<sub>3</sub>); 3.5 (s, br, 2H, NH<sub>2</sub>); 4.1 (q, 2H, CH<sub>2</sub>); 4.8 (s, 1H, pyran H-4); 5.5 (s, br, 2H, NH<sub>2</sub>); 6.6 (dd, 1H, thiophene H-4); 7.0 (m, 1H, thiophene H-3); 7.7 (m, 1H, thiophene H-5). – C<sub>17</sub>H<sub>16</sub>N<sub>4</sub>O<sub>3</sub>S (356) Calcd. C 57.3 H 4.5 N 15.7 Found C 56.8 H 4.2 N 15.5.

#### Ethyl 5-amino-6-cyano-7,8-dihydro-4-furanyl-2-methyl-7-oxo-4H-pyranol 2,3-bispyridine-3-carboxylate (**6c**)

Yellow crystals from dioxane; 66 %; m.p. 222–4 °C. – IR: 3450–3320 (OH); 2220 (CN); 1695 (CO) cm<sup>-1</sup>. – C<sub>17</sub>H<sub>13</sub>N<sub>3</sub>O<sub>3</sub> (341) Calcd. C 59.8 H 4.4 N 12.3 Found C 39.6 H 4.2 N 12.0.

#### Ethyl 5-amino-6-cyano-7,8-dihydro-2-methyl-7-oxo-4-thienyl-4H-pyranol 2,3-bispyridine-3-carboxylate (**6c**)

Yellow crystals from ethanol; 55 %; m.p. 228–30 °C. – IR: 3420–3200 (OH); 2210 (CN); 1700 (CO) cm<sup>-1</sup>. – C<sub>17</sub>H<sub>13</sub>N<sub>3</sub>O<sub>4</sub>S (357) Calcd. C 57.1 H 4.2 N 11.8 Found C 56.8 H 4.0 N 11.6.

### Reaction of **3** with trichloroacetonitrile

**3** (0.01 mole) and trichloroacetonitrile (0.01 mole) were refluxed in dry benzene (30 ml) with a catalytic amount of triethylamine for 8 h. The solid product was collected and crystallised from the proper solvent.

#### Ethyl 4-amino-5-furanyl(2)-7-methyl-2-trichloromethyl-4H-pyranol 2,3-dipyrimidine-6-carboxylate (**10a**)

Buff crystals from benzene; 68 %; m.p. 247–9 °C. – IR: 3480, 3360 (NH<sub>2</sub>); 1700 (CO) cm<sup>-1</sup>. – <sup>1</sup>H-NMR: 1.1 (t, 3H, CH<sub>3</sub>); 2.6 (s, 3H, CH<sub>3</sub>); 4.1 (q, 2H, CH<sub>2</sub>); 4.8 (s, 1H, pyran H-4); 6.2 (s, br, 2H, NH<sub>2</sub>); 6.8 (m, 1H, furan H-4); 7.3 (m, 1H, furan H-3); 7.6 (m, 1H, furan H-5). – C<sub>16</sub>H<sub>14</sub>N<sub>3</sub>O<sub>4</sub>Cl<sub>3</sub> (417) Calcd. C 46.0 H 3.4 N 10.1 Found C 45.8 H 3.0 N 9.8.

#### Ethyl 4-amino-7-methyl-5-thienyl(2)-2-trichloromethyl-4H-pyranol 2,3-dipyrimidine-6-carboxylate (**10b**)

Brown crystals from dioxane; 70 %; m.p. 244–5 °C. – IR: 3400, 3360. (NH<sub>2</sub>); 1695 (CO) cm<sup>-1</sup>. – <sup>1</sup>H-NMR: 1.2 (t, 3H, CH<sub>3</sub>); 2.8 (s, 3H, CH<sub>3</sub>); 4.2 (q, 2H, CH<sub>2</sub>); 4.6 (s, 1H, pyran H-4); 6.4 (s, br, 2H, NH<sub>2</sub>); 6.7–7.5 (m, 3H, thiophene H-3, 4, 5). – C<sub>16</sub>H<sub>14</sub>N<sub>3</sub>O<sub>3</sub>SCl<sub>3</sub> (433) Calcd. C 44.3 H 3.2 N 9.7 Found C 44.0 H 3.0 N 9.6.

### Reaction of **3** with benzoyl isothiocyanate

A solution of **3** (0.01 mol) in dry dioxane (30 ml) and the appropriate amount of benzoyl isothiocyanate was refluxed for 5 h. The reaction mixture was cooled and poured onto water. The oily product, so formed was left over night, the precipitated solid was collected and crystallised from the proper solvent.

#### Ethyl 4-amino-1,2-dihydro-4-furanyl(2)-7-methyl-2-thioxo-5H-pyranol 2,3-dipyrimidine-6-carboxylate (**8a**)

Orange crystals from methanol; 75 %; m.p. 196 °C. – IR: 3480–3680 (NH<sub>2</sub>), 1710 (CO) cm<sup>-1</sup>. – <sup>1</sup>H-NMR: 1.2 (t, 3H, CH<sub>3</sub>); 2.6 (s, 3H, CH<sub>3</sub>); 4.1 (q, 2H, CH<sub>2</sub>); 5.2 (s, 1H, pyran H-4); 6.7–7.8 (m, 3H, furan H-3,4,5); 8.0 (s, br, 2H, NH<sub>2</sub>). – C<sub>15</sub>H<sub>13</sub>N<sub>3</sub>O<sub>4</sub>S (333) Calcd. C 54.1 H 4.5 N 12.6 Found C 53.8 H 4.2 N 12.4.

#### Ethyl 4-amino-1,2-dihydro-7-methyl-4-thienyl(2)-2-thioxo-5H-pyranol 2,3-dipyrimidine-6-carboxylate (**8b**)

Yellow crystals from ethanol; 70 %; m.p. 191 °C. – IR: 3450, 3400, 3750 (NH<sub>2</sub>); 1705 (CO) cm<sup>-1</sup>. – <sup>1</sup>H-NMR: 1.1 (t, 3H, CH<sub>3</sub>); 2.4 (s, 3H, CH<sub>3</sub>); 4.2 (q, 2H, CH<sub>2</sub>); 5.4 (s, 1H, pyran H-4); 6.8–7.9 (m, 3H, thiophene H-3,4,5); 8.2 (s, br, 2H, NH<sub>2</sub>). – C<sub>15</sub>H<sub>13</sub>N<sub>3</sub>O<sub>3</sub>S<sub>2</sub> (349) Calcd. C 51.6 H 4.3 N 12.0 Found C 51.8 H 4.2 N 11.7.

### Reaction of **3** with acetic anhydride

A solution of **3** (0.01 mol) in acetic anhydride (30 ml) was refluxed for 4 h, the solvent was then evaporated under vacuum to about 25 % of its original volume and left to cool, the solid product, so formed, was collected and crystallised from the proper solvent.

*Ethyl 3,4-dihydro-4-furanyl-2,2,7-dimethyl-4-oxo-5H-pyrano[2,3-d]pyrimidine-6-carboxylate (11a)*

Colourless crystals from ethanol, 82 %; m.p. 279–80 °C. – IR: 3200–2850 (OH), 1685 (CO)  $\text{cm}^{-1}$ . –  $^1\text{H-NMR}$ : 1.1 (t, 3H,  $\text{CH}_3$ ); 2.1 (s, 3H,  $\text{CH}_3$ ); 2.6 (s, 3H,  $\text{CH}_3$ ); 4.1 (q, 2H,  $\text{CH}_2$ ); 4.6 (s, 1H, pyran H-4). 6.8–7.8 (m, 3H, furan H-3,4,5), 12.6 (s, br, 1H, NH). –  $\text{C}_{16}\text{H}_{16}\text{N}_2\text{O}_5$  (316) Calcd. 60.8 H 5.1 N 8.9 Found C 60.6 H 4.8 N 8.8.

*Ethyl 3,4-dihydro-2,7-dimethyl-4-oxo-4-thienyl-5H-pyrano[2,3-d]pyrimidine-6-carboxylate (11b)*

Yellow crystals from methanol, 60 %; m.p. 272–4 °C. – IR: 3350–3200 (OH), 1680 (CO)  $\text{cm}^{-1}$ . –  $^1\text{H-NMR}$ : 1.2 (t, 3H,  $\text{CH}_3$ ); 2.2 (s, 3H,  $\text{CH}_3$ ); 2.8 (s, 3H,  $\text{CH}_3$ ); 4.2 (q, 2H,  $\text{CH}_2$ ); 4.8 (s, 1H, pyran H-4); 6.7–7.9 (m, 3H,

thiophene H-3,4,5); 12.9 (s, br, 1H, NH). –  $\text{C}_{16}\text{H}_{16}\text{N}_2\text{O}_4\text{S}$  (332) Calcd. C 57.8 H 4.8 N 8.4 Found C 57.6 H 4.6 N 8.0.

## References

- 1 G. E. H. Elgemeie, H. A. Elfahham, S. Elgamal, and M. H. Elnagdi, *Heterocycles* **23**, 1999 (1985).
- 2 G. E. H. Elgemeie, M. M. Sallam, S. Mourad, and M. H. Elnagdi, *Heterocycles* **23**, 3107 (1985).
- 3 G. E. H. Elgemeie and F. A. Aal, *Heterocycles* **24**, 349 (1986).
- 4 G. E. H. Elgemeie, B. Y. Riad, G. A. Nawwar, and S. Elgamal, *Arch. Pharm. (Weinheim)* **320**, 223 (1987).

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