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A practical two-step synthesis of imidazo[1,2-*a*]pyridines from *N*-(prop-2-yn-1-yl)pyridin-2-amines[†]

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The Sandmeyer reaction of differently C-2 substituted N-(prop-2yn-1-ylamino)pyridines is an efficient, mild, new and practical method for the stereospecific synthesis of (*E*)-*exo*-halomethylene bicyclic pyridones bearing the imidazo[1,2-*a*]pyridine heterocyclic ring system.

Two recent communications by Chernyak and Thiel reporting the transition metal-catalyzed synthesis of imidazoheterocycles,¹ and triazolopyridines,² respectively, prompt us to report here a practical two-step synthesis of imidazo[1,2-*a*]pyridines (**3**) from *N*-(prop-2-yn-1-yl)pyridin-2-amines (**2**), prepared from appropriate precursors **1**, by a Sandmeyer reaction (SR),³ transition metal [CuCl₂, CuCl, Pd(OAc)₂, NaAuCl₄ or PtCl₂], *N*-iodosuccinimide (NIS), or acid promoted heterocyclization (Scheme 1).

In the context of our current work on the synthesis of 2-chloro(4*H*-pyrano)pyridines from the corresponding 2-amino substituted heterocycles,⁴ when 2-amino-6-(methyl(prop-2-yn-1-yl)-amino)pyridine-3,5-dicarbonitrile (**2a**), prepared from 2-amino-6-chloropyridine-3,5-dicarbonitrile (**1a**),^{6a} was submitted to SR⁵ (ESI†), instead of the expected product, a new compound was isolated in 62% yield. This product was fully characterized by its analytical, spectroscopic data, and X-ray diffraction analysis (ESI†)⁷ as the diastereomerically pure, chlorinated bicylic pyridone **3a**, bearing an imidazo[1,2-*a*]pyridine heterocyclic ring system (Scheme 2, eqn (1)).

Bicyclic pyridones,⁸ and imidazo[1,2-*a*]pyridines,⁹ such as saripidem (Scheme 1), alpidem or zolpidem, are well known pharmacologically active molecules.¹⁰

Consequently, and in view of the interest of these compounds, the increased number of communications describing



Scheme 1 Synthesis of imidazo[1,2-*a*]pyridines (3) *via* 6-(prop-2-yn-1-yl-amino)pyridines (2) from 2-amino-6-chloropyridines (1).

the transition metal promoted heterocyclic ring synthesis,^{1,2,11} and the synthetic potential of our result, the scope and the generality of this process were investigated. Based on our current interest on highly substituted pyridine derivatives for biological purposes,¹² intermediates **2b,c** (Table 1) have been synthesized from easily available precursors,⁶ and submitted to SR experimental conditions.

As shown in Table 1, SR of intermediate 2a, in the presence of CuBr₂ or CuI, afforded similar compounds 3b and 3c in 63% and 41% yields (entries a and b), respectively, as the only reaction products, in clean and highly diastereoselective reactions. SR of intermediates 2b and 2c gave the reaction products 3d and 3e (entries c and d), respectively, in good yields. The configuration at the double bond in compounds 3b-e, as well as in the other related derivatives prepared here by the same or analogous methods, has been unequivocally assigned by comparison of their NMR data with those of compound **3a**. Compound **2d**, 6c bearing an ethoxy group at C2, gave the expected bicyclopyridone 3d in 76% yield (Scheme 2, eqn (2)), identical to the product obtained from compound 2b (Table 1, entry c). With these results in mind, next we investigated the heterocyclization reaction, using intermediate 2a, without any added nitrite, and CuCl₂ or CuCl as catalysts. The reaction was still possible, but very surprisingly afforded the exomethylene product 3f in moderate to good yields (Table 1, entries e and f). Similarly, other transition metal salts, such as Pd(OAc)2, NaAuCl₄ or PtCl₂ promoted the reaction giving compound 3f in

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Scheme 2 Heterocyclization of diversely functionalized prop-2-yn-1-ylamino substituted pyridines (2).

 Table 1
 Heterocyclization reaction of 2-amino-6-(prop-2-yn-1-ylamino)-pyridines (2)

 CN
 P²

 CN
 P²

R ₁ NC 2a R ¹ = 2b R ¹ = 2c R ¹ .	$ \\ \\ NH_2 \\ H, R_2 = H, R_2 = R^2 = H$	$Me \qquad 3a-c R= Me \\ = Me \qquad 3e \qquad R= H$	CN Me N N CI NC	Sf Me
Entry	2	CuX _n	3	Yield (%)
a	2a	CuBr ₂ ^{<i>a</i>}	3b (X = Br)	63
b	2a	CuI ^a	3c(X = I)	41
с	2b	$CuCl_2^a$	3d	67
d	2c	CuCl ₂ ^a	3e(X = Cl)	55
e	2a	$CuCl_2$ (0.5 equiv.)	3f	46
f	2a	CuCl (0.25 equiv.)	3f	80
g	2a	$Pd(OAc)_2$ (0.1 equiv.)	3f	72
h	2a	NaAuCl ₄ (0.1 equiv.)	3f	76
i	2a	$PtCl_2$ (0.2 equiv.)	3f	68
^a Under	Sand	meyer reaction conditions	s.	

good yields (Table 1, entries g-i). However, compound 2c did not react under these experimental conditions with none of these

catalysts. Based on these results, we tested, but without success, the "one-pot" palladium-catalyzed $[Pd_2(dba)_3, (+)-BINAP, NaOtBu, toluene, reflux]$ Buchwald^{13a} or the $[(IPr)Ni(allyl)Cl]^{13b}$ promoted *N*-propargylation of 2-amino-6-halopyridines, followed by *in situ* heterocyclization. Conversely, we have also found that NIS,¹⁴ unlike NBS or NCS, promotes the heterocyclization reaction of intermediate **2a** to give compound **3c** (Table 1), identical to the product obtained in the SR protocol, in higher (68% *vs.* 41%) yield, but similar diastereoselectivity.

Next the scope of the reaction was further investigated by modifying the structure of the intermediates. Unfortunately, SR of N^2 -(prop-2-yn-1-yl)pyridine-2,6-diamine (2e) or 6-fluoro-N-(prop-2-ynyl)pyridin-2-amine (2f) (ESI^{\dagger}) gave complex reaction mixtures, and 6-fluoro-N-methyl-N-(prop-2-yn-1-yl)pyridin-2-amine (2g) did not react. However, when we submitted compound 2g to reaction with aqueous hydrochloric acid in 1,4-dioxane,^{8b} 1,3-dimethylimidazo[1,2-a]pyridin-5(1H)-one (3g) was obtained in 51% yield (Scheme 2, eqn (3)). Very interestingly, under the usual SR conditions, 2-(prop-2-ynylamino)nicotinonitrile (2h) provided stereospecifically imidazo[1,2-a]pyridine 3h in 77% yield (Scheme 2, eqn (4)); similarly, when only CuCl was used as catalyst, compound 3i was the sole product isolated in moderate yield (Scheme 2, eqn (5)). Finally, SR of tert-butyl prop-2-yn-1-yl(pyridin-2-yl)carbamate (2i), and N-(prop-2-yn-1-yl)pyridin-2-amine (2j),^{11c} prepared from 2-aminopyridine,¹⁵ afforded the same reaction products 3j and 3k (Scheme 2, eqn (6) and (7)) in moderate chemical yields, but still competitive regarding the methods previously described in the literature for the synthesis of aldehyde $3j^{16a}$ and nitrile 3k.^{16b} The use of CuCl₂ or NIS to promote the heterocyclization reaction did not improve these results, but only aldehvde 3i was isolated in 8% and 10% yields, respectively. The unexpected formation of compounds 3 and 3k shows that SR results on N-(prop-2-yn-1-ylamino)pyridines are very substrate-dependent, but has precedent in the I2-mediated heterocyclization of ethyl 2-(pyridin-2-yl)pent-4vnoates leading to ethyl 3-formylindolizine-1-carboxylates.¹⁷ In fact, under these reaction conditions, N-(prop-2-yn-1-yl)pyridin-2-amine (2j) gave (E)-3-(iodomethylene)-2,3-dihydro-1H-imidazo-[1,2-a] pyridin-4-ium iodide 4 (40%), and aldehyde 3i (11%) which after a Heck reaction [DIPEA (4 equiv.), Ag₂CO₃ (0.7 equiv.), PPh₃ (0.11 equiv.), Pd(OAc)₂ (0.05 equiv.), acetonitrile, reflux, 48 hl,^{11d} with 1-chloro-4-iodobenzene, afforded compound 5 in 15 % yield (not optimized) (Scheme 3), a known intermediate in the synthesis of saripidem (Scheme 1).¹⁸

The transition metal promoted (or the NIS-mediated) heterocyclization of 2-amino-6-(prop-2-yn-1-ylamino)pyridines seems to proceed through an alkyne-metal complexation followed by a pyridinic nitrogen NH_2 -(C2) assisted ring closure, ending by protonolysis/demetallation, with recovery of the catalyst,¹ affording compound **3f** (Table 1).



Scheme 3 Formal total synthesis of saripidem from imidazo-[1,2-*a*]pyridine-3-carbaldehyde (3j).



Scheme 4 A tentative free radical based mechanism for the formation of bicyclo pyridone **3a** from 2-amino-4-phenyl-6-(prop-2-yn-1-ylamino)-pyridine-3,5-dicarbonitrile (**2a**).

In Scheme 4 we show a tentative reaction mechanism for SR heterocyclization of 2-amino-6-(prop-2-yn-1-ylamino)pyridines, exemplified for intermediate **2a**, but extensive to other related intermediates. As proposed for the SR mechanism,^{5,19} the formation of nitrosyl complexes of anhydrous copper halides, strongly coordinated to the alkyne moiety, should be the first step.

Then, the weak RO-NO bond of the nitrite should be broken, leading to an alkoxy radical that would trap a hydrogen, breaking the weak H-NH(C2) bond, starting then the free radical chain reaction by a 5-exo-dig free radical heterocyclization of the nitrogen centered radical at N(1) into the activated alkyne. Next, and possibly due to the steric hindrance, the resulting vinyl radical stereospecifically would afford the chlorinated *E-exo*-chloromethylene intermediate. Final hydrolysis of the HN=(C2)N moiety would afford the bicyclic pyridone **3a**.²⁰ The formation of imidazo[1,2-a]pyridines 3i and 3k could be explained assuming that the presumed (E)-exo-halomethylene intermediate is too unstable, and decomposes in situ in the presence of traces of an external water source to the corresponding aldehyde, that evolves to aldehyde 3j by oxidation, and to nitrile 3k, by reaction with hydroxylamine, or nitrous acid, followed by dehydration.

In conclusion, we have discovered that SR, in the presence of organic nitrites and copper halides, of *N*-(prop-2-yn-1-ylamino)pyridines synthesized by reacting easily available or commercial 2-halopyridines and *N*-propargylamines is a mild, new and practical method for the stereospecific synthesis of highly substituted (*E*)-*exo*-halomethylene bicyclic pyridones bearing the imidazo[1,2-*a*]pyridine heterocyclic ring system of potential biological interest. These are suitable intermediates for further synthetic transformations and modulation. In the course of this prospective study we have also found that 2-amino-6-(prop-2-yn-1-ylamino)pyridines afford *exo/endo*methylene bicyclopyridones in transition metal-catalyzed [CuCl₂, Pd(OAc)₂, PtCl₂], NIS, or HCl/H₂O-mediated heterocylization reactions in convenient chemical yields.

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