

PREPARATION OF DIFFERENTIATED DIAMIDES OF 4,5,6,7-TETRAHYDROPYRAZOLO[1,5-*a*]PYRIDINE- 2,6- AND -3,6-DICARBOXYLIC ACIDS SUITABLE FOR PARALLEL SYNTHESIS

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*1,3-Dipolar cycloaddition of a bicyclic sydnone to a propargylic acid amide affords 4,5,6,7-tetrahydropyrazolo[1,5-*a*]pyridine-2,6- and -3,6-dicarboxylic acid monoamides which are further converted to corresponding differentiated diamides.*

Keywords: sydrones, 4,5,6,7-tetrahydropyrazolo[1,5-*a*]pyridines, 1,3-dipolar cycloaddition.

The pyrazole core represents an important class of nitrogen-containing heterocycles since it is a structural element of natural products [1] and pharmaceutically active compounds [2–4]. The bicyclic pyrazole motif is a fragment of numerous biologically active compounds [5–8]. The partially saturated bicyclic pyrazole system, 4,5,6,7-tetrahydropyrazolo[1,5-*a*]pyridine, has found use as a scaffold in drug research [9, 10] and agrochemicals [11].

Within the medicinal chemistry program in our laboratories, we considered a diversity-oriented library synthesis around this scaffold, specifically, the synthesis of differentiated diamides of 4,5,6,7-tetrahydropyrazolo[1,5-*a*]pyridine-2,6-dicarboxylic acid **1a** and 4,5,6,7-tetrahydropyrazolo[1,5-*a*]pyridine-3,6-dicarboxylic acid **1b**. The literature search revealed that only few structural modifications of 4,5,6,7-tetrahydropyrazolo[1,5-*a*]pyridine [12], the corresponding 2-carboxylic acid derivatives [13, 14], and boranes [14] are known. Furthermore, 6-substituted 4,5,6,7-tetrahydropyrazolo[1,5-*a*]pyridine derivatives have not been reported.

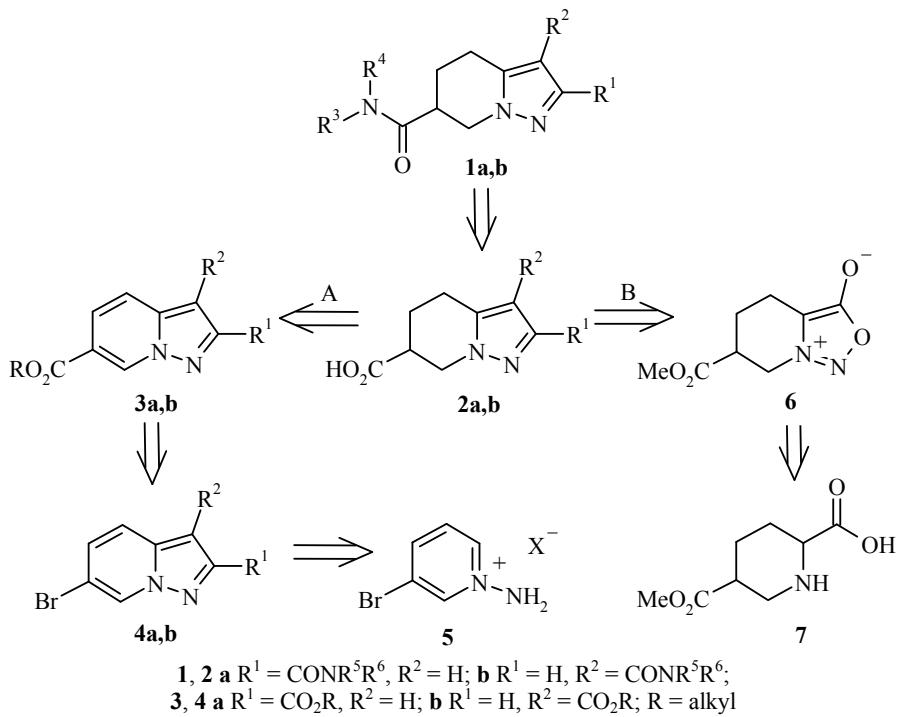
Two principally distinct synthetic routes towards 4,5,6,7-tetrahydropyrazolo[1,5-*a*]pyridine derivatives have been reported. Route A is based on the hydrogenation of pyrazolo[1,5-*a*]pyridines 3a,b [9, 10], and route B – on the pyrazole ring closure by 1,3-dipolar cycloaddition of bicyclic sydnone 6 to propargylic acid derivatives. Retrosynthetic analysis of target compounds showed that for route A, an intermediate pyrazolo[1,5-*a*]pyridine-2- or -3-carboxylic acid derivatives 4a,b would be required. Pyrazolo[1,5-*a*]pyridine-2-carboxylic acid derivatives 4a are synthesized by 1,3-dipolar cycloaddition of 1-aminopyridinium species 5 to acetylene dicarboxylic acid esters and subsequent monodecarboxylation [15], whereas the corresponding

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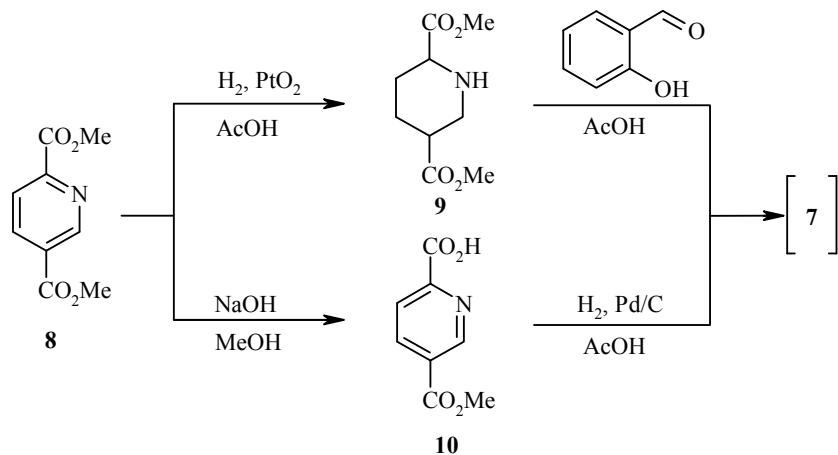
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3-carboxylic acid derivatives **4b** are obtained in the reaction of compound **5** with propiolic acid derivatives [15, 16]. Therefore route A is not appropriate for parallel synthesis. This consideration, together with potential problems associated with differentiation of two carboxylic acids/ester/amide functionalities, prompted us to choose route B as synthetically more attractive.



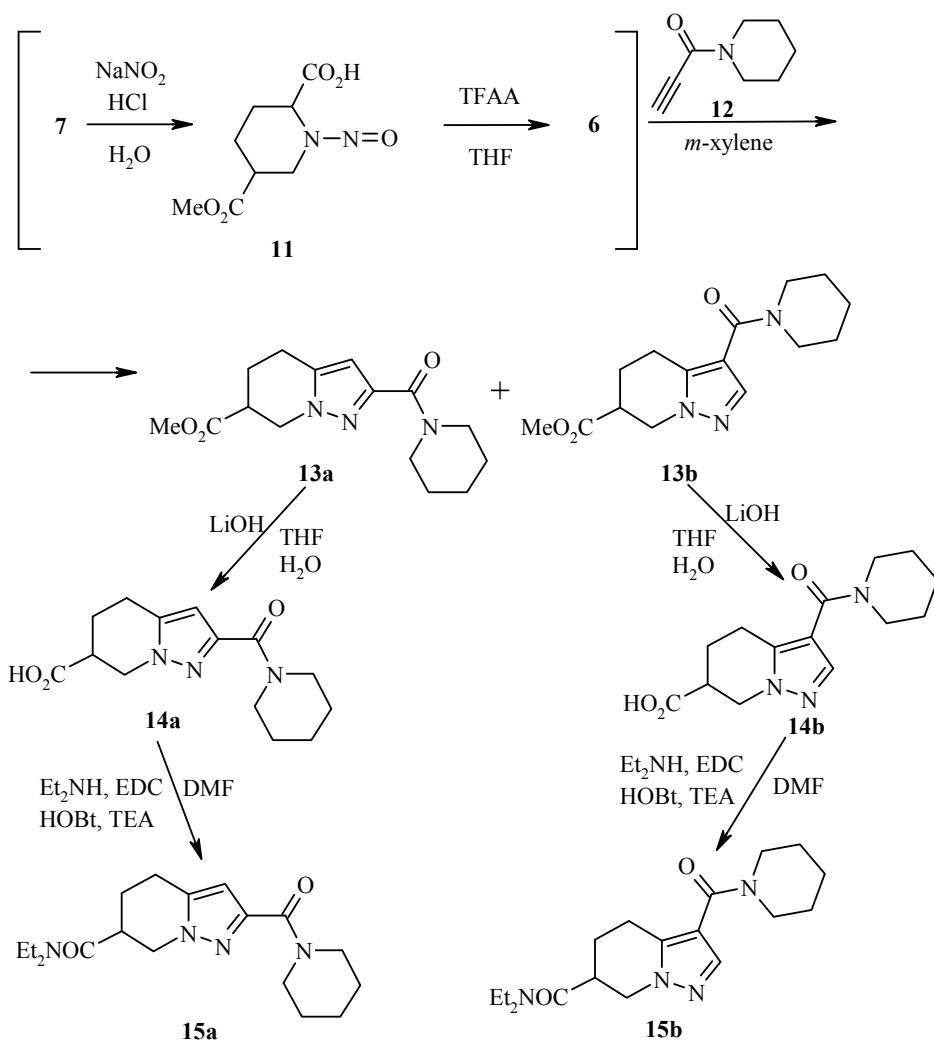
Route B is based on 1,3-dipolar cycloaddition of sydrones **6** to acetylenes. Although successful 1,3-dipolar cycloaddition of sydnone **6** to olefins [17], acetylenes [18], and propargylic acid esters [13, 19] is documented, there is no literature precedent of such reaction with propargylic acid amides. To enable the differentiation of the carboxylic amide at the 2- or 3-position from the ester at 6-position, we chose sydnone **6** 1,3-dipolar cycloaddition to propargylic amides as our approach.

The synthesis of key intermediate **2** began with the preparation of piperidine-2,5-dicarboxylic acid 5-methyl ester (**7**). According to the literature procedure [20], pyridine-2,5-dicarboxylic acid dimethyl ester (**8**) can be hydrogenated to piperidine **9** and then converted to compound **7**.



Although the described method is suitable for preparation of compound **7**, we found that saturation of the pyridine ring of compound **10** in the last step is more convenient. Dimethyl ester **8** was converted to monomethyl ester **10** [21] and then hydrogenated to provide the target piperidine **7**, which was used without purification.

Sydnone **6** was readily prepared in two steps. Piperidine-2-carboxylic acid **7** was converted to *N*-nitroso-piperidine **11**, treatment of which with trifluoroacetic anhydride (TFAA) according to the literature procedure [13, 17] led to sydnone **6**. 1,3-Dipolar cycloaddition of sydnone **6** to 1-piperidin-1-yl-propynone (**12**) led to the formation of two regioisomers **13a** and **13b**, which were easily separated by column chromatography on silica gel to give the desired compounds **13a** and **13b** in ~1:2 ratio in overall yield of 30% in four steps. The alkaline hydrolysis of esters **13a** and **13b** led to the key intermediates **14a** and **14b** in good yield.



The obtained acids **14a** and **14b** can be used in the parallel synthesis of different amides, which we exemplified by preparing diethylamides **15a** and **15b**.

In conclusion, a method for preparing differentiated diamides of 4,5,6,7-tetrahydropyrazolo[1,5-*a*]pyridine-2,6- and -3,6-dicarboxylic acids, suitable for parallel synthesis, has been elaborated.

EXPERIMENTAL

¹H NMR spectra were recorded on a Varian Mercury (200 MHz, compounds **13a,b**, **14a,b**) and Varian MR (400 MHz, compounds **15a,b**) spectrometers in CDCl₃. Chemical shifts are referenced to HMDS as the internal standard ($\delta = 0.05$ ppm) or using the signal of the residual protonated solvent ($\delta = 7.26$ ppm). High-resolution mass spectra analyses were performed using a Micromass Q-TOF Micro quadrupole time-of-flight high-resolution mass spectrometer with ESI ionization. Leucine enkephalin was used as internal lock mass for accurate mass calculation (*m/z* 566.2771). LC-mass spectra analyses were performed on a Shimadzu Prominence system connected to the Applied Biosystems API 2000 mass spectrometer (atmospheric pressure ionization).

Reagents and solvents were purchased from Aldrich, Acros and Alfa Aesar Chemical Industries. Pyridine-2,5-dicarboxylic acid 5-methyl ester hydrogen chloride (**10**) was prepared from pyridine-2,5-dicarboxylic acid dimethyl ester (**8**) according to the literature procedure [21].

2-(Piperidin-1-ylcarbonyl)-4,5,6,7-tetrahydropyrazolo[1,5-*a*]pyridine-6-carboxylic Acid Methyl Ester (13a) and 3-(Piperidin-1-ylcarbonyl)-4,5,6,7-tetrahydro-pyrazolo[1,5-*a*]pyridine-6-carboxylic Acid Methyl Ester (13b). Pyridine-2,5-dicarboxylic acid 5-methyl ester hydrogen chloride (**10**) (2.0 g, 9.2 mmol) was dissolved in a mixture of AcOH (40 ml) and MeOH (5 ml), and 10% Pd/C (0.53 g, 0.5 mmol) was added. Hydrogenation was carried out at room temperature at 10 atm for 24 h. The reaction mixture was filtered through Celite, washed with EtOAc, and evaporated to give 0.77 g of the **piperidine-2,5-dicarboxylic acid 5-methyl ester (7)** as a white solid which was used without purification in the next step. LC-MS, *m/z*: 188 [M+H]⁺.

To a solution of methyl ester **7** in water (5 ml) at 0°C, concentrated hydrochloric acid (0.6 ml) was added, followed by NaNO₂ (0.885 g, 12.8 mmol) in small portions. The mixture was stirred at 0°C for 1 h, extracted with CHCl₃, dried over Na₂SO₄, filtered, and evaporated under reduced pressure at 30–35°C. The residue containing **1-nitroso-piperidine-2,5-dicarboxylic acid 5-methyl ester (11)** was immediately used in the next step.

To a stirred solution of compound **11** in dry THF (20 ml) at 0°C under argon atmosphere, a solution of trifluoroacetic anhydride (1 ml, 7 mmol) in THF (5 ml) was added dropwise. The mixture was stirred at 0°C for 5 h and then 16 h at room temperature. The solvent was evaporated and the crude material diluted with EtOAc and stirred with anhydrous K₂CO₃ (1 g). The mixture was filtered through a pad of silica gel, and the filtrate was evaporated. The obtained brownish oil containing **6-(methoxycarbonyl)-4,5,6,7-tetra-hydro[1,2,3]oxadiazolo-[3,4-*a*]pyridin-8-i um-3-olate (6)** was used in the next step without purification.

To a stirred solution of compound **6** in *m*-xylene (5 ml), propargylic acid amide **12** [22] (1.26 g, 9.2 mmol) was added, the reaction mixture was heated at 140°C for 8 h and evaporated; the residue was purified by flash chromatography on silica gel (eluent hexane–ethyl acetate, 1:1 to 1:4) to obtain 0.55 g of compound **13a** and 0.25 g of compound **13b** as oils. The overall yield of both products together was 30% (over 4 steps).

Compound 13a. ¹H NMR spectrum, δ , ppm (*J*, Hz): 1.49–1.65 (6H, m, 3',4',5'-CH₂ piperidine); 1.85–2.05 (1H, m) and 2.20–2.26 (1H, m, 5-CH₂); 2.74–3.06 (3H, m, 4-CH₂, 6-CH); 3.63–3.74 (4H, m, 2',6'-CH₂ piperidine); 3.71 (3H, s, OCH₃); 4.22 (1H, dd, *J* = 13.2, *J* = 8.8) and 4.37 (1H, dd, *J* = 13.2, *J* = 5.9, 7-CH₂); 6.25 (1H, s, H-3). Found: *m/z* 292.1659 [M+H]⁺. C₁₅H₂₂N₃O₃. Calculated: M 292.1661.

Compound 13b. ¹H NMR spectrum, δ , ppm (*J*, Hz): 1.58–1.63 (6H, m, 3',4',5'-CH₂ piperidine); 1.87–2.06 (1H, m) and 2.25–2.34 (1H, m, 5-CH₂); 2.82–3.21 (3H, m, 4-CH₂, 6-CH); 3.56–3.62 (4H, m, 2',6'-CH₂ piperidine); 3.74 (3H, s, OCH₃); 4.24 (1H, dd, *J* = 13.2, *J* = 8.8) and 4.40 (1H, dd, *J* = 13.2, *J* = 5.9, 7-CH₂); 7.48 (1H, s, H-2). Found: *m/z* 292.1659 [M+H]⁺. C₁₅H₂₂N₃O₃. Calculated: M 292.1661.

2-(Piperidin-1-ylcarbonyl)-4,5,6,7-tetrahydropyrazolo[1,5-*a*]pyridine-6-carboxylic Acid (14a). To a solution of **13a** (58 mg, 0.20 mmol) in THF (1 ml), water (3 ml) was added followed by LiOH·H₂O (9 mg, 0.22 mmol), and the mixture was stirred at room temperature for 2 h. The THF was evaporated, and the water layer was acified with 10% HCl and extracted with EtOAc. The organic layer was dried over Na₂SO₄, filtered, and evaporated to give 55 mg (99%) of the desired acid **14a** as a yellow oil. ¹H NMR spectrum, δ , (ppm)

(*J*, Hz): 1.50–1.74 (6H, m, 3',4',5'-CH₂ piperidine); 2.01–2.28 (2H, m, 5-CH₂); 2.70–3.07 (3H, m, 4-CH₂, 6-CH); 3.69–3.77 (4H, m, 2',6'-CH₂ piperidine); 4.24–4.38 (2H, m, 7-CH₂); 6.30 (1H, s, H-3); 7.78 (1H, br. s, OH). Mass spectrum, *m/z*: 278 [M+H]⁺.

3-(Piperidin-1-ylcarbonyl)-4,5,6,7-tetrahydropyrazolo[1,5-*a*]pyridine-6-carboxylic Acid (14b) was synthesized in the same manner as compound **14a** from ester **13b** (65 mg, 0.22 mmol) to give 48 mg (78%) of acid **14b** as a yellow oil. ¹H NMR spectrum, δ , ppm (*J*, Hz): 1.50–1.75 (6H, m, 3',4',5'-CH₂ piperidine); 1.98–2.29 (2H, m, 5-CH₂); 2.81–3.18 (3H, m, 4-CH₂, 6-CH); 3.51–3.70 (4H, m, 2',6'-CH₂ piperidine); 4.26–4.44 (2H, m, 7-CH₂); 7.51 (1H, s, H-2); 7.78 (1H, br. s, OH). Mass spectrum, *m/z*: 278 [M+H]⁺.

2-(Piperidin-1-ylcarbonyl)-4,5,6,7-tetrahydropyrazolo[1,5-*a*]pyridine-6-carboxylic Acid Diethylamide (15a). To a solution of acid **14a** (55 mg, 0.20 mmol), HOBr·H₂O (41 mg, 0.24 mmol), EDC (46 mg, 0.24 mmol) in DMF (1.5 ml), diethylamine (25 μ l, 0.24 mmol) and TEA (30 μ l, 0.24 mmol) were added. The mixture was stirred at room temperature overnight. The reaction mixture was diluted with H₂O and extracted with EtOAc. The organic layer was washed with brine, dried over Na₂SO₄, filtered and evaporated. The residue was purified by flash chromatography on silica gel (eluent hexane–ethyl acetate, 1:1) to give 18 mg (27%) of the desired compound **15a** as an oil. ¹H NMR spectrum, δ , ppm (*J*, Hz): 1.14 (3H, t, *J* = 7.4, NCH₂CH₃); 1.23 (3H, t, *J* = 7.4, NCH₂CH₃); 1.57–1.70 (6H, m, 3',4',5'-CH₂ piperidine); 2.03–2.09 (2H, m, 5-CH₂); 2.73–2.82 (1H, m, 6-CH); 3.02 (1H, dt, *J* = 16.8, *J* = 4.3, 4-CH); 3.07–3.15 (1H, m, 4-CH); 3.31–3.53 (4H, m, 2NCH₂CH₃); 3.64–3.72 (2H, m) and 3.78–3.85 (2H, m, 2',6'-CH₂ piperidine); 4.24–4.33 (2H, m, 7-CH₂); 6.32 (1H, s, H-3). Found: *m/z* 333.2281 [M+H]⁺. C₁₈H₂₉N₄O₂. Calculated: M 333.2291.

3-(Piperidin-1-ylcarbonyl)-4,5,6,7-tetrahydropyrazolo[1,5-*a*]pyridine-6-carboxylic Acid Diethylamide (15b) was synthesized in the same manner as compound **15a** from acid **14b** (48 mg, 0.17 mmol). Compound **15b** was obtained as a yellow oil, yield 21 mg (36%). ¹H NMR spectrum, δ , ppm (*J*, Hz): 1.13 (3H, t, *J* = 7.4, NCH₂CH₃); 1.22 (3H, t, *J* = 7.4, NCH₂CH₃); 1.59–1.71 (6H, m, 3',4',5'-CH₂ piperidine); 1.97–2.11 (2H, m, 5-CH₂); 2.89–2.96 (1H, m, 6-CH); 3.06–3.14 (1H, m, 4-CH); 3.19 (1H, ddd, *J* = 18.0, *J* = 5.5, *J* = 2.4, 4-CH); 3.27–3.46 (4H, m, 2NCH₂CH₃); 3.60–3.62 (4H, m, 2',6'-CH₂ piperidine); 4.23–4.33 (2H, m, 7-CH₂); 7.48 (1H, s, H-2). Found: *m/z* 333.2281 [M+H]⁺. C₁₈H₂₉N₄O₂. Calculated: M 333.2291.

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