

Diastereoselective Synthesis of 3-Hydroxyindanones via N-Heterocyclic Carbene Catalyzed [4+1] Annulation of Phthalaldehyde and 1,2-Diactivated Michael Acceptors

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Dedicated to Professors Lixin Dai and Xiyan Lu for their inspiration and encouragement on my personal (S. Ye) career in organic chemistry

Abstract: The diastereoselective synthesis of 2,2-disubstituted-3-hydroxy-indanones was realized by the N-heterocyclic carbene catalyzed [4+1] annulation of phthalaldehyde and 1,2-diactivated Michael acceptors, which involves a tandem process of Stetter reaction, proton shift, and aldol reaction.

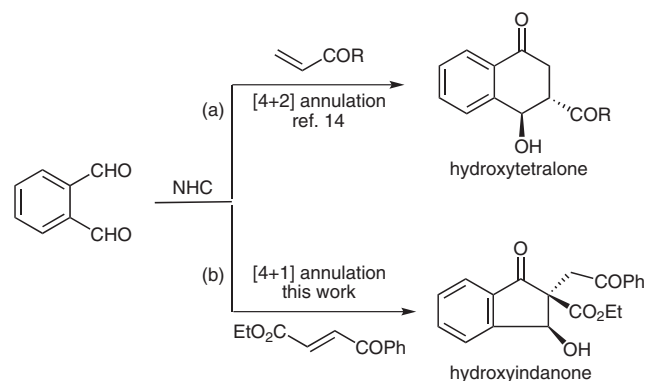
Key words: [4+1] annulation, tandem reaction, N-heterocyclic carbenes, indanone, Stetter reaction

Indanones are presented widely in many natural and unnatural bioactive compounds, various drugs and pharmaceutical candidates.¹ Thus, many methods have been developed for their construction,² including the classic intramolecular Friedel–Crafts acylation³ and Nazarov cyclization.⁴ In addition, many interesting approaches have been reported in recent years, such as the Pd-catalyzed cyclization of 3-(2-iodoaryl)propanenitriles,⁵ Rh-catalyzed [5+1+2+1] cycloaddition of vinylcyclopropane, alkynes, and CO,⁶ Rh-catalyzed hydroacylation of 2-vinyl benzaldehyde,⁷ and other reactions.⁸

As a subunit of indanones, 3-hydroxyindanones show interesting bioactivity and are useful intermediates for organic synthesis.⁹ However, the efficient methods for rapid construction of 3-hydroxyindanones are far less developed. There are only a few examples which synthesize 3-hydroxyindanones from simple starting materials.¹⁰ Representative examples include Hallberg et al.'s Heck–aldol reaction of salicylic aldehydes with vinyl ether¹¹ and Grée et al.'s intramolecular isomerisation–aldolization of allylic alcohols.¹²

Tandem reactions, which allow two or more reactions occur consecutively in one pot, are of great interest in modern synthesis.¹³ Recently, Gravel et al. reported a NHC-catalyzed tandem Stetter–Michael reaction,¹⁴ and You et al. reported a tandem aza-benzoin–Michael reaction with interesting on-site reversal of the reactivity of *N*-Boc imines.^{15,16} We reported a N-heterocyclic carbene (NHC)-catalyzed tandem Stetter–aldol reaction of phthalaldehyde and Michael acceptors to give hydroxytetralones as the [4+2]-annulation products (Scheme 1, a).¹⁷ Interestingly,

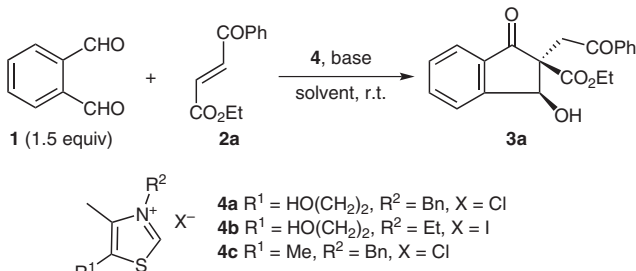
when we try to expand the substrate to 1,2-diactivated Michael acceptors, the corresponding hydroxyindanone was formed exclusively instead of hydroxytetralones (Scheme 1, b). Here we wish to report the primitive results of this tandem [4+1]-annulation reaction.



Scheme 1 NHC-catalyzed tandem reactions of phthalaldehyde and Michael acceptors

It was found that thiazolium NHC **4a'**, generated in situ from thiazolium salt **4a** in the presence of Cs₂CO₃, could catalyze the reaction of phthalaldehyde (**1**) and 1,2-diactivated Michael acceptor **2a** to give the [4+1]-annulation product of 3-hydroxyindanone **3a** in 96% yield with exclusive *cis* selectivity (Table 1, entry 1). It is interesting that thiazolium salt **4b** with *N*-ethyl substituent showed similar catalytic activity with **4a**, but salt **4c** without a free hydroxy group resulted in very low yield (entries 2 and 3). Solvent screening revealed that reaction worked well in THF, toluene, or dichloromethane but did not in diethyl ether (Table 1, entries 1, 4–6). Base screening showed K₂CO₃ and DBU worked as well as Cs₂CO₃ (Table 1, entries 7–9).

With the optimized reaction conditions in hand, the scope of the 1,2-diactivated Michael acceptors was briefly investigated (Table 2). Michael acceptors **2b,c** with electron-withdrawing group (4-O₂NC₆H₄, 4-ClC₆H₄) worked much better than **2d,e** with electron-donating group (4-MeC₆H₄, 4-MeOC₆H₄) (Table 2, entries 1–5). Michael acceptors **2f,g** with bulky substituent (2-ClC₆H₄, 2-naphthyl) also worked well (Table 2, entries 6 and 7). Considering the low yield of reaction of Michael acceptors **2d,e**

Table 1 Optimization of Reaction Conditions


$\text{1 (1.5 equiv)} + \text{2a} \xrightarrow[\text{solvent, r.t.}]{\text{4, base}}$ 3a

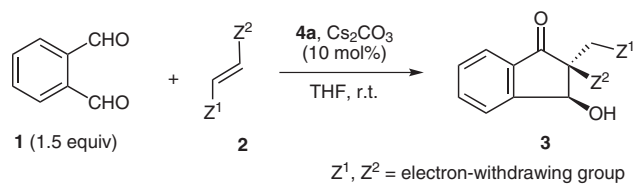
$\text{4a R}^1 = \text{HO(CH}_2)_2, \text{R}^2 = \text{Bn, X} = \text{Cl}$
 $\text{4b R}^1 = \text{HO(CH}_2)_2, \text{R}^2 = \text{Et, X} = \text{I}$
 $\text{4c R}^1 = \text{Me, R}^2 = \text{Bn, X} = \text{Cl}$

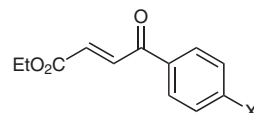
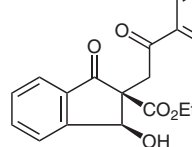
Entry	4	Base (mol%) ^a	Solvent	Yield (%) ^b
1	4a	Cs ₂ CO ₃ (20)	THF	96
2	4b	Cs ₂ CO ₃ (20)	THF	95
3	4c	Cs ₂ CO ₃ (20)	THF	20
4	4a	Cs ₂ CO ₃ (20)	toluene	89
5	4a	Cs ₂ CO ₃ (20)	CH ₂ Cl ₂	94
6	4a	Cs ₂ CO ₃ (20)	Et ₂ O	24
7	4a	Cs ₂ CO ₃ (10)	THF	95
8	4a	K ₂ CO ₃ (10)	THF	94
9	4a	DBU (10)	THF	95

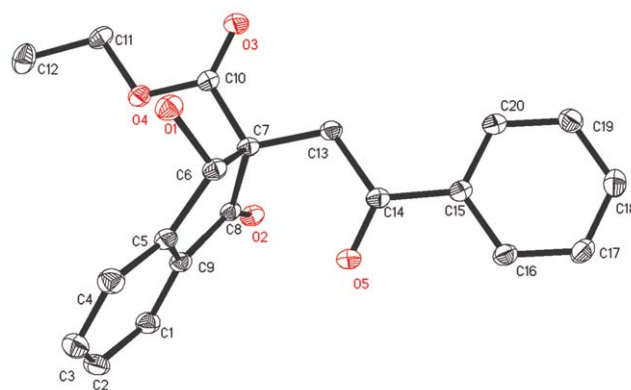
^a The NHC **4** was generated from the corresponding thiazolium salt **4** in situ in the presence of the noted base.

^b Isolated yields of pure *cis* isomer of **3a**. No *trans*-**3a** was detected.

with an electron-rich aryl group, it is better than expected that Michael acceptors with 2-furyl and 2-thienyl gave the corresponding hydroxyindanone in very good yields (entries 8 and 9). Michael acceptor **2j** with diketo group also showed good reactivity (entry 10). The spirocyclic hydroxyindanone **3k** could be obtained in reasonable yield (46%) when *N*-phenylmaleimide was used as the substrate (Table 2, entry 11).

Table 2 Synthesis of 3-Hydroxyindanones via NHC-Catalyzed Tandem Reaction of Phthalaldehyde and Michael Acceptors **2**

Entry	2	3	Yield (%) ^a
1	 2a X = H	 3a X = H 3b X = NO ₂ 3c X = Cl	95
2	2b X = NO ₂		97
3	2c X = Cl		95

**Figure 1** X-ray crystal structure of hydroxyindanone **3a**

The *cis* stereochemistry of hydroxyindanone **3a** was unambiguously assigned by the X-ray analysis of its crystal (Figure 1). It is worthwhile to note that only *cis* isomer of the hydroxyindanone was detected for all the reactions examined in Table 2.

The resulted highly functionalized indanones offer opportunities for further chemical transformations. For example, indanone **6a** could be prepared smoothly from hydroxyindanone **3a** via a two-step dehydroxy process (Scheme 2).

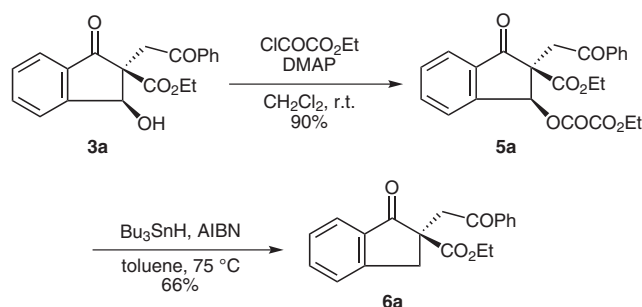
**Scheme 2** Synthesis of indanone **6a** from hydroxyindanone **3a**

Table 2 Synthesis of 3-Hydroxyindanones via NHC-Catalyzed Tandem Reaction of Phthalaldehyde and Michael Acceptors **2** (continued)

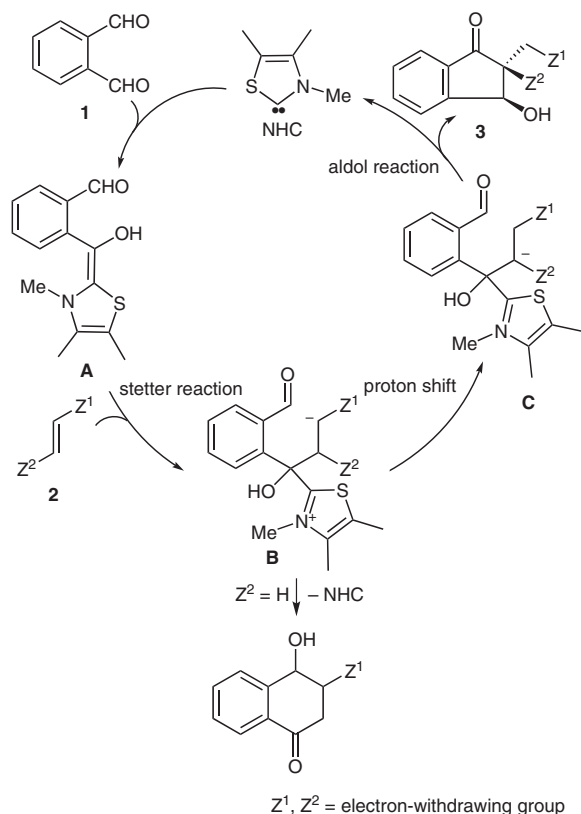
1 (1.5 equiv) + 2 $\xrightarrow[\text{THF, r.t.}]{\text{4a, Cs}_2\text{CO}_3 \text{ (10 mol\%)}}$ 3 $Z^1, Z^2 = \text{electron-withdrawing group}$			
Entry	2	3	Yield (%) ^a
4	2d X = Me	3d X = Me	58
5	2e X = OMe	3e X = OMe	50
6	 2f	 3f	92
7	 2g	 3g	93
8	 2h X = O	 3h X = O	98
9	 2i X = S	 3i X = S	95
10	 2j	 3j	65
11	 2k	 3k	46

^a Isolated yields of pure *cis* isomer of **3** and no *trans*-**3** was detected for all entries.

A possible catalytic cycle of this NHC-catalyzed [4+1]-annulation reaction is depicted in Scheme 3. The addition of NHC to phthalaldehyde gives the Breslow intermediate **A**, which reacts with Michael acceptor **2** affording zwitterion **B**. As we reported previously, zwitterion **B** reacts via an intramolecular aldol reaction to give [4+2]-annulation product if no additional withdrawing group is installed in the Michael acceptor ($Z^2 = \text{H}$). However, when the additional withdrawing group is present in the β -position of the Michael acceptor ($Z^2 = \text{electron-withdrawing group}$),

proton shift occurs to provide another zwitterionic **C** which is stabilized by the additional electron-withdrawing group (Z^2), hydroxy and thiazolium group. The intramolecular aldol reaction and fragmentation of intermediate **C** affords final [4+1]-annulation product of hydroxyindanone and regenerates the N-heterocyclic carbene.¹⁸

In conclusion, the synthesis of 2,2-disubstituted 3-hydroxyindanones was realized by a NHC-catalyzed tandem reaction of phthalaldehyde and 1,2-diactivated Michael acceptors. The additional electron-withdrawing group in



Scheme 3 Proposed catalytic cycle

the Michael acceptor facilitates the proton shift of the intermediate of the Stetter reaction and thus results in a [4+1]-annulation product. This tandem reaction features several advantages, including exclusive *cis* diastereoselectivity, readily available starting materials, and catalyst, and construction of two C–C bonds in one pot, which makes it potentially useful for the synthesis of the 3-hydroxyindanones.

Typical Procedure of the NHC-Catalyzed [4+1] Annulation

To an oven-dried 50 mL Schlenk tube equipped with a stir bar was charged with thiazolium salt **4a** (12.7 mg, 0.047 mmol) and phthalaldehyde (94.9 mg, 0.71 mmol). The tube was closed with a septum, evacuated, and back-filled with argon. To this mixture was added solvent THF (4.7 mL) and Michael acceptor **2a** (96.3 mg, 0.47 mmol). Then, Cs_2CO_3 (15.4 mg, 0.047 mmol) was added to the tube. The mixture was further stirred overnight, then diluted with EtOAc and passed through a short silica pad. The solvent was removed under reduced pressure, and the residue was purified by chromatography on silica gel (EtOAc–PE, 1:3) to give 151.6 mg (95%) of hydroxyindanone **3a** as a white solid; $R_f = 0.22$ (PE–EtOAc, 3:1); mp 125–126 °C. ^1H NMR (300 MHz, CDCl_3): $\delta = 8.05$ (d, $J = 7.2$ Hz, 2 H), 7.87–7.83 (m, 2 H), 7.79–7.74 (m, 1 H), 7.63 (t, $J = 7.2$ Hz, 1 H), 7.57–7.48 (m, 3 H), 5.46 (s, 1 H), 4.33 (d, $J = 18.3$ Hz, 1 H), 4.15–4.08 (m, 3 H), 3.53 (d, $J = 18.3$ Hz, 1 H), 1.11 (t, $J = 7.2$ Hz, 3 H). ^{13}C NMR (75 MHz, CDCl_3): $\delta = 198.4$, 198.3, 168.4, 153.5, 136.3, 135.8, 135.5, 133.7, 129.2, 128.7, 128.3, 125.2, 123.8, 76.4, 65.5, 62.1, 40.8, 13.8. IR (KBr): $\nu = 1738$, 1717, 1684, 1210. MS (EI): m/z (%) = 338 (3.0), 105 (100). HRMS (EI): m/z [M^+] calcd for $\text{C}_{20}\text{H}_{18}\text{O}_5$: 338.1154; found: 338.1150.

Supporting Information for this article is available online at <http://www.thieme-connect.com/ejournals/toc/synlett>.

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- (18) At this stage, we don't know whether the NHC leaves before or after the aldol reaction.

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