

γ -Turn Mimicry with Benzodiazepinones and Pyrrolobenzodiazepinones Synthesized from a Common Amino Ketone Intermediate

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Supporting Information

$$(S) - \textbf{6f} : \phi = -72; \ \psi = 69 \\ \text{inverse } \gamma \text{-turn} : \phi = -75; \ \psi = 65$$

ABSTRACT: To investigate diazepinone analogues as γ -turn mimics, seven 1,4-benzodiazepin-2-ones 6 and fourteen pyrrolo[1,2-d][1,4]benzodiazepin-6-ones 4 and 5 were synthesized from 1-(2-aminophenyl)pent-4-en-1-one (7). Acylation of aniline 7 with N-Boc-amino acids, olefin oxidation, Boc removal, and intramolecular Paal–Knorr condensation gave 4 and 5. Alternatively, Boc removal prior to oxidation gave benzodiazepinones 6, which were converted to 4 by ozonolysis and cyclization. Comparison of dihedral angle values for the amino acid component from X-ray analyses of 4g, 5f, and 6f and related diazepinones has catalogued the manner by which ring substituents affect the component's ability to mimic the central residues of γ -turns.

Turn mimicry may lie at the heart of the wide range of biological, pharmacological, and clinical activities of 1,4-diazepin-2-ones and their aryl-fused ring analogues. Considered "privileged structures" because they exhibit high affinity to multiple receptors, $^{1-3}$ diazepinones have the capacity to mimic natural γ - and β -turn peptide secondary structures (Figure 1). 4,5

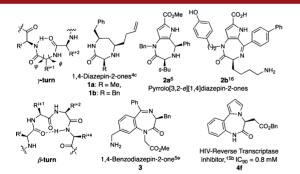


Figure 1. Representative diazepinones and γ - and β -turns.

X-ray structural analyses of diazepin-2-ones (e.g., 1 and 2) have shown that their amino acid components adopt dihedral angle values similar to those of the central residue in a γ -turn (Figure 1). Ac,6,16 Moreover, spectroscopic and computational analyses have revealed that benzodiazepin-2-one scaffolds (e.g., 3) can mimic β -turn backbone and side-chain geometry. Se,f Knowledge of ring substituent influences on turn geometry is thus valuable for rational application of such scaffolds in drug discovery.

1,4-Benzodiazepines are CNS drugs that show sedative, anxiolytic, anticonvulsant, antihypnotic, muscle relaxing, and anterograde amnesia activities. They modulate hormone receptors, block ion channels, and inhibit enzymes. They exhibit antimalarial, sh,100 antible by inhibiting protein—DNA interactions.

Pyrrolo[2,1-c][1,4]benzodiazepinones exhibit anticancer and antitumor activity due to their ability to sequence-selectively bind and alkylate double-stranded DNA. Their synthetic dimers are potent DNA cross-linking agents that have entered phase II clinical trials. Pyrrolobenzodiazepinones have also shown anti-ischemic and herbicidal activity. Pyrrolo[1,2-d][1,4]benzodiazepinones (e.g., 4f) act as non-nucleoside HIV-1 reverse transcriptase inhibitors, albeit few publications have reported on this tricyclic system.

Interested in their potential as turn mimics, 4c,6,16 we have pursued diazepinone analogues (e.g., 1 and 2) using solution- 4c,6,18a and solid-phase 16,17,18b methods featuring diastereoselective Pictet—Spengler reactions 6,16,17 and coppercatalyzed cascade addition of vinyl Grignard reagents to α -aminoacyl- β -amino esters to afford γ , δ -unsaturated ketone precursors. Employing the latter reaction on methyl anthranilate, we synthesized effectively 1-(2-aminophenyl)-pent-4-en-1-one (7), which has previously served to make

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aminophenylpyrroles, ¹⁹ pyrrolo[1,2-c]quinazolin-5(6H)-ones, ¹⁹ and 2,3,4-trisubstituted quinolines (e.g., 8-10, Figure 2). ²⁰ To gain more insight into tools for γ -turn mimicry, ²¹ benzodiazepinone and pyrrolobenzodiazepinone analogues 4-6 have now been made from amino ketone 7 and examined by X-ray diffraction.

Figure 2. Amino ketone 7 as common precursor for heterocycle synthesis.

Among benzodiazepine syntheses, 22 the preparation of 1,4-benzodiazepin-2-ones has been less widely explored. 1a,5a,b,8,9a,c,10a,23,24 We perceived ring annulation by an N(4)-C(5) bond connection as an attractive route to 1,4-benzodiazepin-2-ones (e.g., 6) because the requisite precursor may be synthesized by amino acylation of 2-aminophenones with amino acid derivatives. We have now validated this synthetic strategy by surmounting issues, low yield and racemization, in the amino acylation of electron-poor aniline 7. Pyrrolo[1,2-d][1,4]benzodiazepinones 4 and 5 were also made by using 2-aminophenone 7 as a common precursor.

Electron-deficient aniline 7 was coupled to *N*-Boc-L-α-amino acids **11a**–**g** using DCC and DMAP in DCM at room temperature for 40 h to give amides **12** after chromatography on silica gel in 67–97% yields (Scheme 1). Fears of racemization from employing DMAP during peptide coupling were, however, realized by examining acylation products **12** by chiral SFC. For example, aspartate analogue **12f** exhibited a 56:44 enantiomeric ratio of (*S*)-**12f**:(*R*)-**12f**.^{25,26} To overcome racemization, HOBt was added, and the DCC and DMAP procedure was run at 0 °C. In addition, the mixed anhydride method employing isobutyl chloroformate and *N*-methylmorpholine in THF at –15 °C was used to synthesize amide **12f**. The modified DCC coupling and mixed anhydride activation

protocols provided material exhibiting, respectively, >94% and >93% enantiomeric excess by SFC analysis, albeit in 47% and 36% yields (92% and 71% yields based on recovered starting material, Scheme 1). With amino acids that are less susceptible to epimerization, such conditions may similarly yield high enantiomeric purity.

Benzodiazepinones 6 were synthesized in 65-97% yields from ketones 12 by Boc group removal with HCl gas in DCM, free-basing with triethylamine in the presence of 4 Å molecular sieves in DCM, and chromatography on silica gel (Scheme 1). Pyrrolo [1,2-d] [1,4] benzodiazepinones 4 and 5 were made from olefin 12 as the common starting material. 4-Ketoaldehydes 13 and 1,4-diones 14 were obtained, respectively, by olefin oxidation using OsO₄/NaIO₄²⁷ and Tsuji–Wacker conditions.²⁸ 4-Ketoaldehydes 13 were isolated in 61-74% yields by chromatographic purification after oxidative cleavage of olefins 12 using sodium periodate and catalytic osmium tetraoxide in dioxane/H₂O at room temperature for 45 h. Chromatography of the products after treatment of olefins 12 with PdCl₂ and CuCl in DMF/H₂O at room temperature under oxygen atmosphere for 38 h afforded 1,4-diones 14 in 58-90% yields.

Various conditions were examined to synthesize pyrrolobenzodiazepinones from the dicarbonyl compounds by intramolecular Paal-Knorr condensations (Scheme 1). 29 For example, the Boc group of 1,4-diones 14 was removed with 1:2 TFA/DCM, and the resulting trifluoroacetates were freebased using Amberlyst A-21 ion-exchange resin.³⁰ Catalytic ptoluenesulfonic acid in dilute DCM at 45 °C provided pyrrolo[1,2-d][1,4]benzodiazepinones 5 in 46-63% yields after chromatography on silica gel. Under the same protocols, however, 4-ketoaldehydes 13 gave lower yields (10-39%) of pyrrolobenzodiazepinones 4. Prior to chromatographic purification, the TLC profiles of crude 4 exhibited significant amounts of impurities. Attempts were unsuccessful to improve the yield of 4 by treating 13 with HCl gas in DCM to remove the Boc group prior to free-basing with triethylamine and Paal— Knorr cyclization. Limited success was also obtained by employing microwave irradiation of 4-ketoaldehyde 13 supported on silica gel at 100 °C to both remove the Boc group and effect in situ Paal-Knorr cyclization. On the other hand, pyrroles 4 were isolated in 52-83% yields from microwave irradiation of an aqueous mixture of 1,4-

Scheme 1. Synthesis of Benzodiazepinones 6a-g and Pyrrolobenzodiazepinones 4a-g and 5a-g from Amino Ketone 7

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ketoaldehydes 13 at 150 $^{\circ}$ C for 5–10 min and chromatography. 31

Pyrrolobenzodiazepinones 4 were also prepared from benzodiazepinones 6 in 44–77% yields by ozonolysis in CH₂Cl₂/MeOH (1:1) at -78 °C, reduction of the ozonide with excess dimethyl sulfide, and chromatography on silica gel (Scheme 1). Attempts failed, however, to prepare pyrrolobenzodiazepinone 4f by oxidation of olefin 13f using sodium periodate/osmium tetraoxide. Employing enantiomerically enriched amide 12f (>93% ee) in routes to pyrrolo[1,2-d][1,4]benzodiazepinones 4f and 5f featuring ozonolysis of benzodiazepinone 6f and Tsuji–Wacker oxidations prior to Paal–Knorr cyclization of 1,4-dione 14f gave products with high enantiomeric purity as assessed by SFC analysis (Scheme 1).

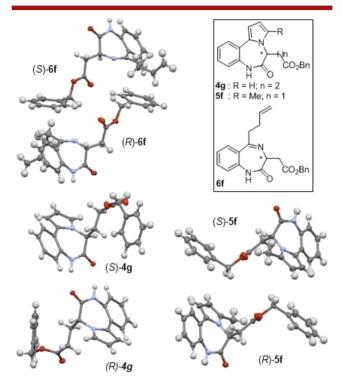


Figure 3. X-ray crystal structures of (RS)-4g, -5f, and -6f.

For X-ray analyses, crystals of benzodiazepinone (RS)-6f and pyrrolobenzodiazepinone (RS)-5f were grown by diffusion of hexanes into an ethyl acetate/hexane mixture. Crystals of pyrrolobenzodiazepinone (RS)-4g were grown by diffusion of hexanes into acetone. In both crystal structures, the enantiomers paired up in the unit cell. The amino acid side chain adopted, respectively, pseudoequatorial and pseudoaxial orientations for 6f and for 4g and 5f in the solid state. The ϕ and ψ dihedral angles of the amino acid component of (S)-4g, (S)-5f, and (R)-6f and those of their enantiomers (R)-4g, (R)-5, and (S)-6f correlated, respectively, with the central residues of classical $(75^{\circ} \pm 16$ and $-65^{\circ} \pm 10)$ and inverse $(-75^{\circ} \pm 15$ and $65^{\circ} \pm 13)$ γ -turns (Table 1).³² The subtle changes of dihedral angle geometry may affect differences in receptor affinity and activity of related diazepinone ligands.

Comparing their respective crystal structures, substituent effects were observed on the dihedral angles of the amino acid component in benzodiazepinone 6f, pyrrolobenzodiazepinones 4g and 5f, diazepinones 1a and 1b, 4c and pyrrolodiazepinone

Table 1. Diazepinone Crystal Analyses with ϕ and ψ Dihedral Angles Compared with Ideal γ -Turns

type of turn	ϕ	Ψ
γ -turn 31	75	-65
inverse γ -turn ³¹	-75	65
diazepinone 1a ^{4c}	-80	70
diazepinone 1b ^{4c}	-83	67
pyrrolodiazepinone 2a ⁶	-93	72
pyrrolodiazepinone 2b ¹⁶	-79	74
pyrrolobenzodiazepinone (S)-4g	59	-55
pyrrolobenzodiazepinone (R)-4g	-60	52
pyrrolobenzodiazepinone (S)-5f	61	-57
pyrrolobenzodiazepinone (R)-5f	-61	57
benzodiazepinone (S)-6f	-72	69
benzodiazepinone (R)-6f	75	-68

2a and 2b. 6,16 Although all of the analogues presented torsion angle values indicative of the central residue of an ideal γ -turn (75° \pm 16 and -65° \pm 10) and an ideal inverse γ -turn (-75° \pm 18 and 65° \pm 13), benzodiazepinone 6f exhibited the best fit. Relatively smaller and larger dihedral angle values were observed, respectively, for pyrrolobenzodiazepinones 4g and 5f and diazepinones 1 and pyrrolodiazepinones 2 (Table 1).

The NMR spectra of benzodiazepinones 6 and pyrrolobenzodiazepinones 4 and 5 exhibited diastereomeric as well as broad signals due to atropisomerism contingent on ring substituent and environment. Typically, benzodiazepinones 6b-g and pyrrolobenzodiazepinones 5b-g and 4c exhibited resolved spectra, with doubling of certain signals, such as the amide NH protons in chloroform. On the other hand, pyrrolobenzodiazepinones 4b-g exhibited broad proton and missing carbon signals in chloroform, due likely to equilibrating atropisomers. Improved resolution of the NMR spectra of the benzodiazepinones was obtained in acetone, which may hydrogen bond with the amide proton. Temperature could also be varied to obtain better spectral resolution. For example, the ¹H and ¹³C NMR spectra of **4b** were resolved by heating to 313 K to accelerate interconversion of the atropisomers. On the other hand, 1:1 to 3:2 diastereomeric mixtures of 4d, 4e, and 4g were well resolved at low temperatures (e.g., 193 K), which slowed interconversion of the atropisomers. Finally, the ¹H NMR spectrum of 4f at 193 K indicated a single set of resolved signals in acetone.

Insight has been gained into the capacity of diazepinone structures to mimic γ -turn conformations. Furthermore, by employing methyl anthranilate as common starting material, practical methods were conceived to synthesize benzodiazepinone and pyrrolobenzodiazepinone scaffolds. X-ray analyses demonstrated subtle effects of ring fusion and unsaturation in the 1,4-diazepinones on the capacity of their amino acid component to mimic the central residue of peptide γ -turns. This method offers interesting potential for applications in peptide mimicry and medicinal chemistry because of the importance of γ -turns in peptide biology³³ and the activity of related small molecules. Testing is underway to assess the biological activity of these novel diazepine analogues.

ASSOCIATED CONTENT

S Supporting Information

Experimental details, spectroscopic characterization (¹H and ¹³C NMR) for all compounds, SFC traces, and X-ray crystallographic data for **4g**, **5f**, and **6f**. The Supporting

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Notes

The authors declare no competing financial interest.

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