

## ENANTIOSELECTIVE SYNTHESIS OF THE 6,8-DIOXABICYCLO-[3.2.1]OCTANE SKELETON BY ASYMMETRIC DIHYDROXYLATION

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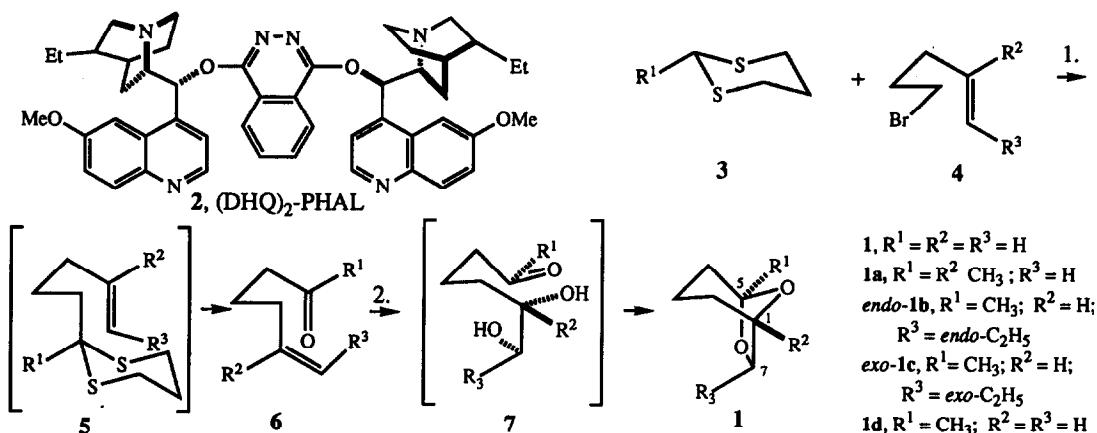
**Summary (Keywords):** Sharpless asymmetric dihydroxylation (AD) of 6-hepten-2-one with AD-mix- $\alpha$ <sup>®</sup> followed by acidification gives (1*S*,5*R*)-5-methyl-6,8-dioxabicyclo[3.2.1]octane in moderate ee. This methodology defines a two-pot enantioselective construction of the (-)-frontalin and brevicomin skeleton from the corresponding 2-alkyl-1,3-dithiane and 5-bromo-1-hexene derivatives.

The 6,8-dioxabicyclo[3.2.1]octane ring system **1** ( $R^1 = R^2 = R^3 = H$ , see Scheme) defines the basic skeleton of the beetle aggregation pheromone (-)-frontalin (**1a**,  $R^1 = R^2 = CH_3$ ;  $R^3 = H$ ) and *endo/exo*-brevicomin (*endo*-**1b**,  $R^1 = CH_3$ ,  $R^2 = H$ ;  $R^3 = \textit{endo}$ -C<sub>2</sub>H<sub>5</sub>; *exo*-**1c**,  $R^1 = CH_3$ ,  $R^2 = H$ ,  $R^3 = \textit{exo}$ -C<sub>2</sub>H<sub>5</sub>).<sup>1</sup> Published enantioselective syntheses<sup>2</sup> of derivatives of **1** are lengthy or produce racemic derivatives.<sup>3</sup> Since the absolute stereochemistry of **1** can be set by a 1,2-diol, the adaptation of ligand accelerated catalytic (LAC) asymmetric dihydroxylation (AD)<sup>4</sup> appeared well-suited to the construction of this ring system. Recently, Sharpless disclosed improvements to the AD process<sup>5</sup> mediated by the 1,4-bis-dihydroquinine phthalazine ligand (DHQ)<sub>2</sub>-PHAL (**2**). A premix of **2** with potassium ferricyanide, potassium carbonate and potassium osmate ("AD-mix- $\alpha$ "<sup>6a,b</sup>) makes the oxidation protocol exceptionally simple.

The monosubstituted olefin 6-hepten-2-one (**6**,  $R^1 = CH_3$ ,  $R^2 = R^3 = H$ ) was selected as an appropriate substrate for testing this LAC transformation into the (-)-frontalin framework. Thus, dithiane **3** ( $R^1 = CH_3$ ) was converted into the 2-lithio-1,3-dithiane derivative **7** by reaction with *n*-BuLi in THF-hexane (0.5 M, -20 °C, 2 h) and alkylated with 5-bromo-1-hexene (-78 to -25 °C). Intermediate dithiane **5** ( $R^1 = CH_3$ ,  $R^2 = R^3 = H$ ) could be isolated and purified<sup>8</sup> (92%) but generally was reacted directly with 5 equiv of methyl iodide and 3 equiv of calcium carbonate in 10% aqueous acetonitrile (5 ml/g **5**, 25 °C, 96 h) affording 6-hepten-2-one (**6**,  $R^1 = CH_3$ ,  $R^2 = R^3 = H$ ; bp 40 °C at 10 torr) in 83% yield from **3**. Oxidation of **6** with AD-mix- $\alpha$  (1.4 g/mmol **6**)<sup>5, 6a</sup> in 1:1 *tert*-butyl alcohol-water (0.05 M, 0 °C, 18 h) provided the *S* diol **7** ( $R^1 = CH_3$ ,  $R^2 = R^3 = H$ ) in situ. Direct treatment of the reaction mixture with hydrochloric acid (pH = 2, 25 °C) gave (1*S*,5*R*)-5-methyl-6,8-dioxabicyclo[3.2.1]octane (**1d**,  $R^1 = CH_3$ ,  $R^2 = R^3 = H$ ) in 84% yield admixed with approx 15-20 % of the corresponding (1*R*, 5*S*)<sup>8</sup> enantiomer (*ent*-**1d**).

Adaptation of the improved Sharpless LAC-AD methodology allows rapid and facile synthesis of the carbon-oxygen framework of (-)-frontalin and brevicomin in 60-70% ee. The modest ee might be due to the disruption

## Scheme



(1). 1.05 equiv *n*-BuLi, 3 (R<sup>1</sup> = CH<sub>3</sub>), -20 °C, 2 h; 1.15 equiv 4 (R<sup>2</sup> = H), -78 to -25 °C, 72 h; 5 equiv CH<sub>3</sub>I, CH<sub>3</sub>CN, 3 equiv CaCO<sub>3</sub>, 25 °C, 94 h. (2). AD-mix-α, 1:1 *tert*-BuOH-H<sub>2</sub>O, 0 °C, 18 h; HCl, 25 °C.

of the LAC process by the carbonyl of 6. However, ee's in this range would prove useful in the synthesis of this class of pheromones since partially racemic mixtures remain biologically active in practical applications.<sup>3,9</sup>

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## REFERENCES AND NOTES

- Payne, T. L.; Andryszak, N. A.; Wieser, H.; Dixon, E. A.; Ibrahim, N.; Coers, J. *J. Chem. Ecol.* **1988**, *14*, 1217.
- A recent account: Asaoka, M.; Hayashibe, S.; Sonoda, S.; Takei, H. *Tetrahedron*, **1991**, *47*, 6967.
- Yus, M.; Ramón, D. J. *J. Org. Chem.* **1992**, *57*, 750.
- Jacobsen, E. N.; Markó, I.; Mungall, W. S.; Schröder, G.; Sharpless, K. B. *J. Am. Chem. Soc.* **1988**, *110*, 1968.
- Sharpless, K. B. Presented at The Fifth Biennial Lilly Grantee Symposium, Indianapolis, Indiana, February 24-25, 1992; paper 6. Sharpless, K. B.; Amberg, W.; Bennani, Y. L.; Crispino, G. A.; Hartung, J.; Jeong, K.; Kwong, H.; Morikawa, K.; Wang, Z.; Xu, D.; Zhang, X. *J. Org. Chem.* **1992**, *57*, 2768.
- (a) The AD-mixes are now commercially available from Aldrich Chemical Corporation. (b) AD-mix-α is a mixture of potassium ferricyanide (979.9 g), potassium carbonate (411.6 g), 2 (7.73 g) and potassium osmate (0.73 g). AD-mix-β is also available and is of a similar composition with the corresponding 1,4-bis-hydroquinidine phthalazine analog of 2. Oxidation of 6 with AD-mix-β gave *ent*-1d in approx 60 % ee.
- Seebach, D.; Corey, E. J. *J. Org. Chem.* **1975**, *40*, 231.
- All intermediates displayed expected <sup>1</sup>H NMR, <sup>13</sup>C NMR, IR, and mass spectral data. Purity was established by GC on 30 m x 0.25 mm x 0.25 μ DB-1 fused silica column at 50-220 °C and by TLC (silica gel with 4:1 hexane-ethyl acetate). Intermediates 6 and 1d are volatile and yields are corrected for solvent content. Enantiomeric excesses were determined by HPLC of the purified primary mono-benzoate of 7 or *ent*-7 (1 equiv of PhCOCl-pyridine with diol in CH<sub>2</sub>Cl<sub>2</sub> and chromatography) on a Chiralcel® OD column with 200:10:1 hexane-isopropanol-diethylamine as eluent and by direct GC analysis on 1d or *ent*-1d with a 50 m x 0.25 mm LIPODEX®-E at 50-220 °C with He carrier. See König, W. A.; Icheln, D.; Runge, T.; Pforr, I.; Krebs, A. *J. High Resolution Chromatography*. **1990**, *13*, 702.
- Joshi, N. N.; Mamdapur, V. R.; Chadha, M. S. *J. Chem. Soc., Perkin Trans. 1* **1983**, 2963.